

Thin-Film Alchemy: Using Epitaxial Engineering to Unleash the Hidden Properties of Oxides

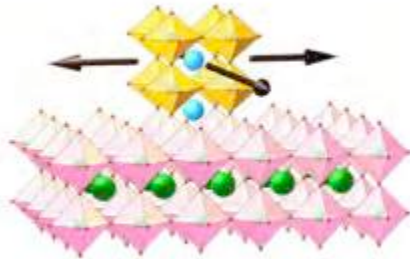
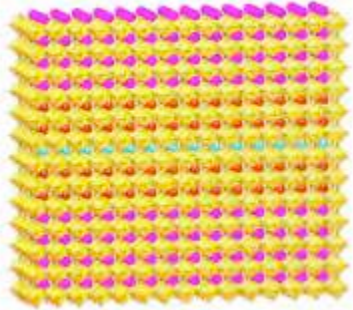
Darrell G. Schlom

*Department of Materials Science and Engineering
Cornell University*

Kavli Institute at Cornell for Nanoscale Science

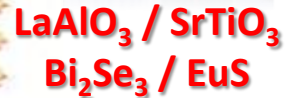
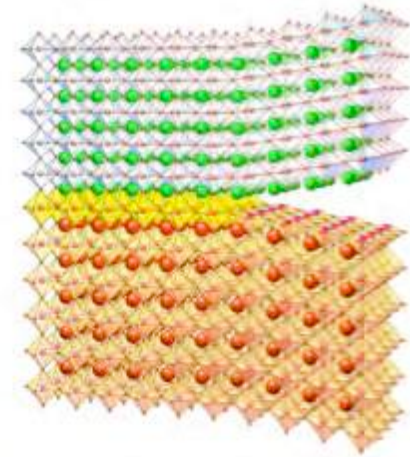
Unleashing Hidden Properties

Interface Engineering



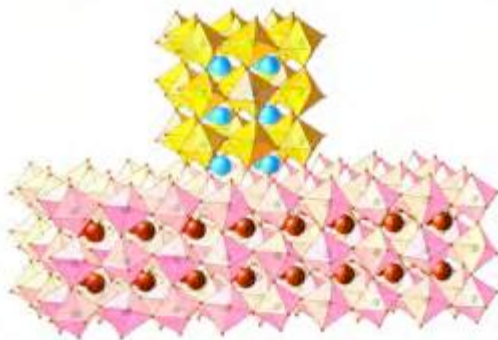
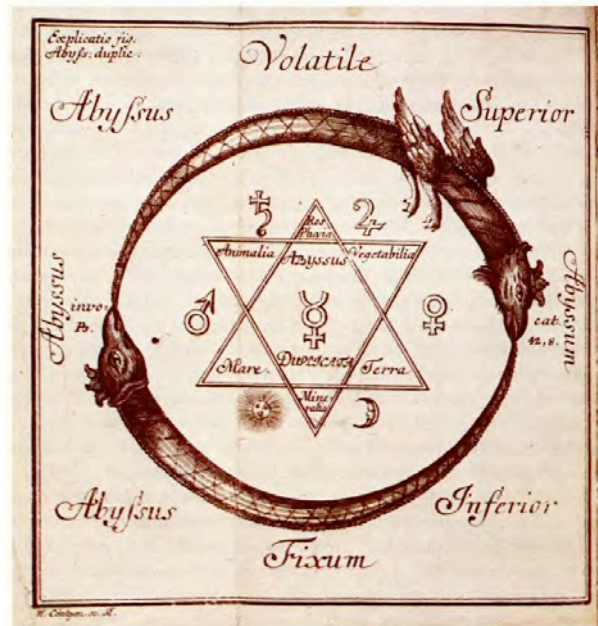
Strain Engineering

$\text{FeSe} / \text{SrTiO}_3$

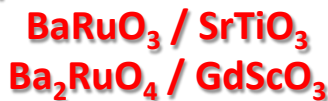


Polarization Doping & Proximity Effects

(from juxtaposed competing ground states)



Epitaxial Stabilization



Dimensional Confinement



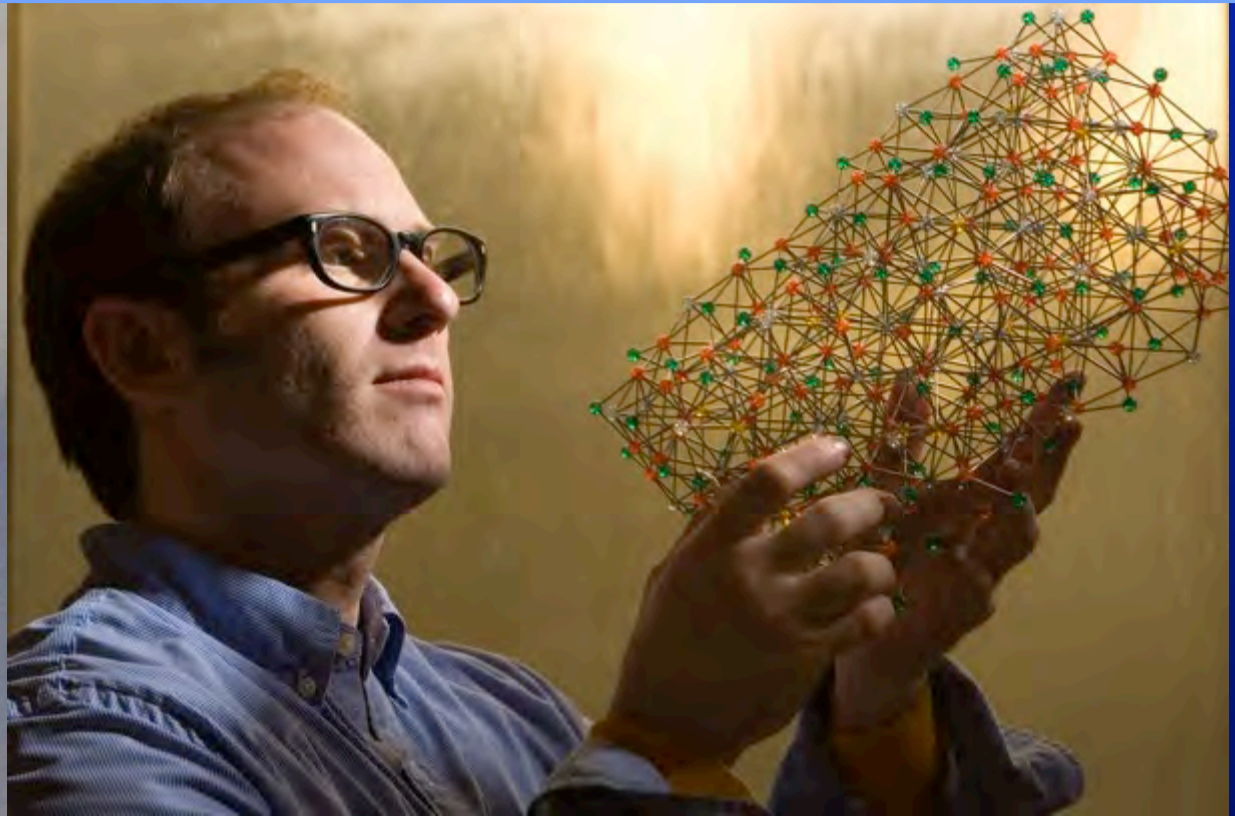
Unleashing Hidden Properties

- Toolkit Enabling Materials-by-Design of Oxide Thin Films
- Ruthenates— BaRuO_3 , $(\text{Ca,Sr,Ba})_2\text{RuO}_4$
altering band structure and properties
- EuTiO_3
strongest ferromagnetic ferroelectric
- $(\text{LuFeO}_3)_9 / (\text{LuFe}_2\text{O}_4)_1$ Superlattices
strongest ferromagnetic ferroelectric
at room temperature

The Sorcerers

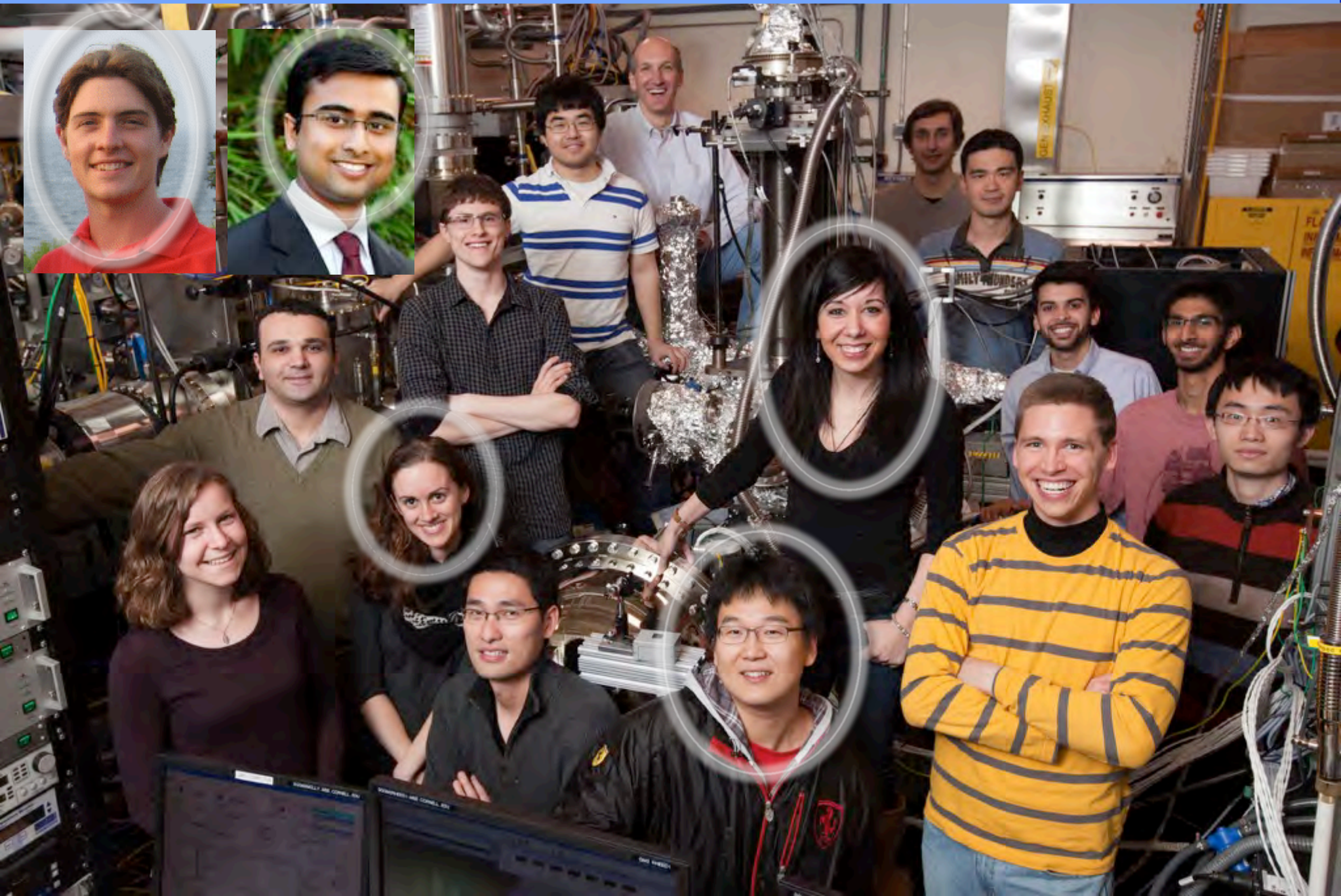


Karin Rabe (Rutgers)



Craig Fennie
(Cornell)

The Sorcerer's Apprentices



In Collaboration with the Groups of:

Craig J. Fennie—*Cornell University*

Karin M. Rabe—*Rutgers University*

Kyle M. Shen—*Cornell University*

David A. Muller—*Cornell University*

Lena F. Kourkoutis—*Cornell University*

Peter Schiffer—*University of Illinois*

Ramamoorthy Ramesh—*University of California, Berkeley*

Andreas Scholl—*Lawrence Berkeley National Laboratory*

Elke Arenholz—*Lawrence Berkeley National Laboratory*

Ezekiel Johnston-Halperin—*Ohio State University*

Venkatraman Gopalan—*Penn State University*

Stanislav Kamba—*Institute of Physics, Czech Republic*

Julie A. Borchers—*NIST*

William D. Ratcliff—*NIST*

Jürgen Schubert—*Forschungszentrum Jülich GmbH*

Reinhard Uecker—*Leibniz Institute für Kristallzüchtung*

Important Synthesis Rules

- Gibbs' Rule
 $\Delta G < 0$ to form stable phases
- Matthias's Rules for Superconductors
... "Stay away from Theorists"
- Pauling's Rules for Crystal Structures
Radius ratio criteria for stability

Rules for QM Synthesis

- Gibbs' Rule
 $\Delta G < 0$ to form stable phases
- Matthias's Rules for Superconductors
... "Stay away from Theorists"
- Pauling's Rules for Crystal Structures
Radius ratio criteria for stability

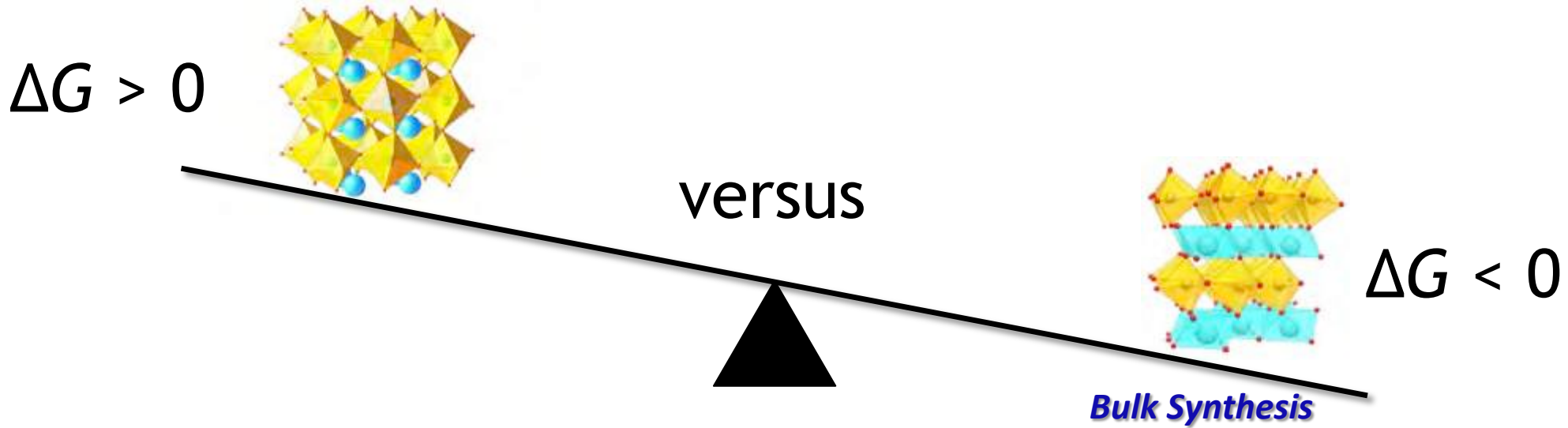
BRILLIANT BUT USELESS!

