

*Intermétalliques thermoélectriques d'hier,
d'aujourd'hui... et de demain*

Collège de France
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Clathrate

Half-Heusler

Skutterudite (C. Uher 15/05 seminar)

Higher Manganese silicides (HMS)

CdI₂ based sulfides and selenides

Bi₂Te₃

PbTe

Zintl Phases

SiGe

Oxyselenides (D. Bérardan/N. Dragoe, ICMMO, Orsay)

La_{3-x}Te₄

Borides

Pseudo-Hollandites

Mg₂Si

Chevrel Phases

A_xMo₉Se₁₁ (M.Potel URennes1)

FeSi₂

TAGS

Oxides (S. Hébert, CRISMAT)

Zn₄Sb₃

Tetrahedrites (Cu₁₂Sb₄S₁₃)₂

Plan du séminaire

1) Introduction

- a) Definitions
- b) Figure of merit
- c) Thermoelectric phenomenon and materials

2) Old thermoelectric materials

- a) Bi_2Te_3
- b) PbTe
- c) SiGe

3) New processes, new techniques in the quest for materials of the future...

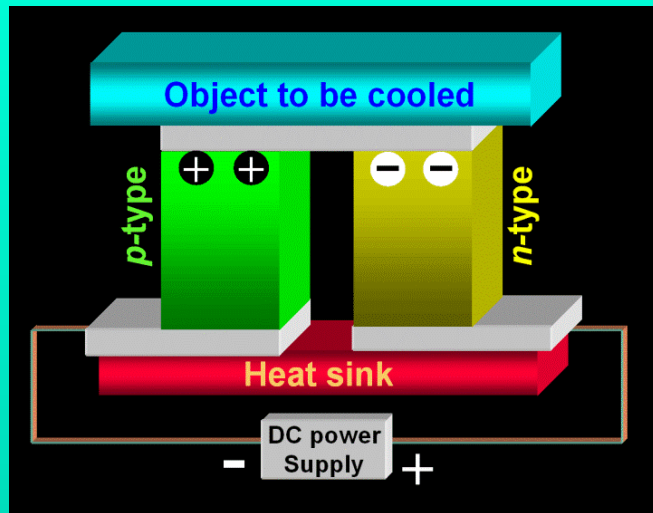
...or how to disrupt the phonons without affecting the charge carriers...

4) Strategy for new TE materials research

- a) Discovery of new TE materials?
- b) Few « good » examples: Zintl, Hollandite, $\text{TiS}_{2-x}\text{Se}_x$
- c) One bad example: AgCrSe_2

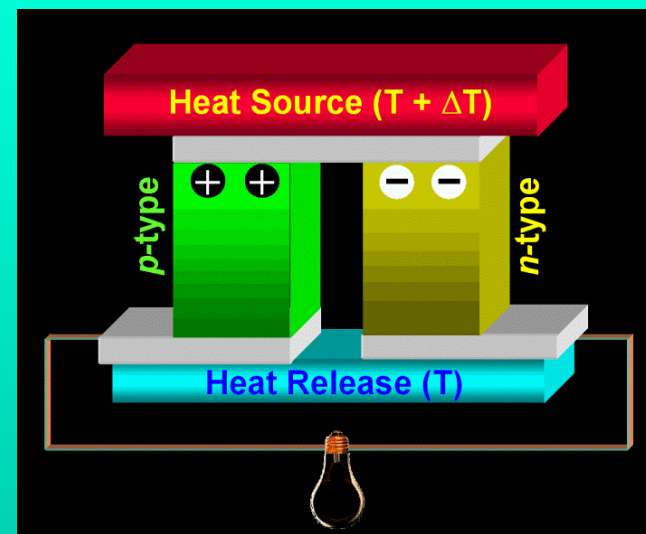
Thermo-Electric Conversion

Refrigeration mode
"Peltier effect"



When a current passes through the circuit, electrons and holes carry heat.

Generation mode
"Seebeck effect"



An applied temperature difference make the materials generate a voltage.

S → High Seebeck coefficient (V/K)

High voltage/power needed !

ρ → Low electrical resistivity (Ω.m)

Limit the thermal loss by Joule effect

κ → Low thermal conductivity (W/(m.K))

Maintain a temperature gradient ΔT high enough

$$ZT = \frac{S^2 T}{\rho \kappa}$$

Thermoelectric figure of merit



$$ZT = \frac{\alpha^2 T}{\rho \kappa}$$

α → coefficient de Seebeck (V/K)
 ρ → conductivité électrique ($\Omega.m$)
 κ → conductivité thermique (W/(m.K))

$$\alpha = \frac{\pi^2 k_B^2 T}{3e} \times \frac{8m^*}{h^2} \times \left(\frac{\pi}{3n}\right)^{2/3}$$

$$\sigma = \frac{1}{\rho} = ne\mu$$

$$\kappa = \kappa_r + \kappa_e$$

- κ_e → contribution des porteurs de charge
- κ_r → contribution du réseau
- k_B → constante de Boltzmann
- e → charge de l'électron
- m^* → masse effective
- h → constante de Planck
- n → Charge carrier concentration
- μ → mobilité des porteurs
- L → facteur de Lorenz

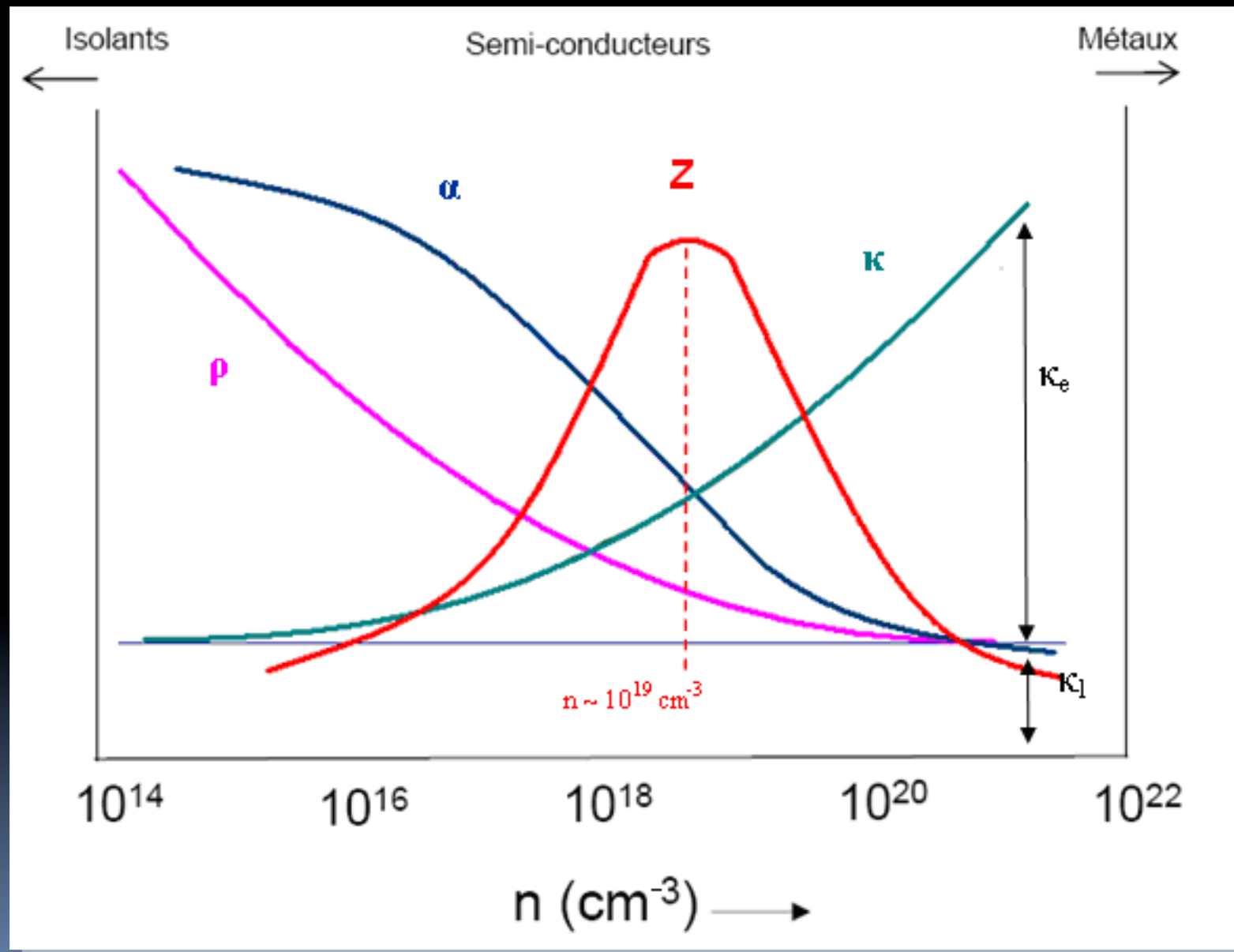
Wiedeman
Franz
Lorenz law

$$\kappa_e = \frac{LT}{\rho}$$



$$ZT = \frac{\alpha^2}{L + \frac{\rho \kappa_r}{T}}$$

Variation with charge carrier concentration



A bit of history...

1823: Seebeck

1835: Peltier

1851: Thomson

1930s: semiconductors with Seebeck $> 100\mu\text{V/K}$

1947: Telkes Thermogenerator with 5% efficiency

1977: NASA _ Voyager 2 (RTG / SiGe)

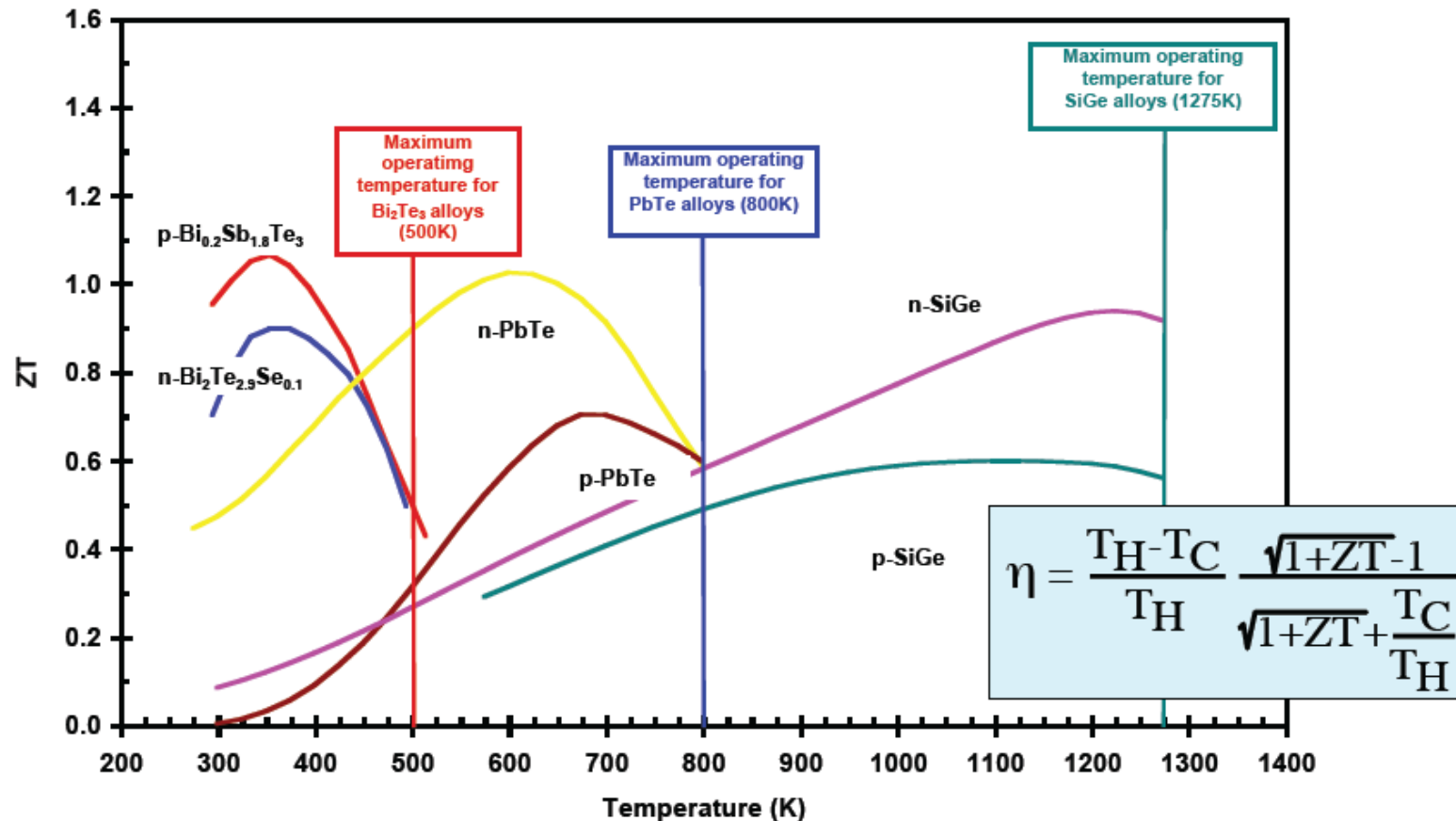
2020: NASA _ RTG (n-La_{3-x}Te₄ / p-Yb₁₄MnSb₁₁) ?

Today:

ONLY commercially available modules made of n and p Bi₂Te₃

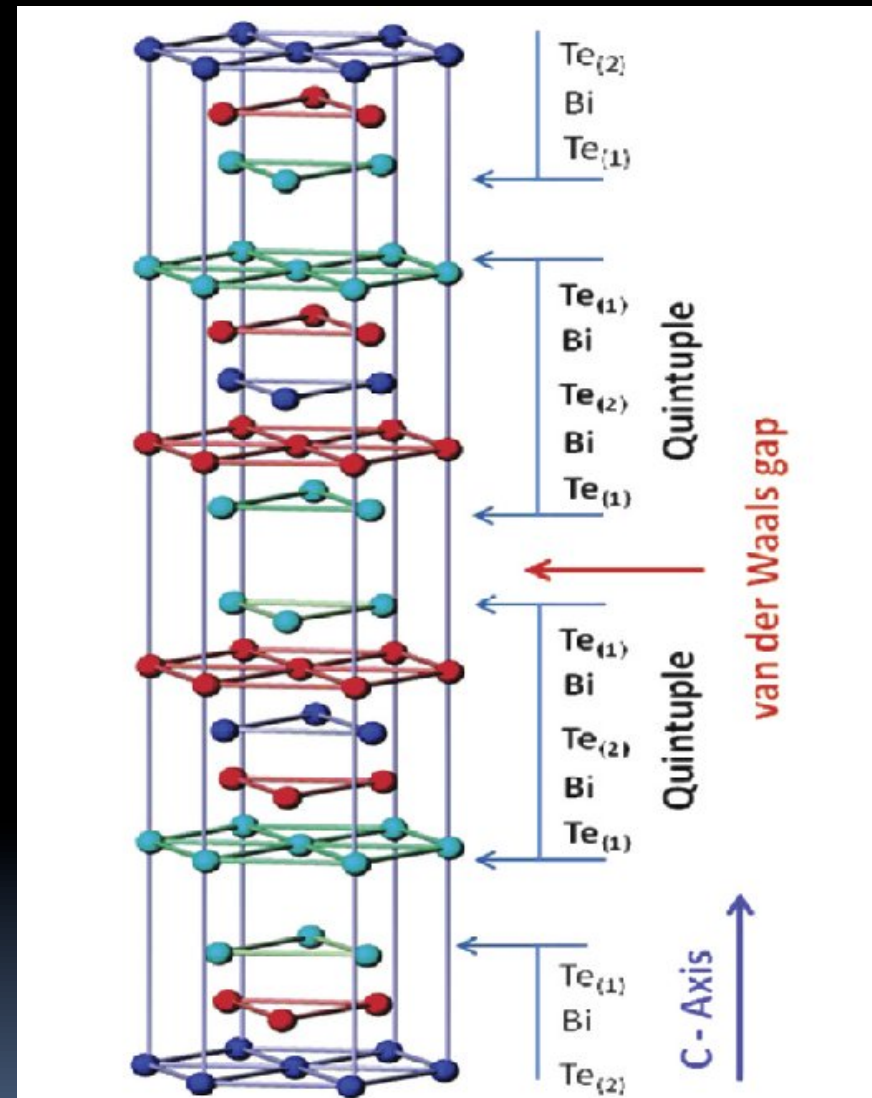
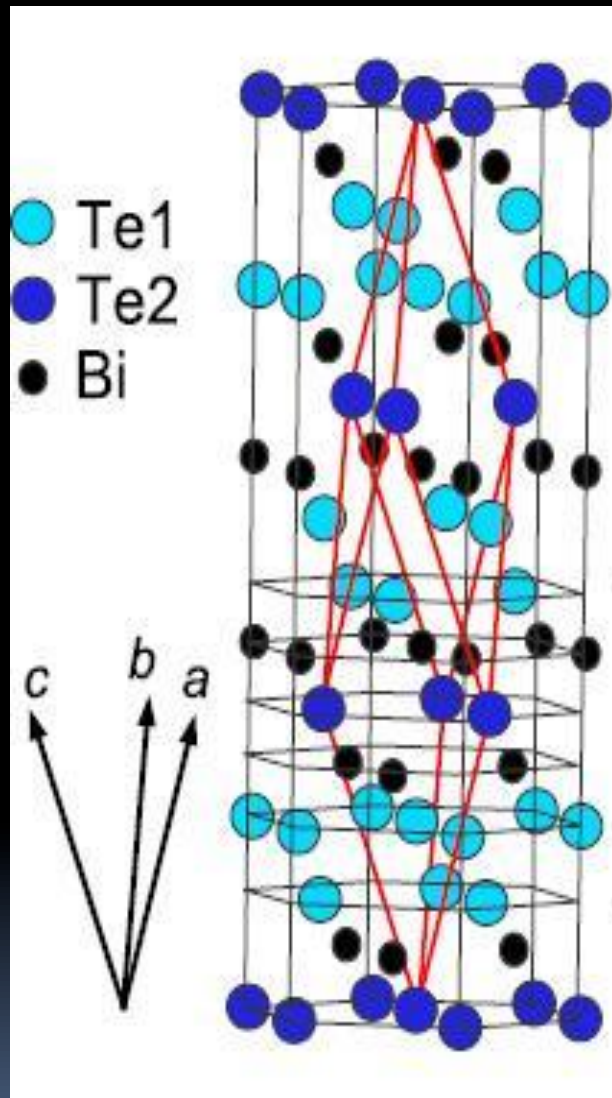


Historique des matériaux thermoélectriques



TE materials used in commercial applications were discovered in the 1960's

Bi₂Te₃ - structure



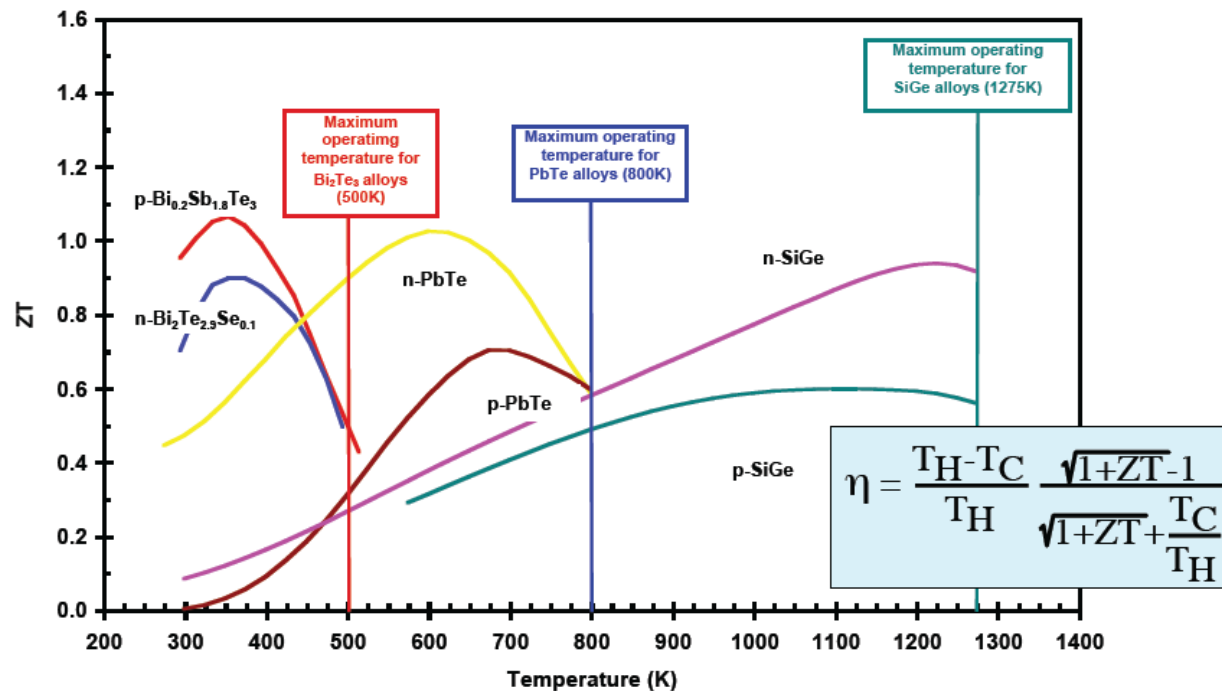
lamellar structure:

- # Clivage plans (mechanically fragile)
- # Anisotropy of transport properties

Bi₂Te₃ - dopage

p-type : solid solution (Bi₂Te₃)_x(Sb₂Te₃)_{1-x} ou Bi_{2-y}Sb_yTe₃

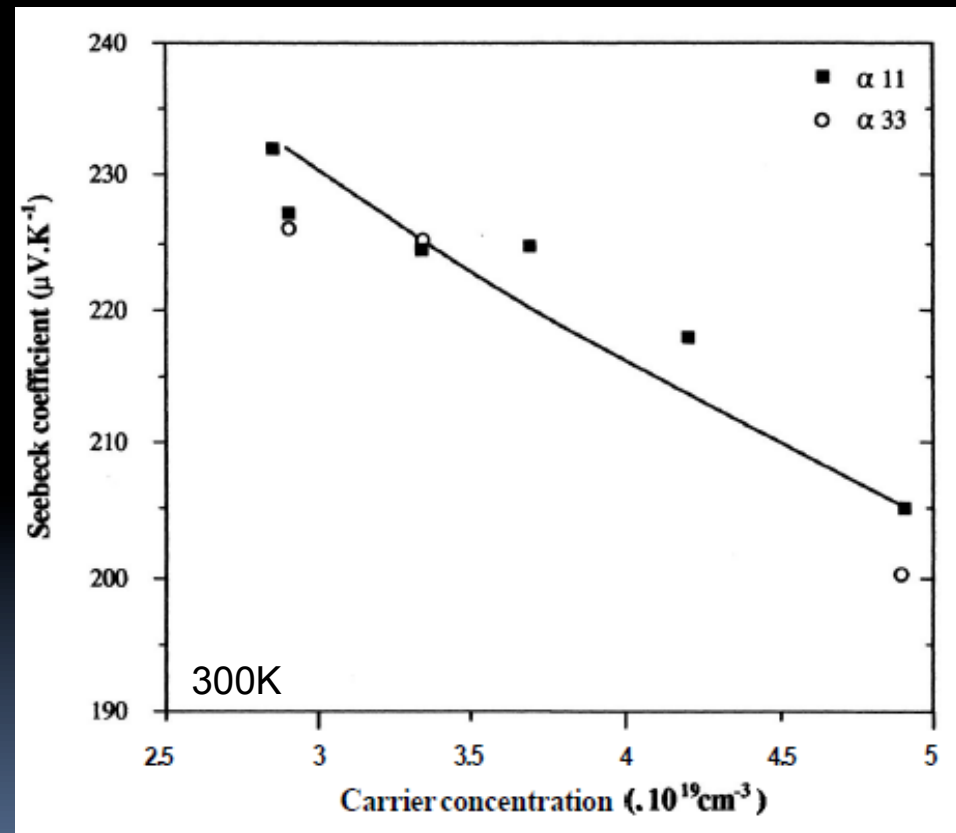
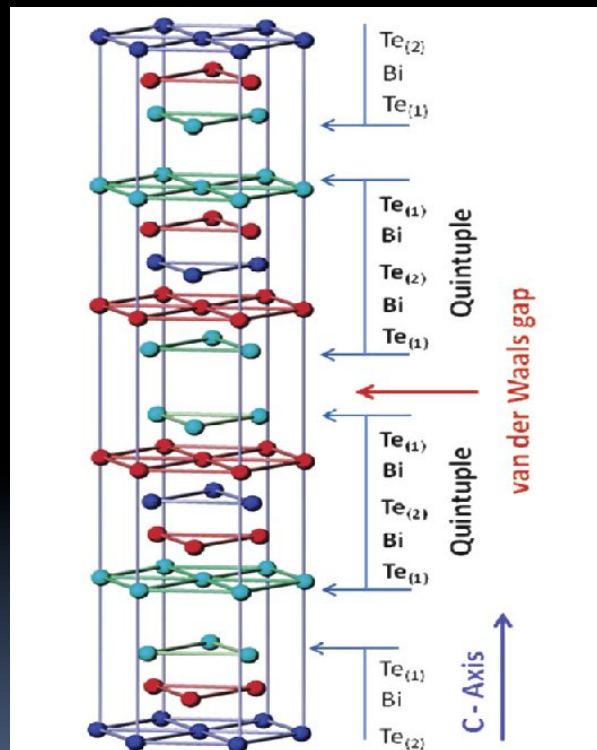
n-type : Solid solution (Bi₂Te₃)_x(Bi₂Se₃)_{1-x} ou Bi₂Se_yTe_{3-y}



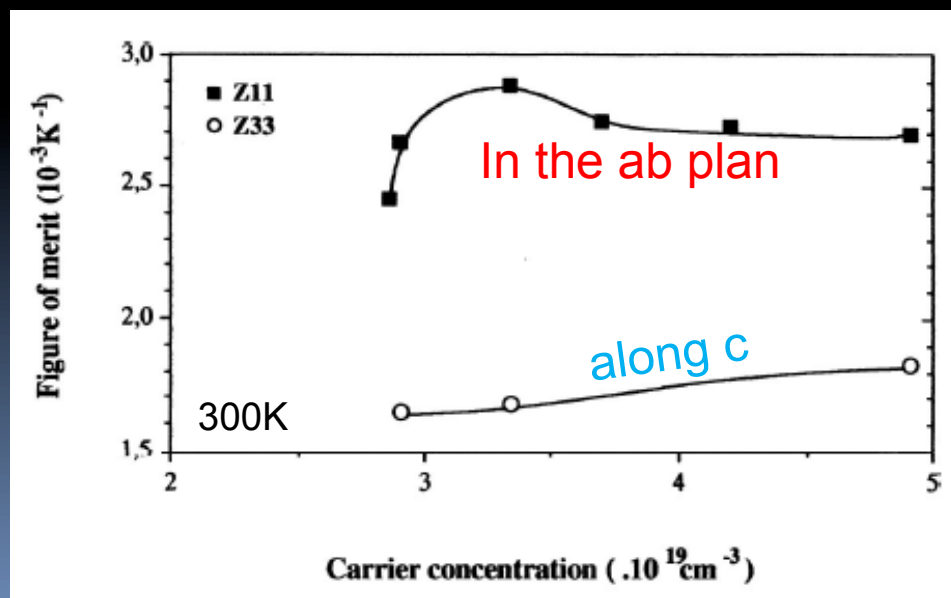
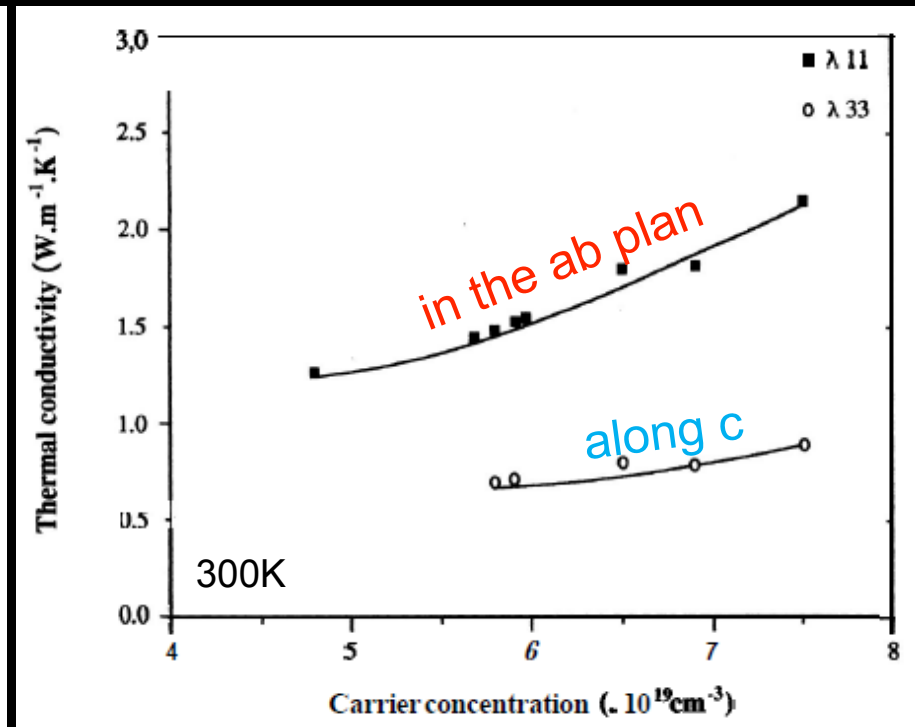
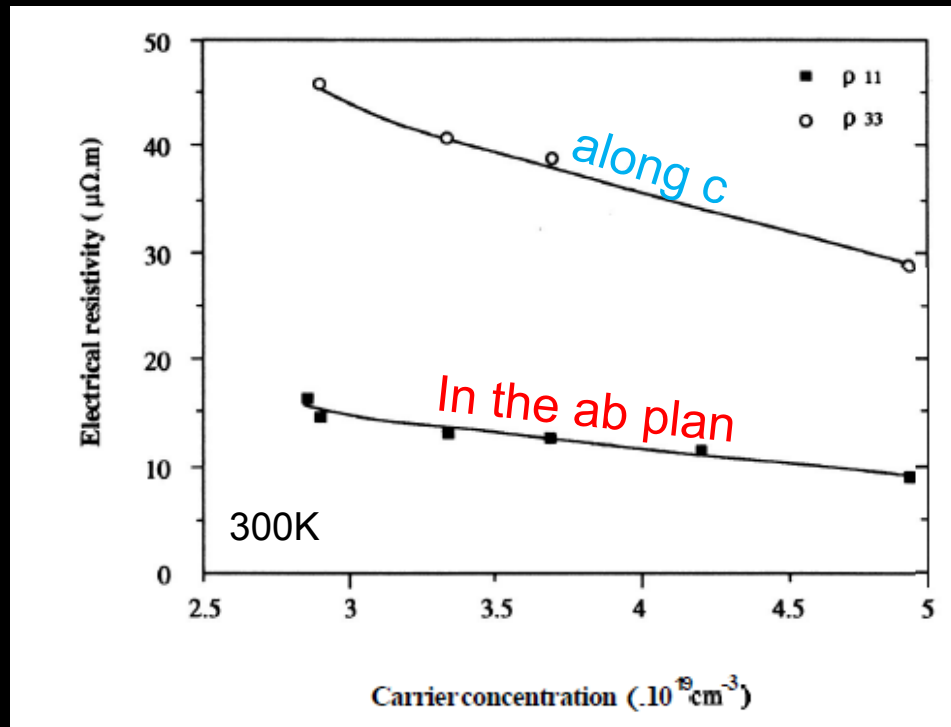
TE materials used in commercial applications were discovered in the 1960's

EXAMPLE: Composition : p- Bi₁₀Sb₃₀Te₆₀

properties TE measured in the ab plan (11) and along the c direction (33)



Seebeck is isotropic



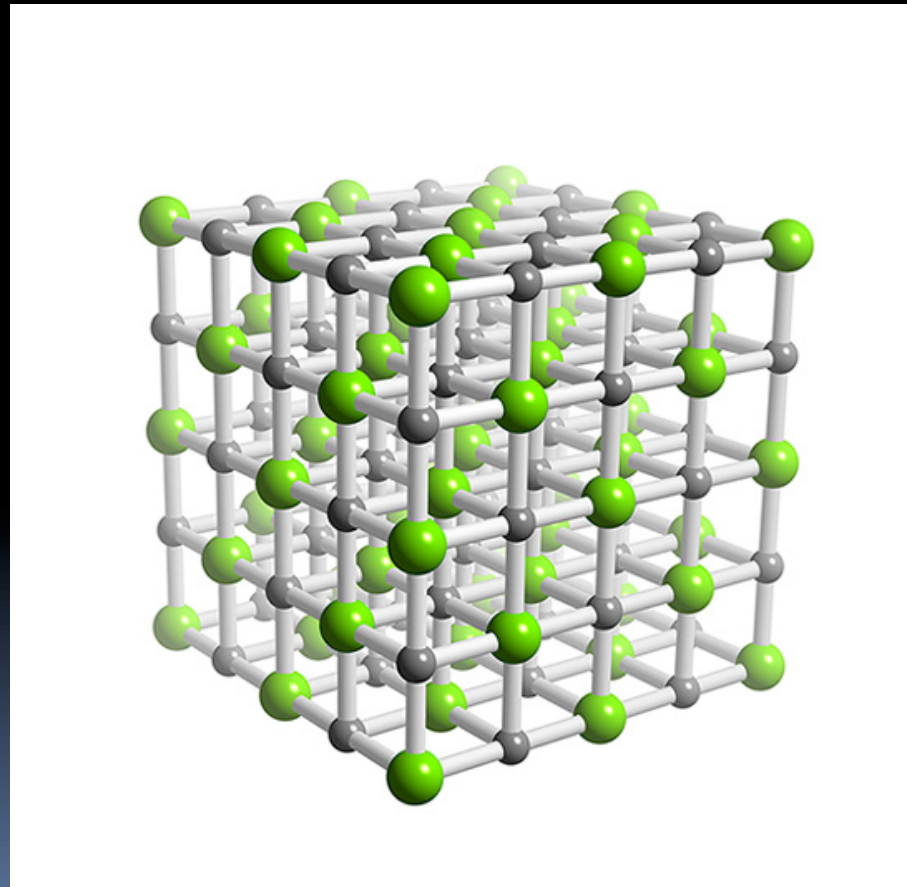
For n or p: $ZT(ab) > ZT(\text{along } c)$!

Must orient the TE legs when making a module
(commercial module : $ZT_p \sim 1-1,1$ and $ZT_n \sim 0.9$)

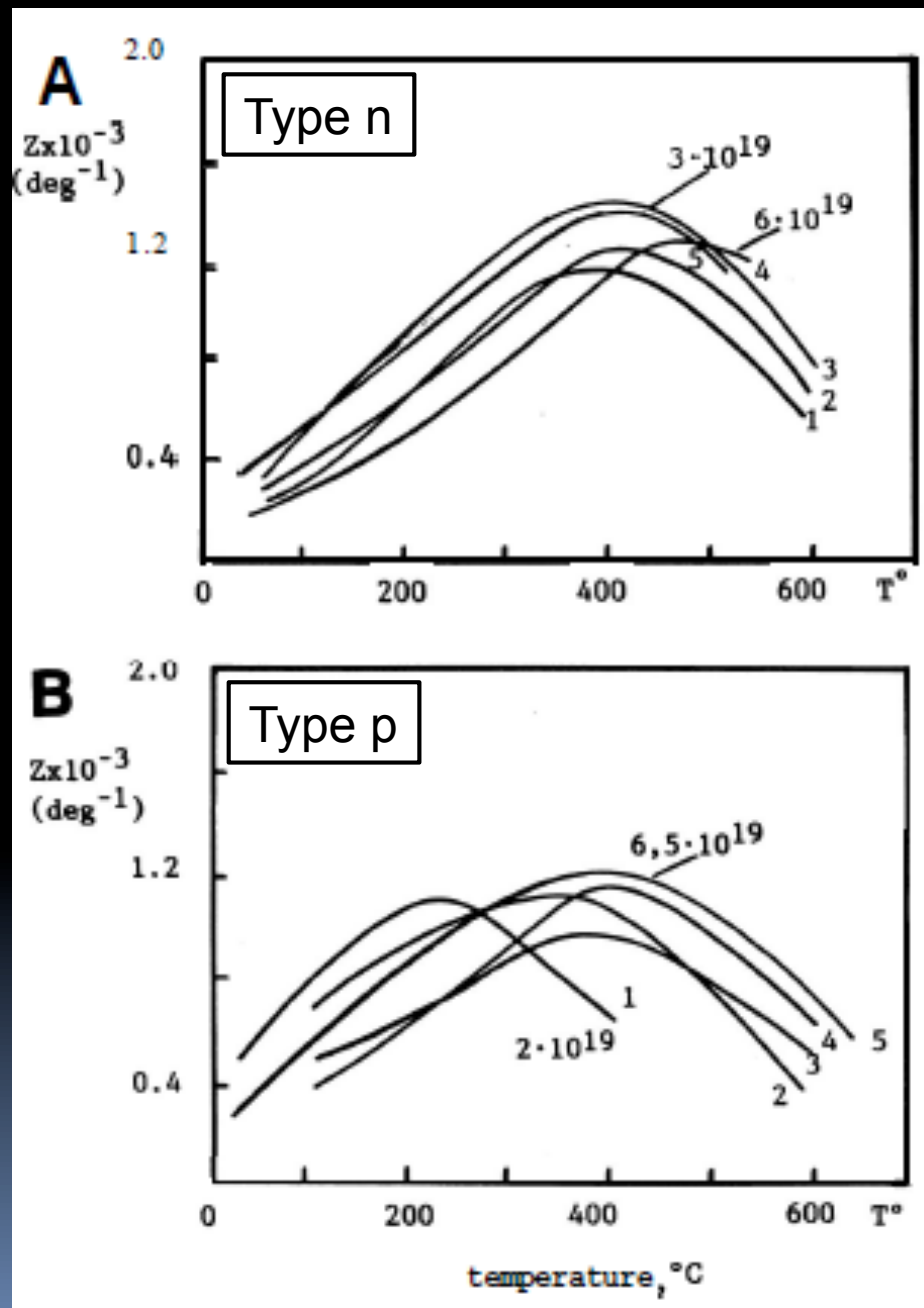
The most studied TE material
...still today...
(Cf 2^{ème} partie)

PbTe

- # a **simple** cubic structure Rock salt type
- # n and p type depending on the dopants
- # single crystals and polycrystalline samples



PbTe: variation of Z with the carrier concentration and temperature



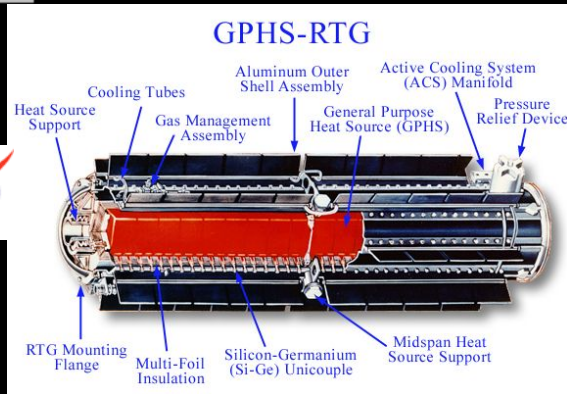
p or n type depending on dopant(s)

ZT about 1 at 800K

Very easy to synthesize

Pb: Forbidden in Europe (REACH...)