BREAK THE RULES

- Gibbs' Rule

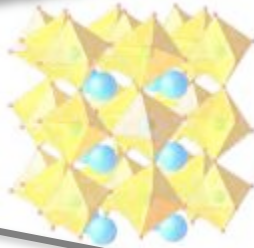
$\Delta G < 0$ to form stable phases

Gibbs' Rule

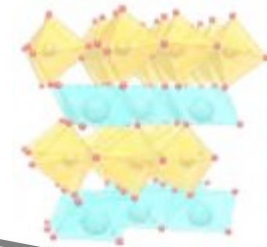


BREAK Gibbs' Rule

$$\Delta G > 0$$



versus

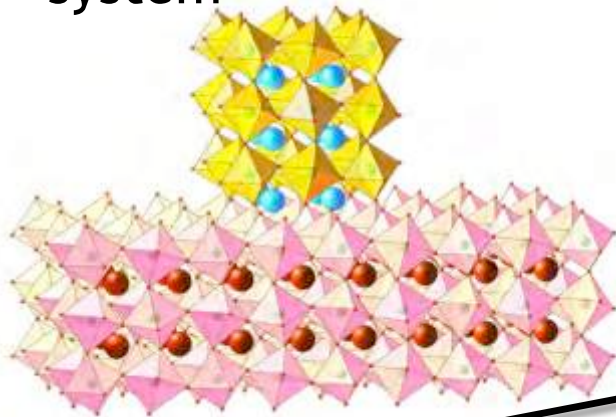


$$\Delta G < 0$$

Bulk Synthesis

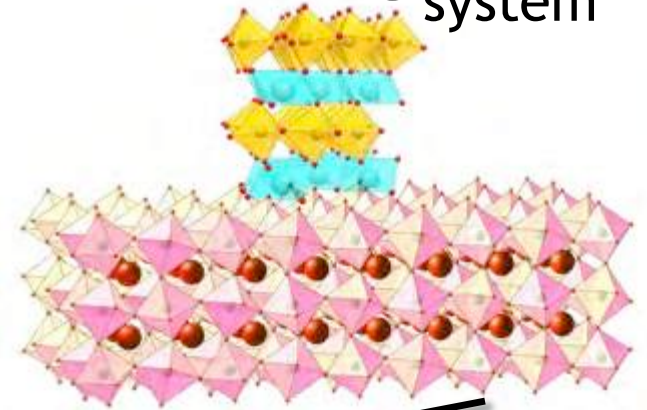
Stable if free energy difference overcome by $\Delta(\text{interface energy}) + \Delta(\text{strain energy}) + \Delta(\text{surface energy})$

$$\Delta G_{\text{system}} < 0$$



Epitaxial Stabilization

$$\Delta G_{\text{system}} > 0$$



E.S. Machlin and P. Chaudhari, in *Synthesis and Properties of Metastable Phases*, edited by E.S. Machlin and T.J. Rowland, (The Metallurgical Society of AIME, Warrendale, 1980), pp. 11-29.

BREAK THE RULES

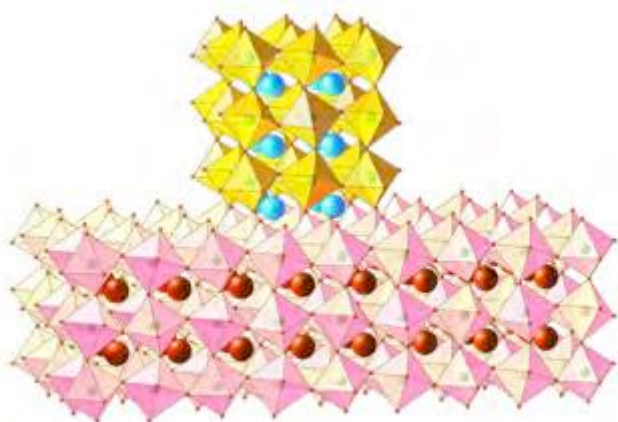
- Gibbs' Rule

$\Delta G < 0$ to form stable phases

Exploit interfacial energy from substrate

Substrates are Important

“Indeed, to achieve the objective of ‘pseudomorphic stabilization,’ the researcher should make the attempt to choose the substrate ...”



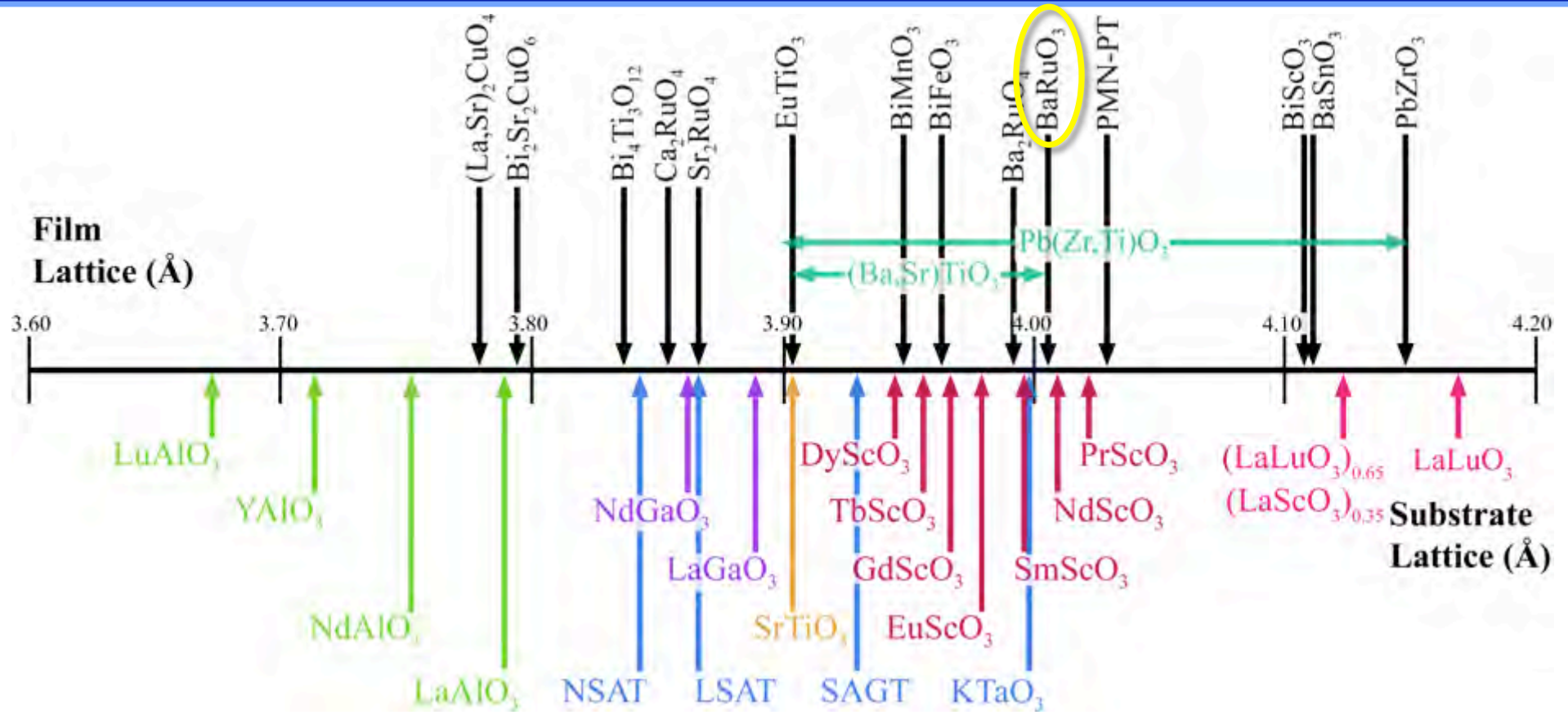
Epitaxial Stabilization



E.S. Machlin and P. Chaudhari,
“Theory of ‘Pseudomorphic Stabilization’ of
Metastable Phases in Thin Film Form,” in
Synthesis and Properties of Metastable Phases,
edited by E.S. Machlin and T.J. Rowland

(The Metallurgical Society of AIME, Warrendale, 1980), pp. 11-29.

Commercial Perovskite Substrates



[110] DyScO₃, $d = 32$ mm

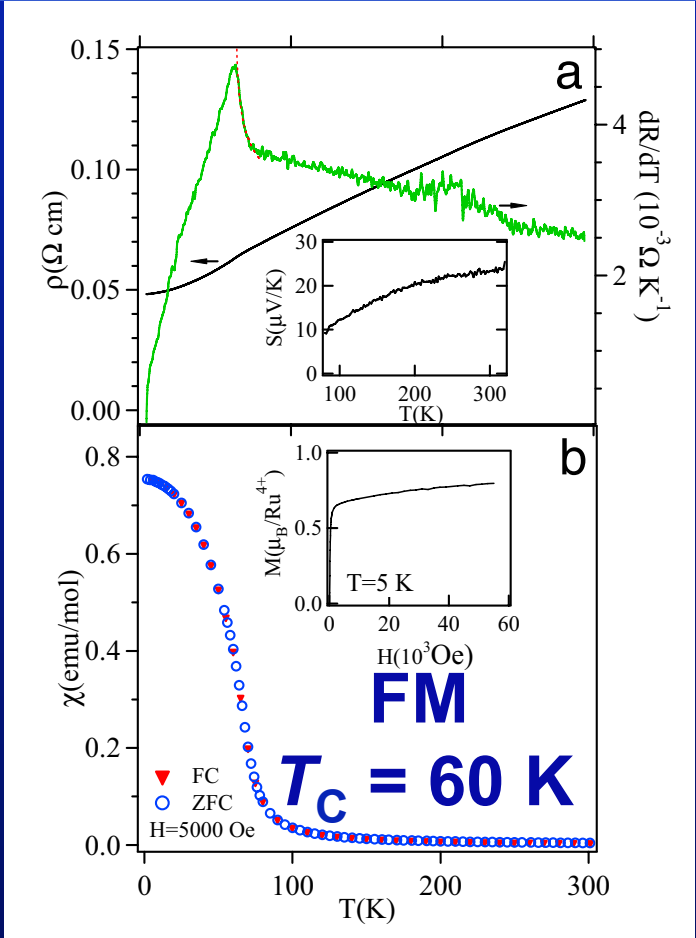
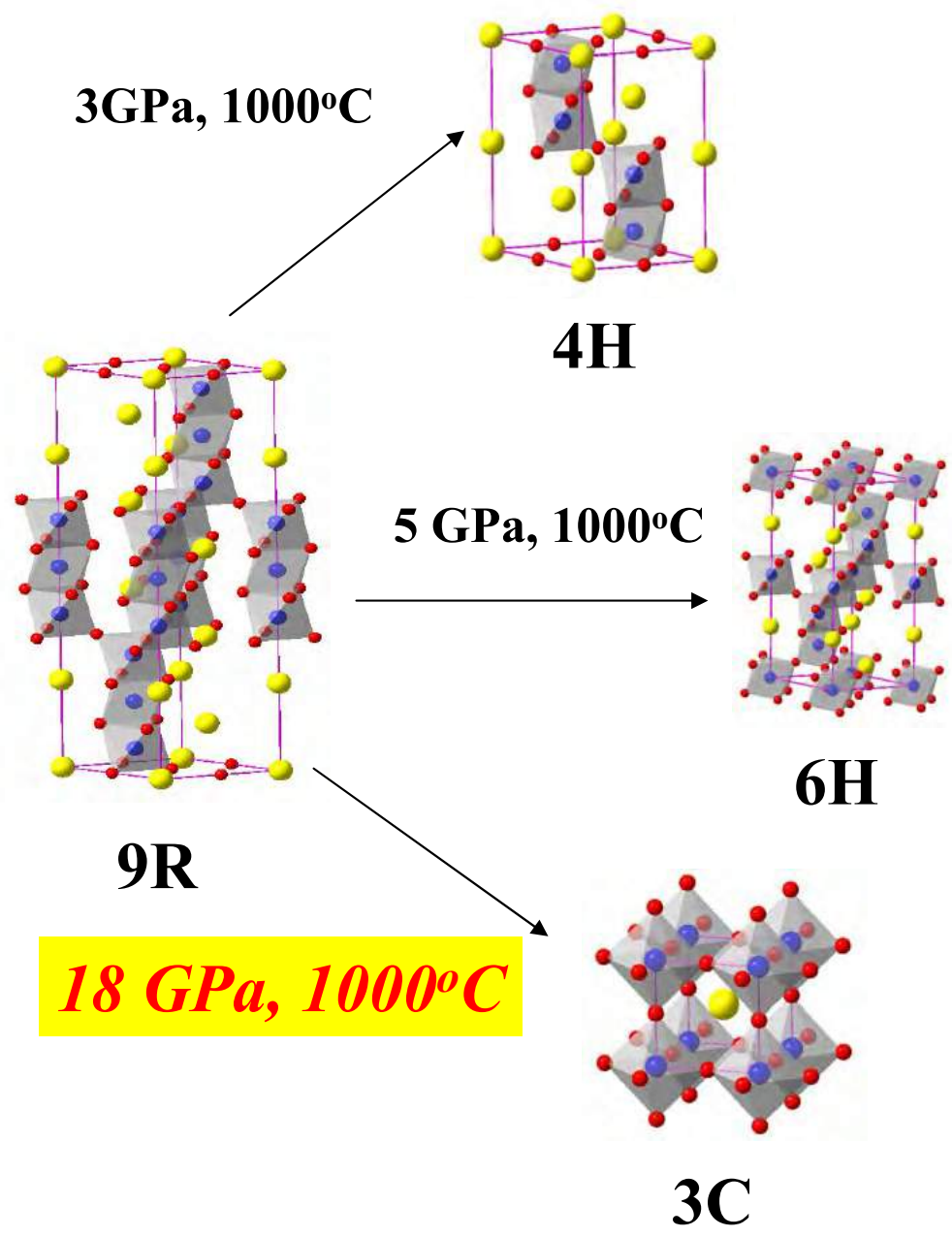


[110] GdScO₃, $d = 32$ mm



D.G. Schlom, L.Q. Chen, C.J. Fennie, V. Gopalan, D.A. Muller, X.Q. Pan, R. Ramesh, and R. Uecker, "Elastic Strain Engineering of Ferroic Oxides," *MRS Bulletin* **39** (2014) 118-130.

BaRuO₃ Polymorphs



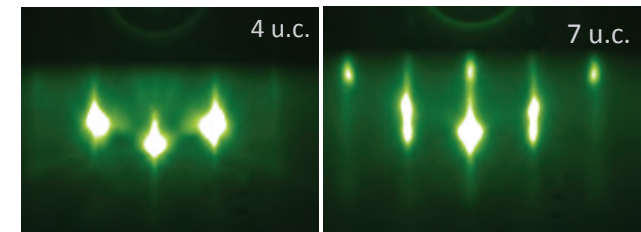
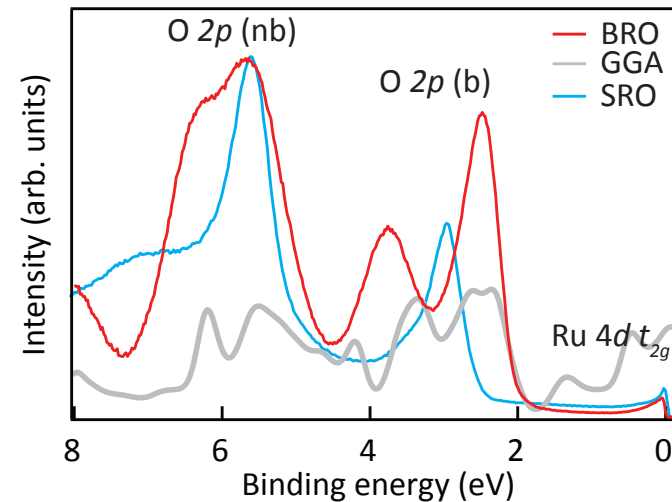
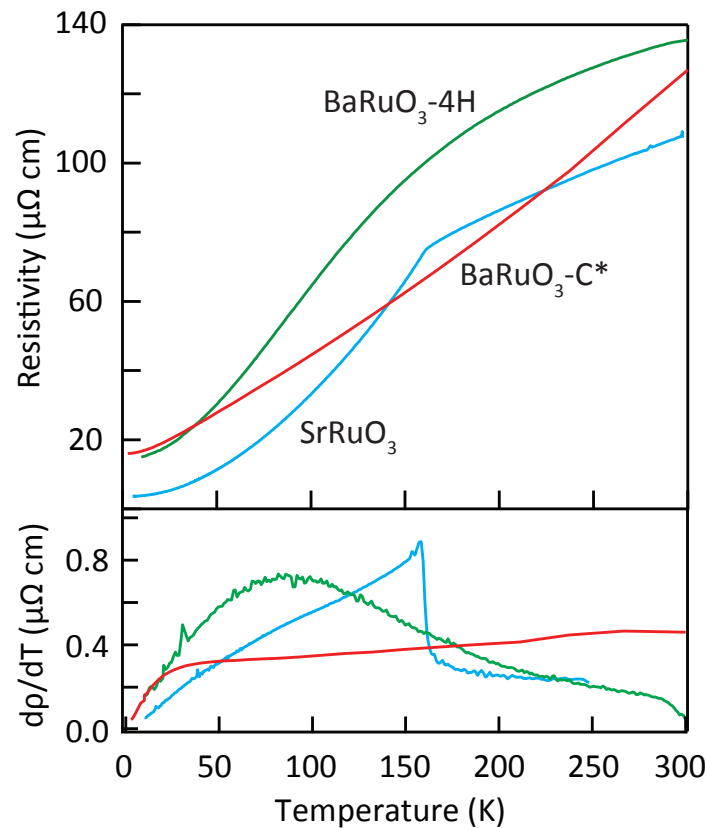
C.Q. Jin, J.S. Zhou, J.B. Goodenough, Q.Q. Liu, J.G. Zhao, L.X. Yang, Y. Yu, R.C. Yu, T. Katsura, A. Shatskiy, and E. Ito, "High-Pressure Synthesis of the Cubic Perovskite BaRuO₃ and Evolution of Ferromagnetism in ARuO₃ (A = Ca, Sr, Ba) Ruthenates," *PNAS* **105** (2008) 7115–7119.

Example – BaRuO₃ / SrTiO₃

- Epitaxially stabilized for ≤ 5 unit cells
- No octahedral rotations
(2.5% compressive strain \rightarrow tetragonal)

- $\frac{\rho_{300\text{ K}}}{\rho_{4\text{ K}}} = 7$

- No FM



BREAK THE RULES

- Gibbs' Rule
 $\Delta G < 0$ to form stable phases
Exploit interfacial energy from substrate
- Matthias's Rules for Superconductors
... "Stay away from Theorists"

Rules of B. Matthias for discovering new superconductors

1. high symmetry is best
2. peaks in density of states are good
3. stay away from oxygen
4. stay away from magnetism
5. stay away from insulators
6. stay away from theorists



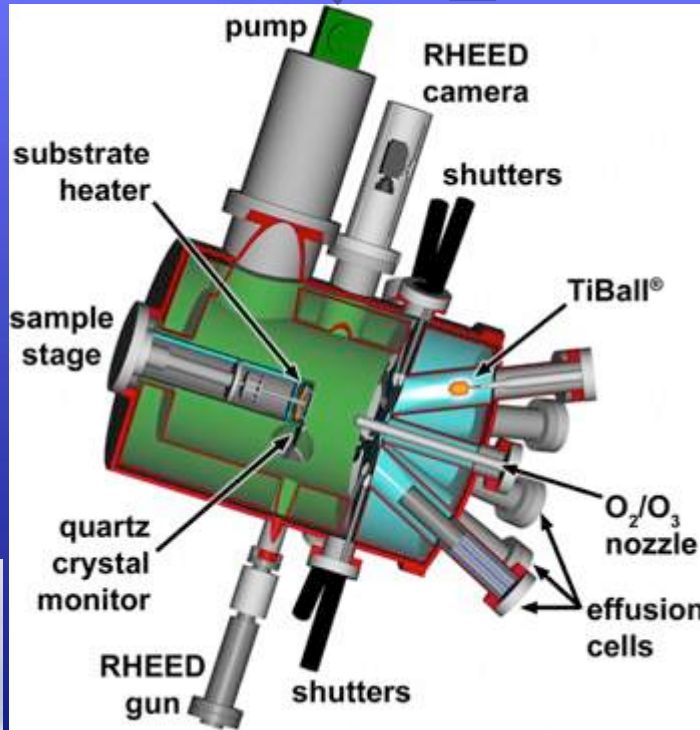
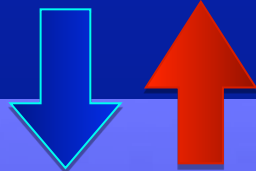
From Steve Girvin's lecture (Boulder Summer School 2000) courtesy of Mike Norman via Matthew Fisher



BREAK

Matthias' Rule

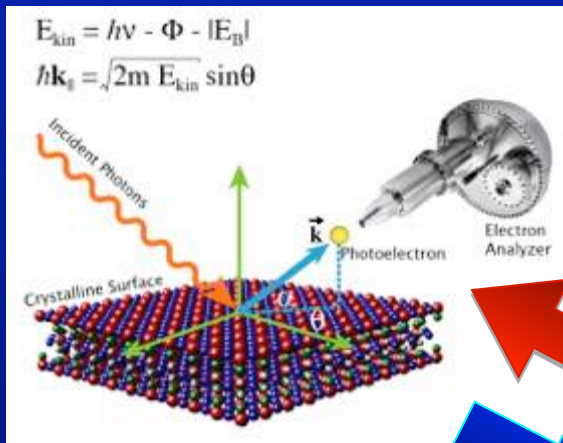
Team up with Theorists



Thin-Film Synthesis

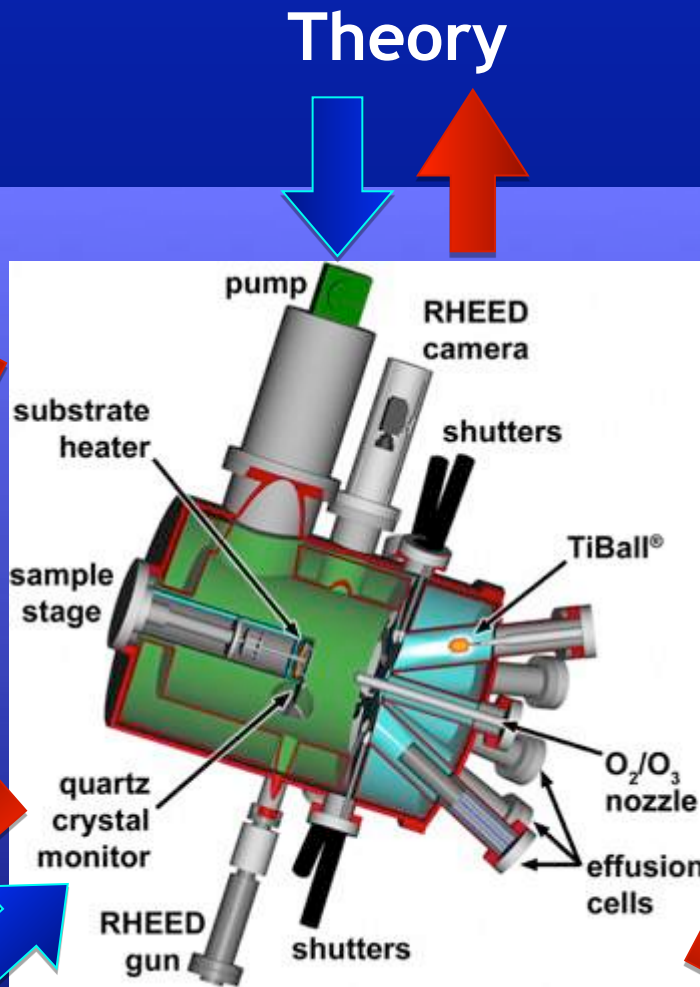
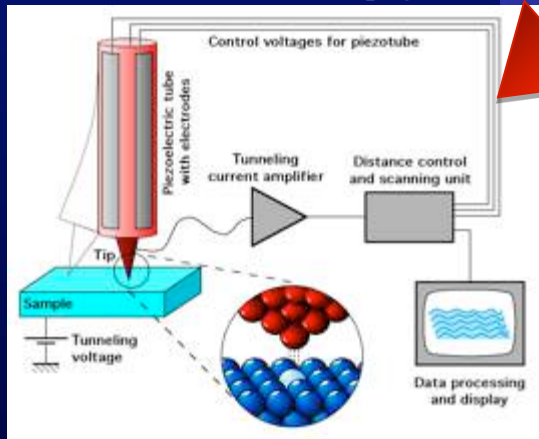


Provide useful Feedback to Theory



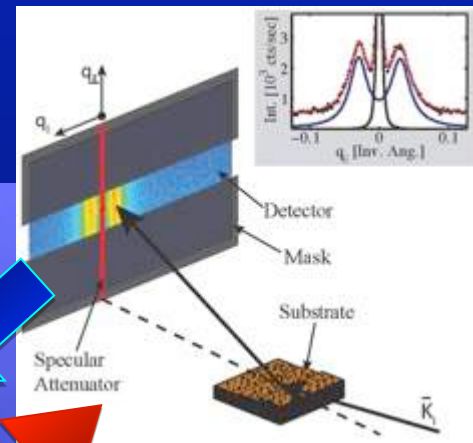
Photoemission Spectroscopy

Scanning Tunneling Microscopy

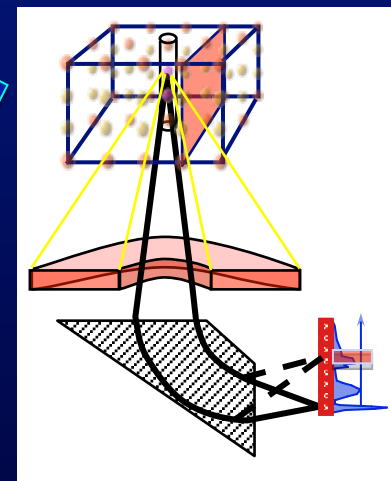


Thin-Film Synthesis

Transmission Electron Microscopy



Synchrotron X-ray Diffraction



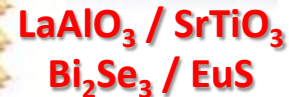
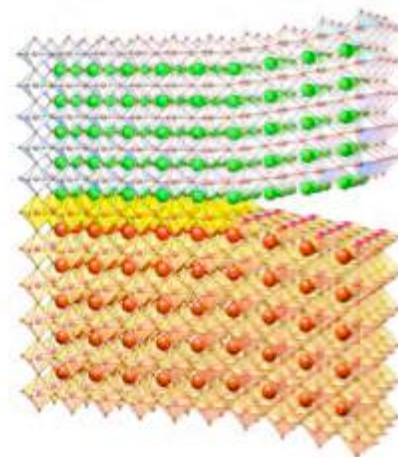
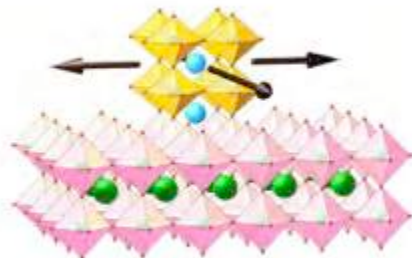
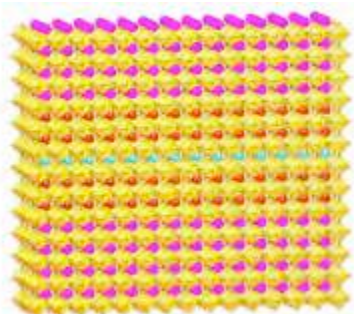
BREAK THE RULES

- Gibbs' Rule
 $\Delta G < 0$ to form stable phases
Exploit interfacial energy from substrate
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... "Stay away from Theorists"
Team up with theorists
(and provide them with useful feedback
e.g., Thin-Film Synthesis + ARPES)

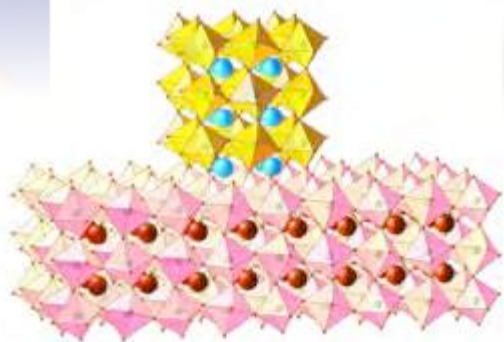
Why Thin-Film Synthesis + ARPES ?

“Artificial” Quantum Materials

Interface Engineering

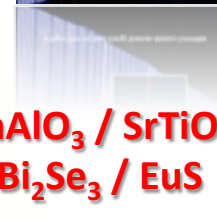


Polarization Doping & Proximity Effects
(from juxtaposed competing ground states)

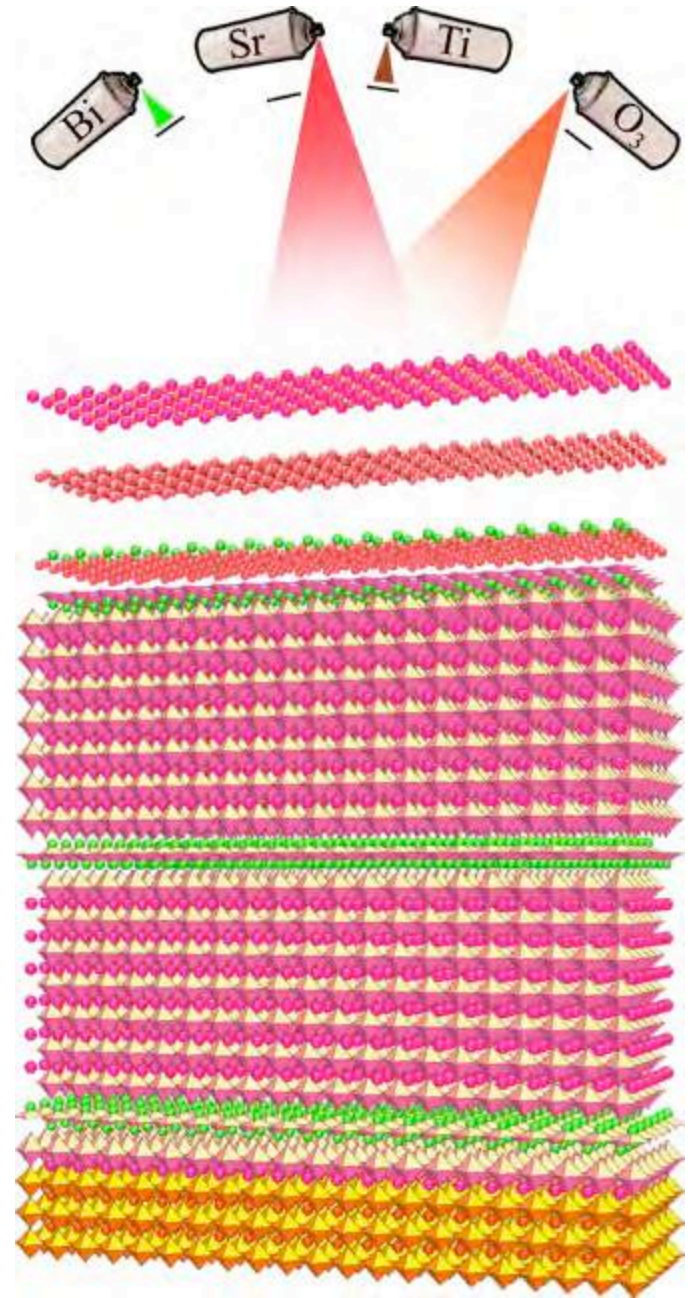
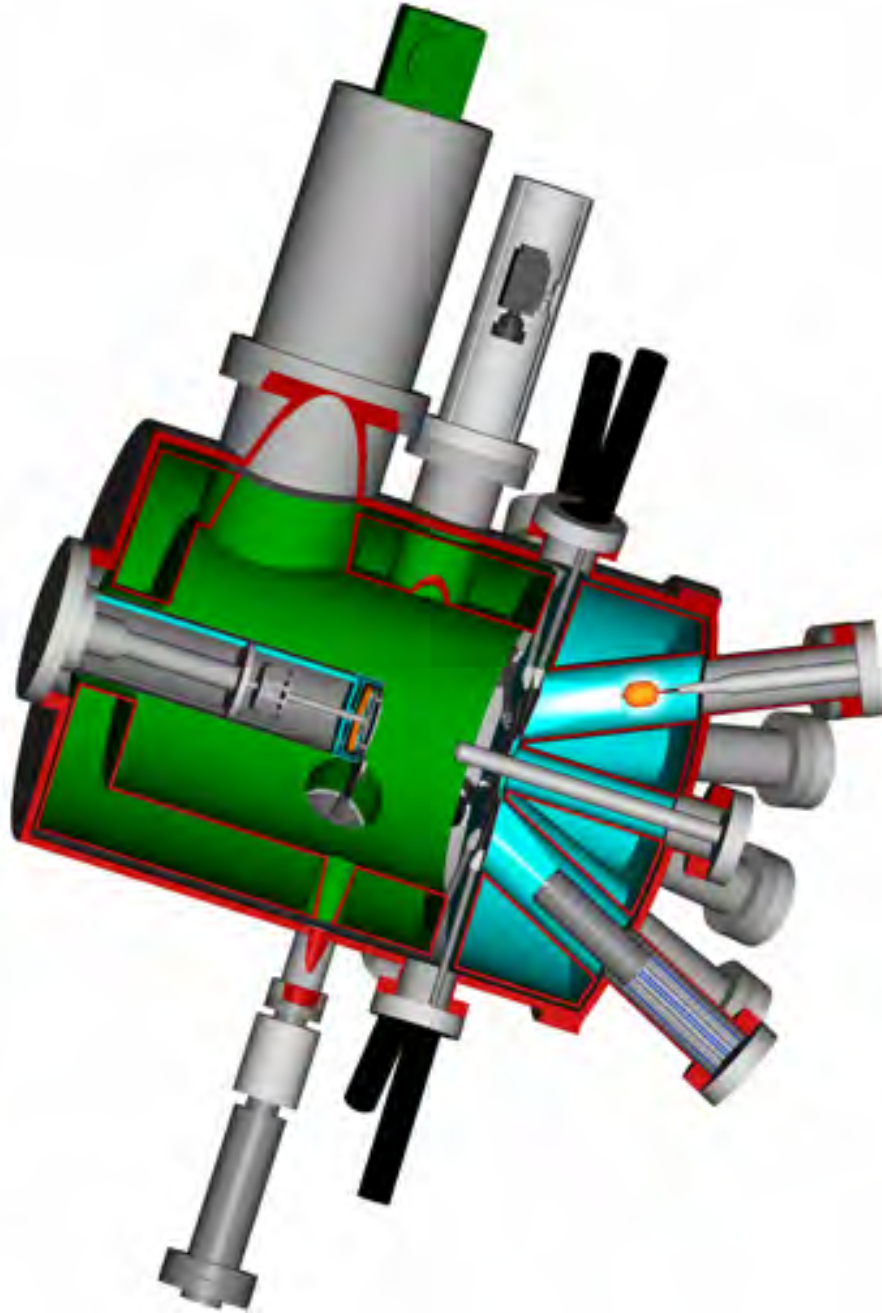