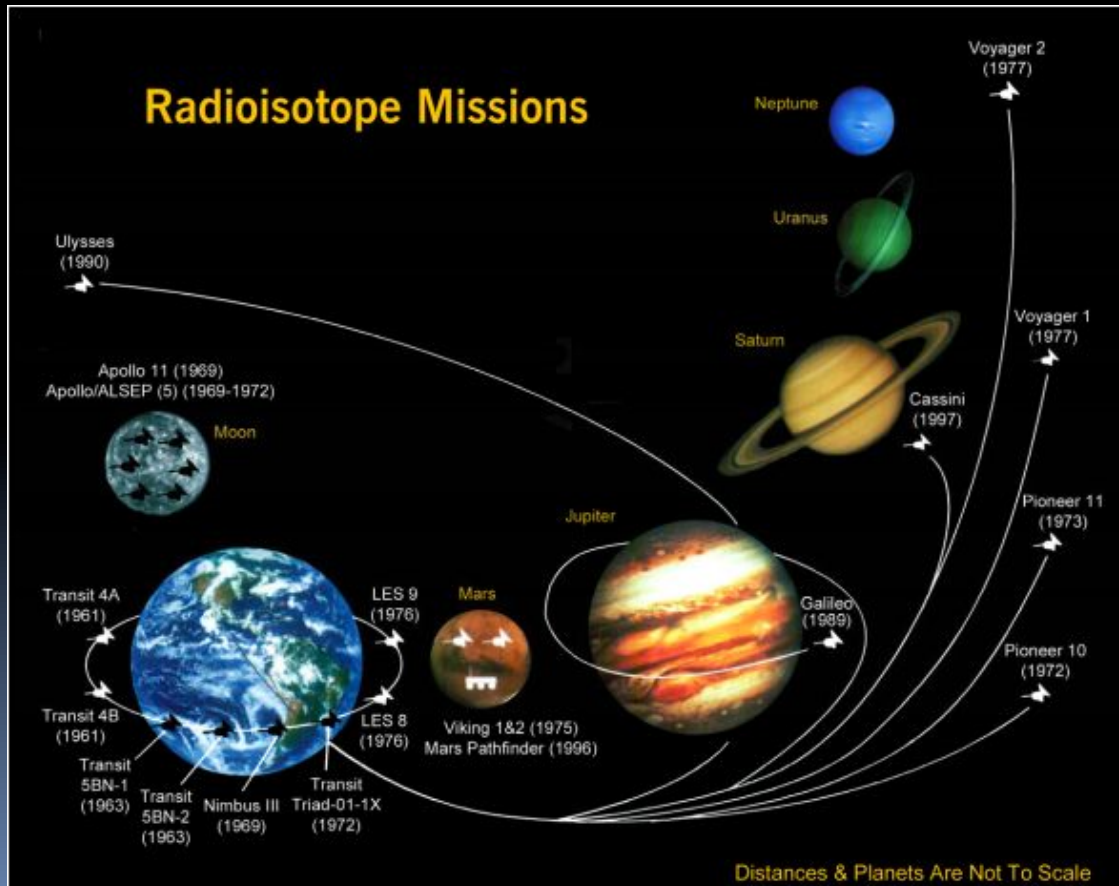
Much research underway in the USA
(Cf 2^{ème} partie)



Used in GPHS RTG since the 70'

n-type : highest ZT ~ 0,9 à 1000°C

p-type : highest ZT ~ 0,6 à 1000°C



Average ZT = 0.53

(couple over the range 473-1273K)

Efficiency ~8.5%

TE materials research strategy

To improve the performances of known TE materials

To find new thermoelectric materials

Improve the TE performances of known TE materials

Example: Bi_2Te_3 , PbTe , SiGe ...



$$\alpha = \frac{\pi^2 k_B^2 T}{3e} \times \frac{8m^*}{h^2} \times \left(\frac{\pi}{3n}\right)^{2/3}$$

$$\sigma = \frac{1}{\rho} = ne\mu$$

$$K = K_r + K_e$$

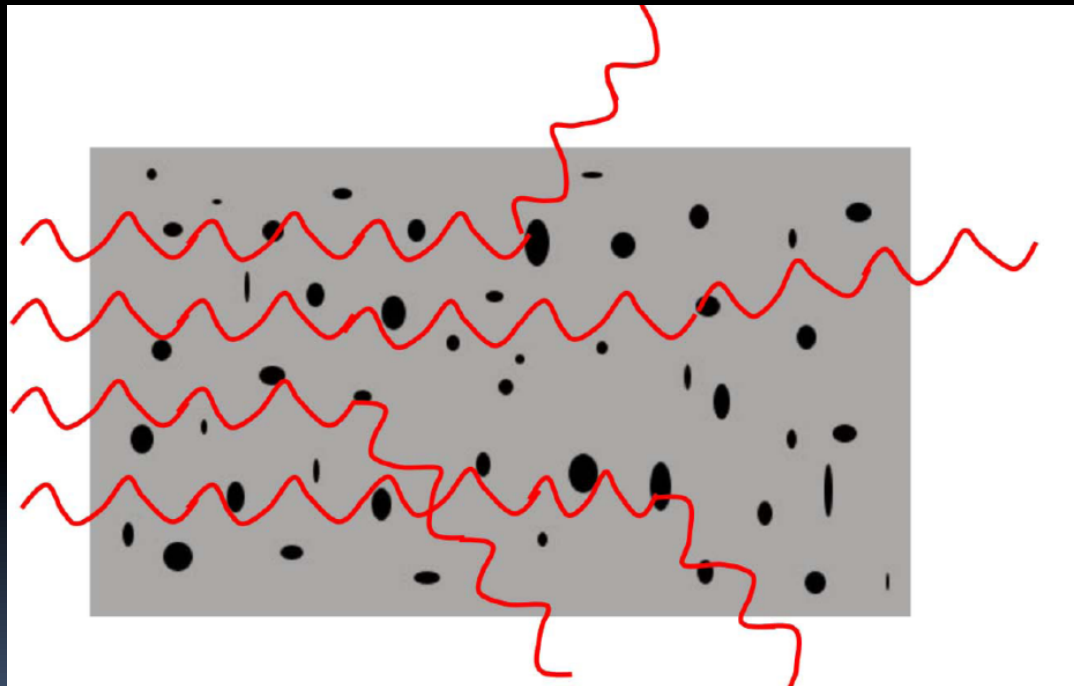
Chemically optimized! (*in terms of charge carrier concentration*),

-BUT K_r can (maybe) be lowered (?)

-BUT can't we decouple the Seebeck coefficient and the resistivity?

How to decrease the lattice thermal conductivity?

- = How to disrupt the phonons path ?
- = How to decrease their mean free path ?



Disrupting the phonons displacement

Structural disorder

Mixed occupations

Positional disorder

"impurities"

Vacancies

Increase the number of grain boundaries
(Decrease the grain size (towards nano...?))

Increase the disorder at the grain boundary

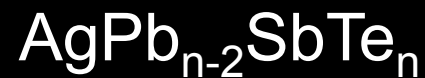
Make (nano) composites

PbTe new generation...



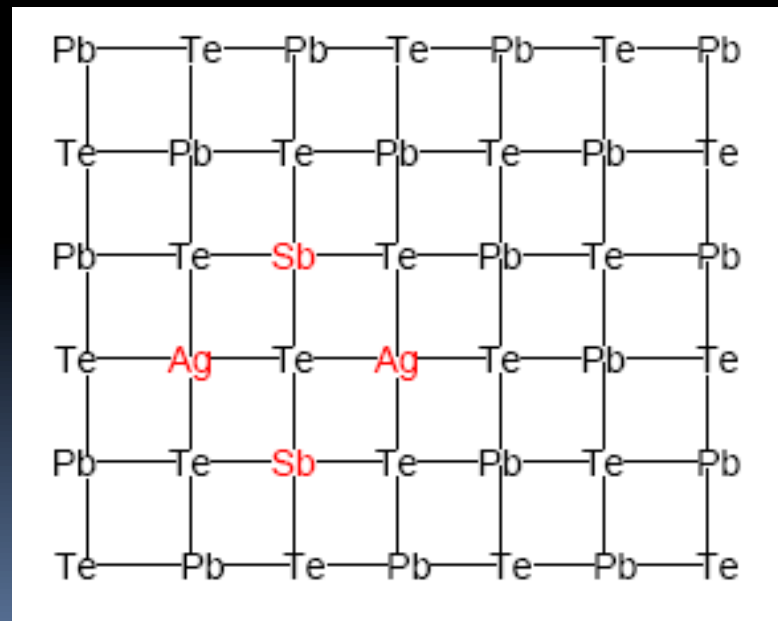
Substitution that leads to the family of **LAST** (M. Kanatzidis)

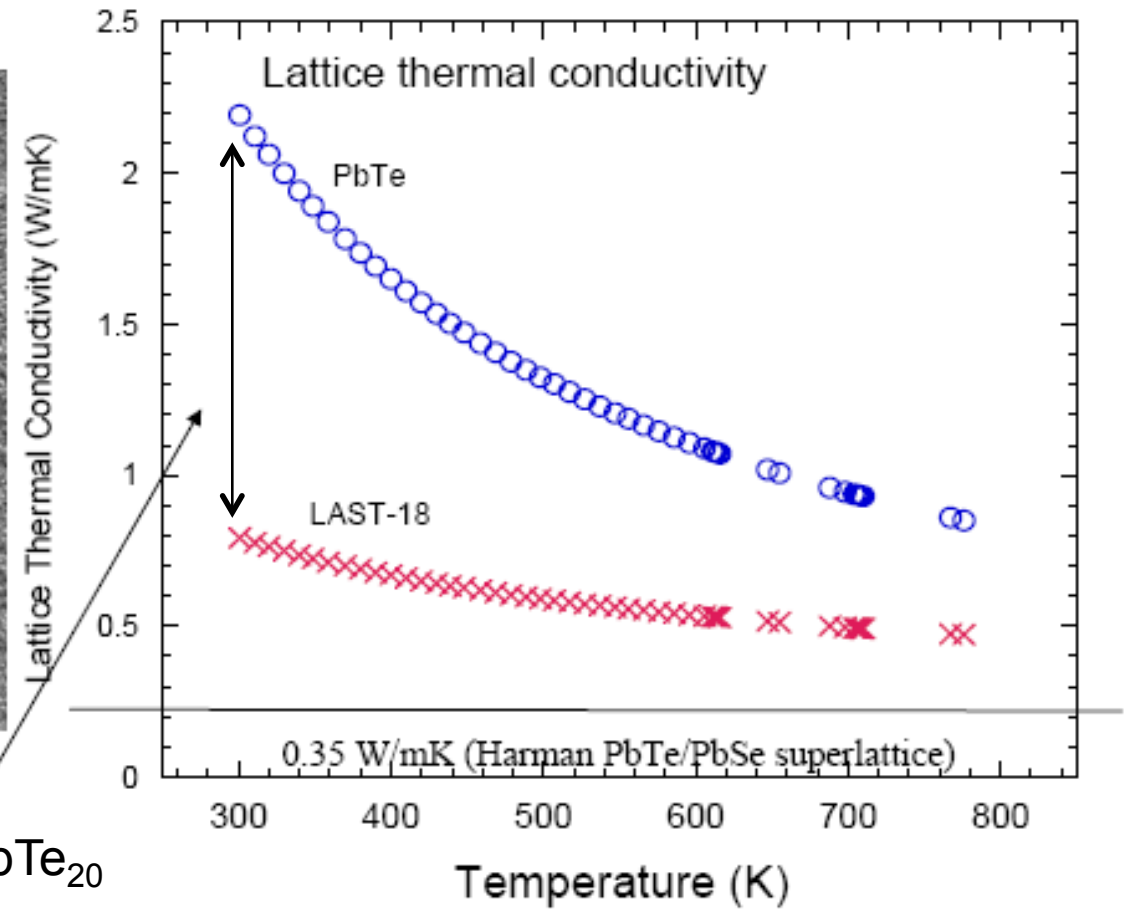
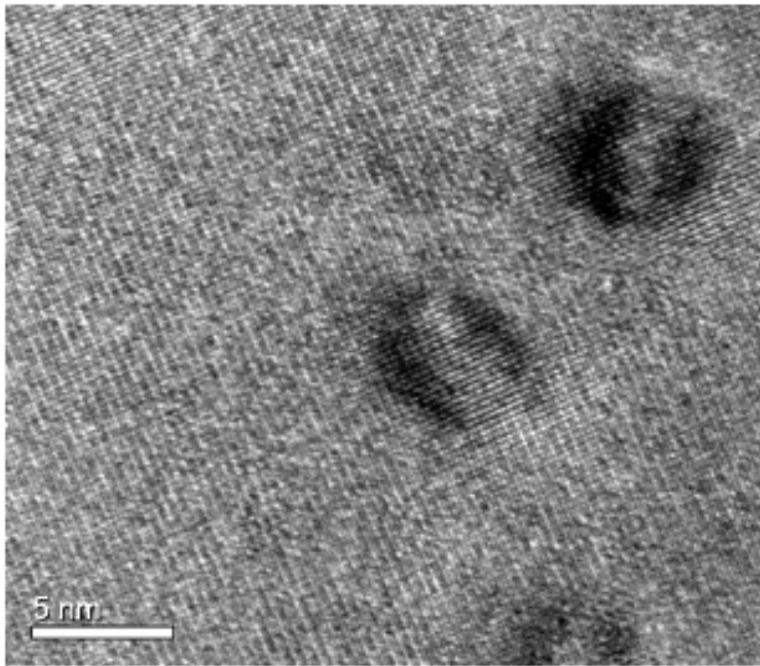
(LAST : Lead Antimony Silver Telluride)



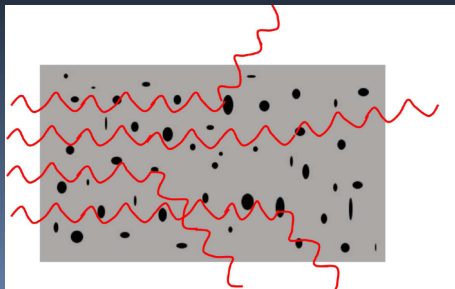
Ag-Sb pockets formed by
electrostatic attraction

Synthesis: fusion followed by
slow cooling





In situ formation of nano-domains Ag-Sb rich.

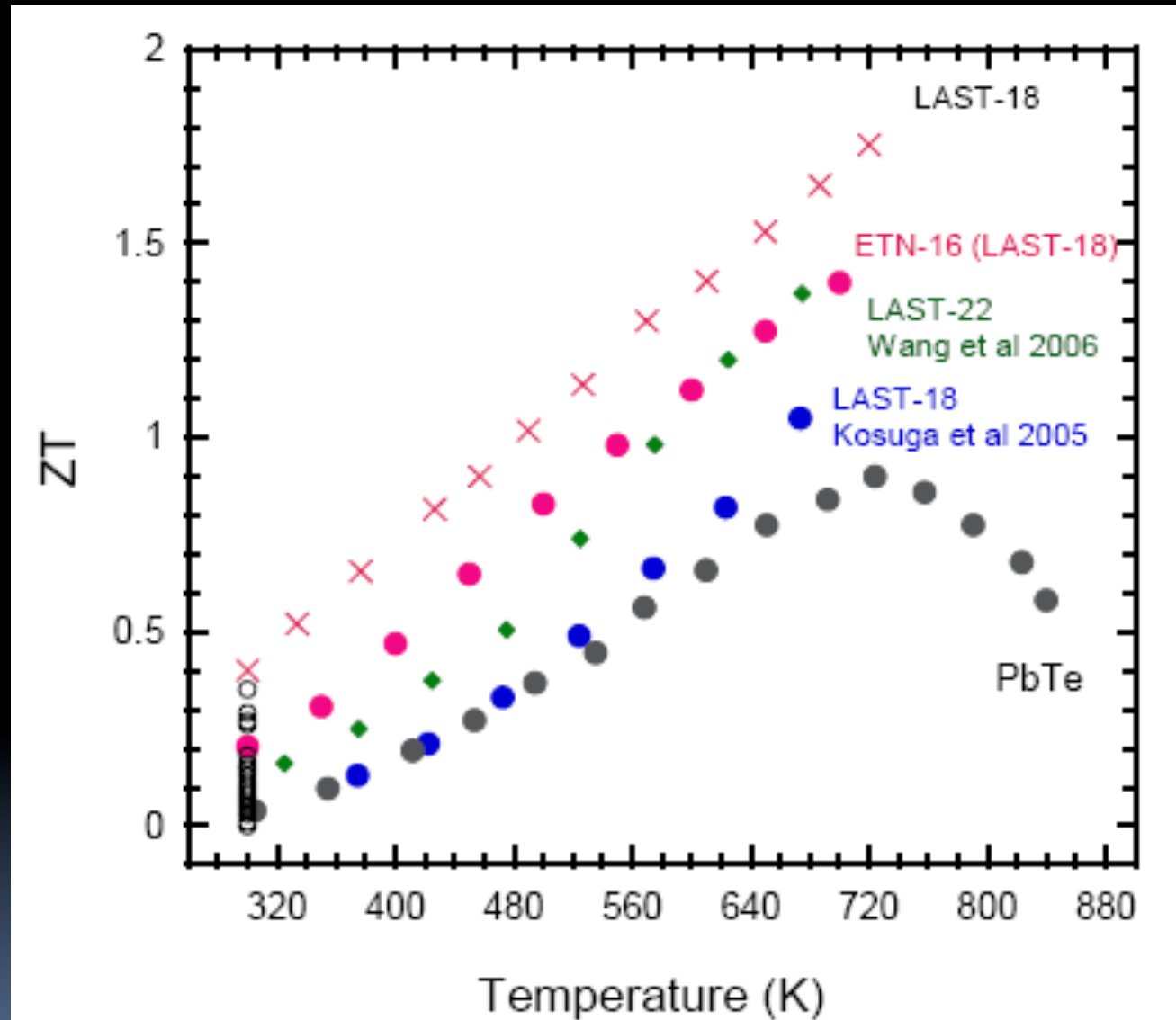


Decrease of the lattice thermal conductivity

Evident consequences on the ZT !



PbTe new generation...



Bi₂Te₃ new generation...

- ~Benefit from the anisotropy
 - # Must utilize the (ab) direction !
- ~Avoid single crystals
 - # Difficult synthesis
 - # Poor mechanical properties



Synthesis of polycrystalline samples

Hot pressing (conventionnal or SPS)

Partial alignement of the grains perpendicularly to the axis of pressing

Measure the properties in both directions

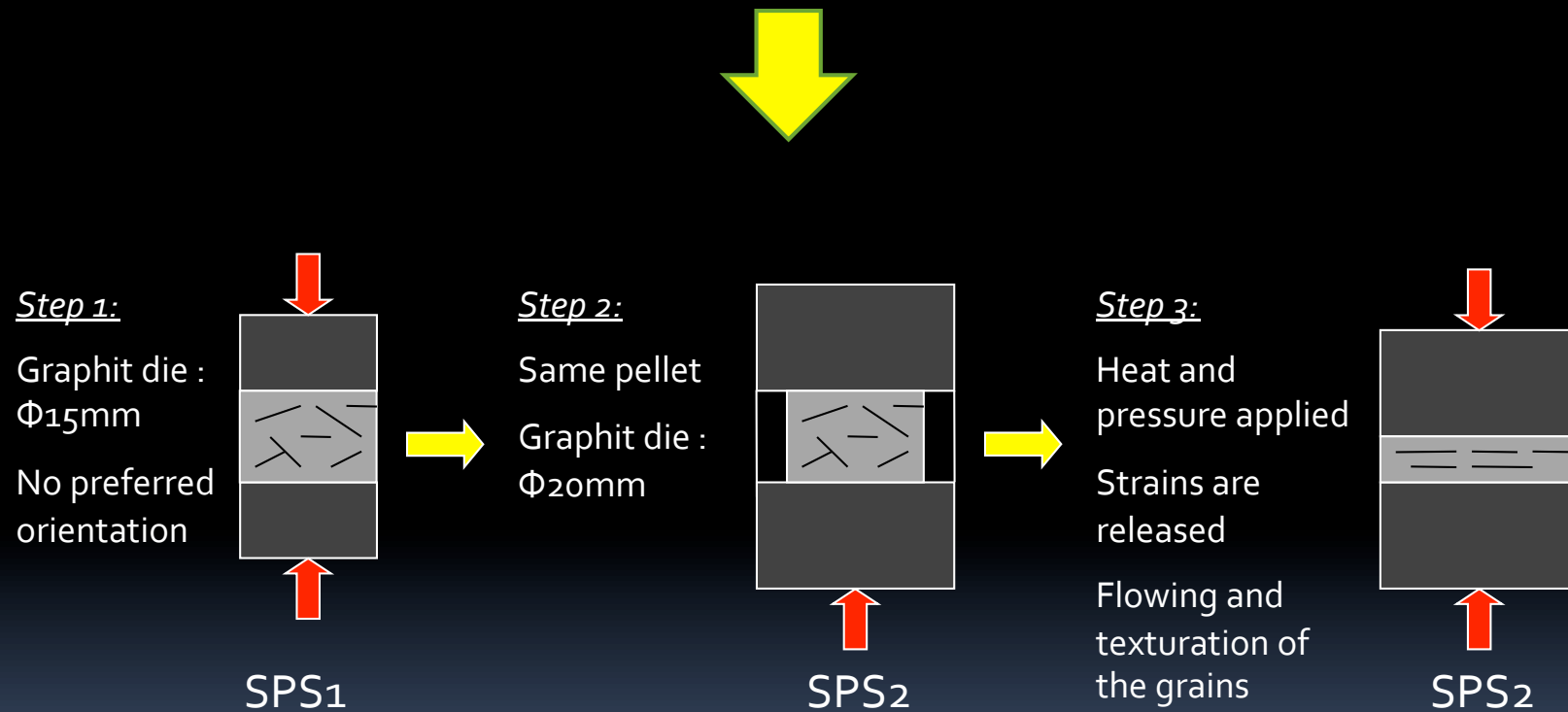
Synthesis: mechanical alloying (broyage réactif)

- # Fast
- # Reproducible
- # Small grain size
- # Low temperature (no evaporation)
- # Fine control of the stoichiometry
- # generate defects formation
- # easy to scale up

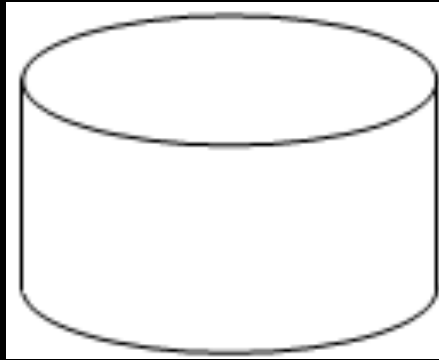
Typical parameters:

30 minutes
700 rpm } 15 grams of Bi₂Te₃

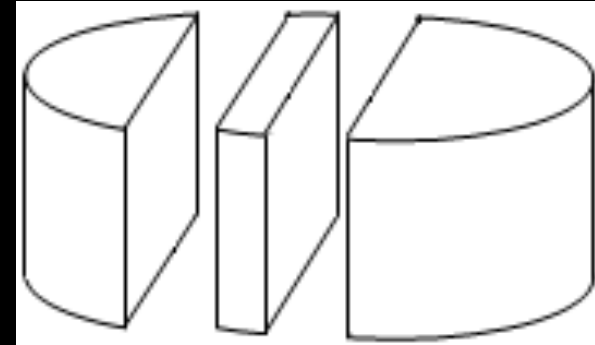
Controlling the degree of texturation is necessary in order to optimize the TE properties



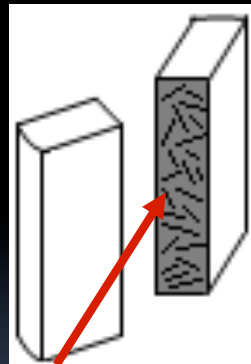
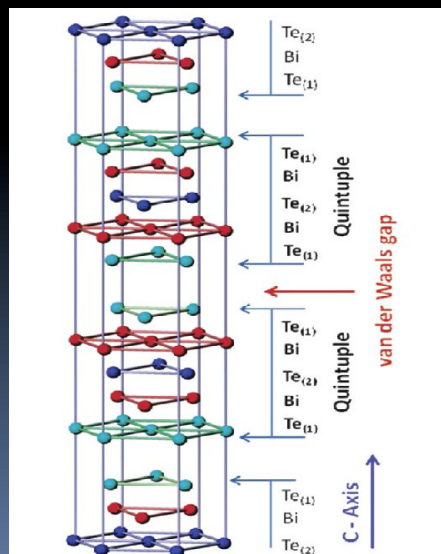
Bi₂Te₃ nouvelle génération... densification et découpes



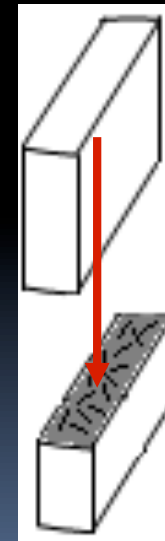
1. Puck after SPS



2. slicing

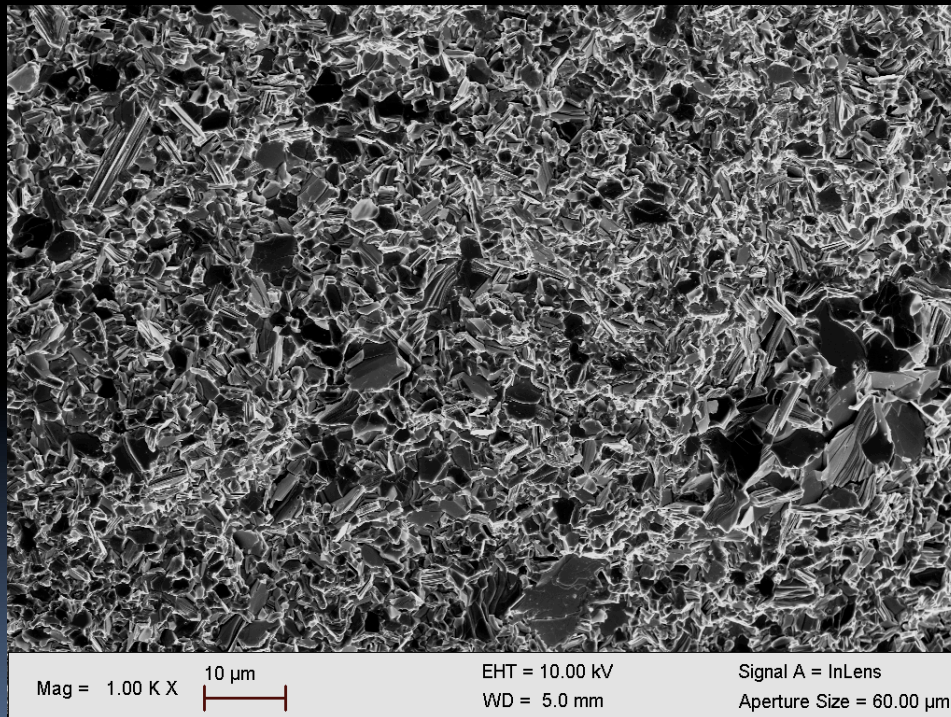
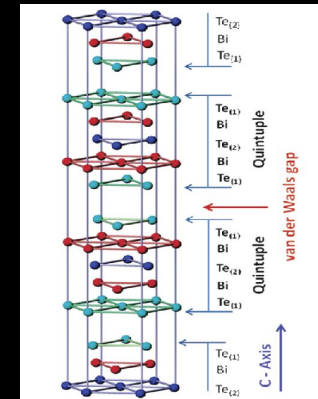


Echantillon coupé verticalement: in plane (ab)

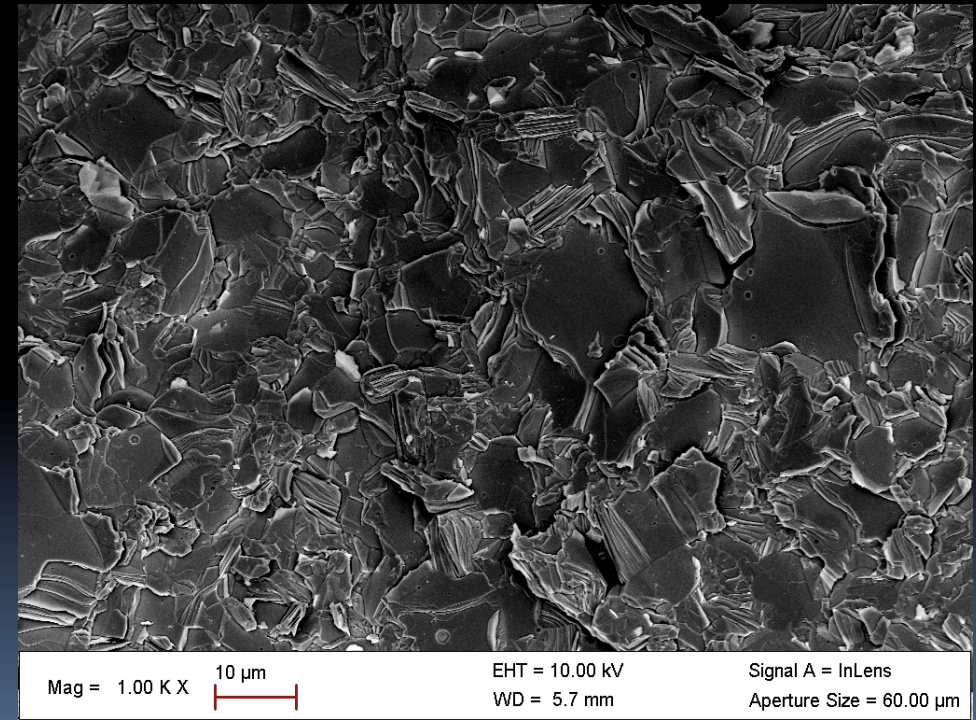


Echantillon coupé horizontalement: cross plane

Bi₂Te₃ nouvelle génération... microstructure

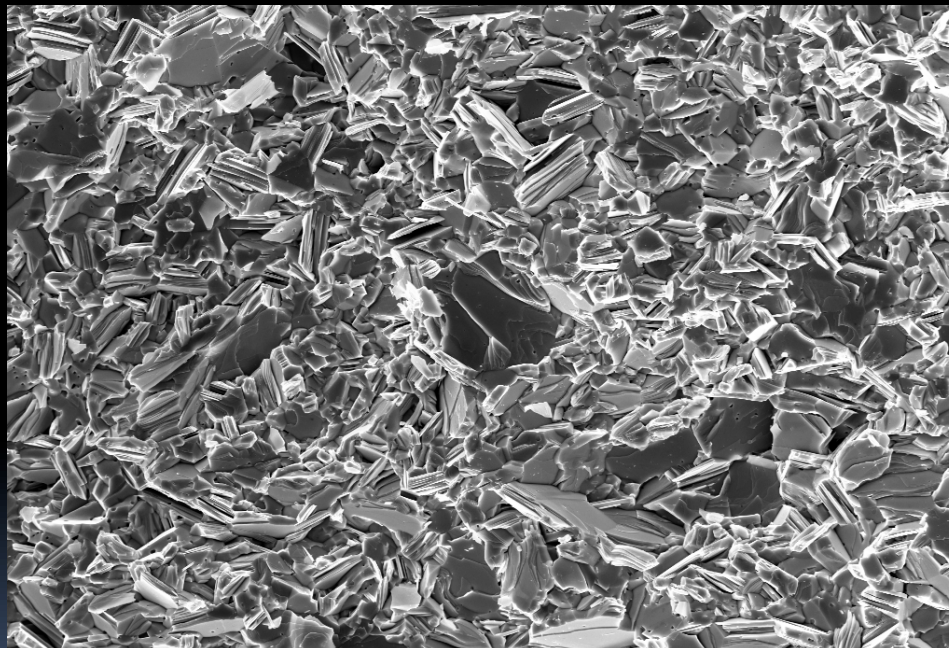
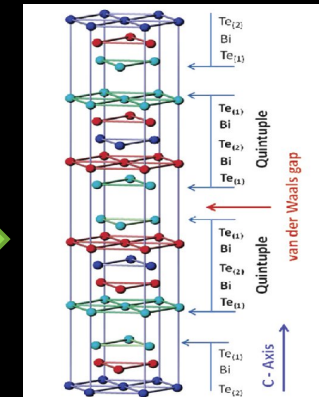


After first SPS densification



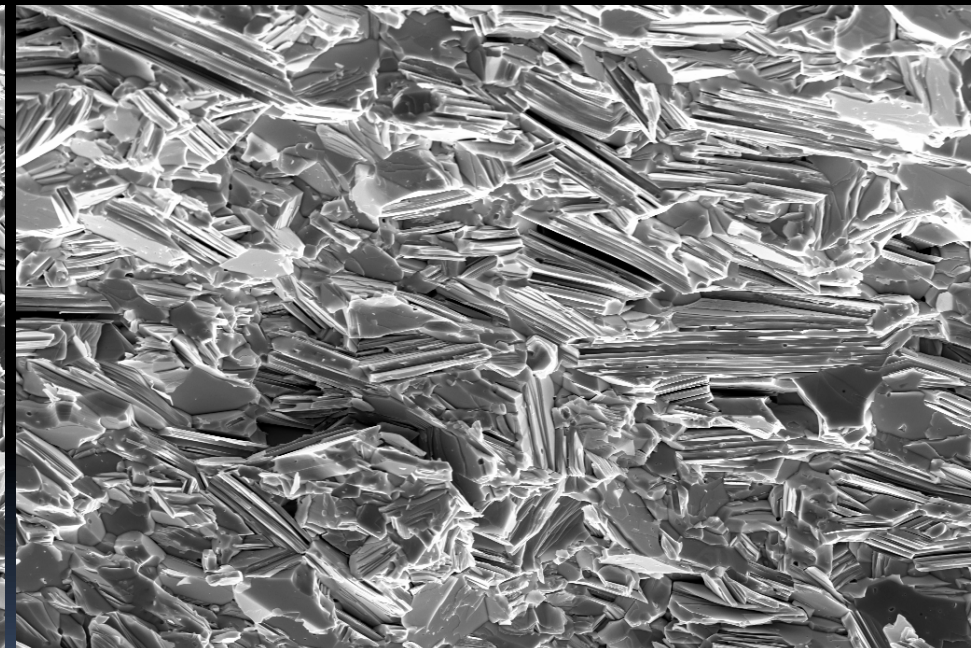
After second densification

Bi₂Te₃ nouvelle génération... microstructure



Mag = 2.00 K X 2 μm
EHT = 10.00 kV Signal A = InLens
WD = 4.8 mm Aperture Size = 60.00 μm

After first SPS densification

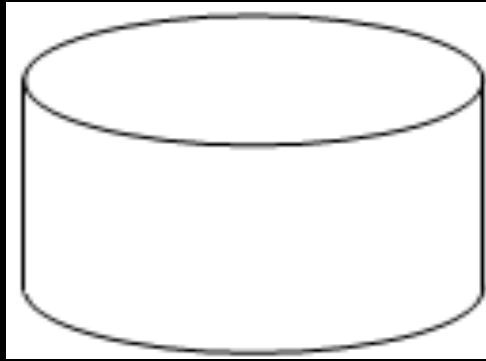


Mag = 2.00 K X 2 μm
EHT = 10.00 kV Signal A = InLens
WD = 3.7 mm Aperture Size = 60.00 μm

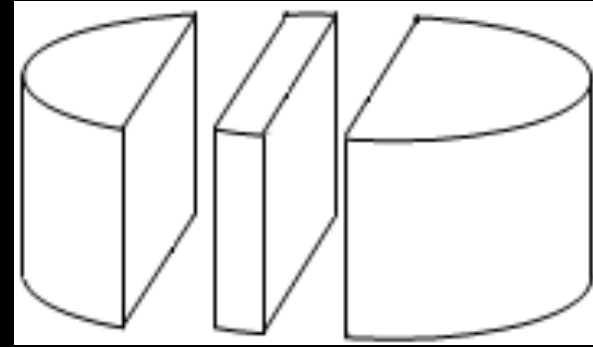
After second densification

Grains plus gros et mieux orientés

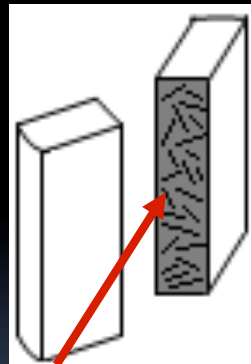
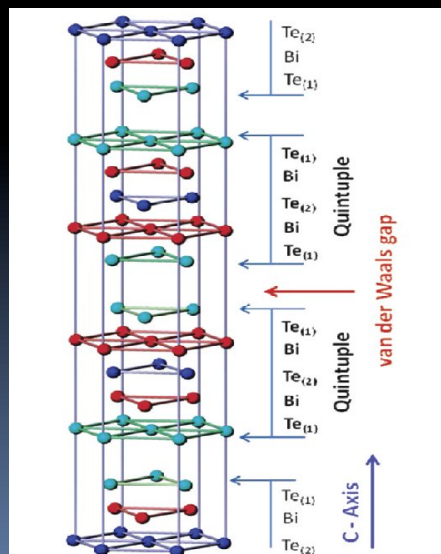
Bi₂Te₃ nouvelle génération... propriétés de transport