Bulk Quantum Materials

ARPES



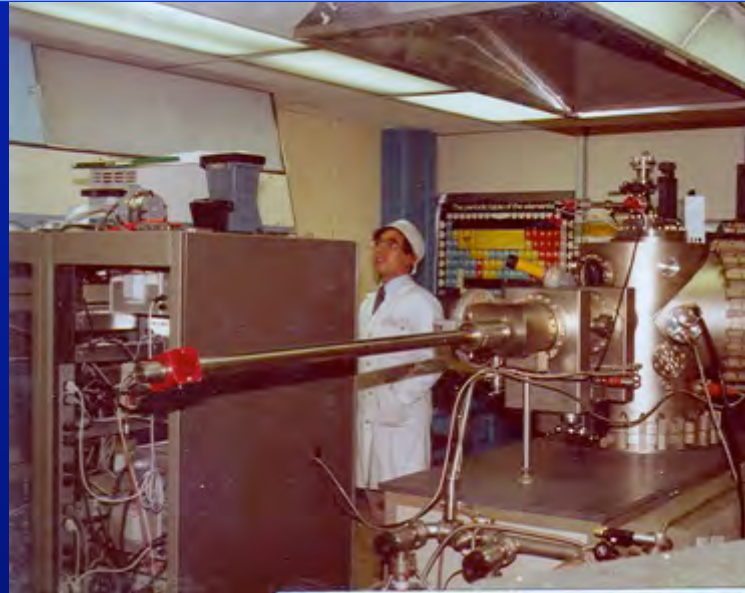
MBE \approx Atomic Spray Painting



Evolution of MBE



1st MBE
Al Cho at Bell Labs, 1972

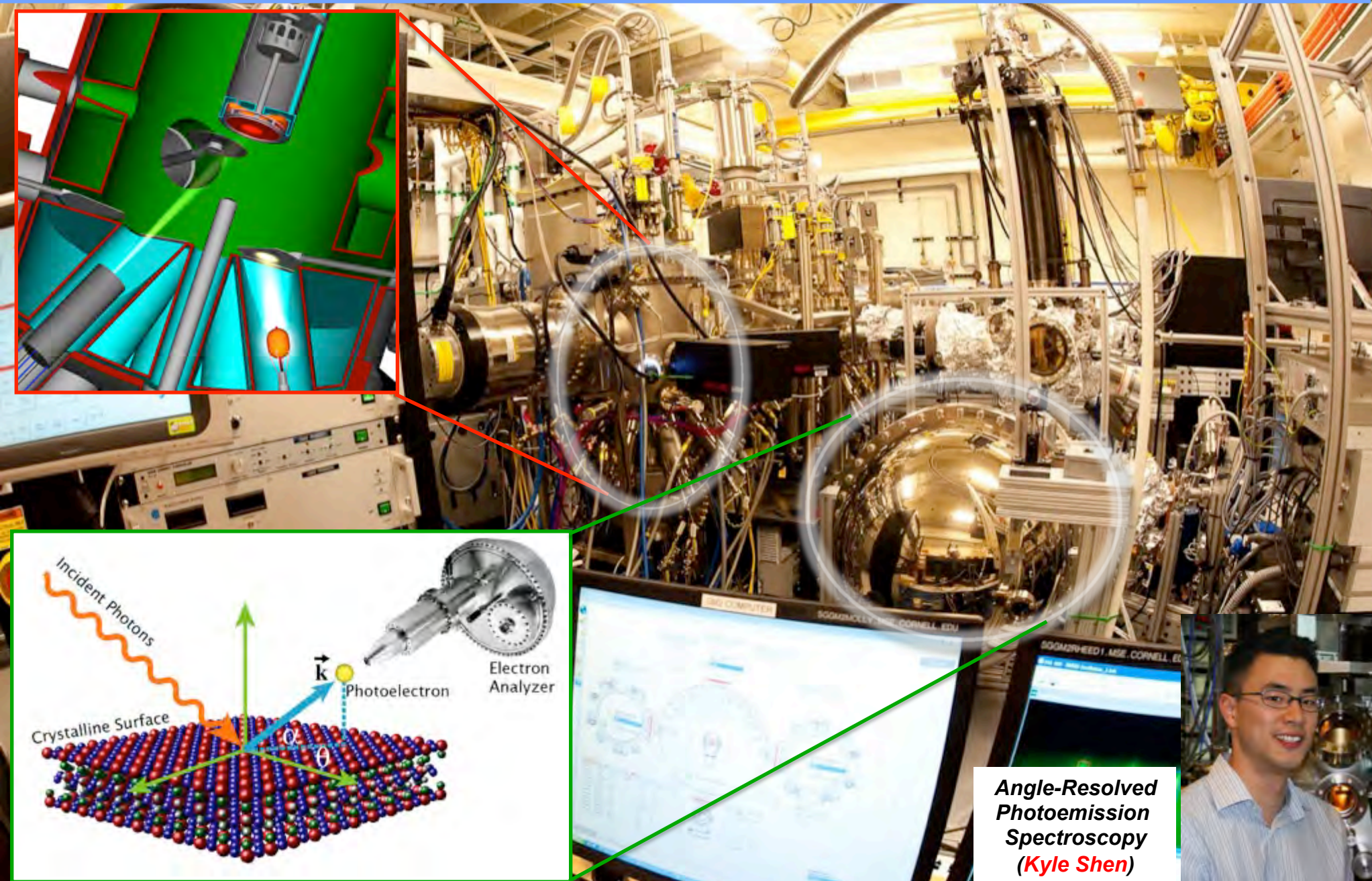


1st
University
MBE
Cornell,
1978



Production
MBE
Today
(courtesy of TRW)

Oxide MBE + ARPES



Angle-Resolved
Photoemission
Spectroscopy
(*Kyle Shen*)

MBE + ARPES

Titanates



Vanadates



Manganites



Nickelates



Cuprates



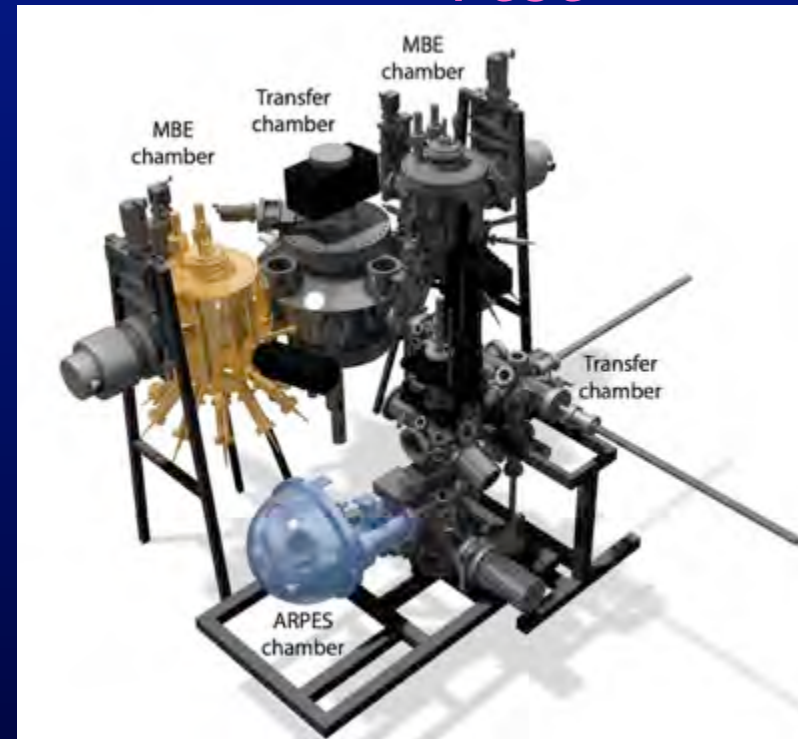
Ruthenates



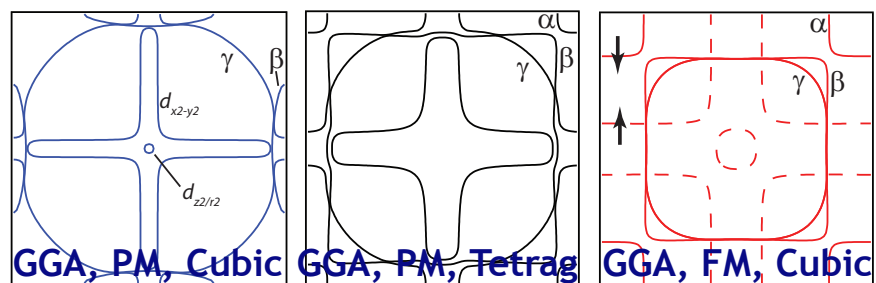
Iridates



Other Materials

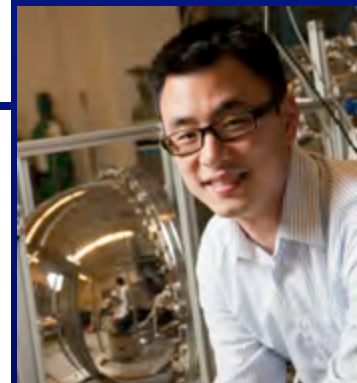
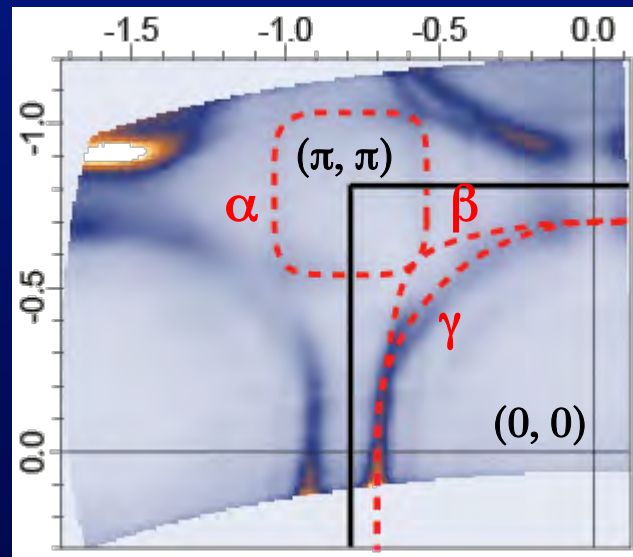
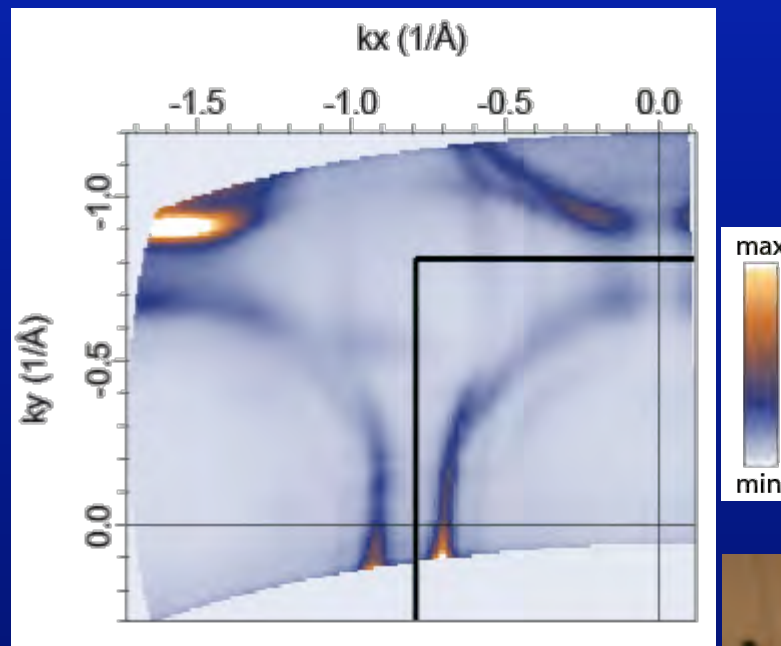
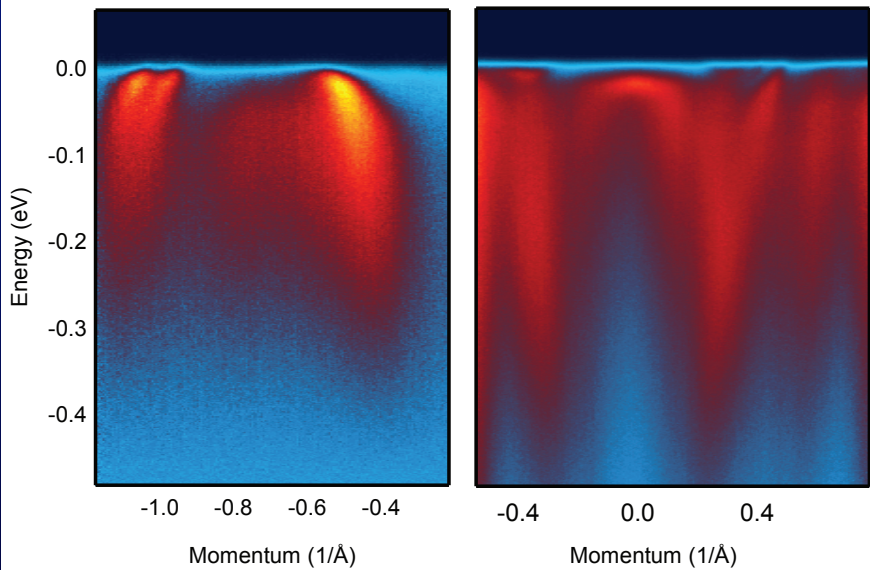
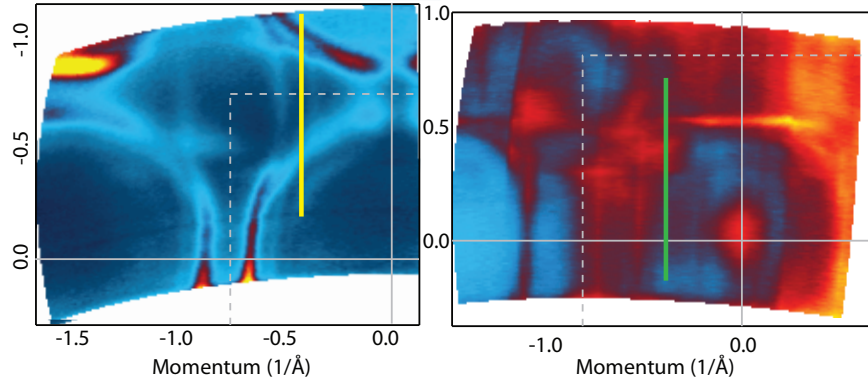