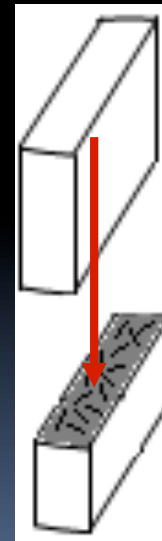
1. Puck après SPS



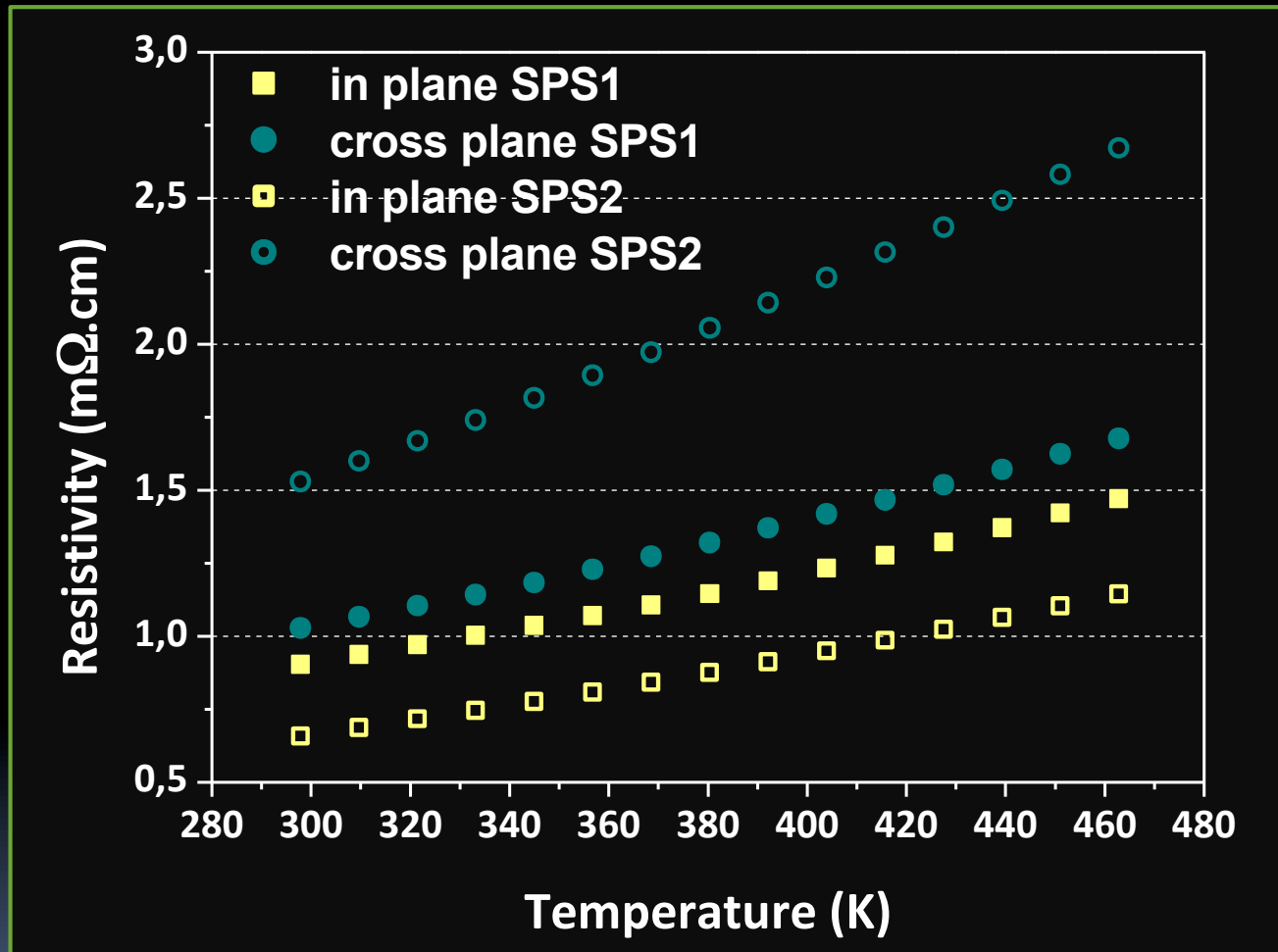
2. Découpe



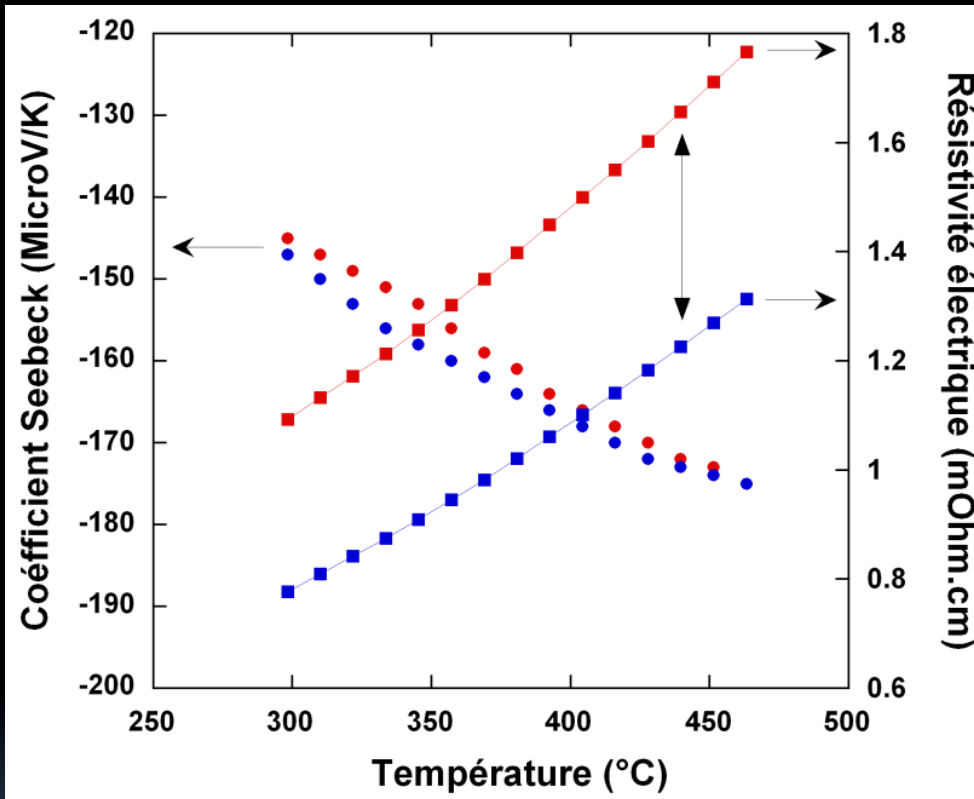
Echantillon coupé verticalement: in plane (ab)



Echantillon coupé horizontalement: cross plane



Bi₂Te₃ nouvelle génération... propriétés de transport



Seebeck is isotropic
(like in single crystals)

Resistivity decreases...

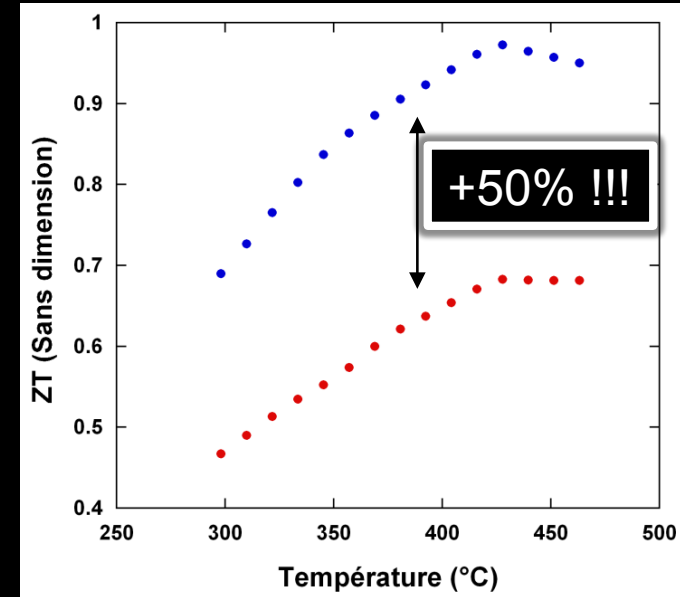
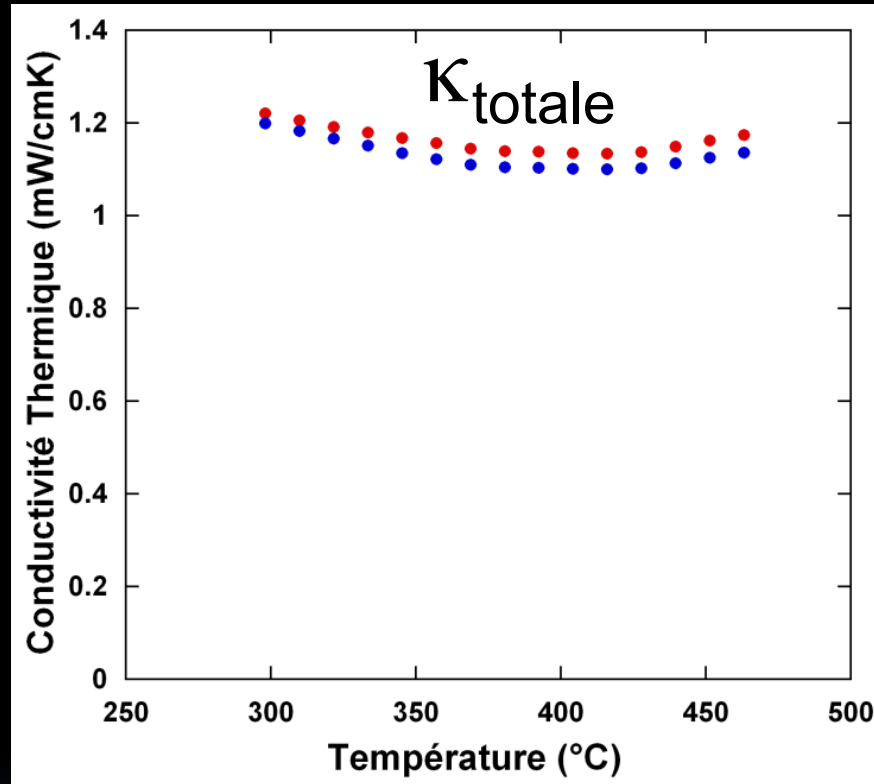


So we expect:

An increased κ_{el}
(WF law: $\kappa_{el} = LT/\rho$)

$$ZT = \frac{\alpha^2 T}{\rho \kappa}$$

Bi₂Te₃ nouvelle génération... propriétés de transport



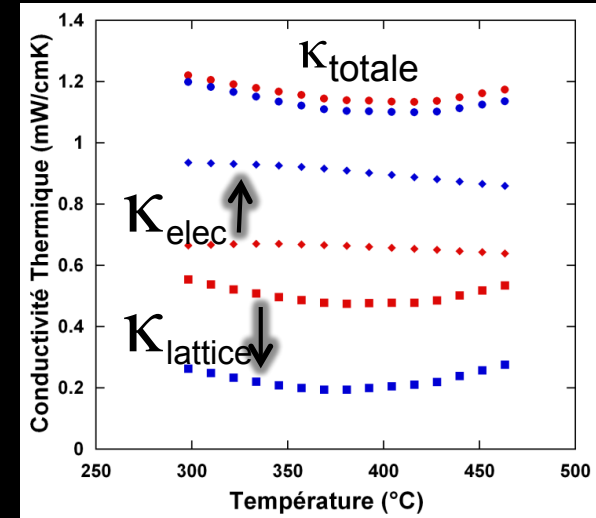
ρ decreases, α is constant, κ unchanged... ZT increase indeed !!!



Bi₂Te₃ nouvelle génération...

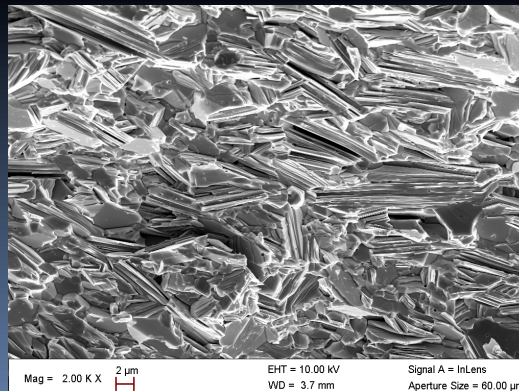
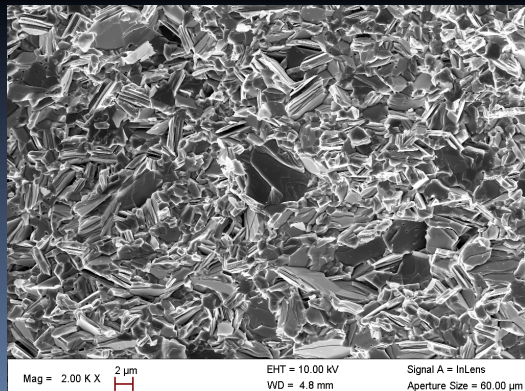
$$\kappa_{\text{totale}} = \kappa_{\text{électronique}} + \kappa_{\text{réseau}} = LT/\rho + \kappa_{\text{réseau}}$$

(Loi de Weidemann Franz)
avec $L_{(\text{Bi}_2\text{Te}_3)} \sim 2 \cdot 10^{-8} \text{ W}\Omega/\text{K}^2$

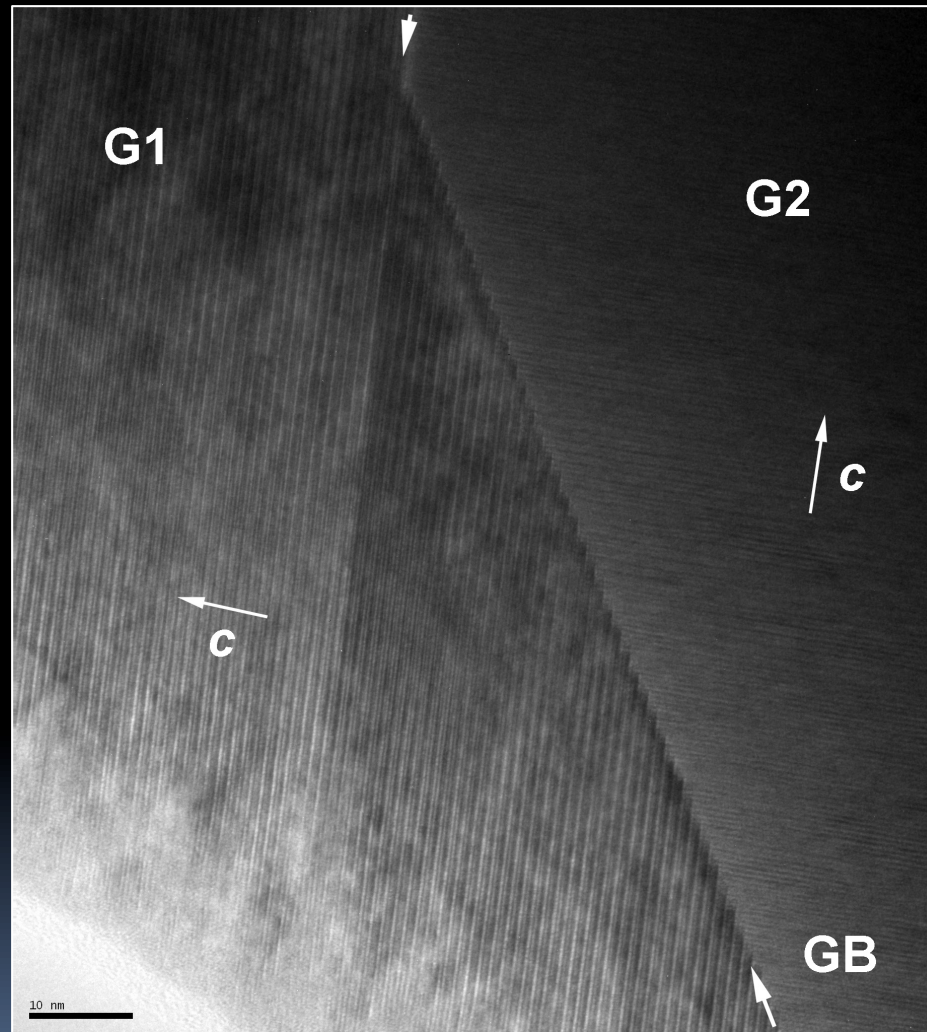


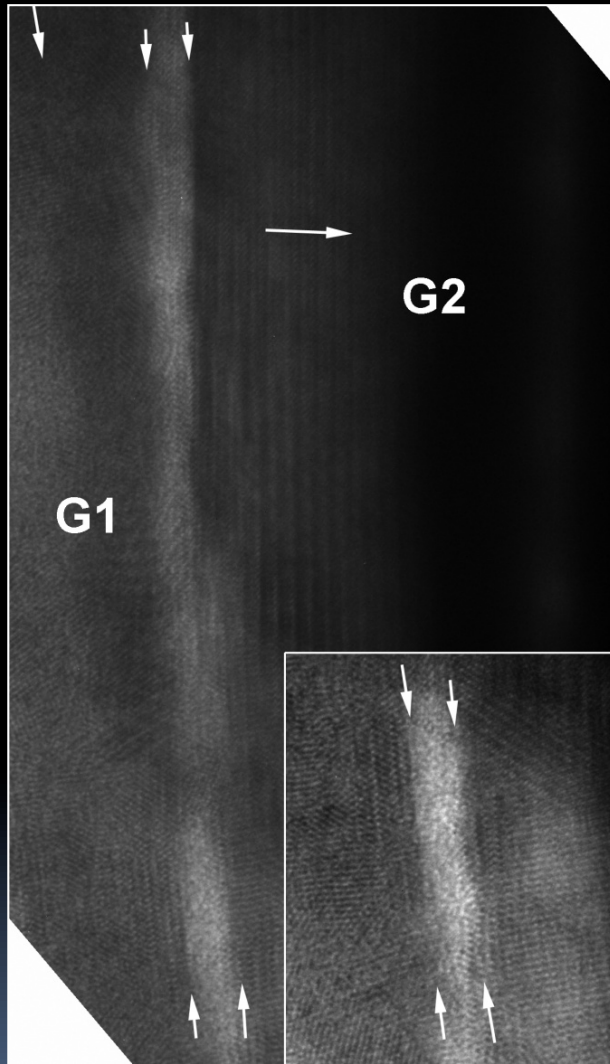
$\kappa_{\text{réseau}} \dots$

Bigger grains should lead to an increase in κ_{lattice}
(less grain boundaries)



Bi₂Te₃ nouvelle génération... TEM after first SPS





Amorphous, disordered grain boundaries...

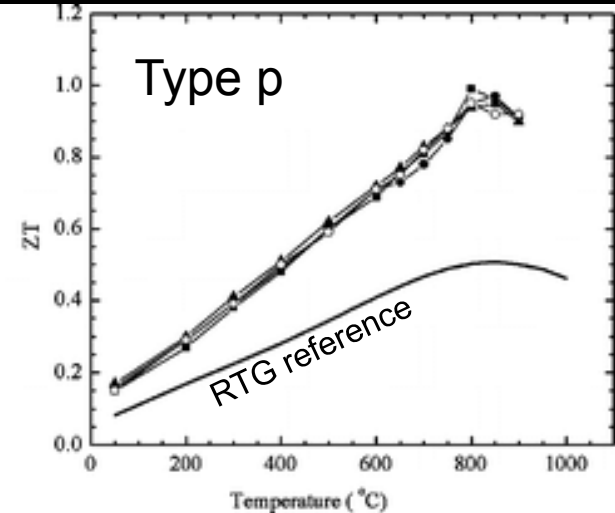
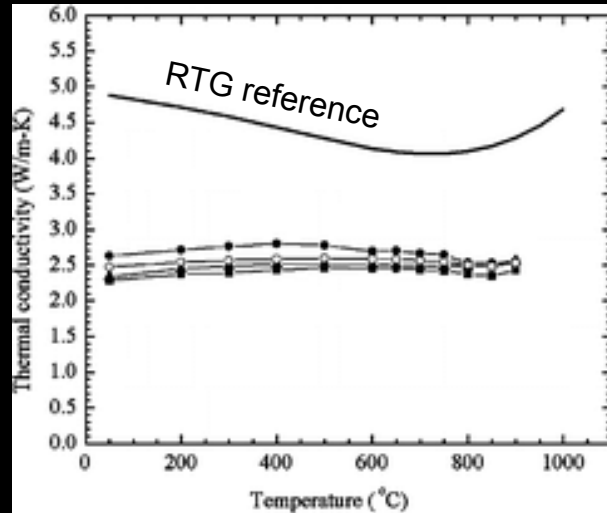
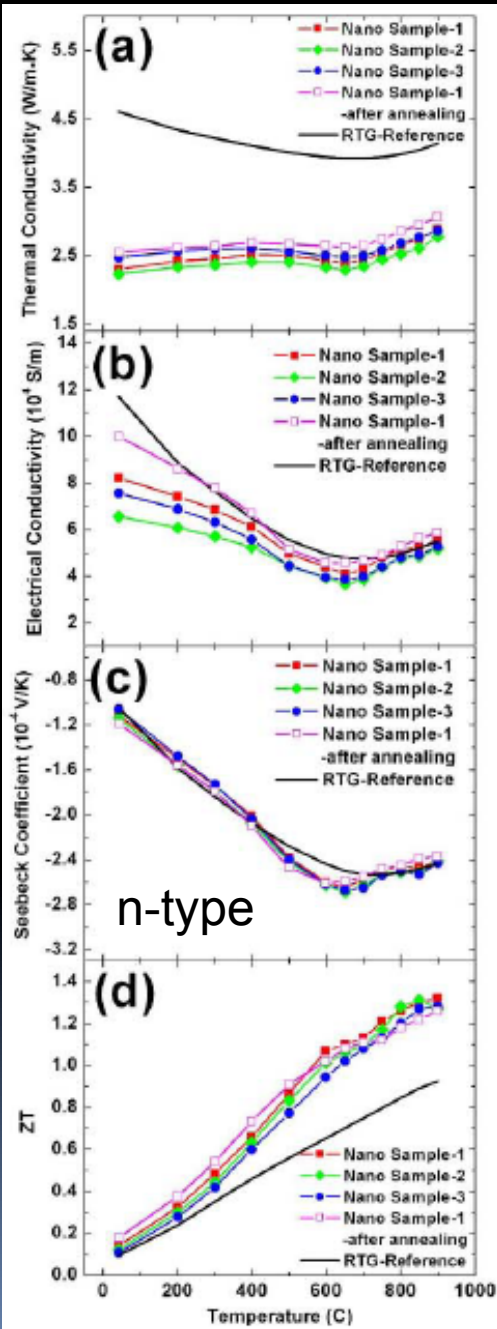
Hypothesis:

Results from the friction engendered by the second SPS, grains force to glide on each other.