ARPES of BaRuO₃ / SrTiO₃



BaRuO₃

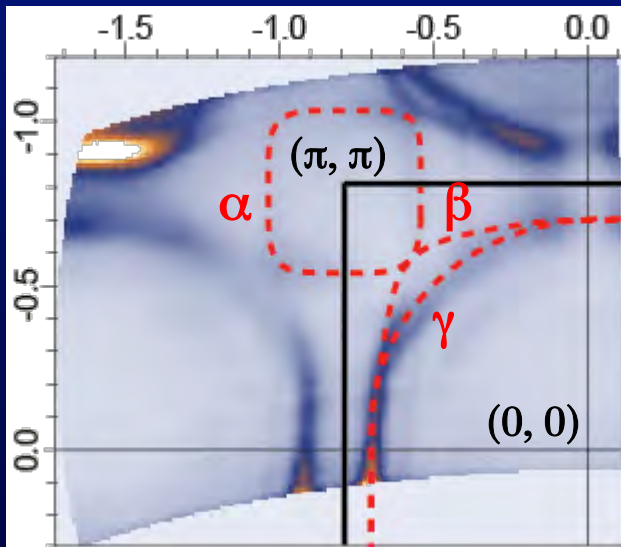
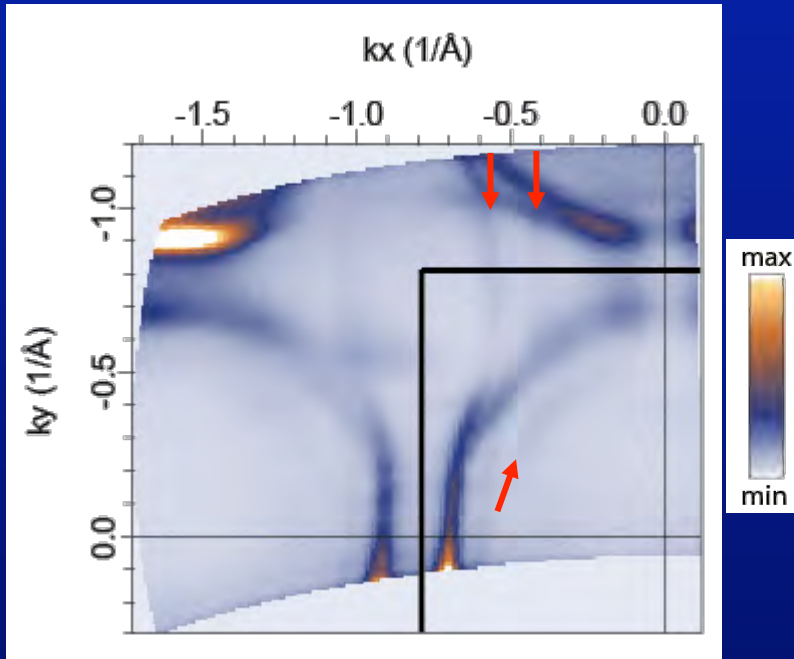
SrRuO₃



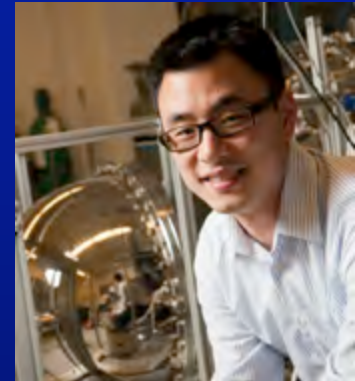
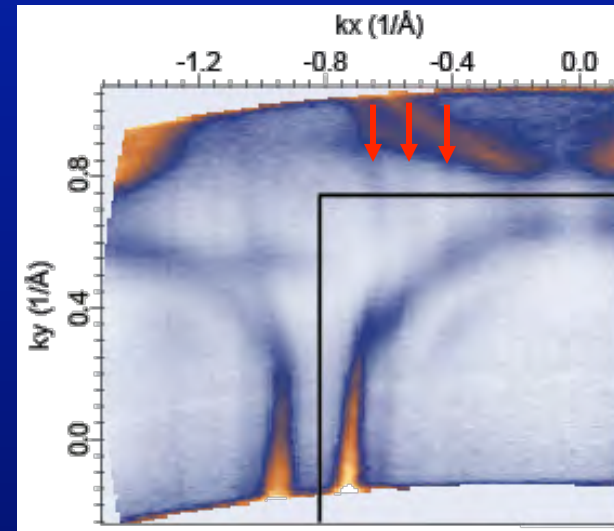
Kyle Shen

Quantum Well States in BaRuO₃

Thickness 3 unit cells



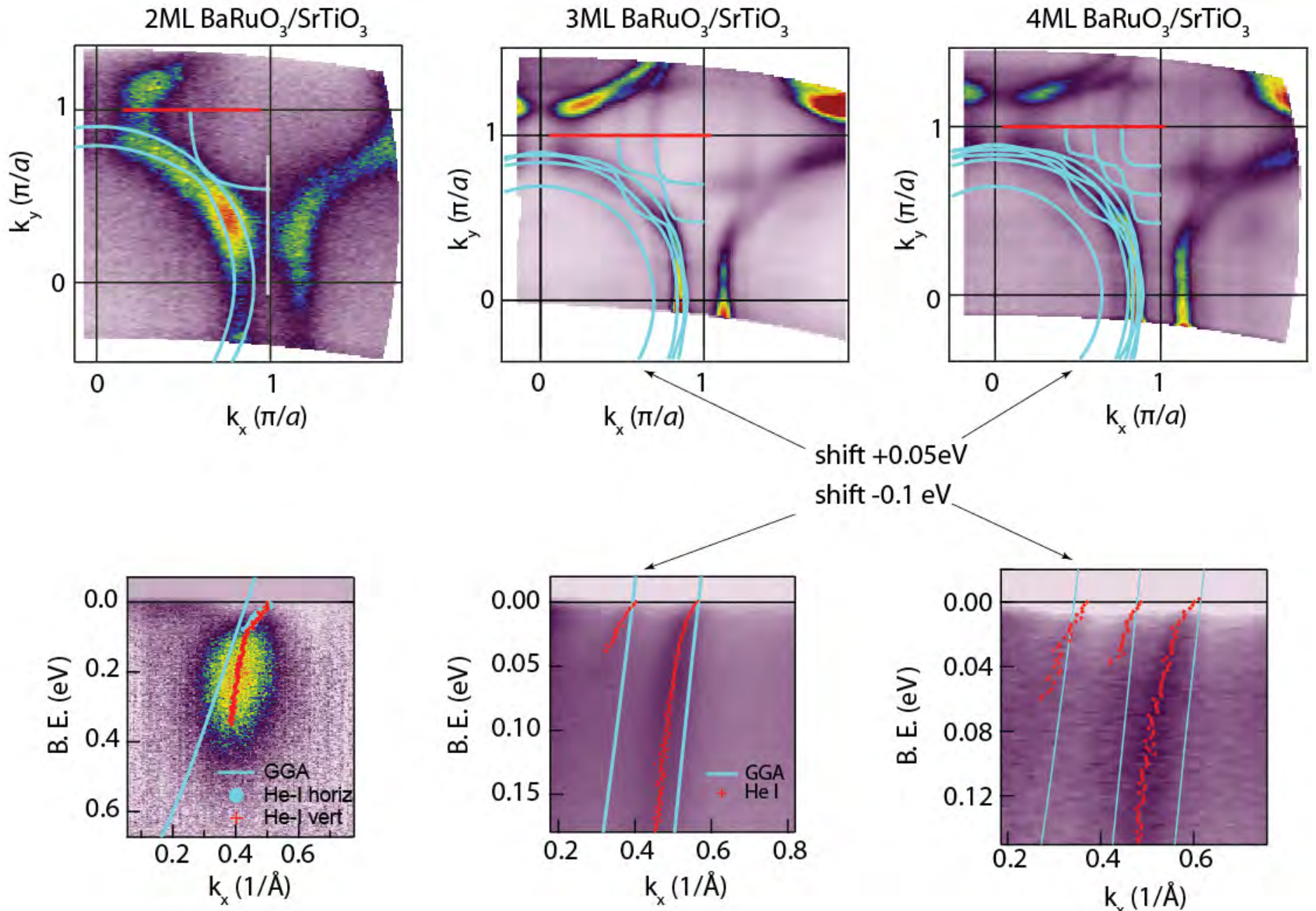
Thickness 4 unit cells



Kyle Shen

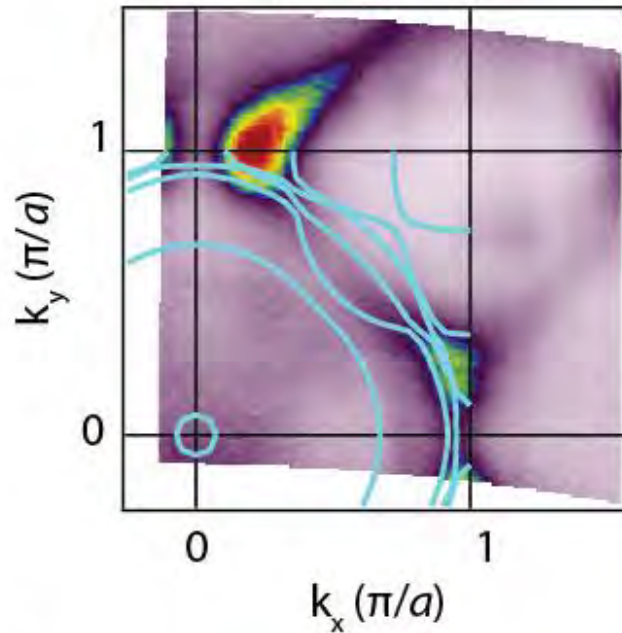
- FM suppressed because of finite thickness?
- Or tetragonal distortion?

Thickness Dependence

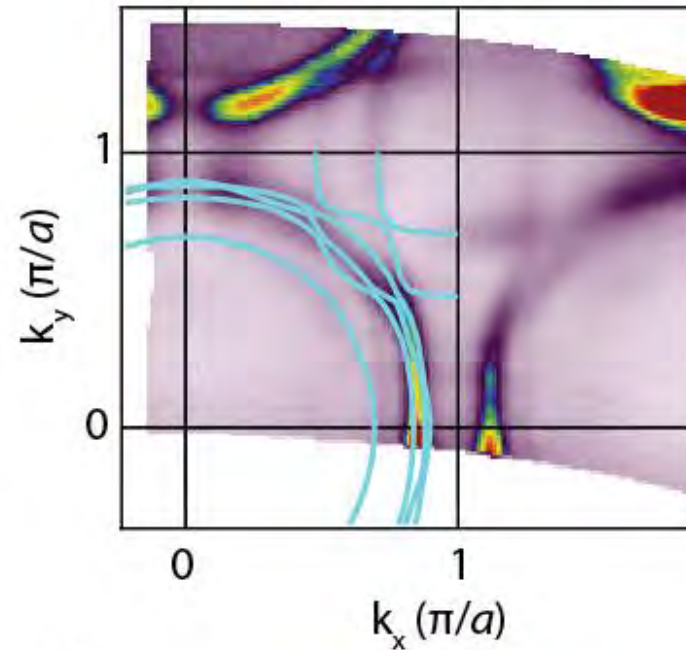


Strain Dependence

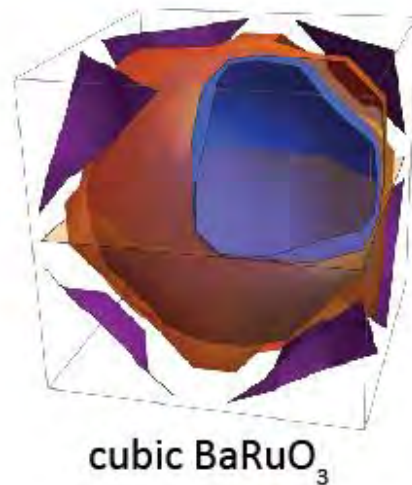
(a) 4ML BaRuO₃/(Nd,Sm)ScO₃
($a \approx 4.00$ Å)



(b) 3ML BaRuO₃/SrTiO₃
($a = 3.905$ Å)