This result has been reproduced numerous times!

Mechanical process results in an increased thermoelectric figure of merit.

SiGe nanostructured



Prepared by mechanical alloying, n and p type SiGe have performances well above that of the heritage SiGe !

The existence of nano domains apparently explains this decreased thermal conductivity whereas the electrical resistivity is virtually unchanged... Remarkably, this nanostructuring can withstand high temperature without evolving much.



Thierry Caillat

Summary

- **To date, thermoelectrics has been mostly applied for niche markets**
 - Using TE materials developed in the 60's
- **New development in TE materials may open up new markets**
 - Automobile, industrial processes waste heat recovery
- **Both low cost and smart system engineering are needed to make these applications viable**
- **ZT ~ 3 cooling materials would open larger markets**

We need:

Low thermal conductivity

- Complex structure
- Heavy elements
- Possible (structural) disorder
 - Mixed occupancy
 - Positional disorder
 - Impurities
 - Vacancies
 - ...

High Seebeck and electrical resistivity = Tunability of the carrier concentration

- flexible structure towards substitution, doping...
 - solid solution
 - flexible framework
 - Intercalation sites
 - Interstitial positions
 - ...

A la recherche de nouveaux matériaux TE ...?

We need:

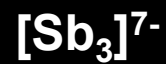
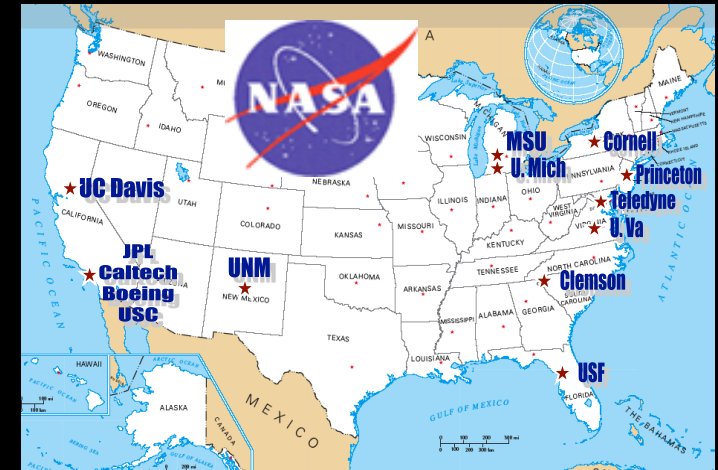
- # To make the material pure in fairly large amount
- # To be able to densify it
- # The material to be mechanically sound
- # The material to be stable towards air and moisture
- # The constituting elements to be cheap enough, available and not toxic
- # Cost / effectiveness



A la recherche de nouveaux matériaux TE ...?

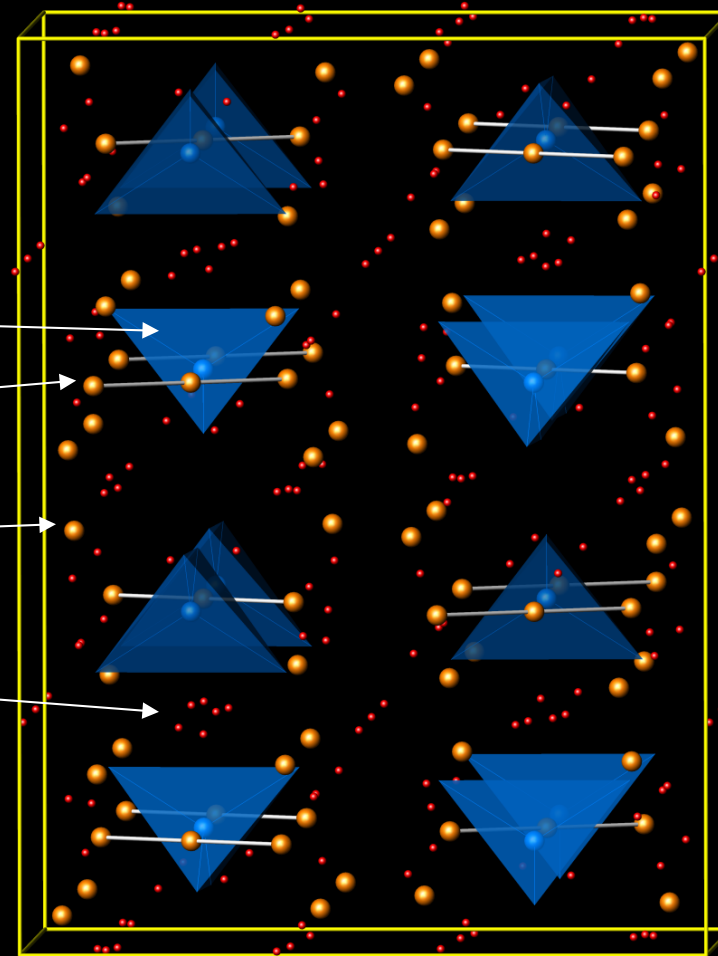
Programme STMC

14 partners, 2 years... to find a solution replacement to the heritage SiGe



$$(10+7+4 \times 3)^- + (14 \times 2)^+$$

$$29(-) + 28(+) + 1h^+$$



Yb₁₄MnSb₁₁ a naturally doped p-type HT TE material

Carrier concentration:

Calculated $[h+] = 8/6000 \cdot 10^{-24} = 1.33 \cdot 10^{21} \text{ cm}^{-3}$
Measured $[h+] \approx 1.1 \pm 0.3 \cdot 10^{21} \text{ cm}^{-3}$

ρ and α OK!

Complex 0D structure

Lattice parameters:

$I4_1/acd$

$a = 16.562(3) \text{ \AA}$

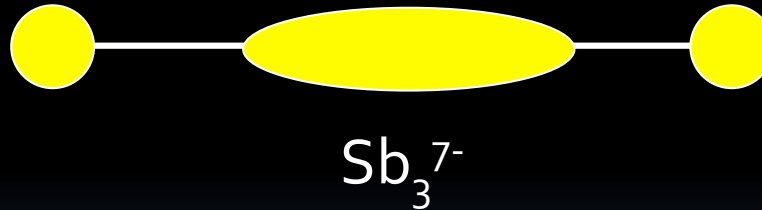
$c = 21.859(2) \text{ \AA}$

$V = 6000 \text{ \AA}^3$

$Z = 8$

+

Highly disordered Sb₃ trimer

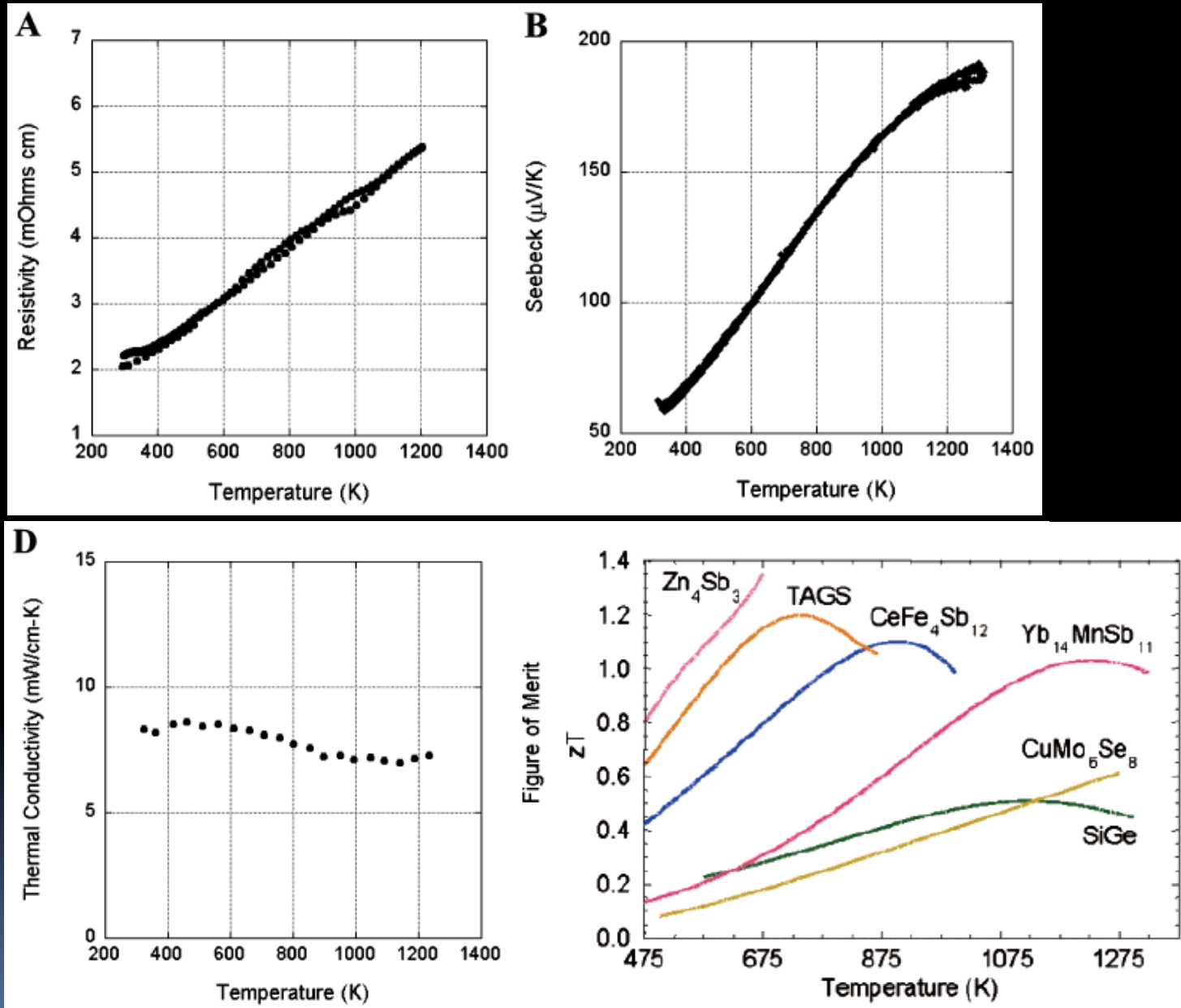


=

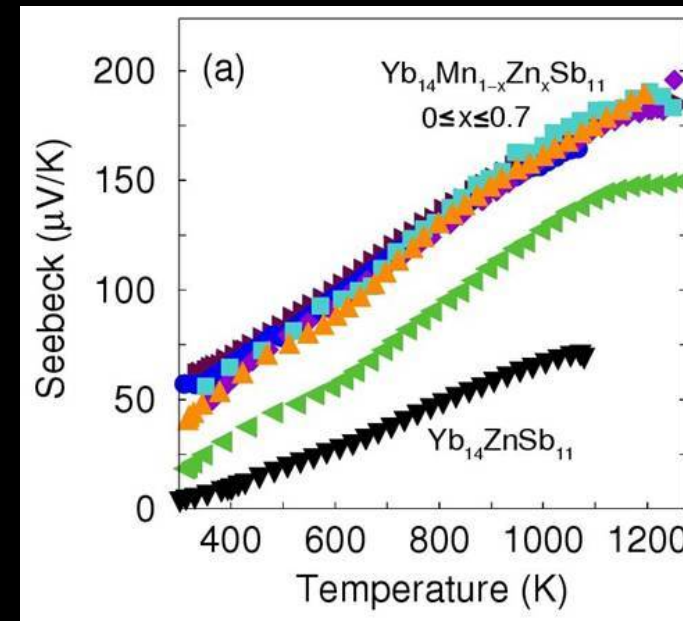
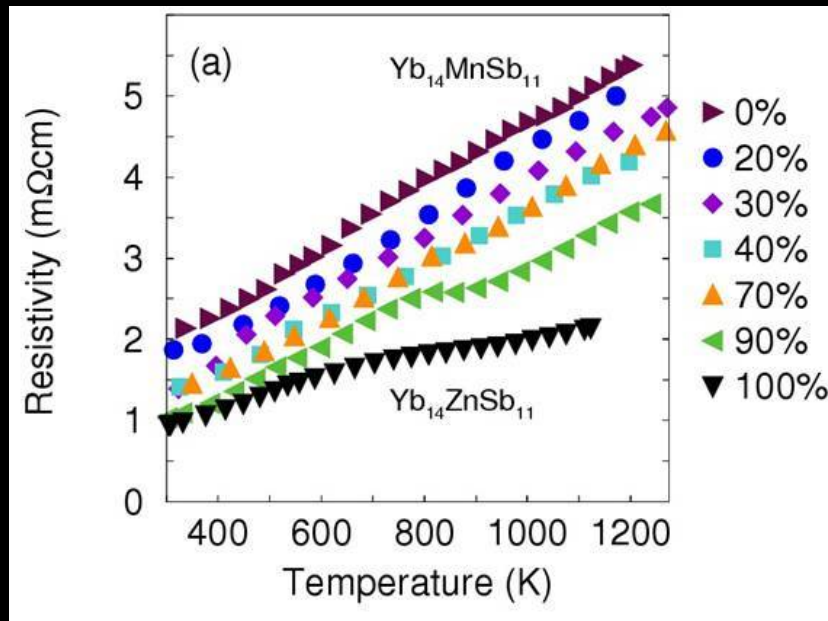
Low κ



$\text{Yb}_{14}\text{MnSb}_{11}$ a naturally doped p-type HT TE material



S. Brown, F. Gascoin, G. J. Snyder, S. M. Kauzlarich, *Chemistry of Materials*, 2006, 18, 1873.
U.S. Patent. Application Serial No. 11/470, 998



Mn and Zn both divalent



No Seebeck variation up to $x = 0.7$
No variation of $[h^+]$

Mn d^5 , Zn d^{10}



lower spin concentration



lower spin disorder scattering



Lower Electrical resistivity

Transition metal chalcogenides as TE materials...

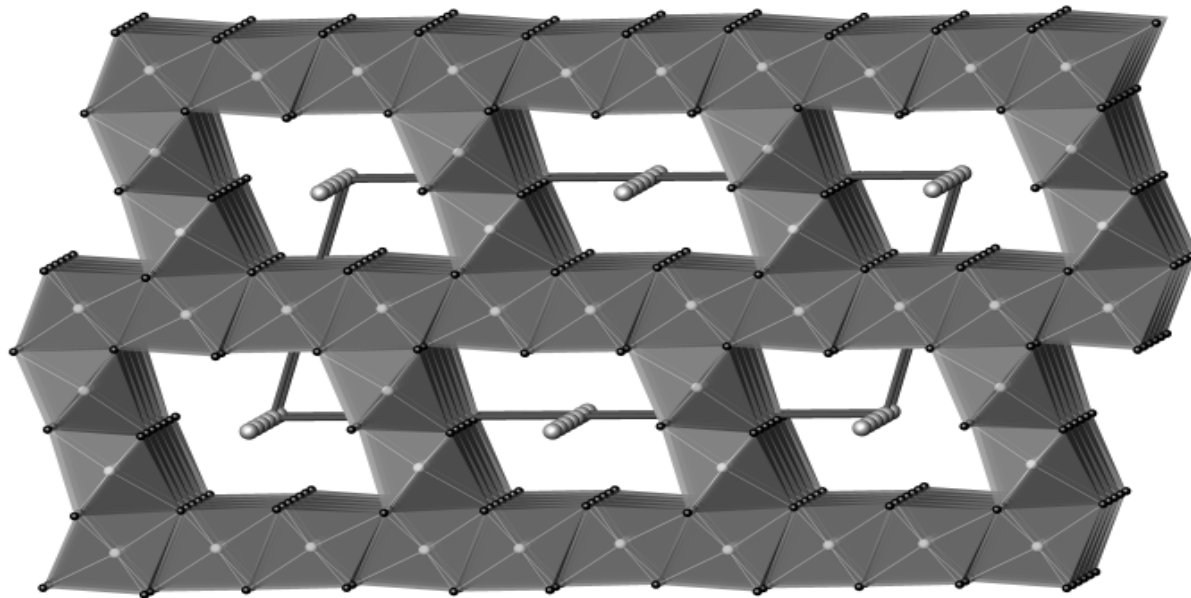
Pseudo Hollandite

CdI₂ type layered : TiX₂ or ACrX₂

Large families of compounds, ideal playground for tuning the transport properties

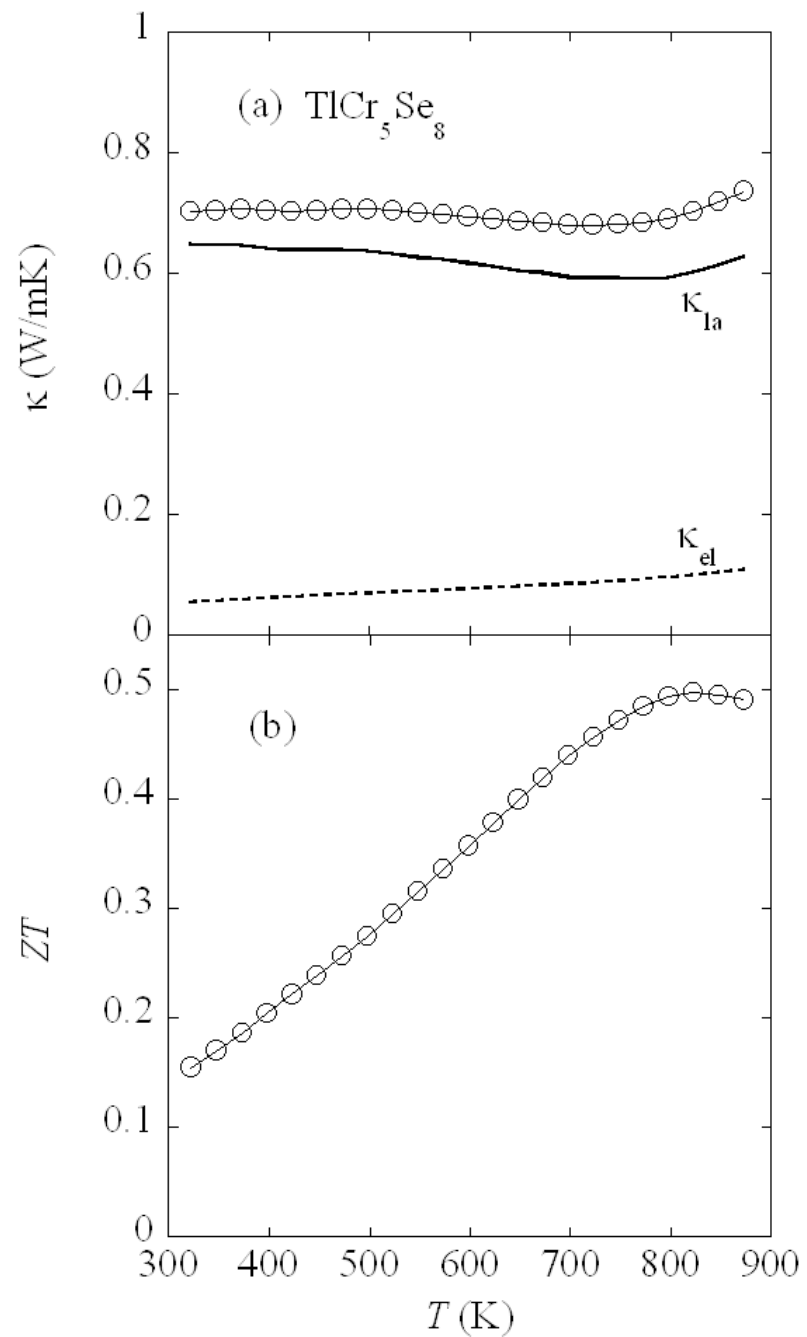
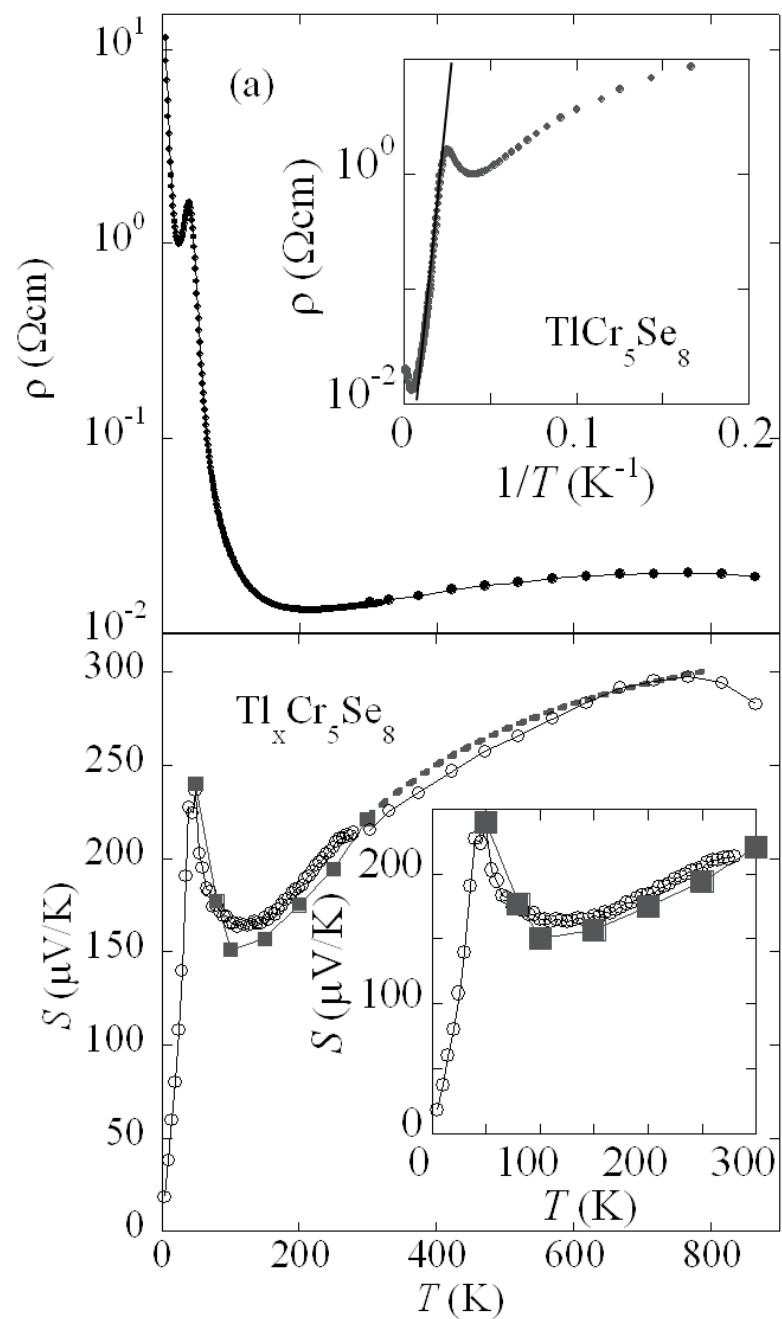
Two distinct subnetworks
Intercalation / Insertion
Solid solutions (TM and S/Se/Te)
Insulator – Metal transition likely...