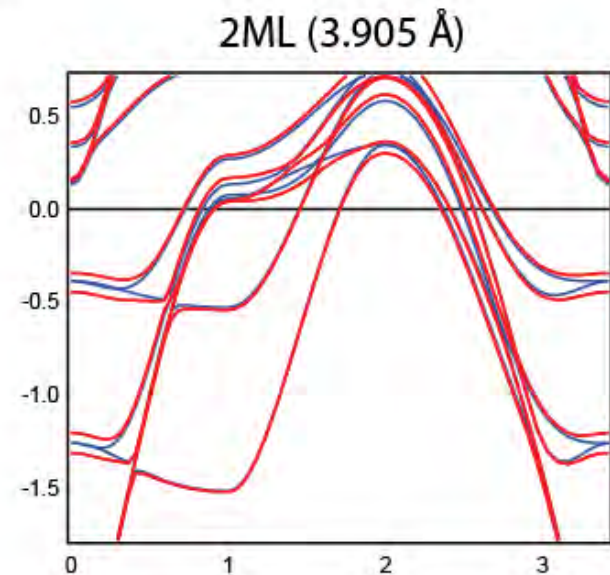
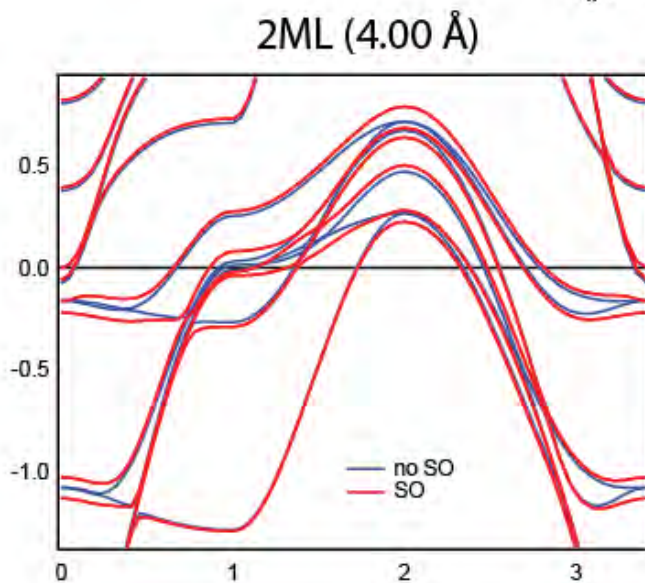
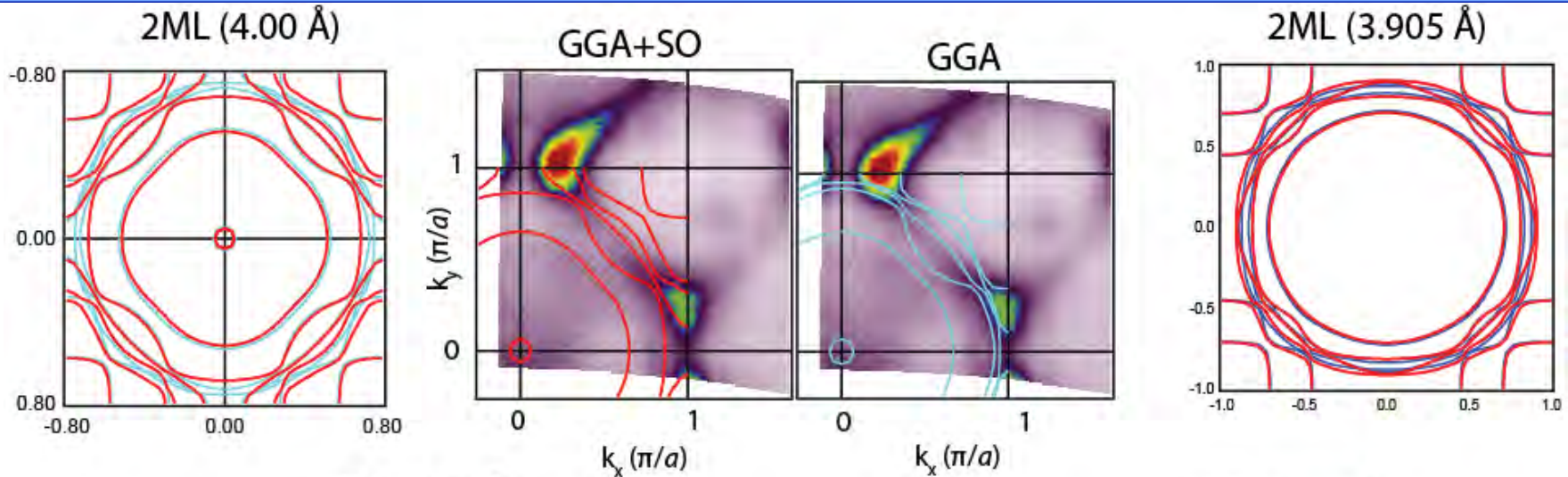
(c)



(d)



Effect of Spin-Orbit (2 ML Cubic)



Spin-Orbit makes little difference except near $(\pi, 0)$

BREAK THE RULES

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 $\Delta G < 0$ to form stable phases
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... "Stay away from Theorists"
Team up with theorists
(and provide them with useful feedback)
- Pauling's Rules for Crystal Structures
Radius ratio criteria for stability

Pauling's Rules

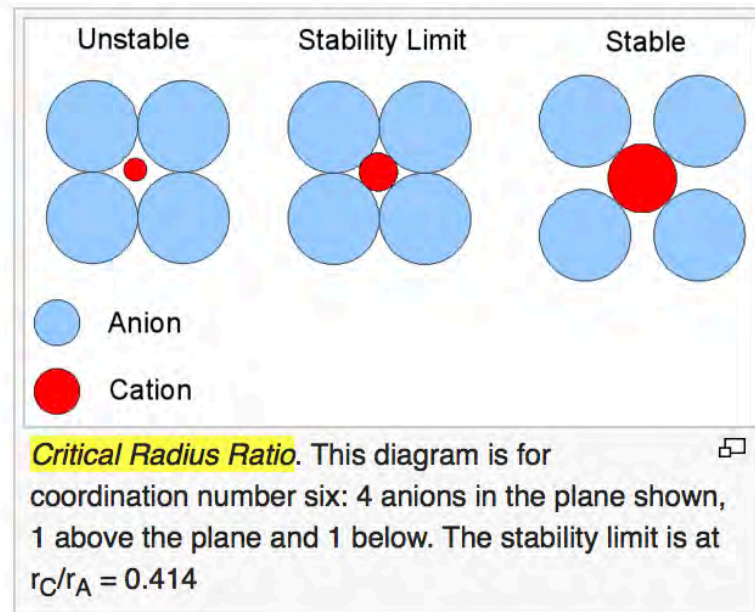
First rule: the radius ratio rule [\[edit \]](#)

For typical ionic solids, the **cations** are smaller than the **anions**, and each cation is surrounded by **coordinated** anions which form a **polyhedron**. The sum of the **ionic radii** determines the cation-anion distance, while the **cation-anion radius ratio** r_+/r_- (or r_c/r_a) determines the **coordination number** (C.N.) of the cation, as well as the shape of the coordinated polyhedron of anions.^{[3][4]}

For the coordination numbers and corresponding polyhedra in the table below, Pauling mathematically derived the *minimum* radius ratio for which the cation is in contact with the given number of anions (considering the ions as rigid spheres). If the cation is smaller, it will not be in contact with the anions which results in instability leading to a lower coordination number.

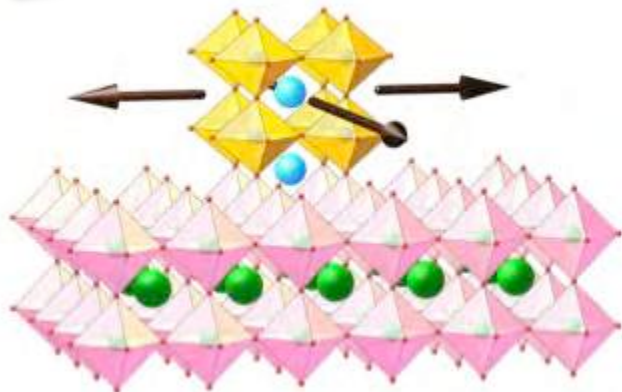
Polyhedron and minimum radius ratio for each coordination number

C.N.	Polyhedron	Radius ratio
3	triangular	0.155
4	tetrahedron	0.225
6	octahedron	0.414
7	capped octahedron	0.592
8	square antiprism (anticube)	0.645
8	cube	0.732
9	triaugmented triangular prism	0.732
12	cuboctahedron	1.00

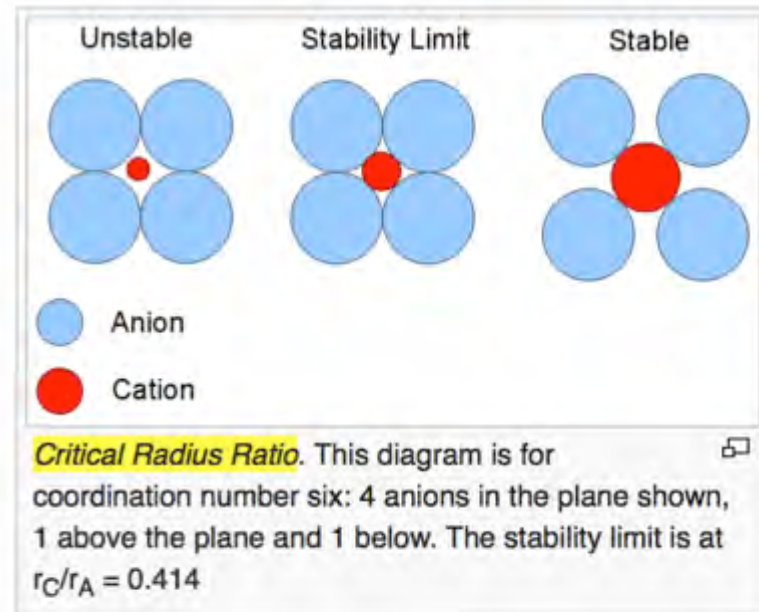
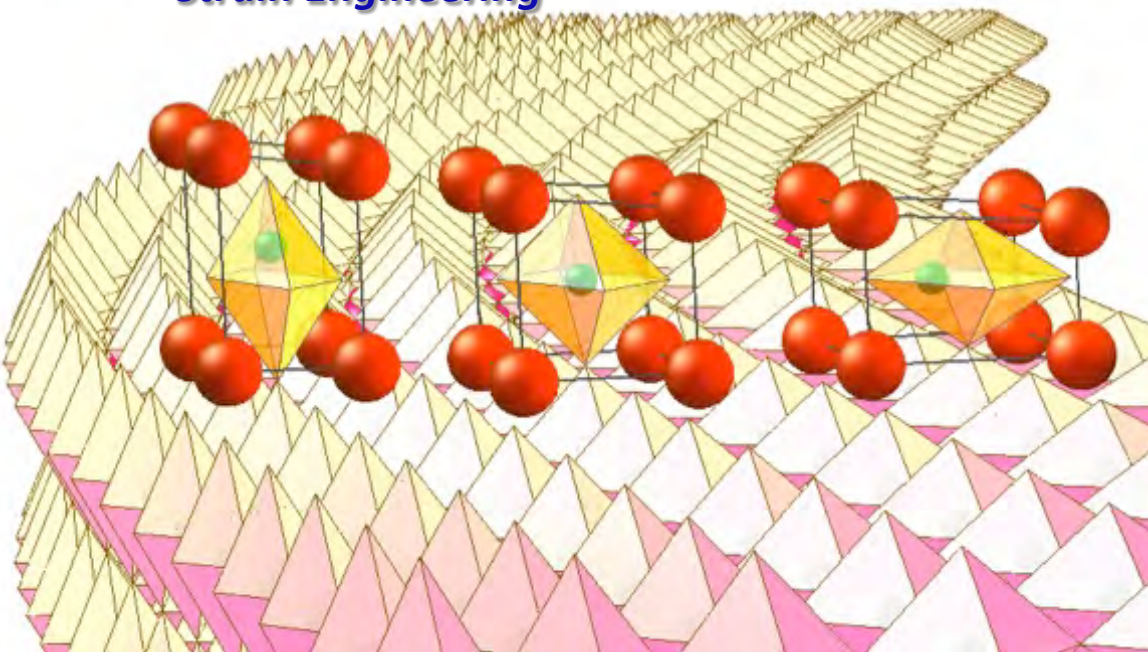


https://en.wikipedia.org/wiki/Pauling%27s_rules

BREAK Pauling's Rules

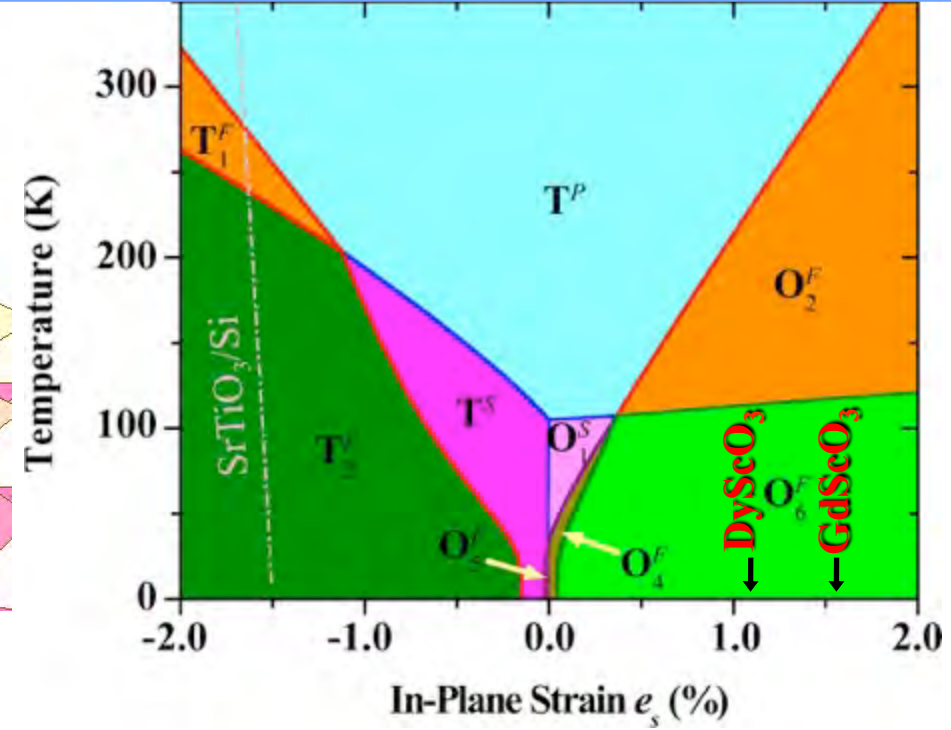
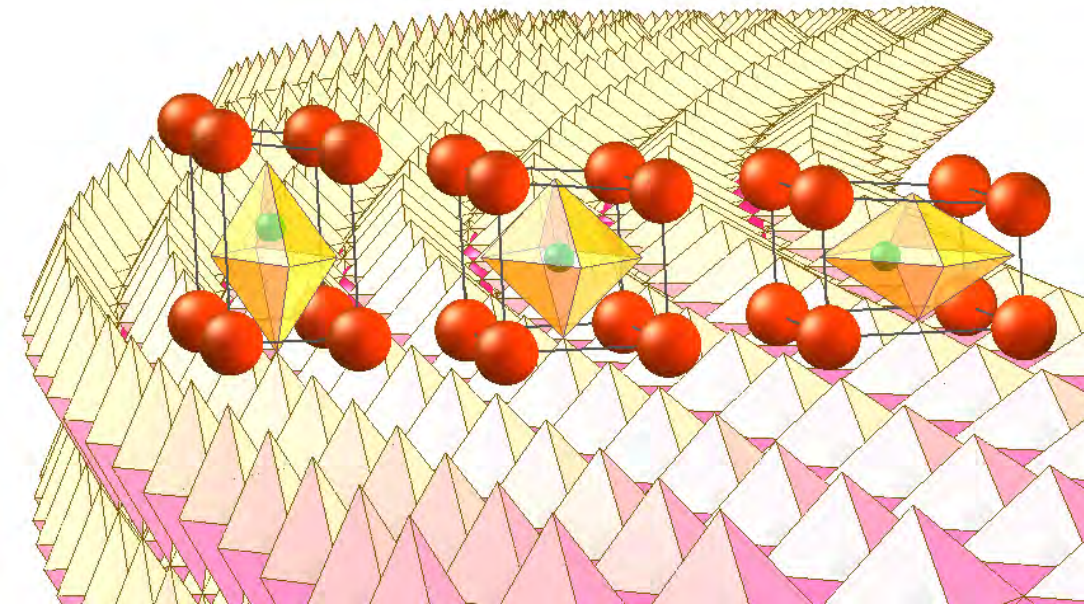


Strain Engineering

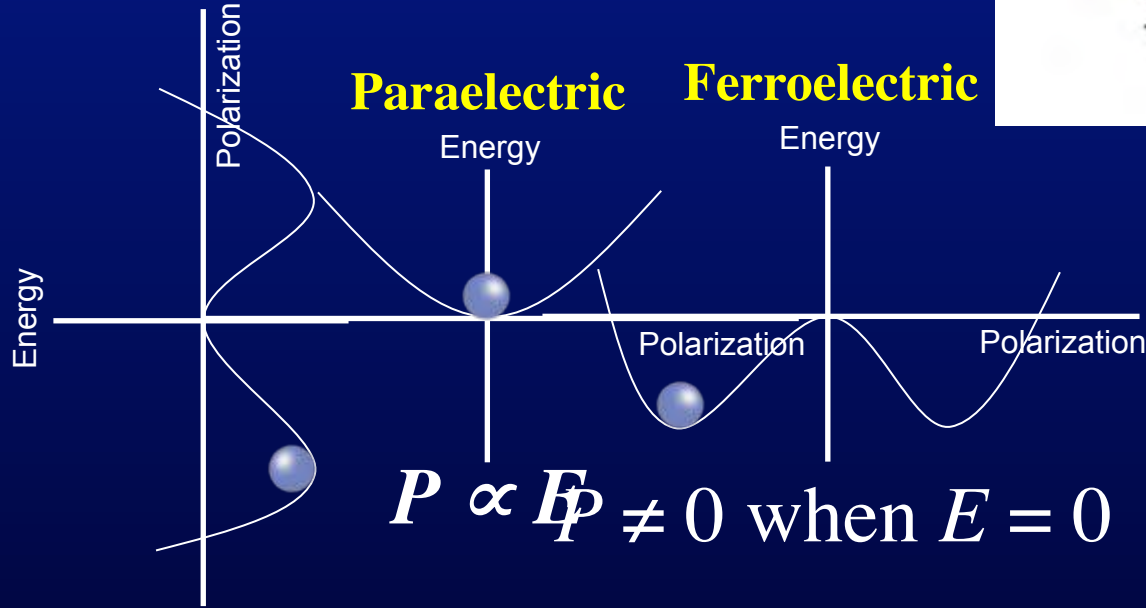


https://en.wikipedia.org/wiki/Pauling%27s_rules

Strained SrTiO_3 – Transmuting a Dielectric into a Ferroelectric

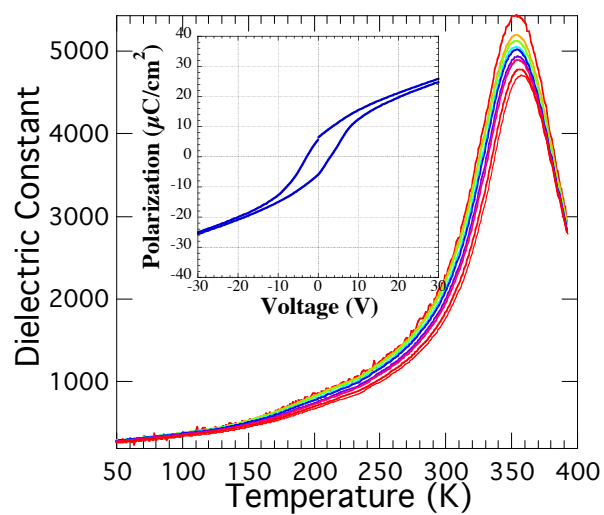
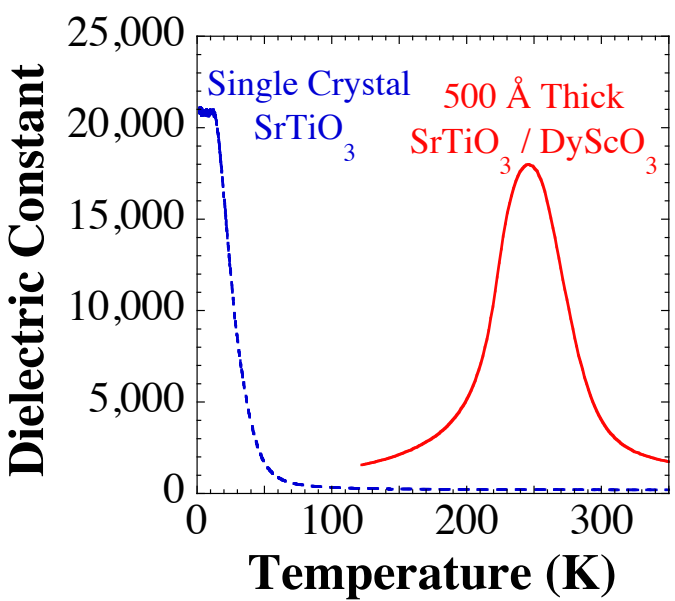
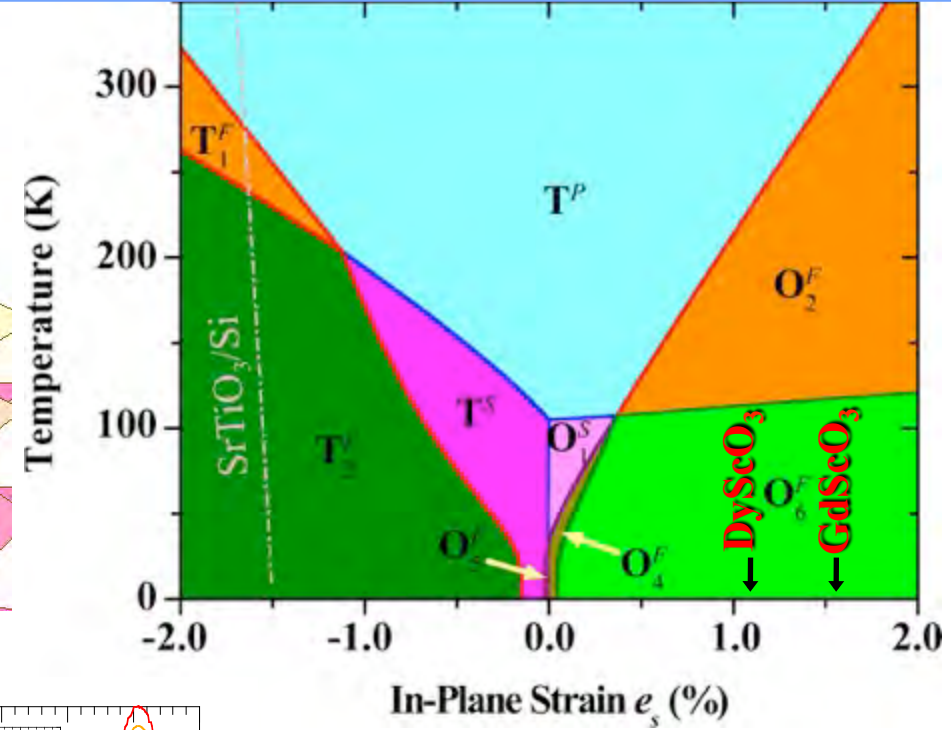
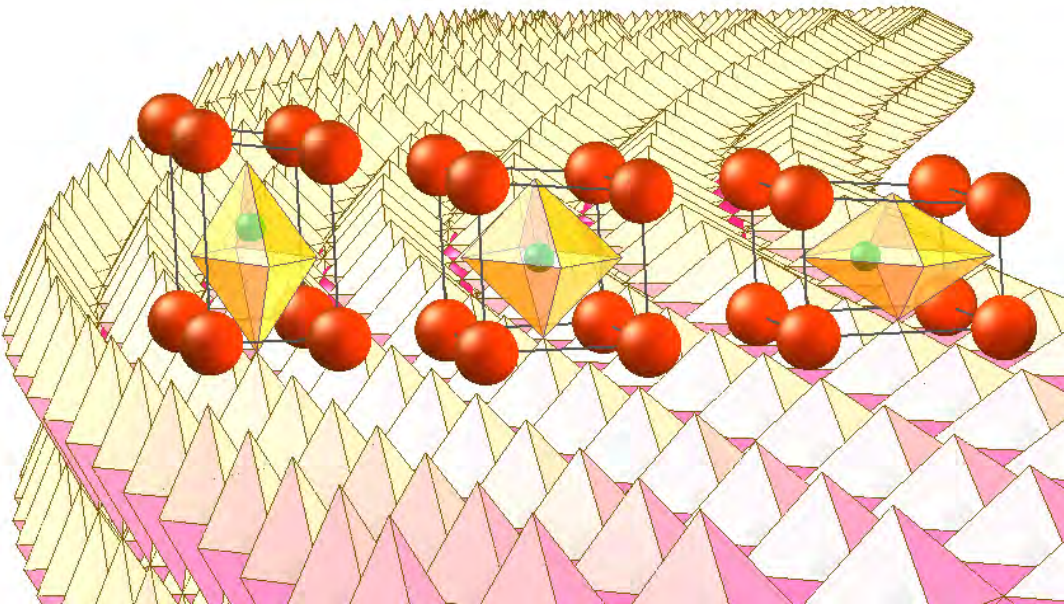


Ferroelectric



N.A. Pertsev, A.K. Tagantsev, and N. Setter, *Physical Review* **61** (2000) 825-829.

Strained SrTiO_3 – Transmuting a Dielectric into a Ferroelectric

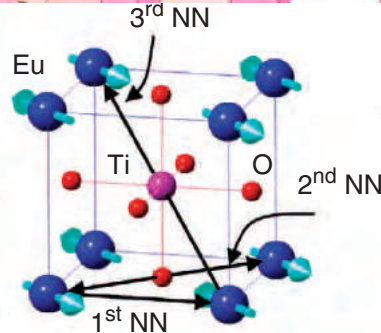
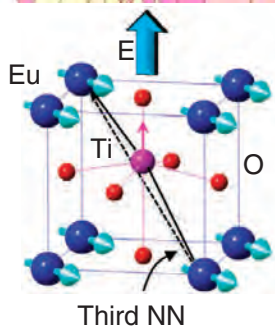
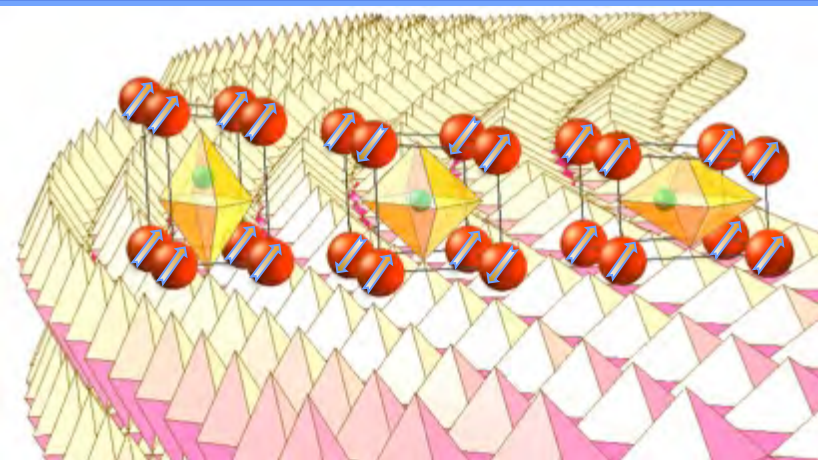


N.A. Pertsev, A.K. Tagantsev, and N. Setter,
Physical Review **61** (2000) 825-829.

Ferroelectric at Room Temperature

J.H. Haeni, P. Irvin, W. Chang, R. Uecker, P. Reiche, Y.L. Li, S. Choudhury, W. Tian, M.E. Hawley, B. Craigo, A.K. Tagantsev, X.Q. Pan, S.K. Streiffer, L.Q. Chen, S.W. Kirchoefer, J. Levy, and D.G. Schlom, *Nature* **430** (2004) 758–761.

Strained EuTiO_3 – Strongest Ferromagnetic Ferroelectric



-1.2

0

+0.75

Biaxial Strain (%), ϵ_s

C.J. Fennie and K.M. Rabe

Physical Review Letters 97 (2006)

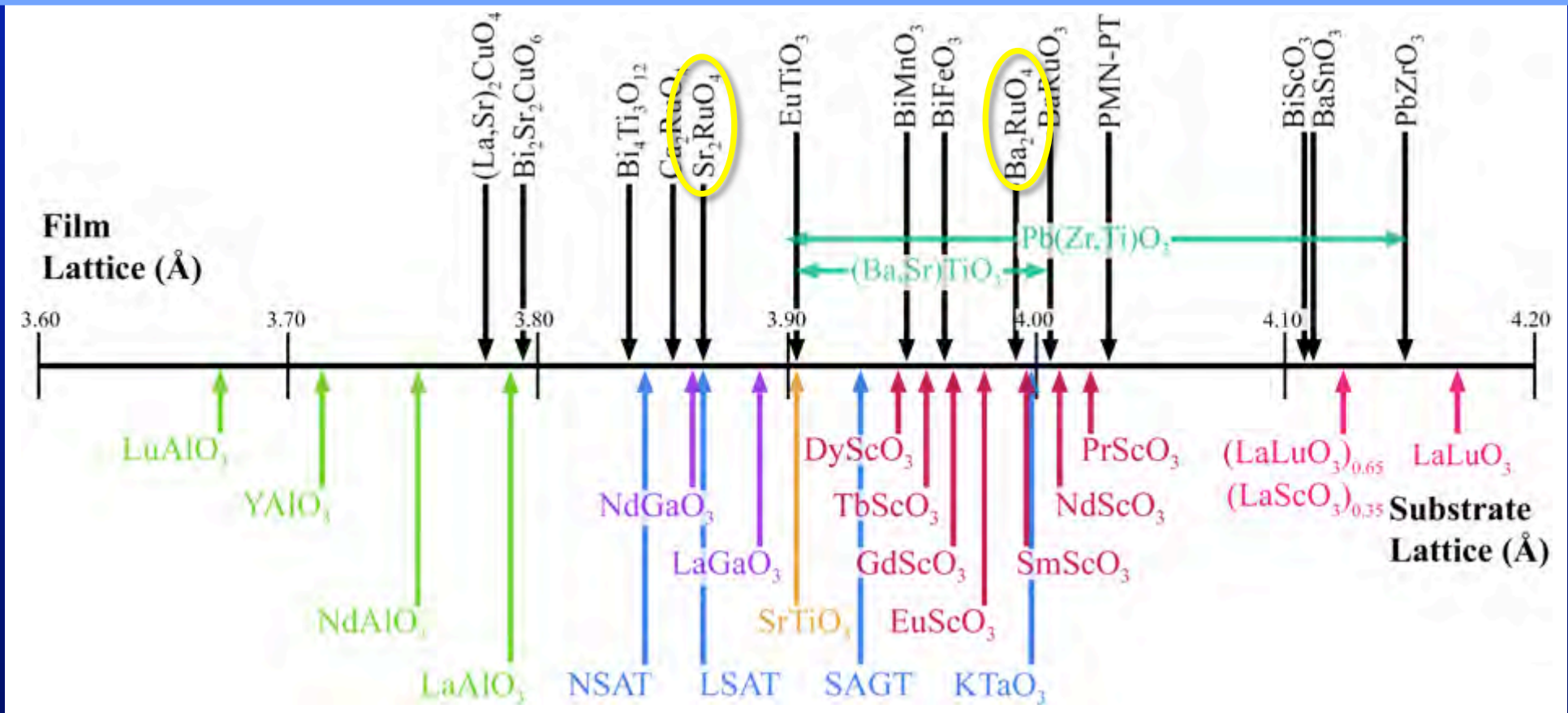
P.J. Ryan, J.-W. Kim, T. Birol, P. Thompson, J.-H. Lee, X. Ke, P.S. Normile, E. Karapetrova, P. Schiffer, S.D. Brown, C.J. Fennie, and D.G. Schlom
Nature Communications 4 (2013) 1334.

Table 1 | Calculated magnetic exchange constants.

ETO-LSAT	J_{1xy}	J_{1z}	J_{2xy}	J_{2z}	J_3
$J/K_B(K)$ -bulk	+ 0.075	- 0.114	+ 0.062	+ 0.083	- 0.031
# Neighbours	4	2	4	8	8
$J/K_B(K)$ -LSAT	+ 0.086	- 0.147	+ 0.06	+ 0.087	- 0.034

Shown are the exchange constants (J) calculated between the Eu ions within the unconstrained bulk $I4/mcm$ ETO and the ETO film on LSAT with $(a^0a^0c^-)$ structure under -0.9% compressive strain, including the first, second and third NN Eu ions describing both the in-plane (xy) and out-of-plane (z) interactions. Positive indicates FM and negative AFM coupling. The second row indicates the number of neighbours for each particular interaction. The first and second NN interactions are mostly FM bar the first NN out-of-plane J_{1z} exchange constant. The calculations indicate the importance of J_3 in determining the G-AFM structure in ETO.