Pseudo-Hollandite



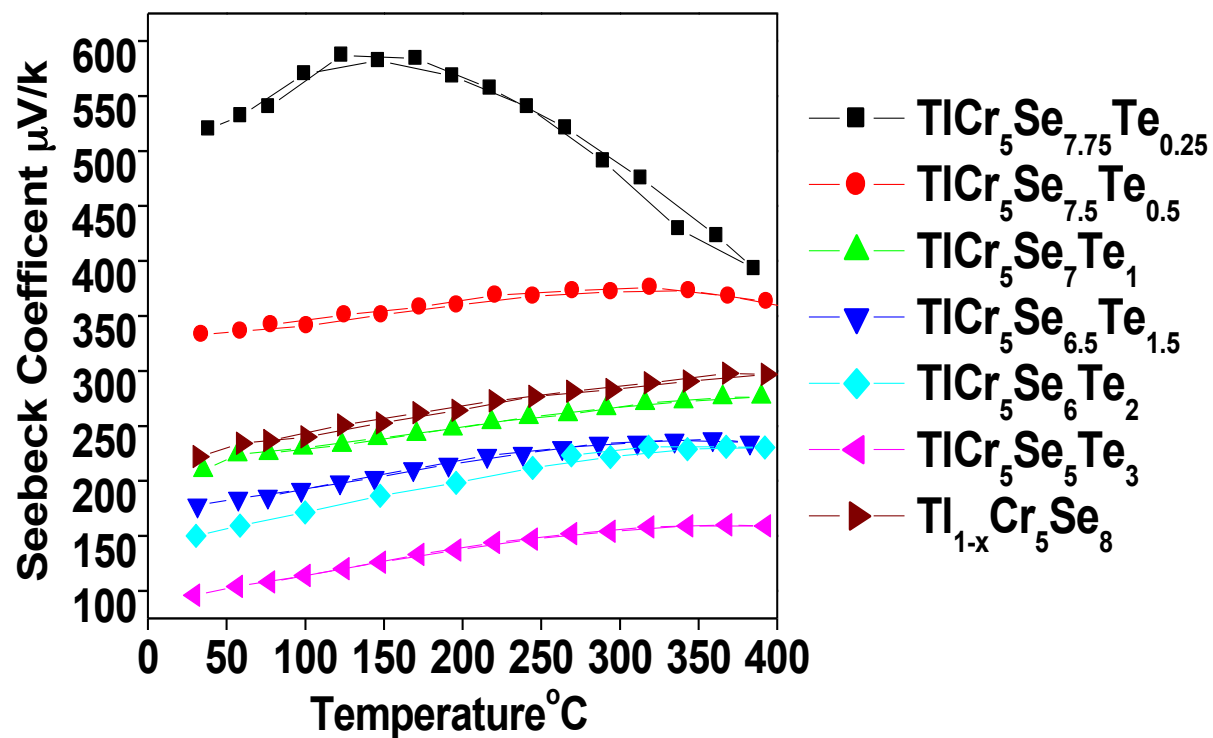
general formula $A_x B_{10} X_{16}$ with $x \leq 2$

Pseudo hollandite: $\text{Tl}_x\text{Cr}_5\text{Se}_8$ with $x < 1$

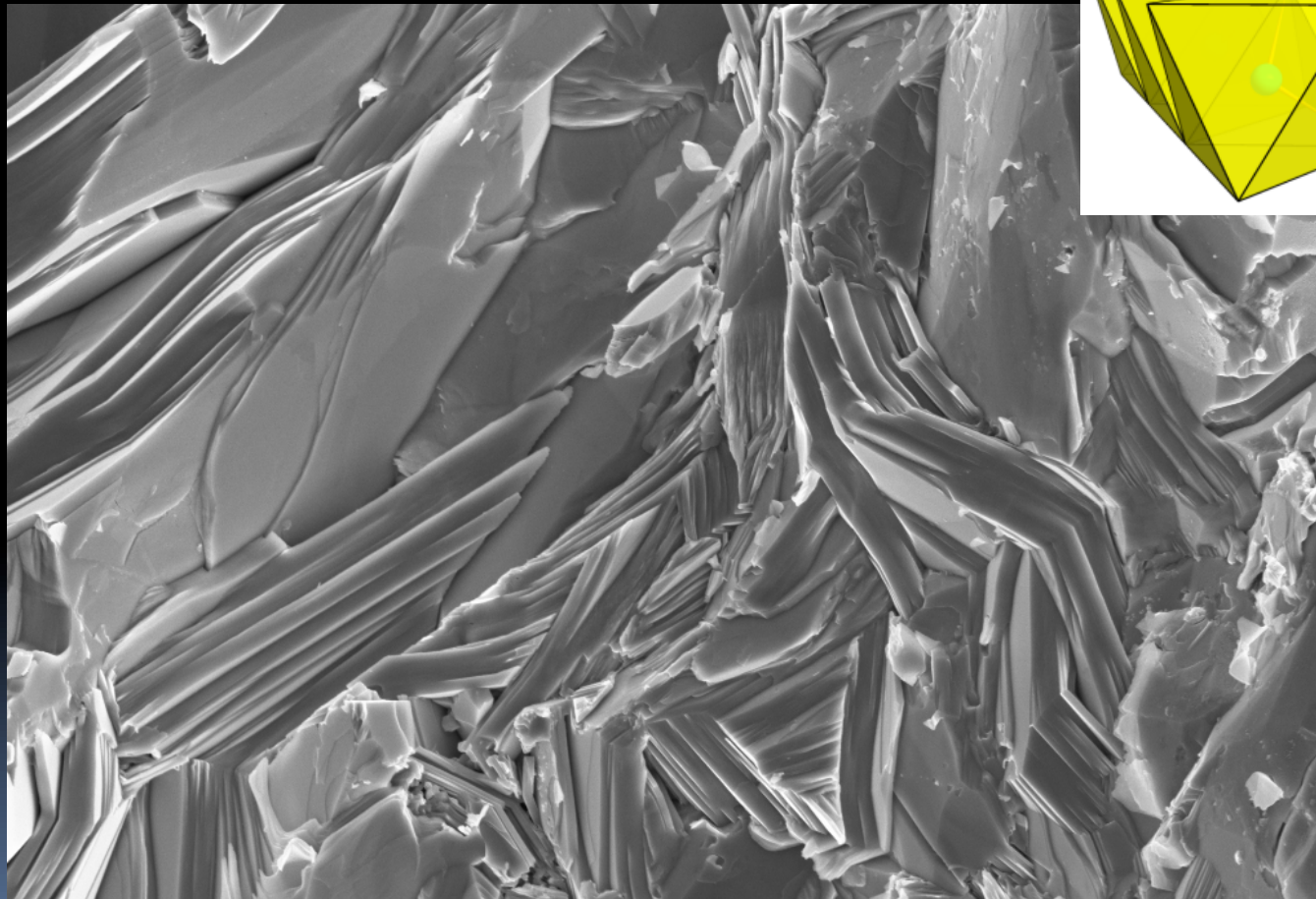
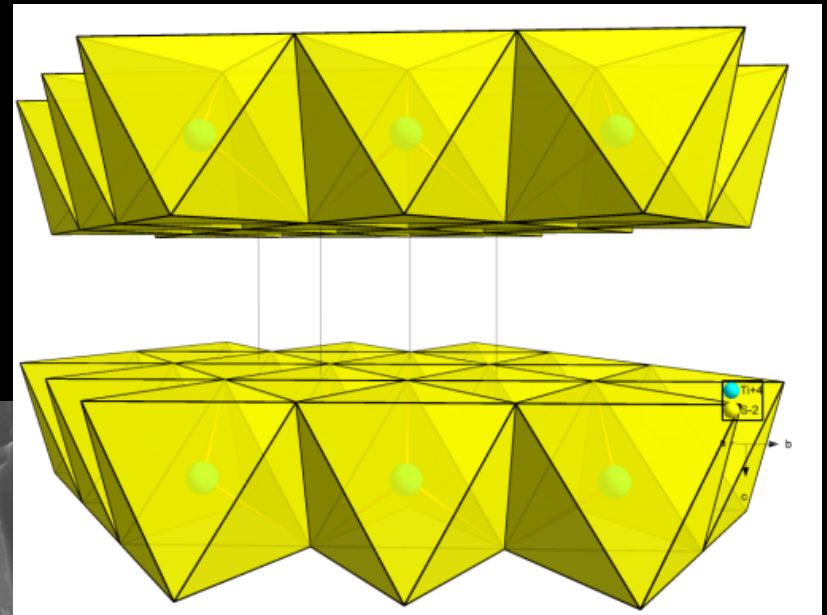



Pseudo hollandite: $Tl_xCr_5Se_{8-y}Te_y$

Going down the column, the metallicity increases...



Solid solution $\text{TiS}_x\text{Se}_{2-x}$



Mag = 6.66 K X 

EHT = 5.00 kV
WD = 5.2 mm

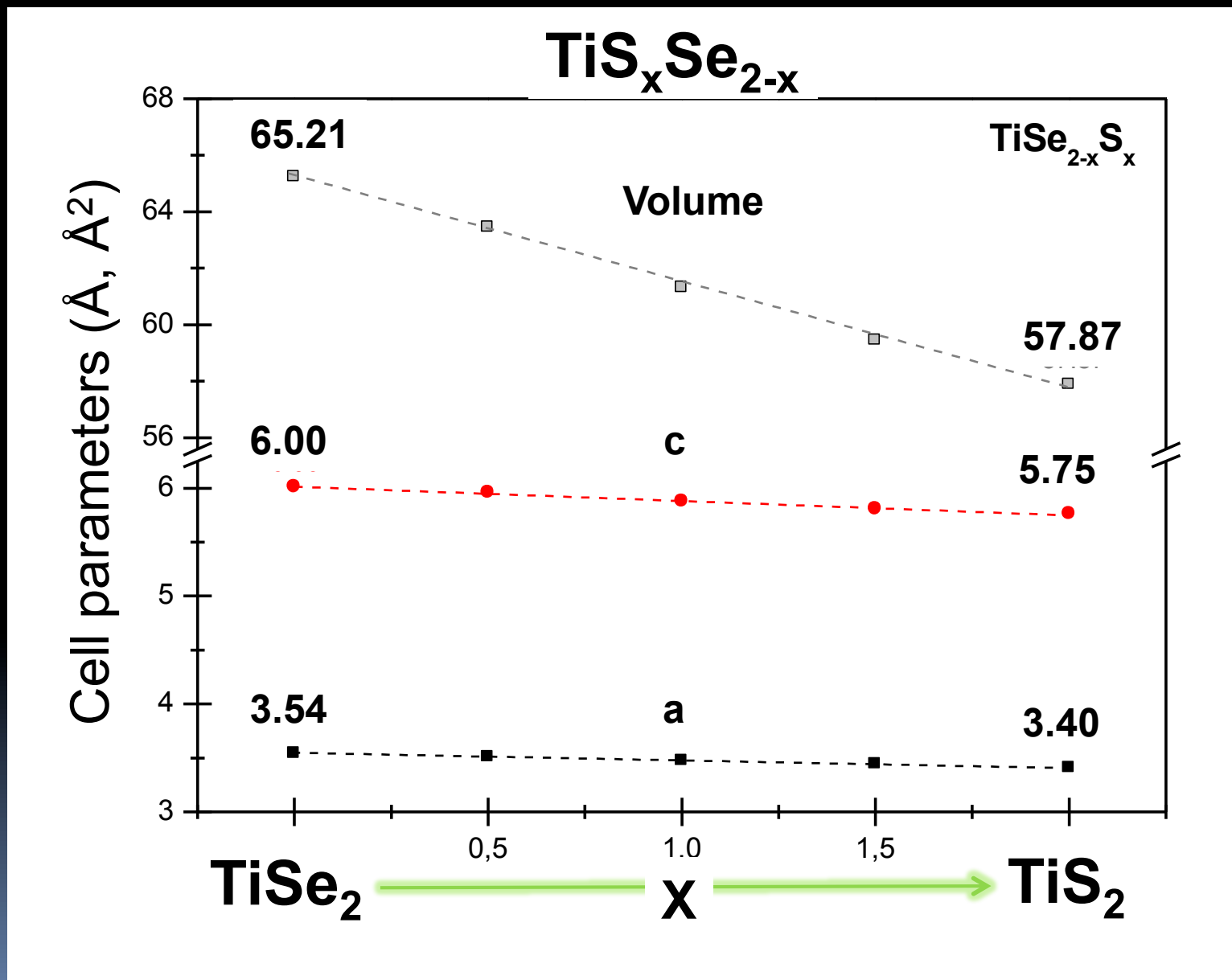
Signal A = InLens
Aperture Size = 30.00 μm

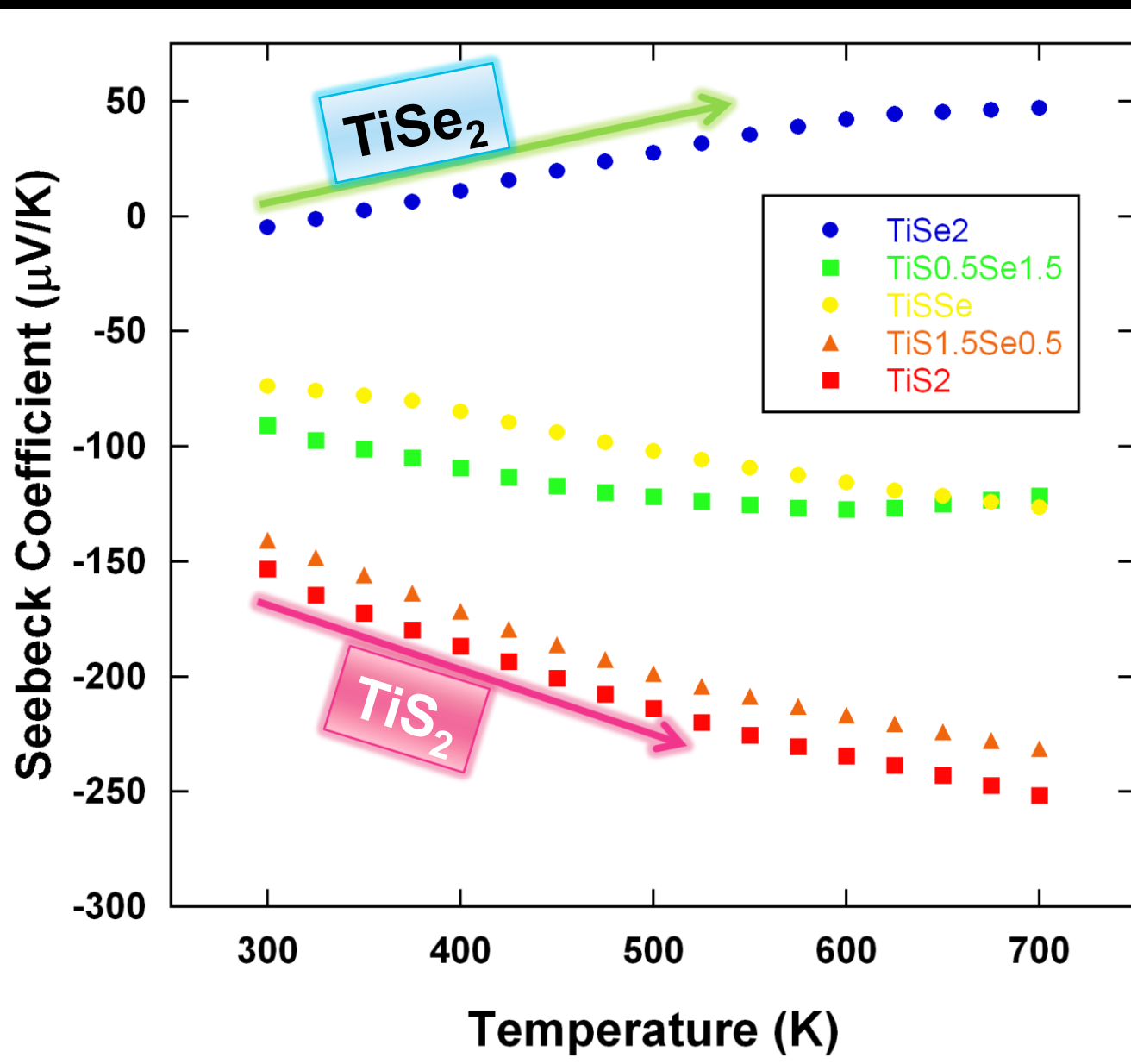
Synthesis:

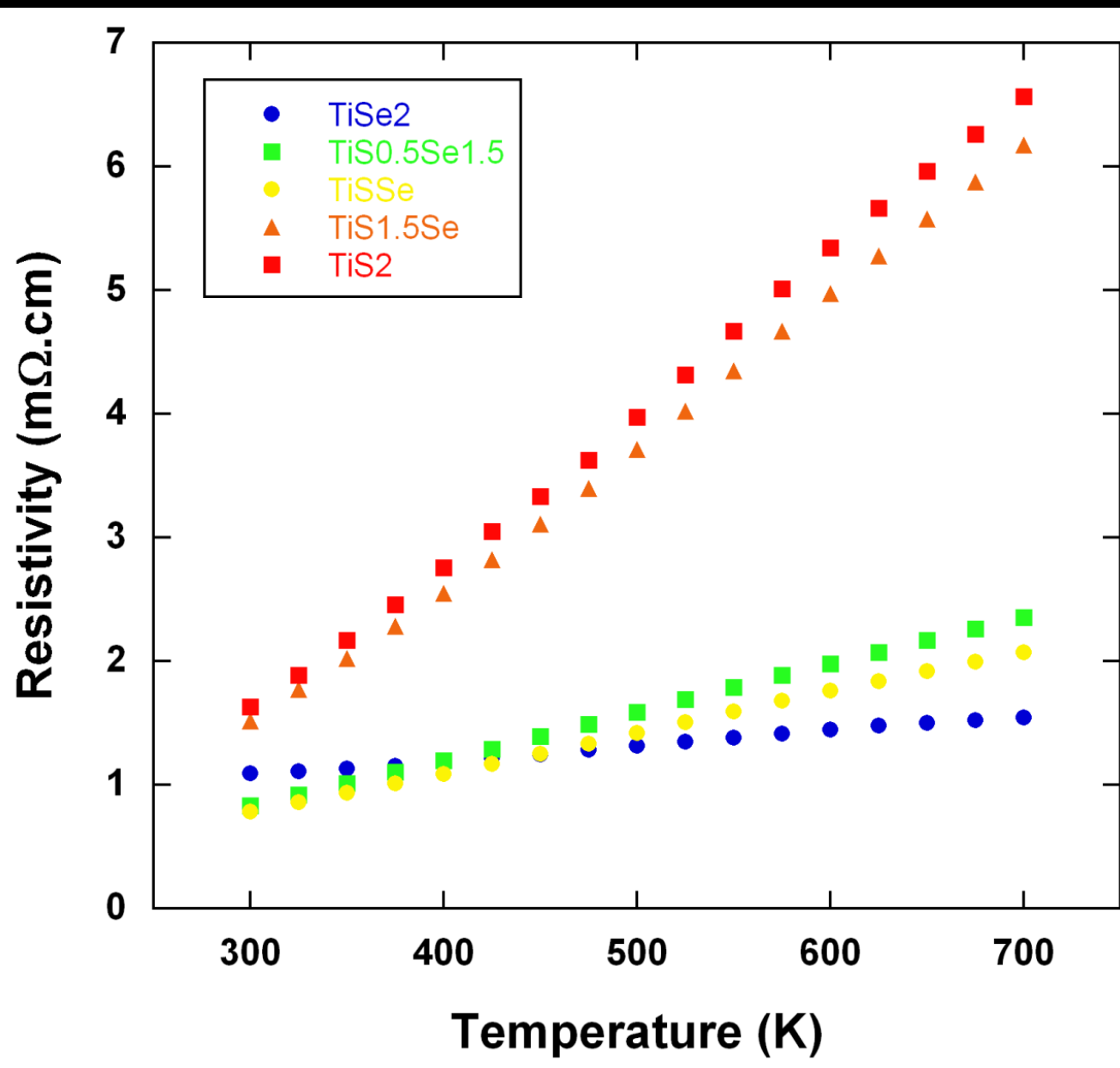
Raw elements in sealed tubes

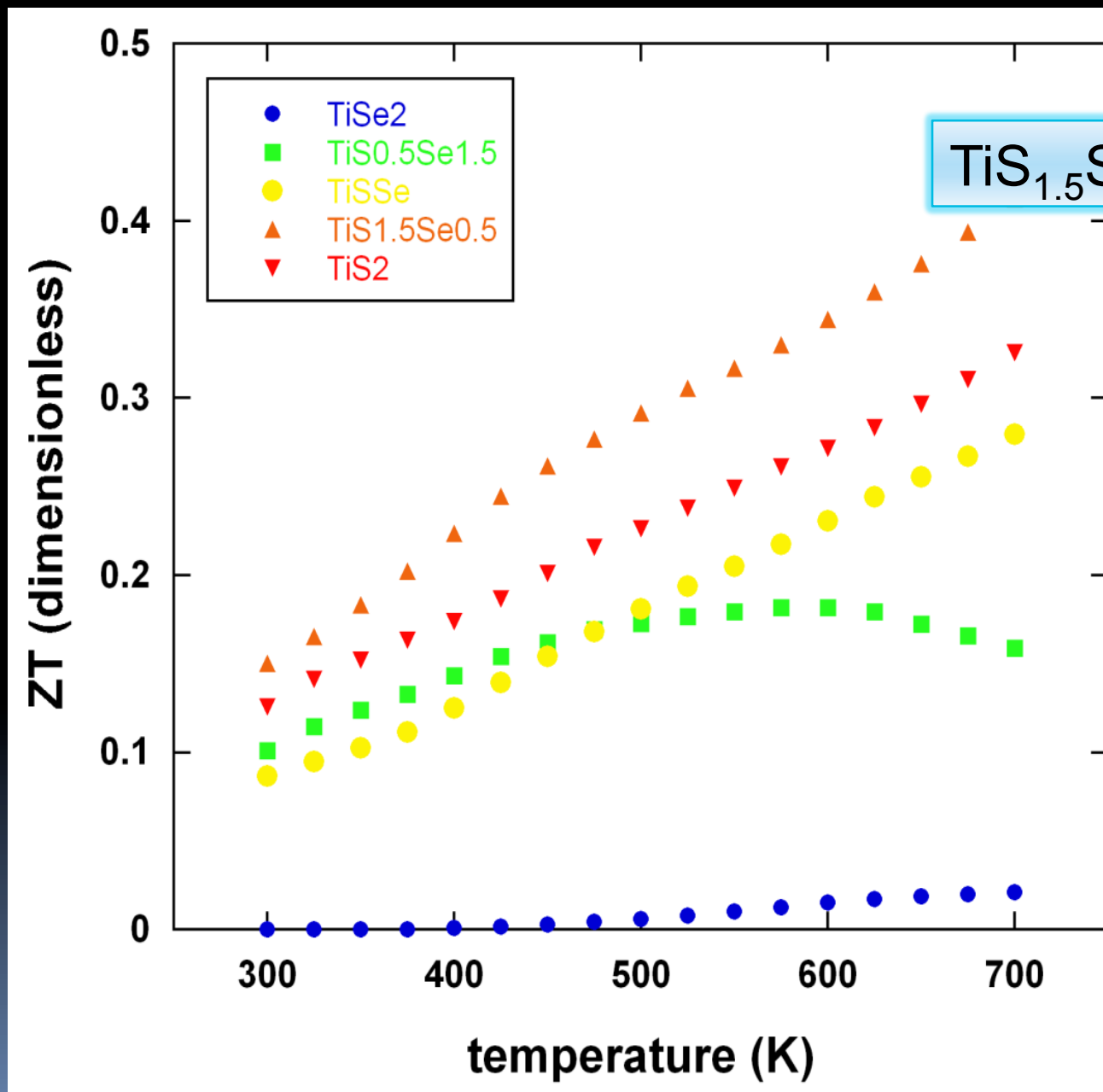
Processing:

SPS densification



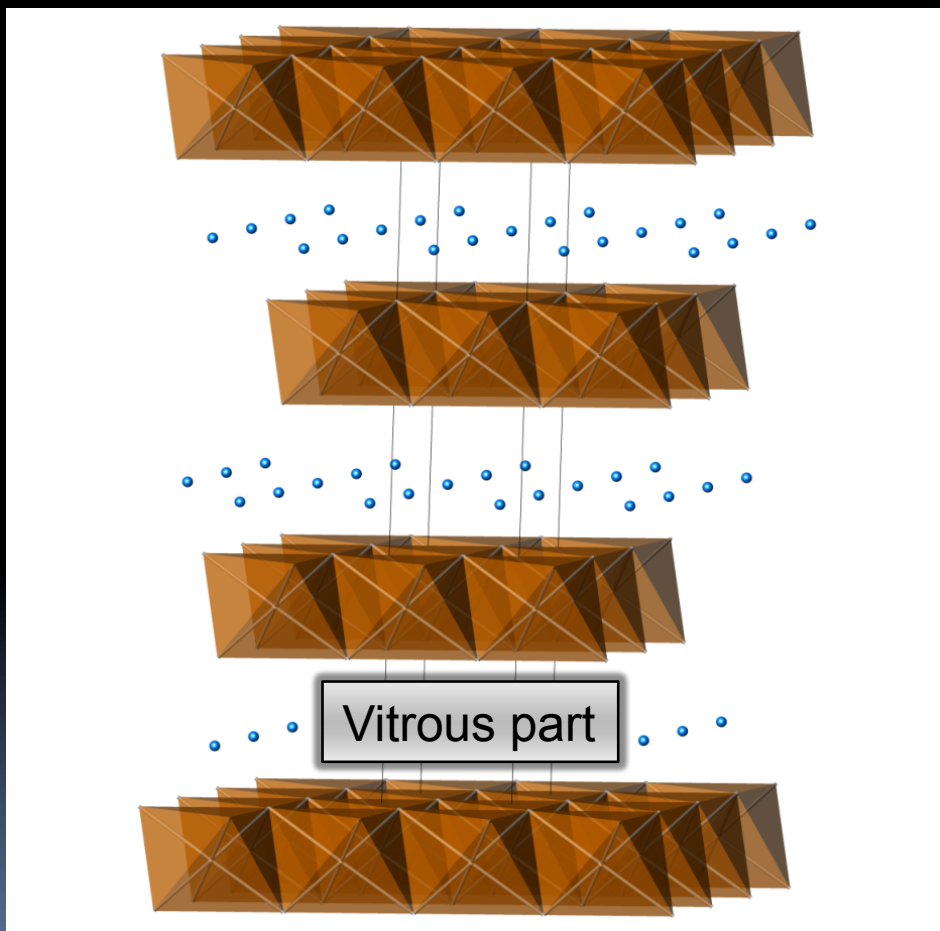






New ideas, new TE materials... ?

AgCrSe_2 , *a priori* a good thermoelectric ...



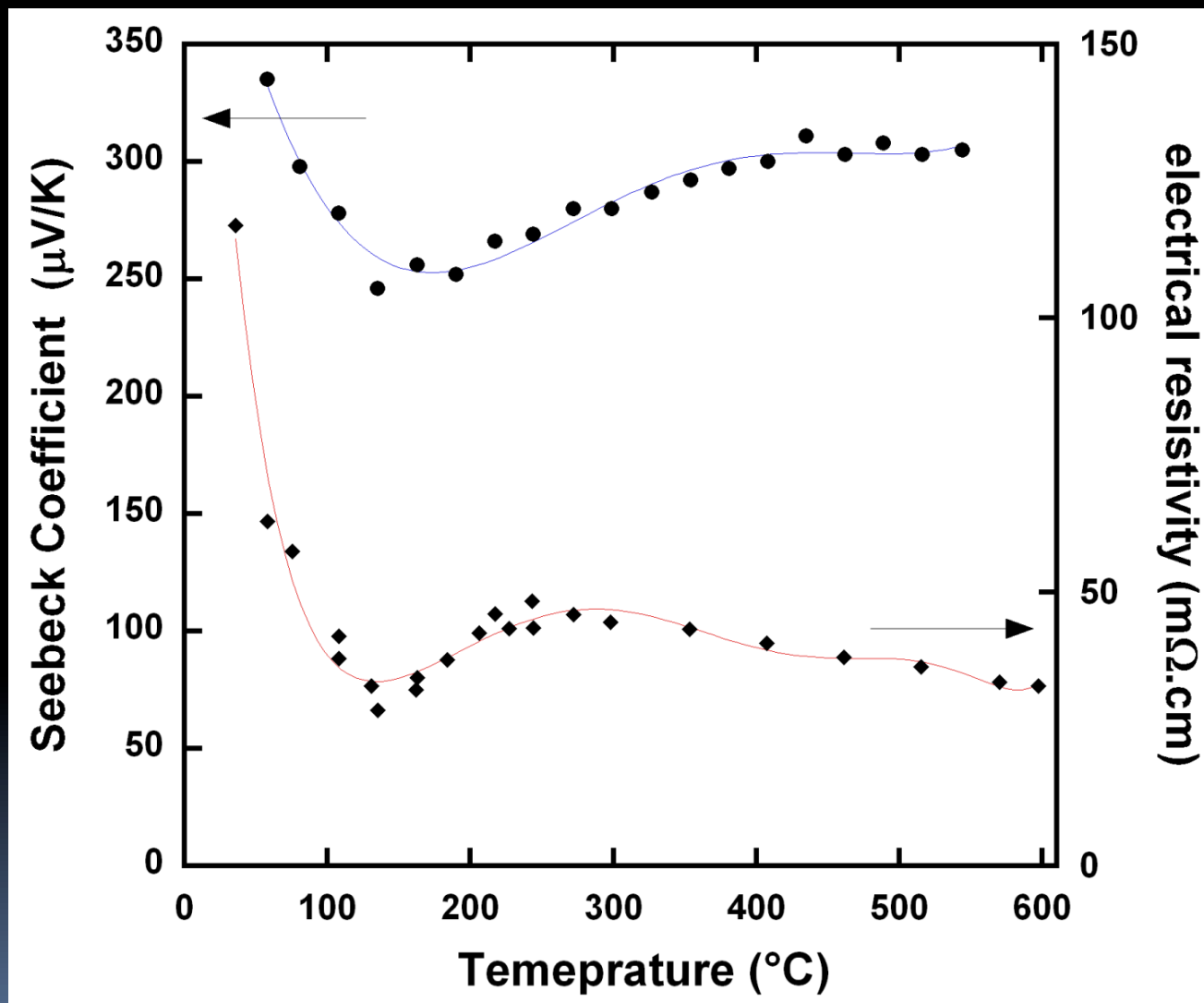
Electronic Conduction in the layer
+
Disorder between the layers
+
Many (!!!) compounds possible
=
Good choice for TE material ...

CdI_2 type CrSe_2 layers

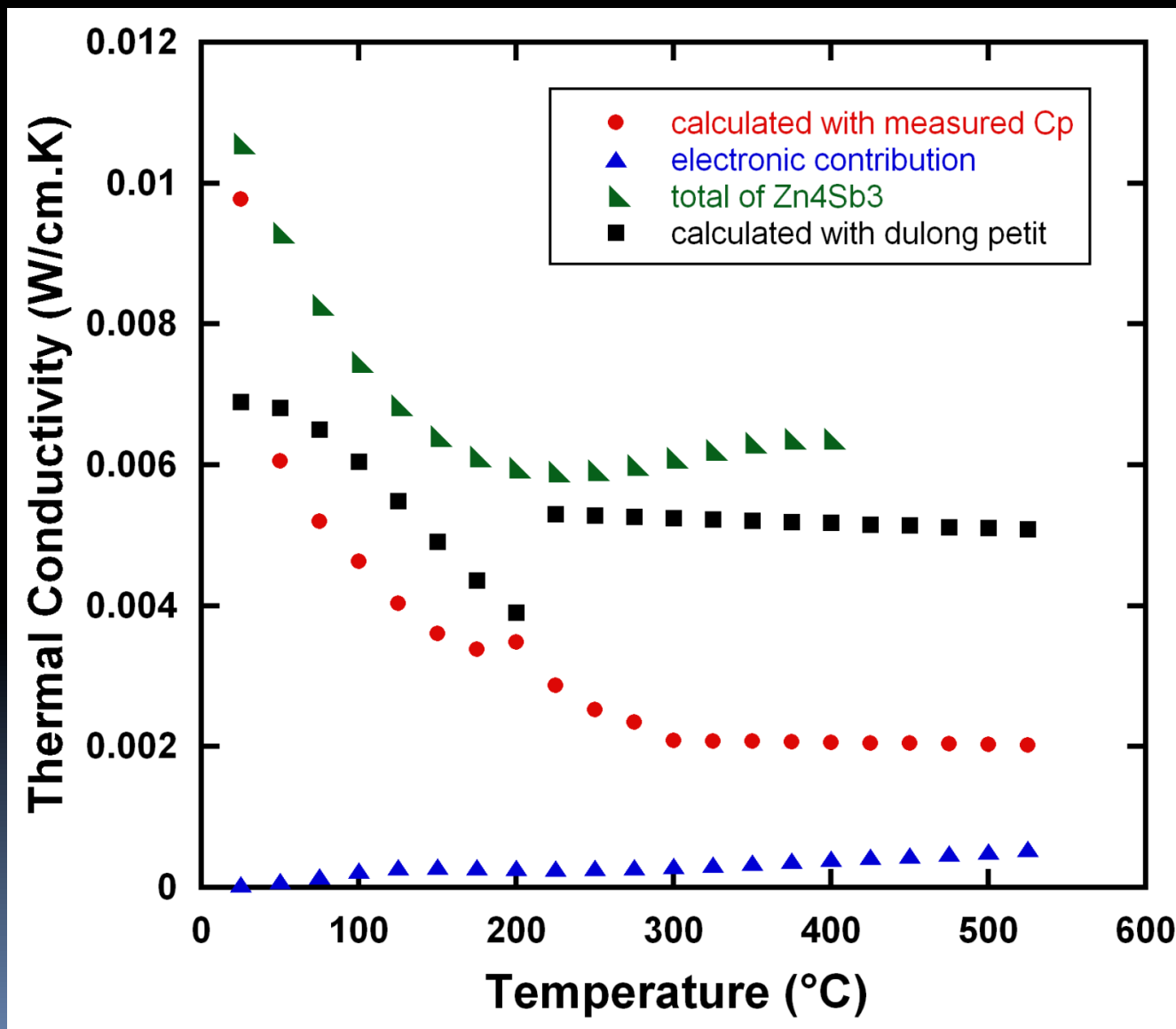
Highly disordered Ag atom in
the interlayer spacing

AgCrSe₂, *a priori* a good thermoelectric ...

Huge Seebeck and « large » resistivity



Extremely low thermal conductivity



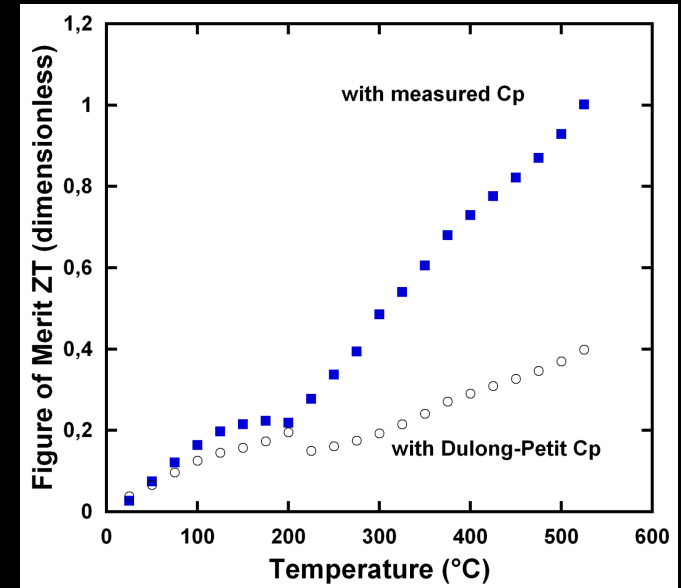
AgCrSe₂, *a priori* a good thermoelectric ...

As expected, very low thermal conductivity
(and that's this only think we could forecast !)

High Seebeck

BUT : Ionic conductor, hence impossible to use in a device...

However, the concept might be fruitful !?



To be continued...

Concluding remarks

There are good TE material out there ... we just need to find them !!!

- $\text{Yb}_{14}\text{MnSb}_{11}$ out of the blue...

Known structure types hide special and interesting compositions
(and chemists know crystallographic structures...)

- Transition metal chalcogenide and oxychalcogenides

New materials (structures!) must be looked for...

Applications/Industrials are waiting for us



Concluding remarks



- # TE Materials research must be accompanied by module conception and system design
- # For each (new) material, the making of a module has to be thought over!
- # A solution replacement for Bi_2Te_3 is far, very far !!! (ZT of 3 ...???)
- # ZTs around 0.5 – 0.8 are easy to find, but not enough...
- # Cheap, non toxic, light, robust, long-life, efficient thermoelectric materials and device are the key to further viable applications