Commercial Perovskite Substrates



[110] DyScO₃, $d = 32$ mm

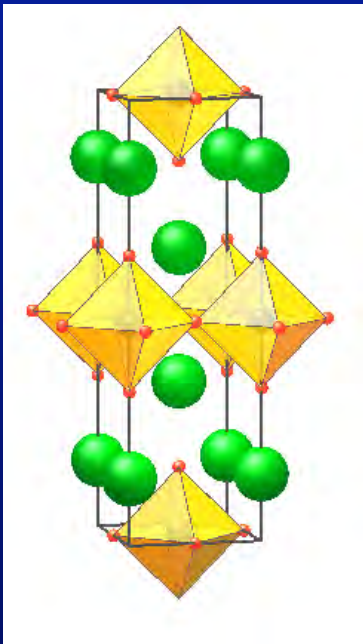


[110] GdScO₃, $d = 32$ mm

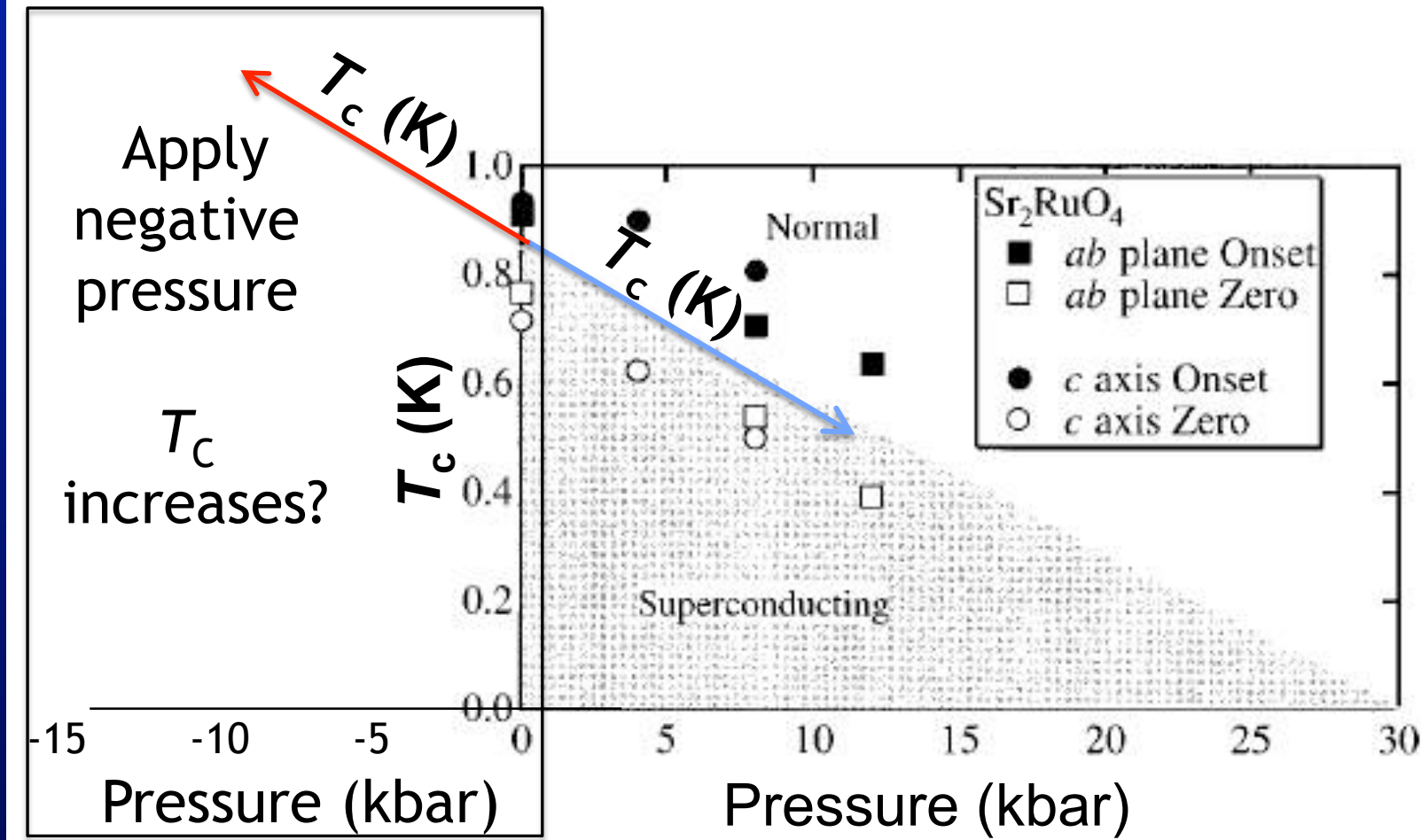


D.G. Schlom, L.Q. Chen, C.J. Fennie, V. Gopalan, D.A. Muller, X.Q. Pan, R. Ramesh, and R. Uecker, "Elastic Strain Engineering of Ferroic Oxides," *MRS Bulletin* **39** (2014) 118-130.

Effect of Strain on Sr_2RuO_4 (a spin-triplet superconductor)



$n = 1$

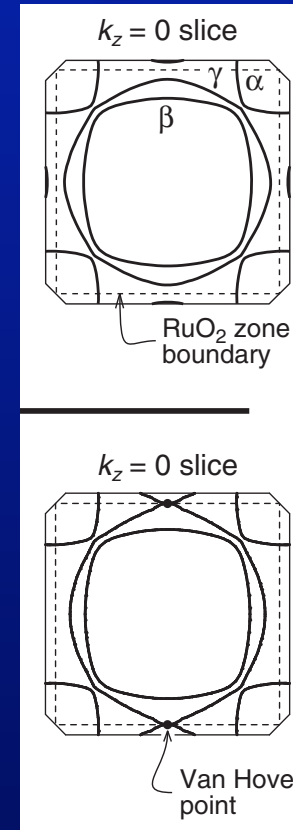
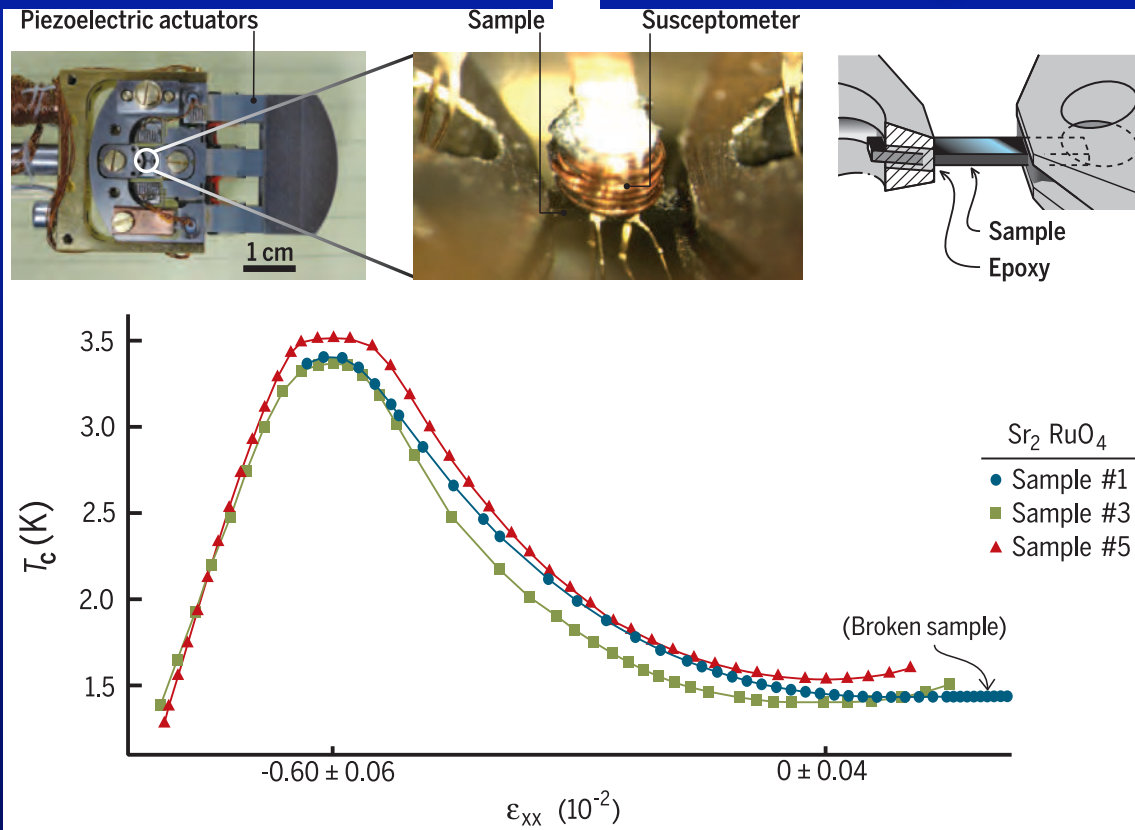


T_c goes down with increasing pressure

→ apply negative pressure

N. Shirakawa, K. Murata, S. Nishizaki, Y. Maeno, T. Fujita,
Phys. Rev. B **56** (1997) 7890-7893.

In-plane Uniaxial Strain Dramatically Increases T_c in Sr_2RuO_4



$$\epsilon_{xx} = 0$$

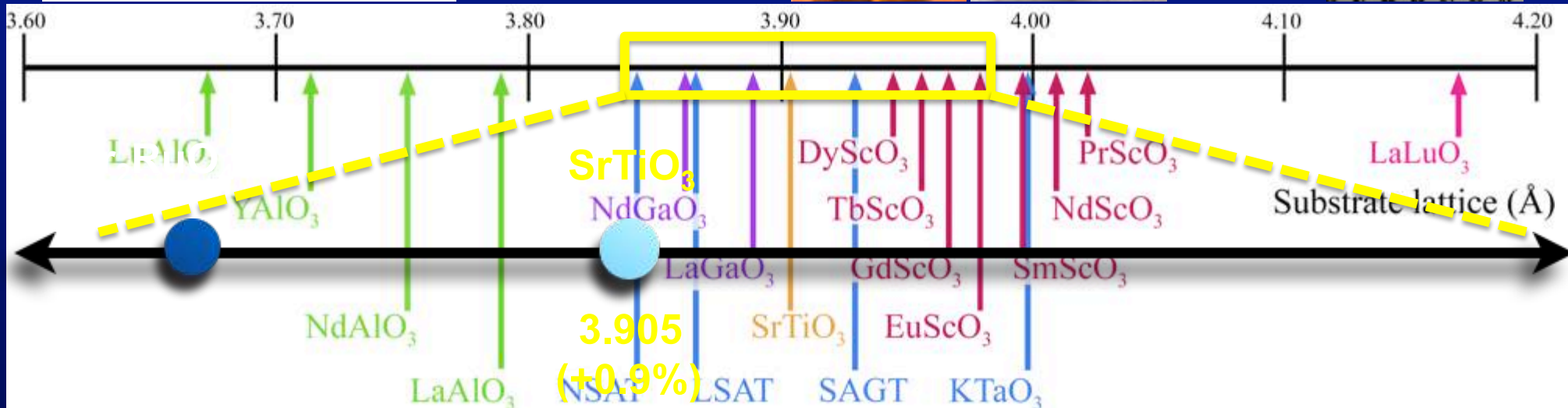
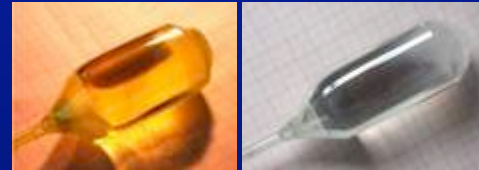
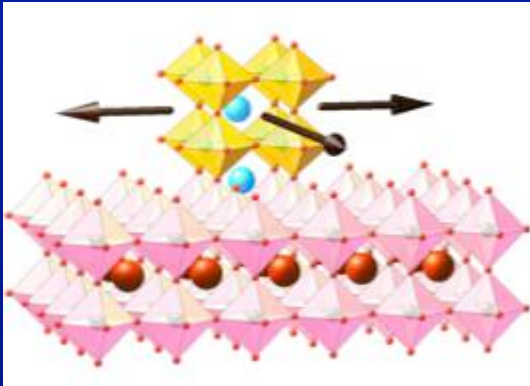
$$\epsilon_{xx} = -0.75\%$$

enhancements in T_c may be tied to proximity of van Hove singularity to E_F , but strains that can be applied to single crystal Sr_2RuO_4 are relatively modest ($\leq 0.8\%$; $T_{c,\text{max}}$ of 3.4 K at 0.6% uniaxial compressive strain)

C.W. Hicks, D.O. Brodsky, E.A. Yelland, A.S. Gibbs, J.A.N. Bruin, M.E. Barber, S.D. Edkins, K. Nishimura, S. Yonezawa, Y. Maeno, and A.P. Mackenzie, *Science* **344** (2014) 283–285.

A. Steppke, L. Zhao, M.E. Barber, T. Scaffidi, F. Jerzembeck, H. Rosner, A.S. Gibbs, Y. Maeno, S.H. Simon, A.P. Mackenzie, and C.W. Hicks, *Science* **355**, 9398 (2017).

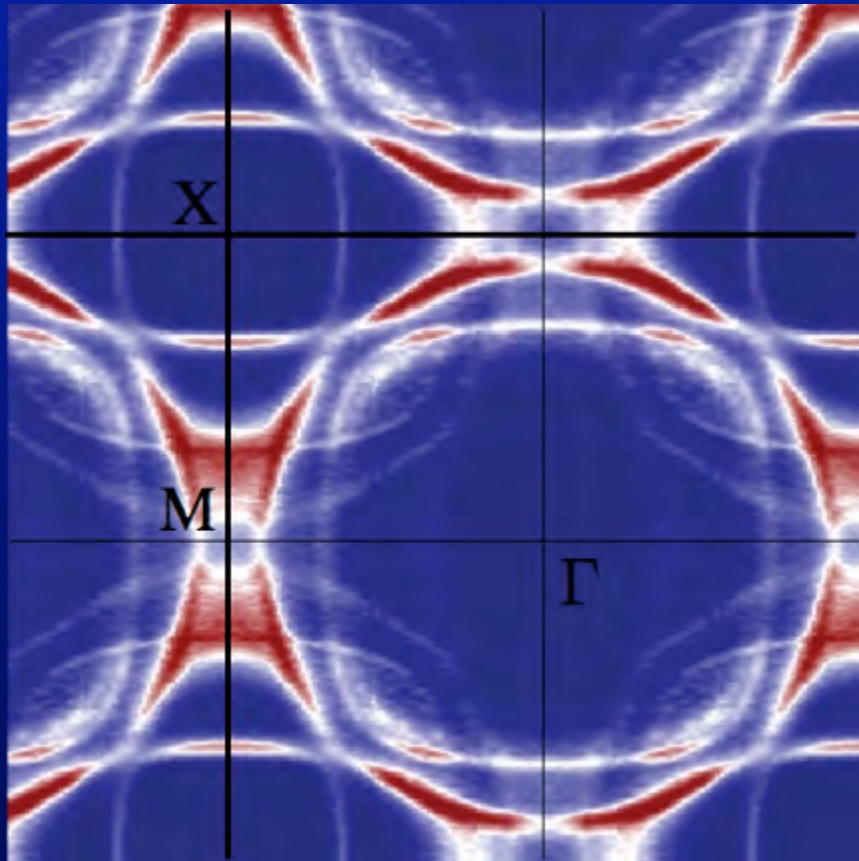
Much Larger Elastic Strains are Possible in Epitaxial Thin Films



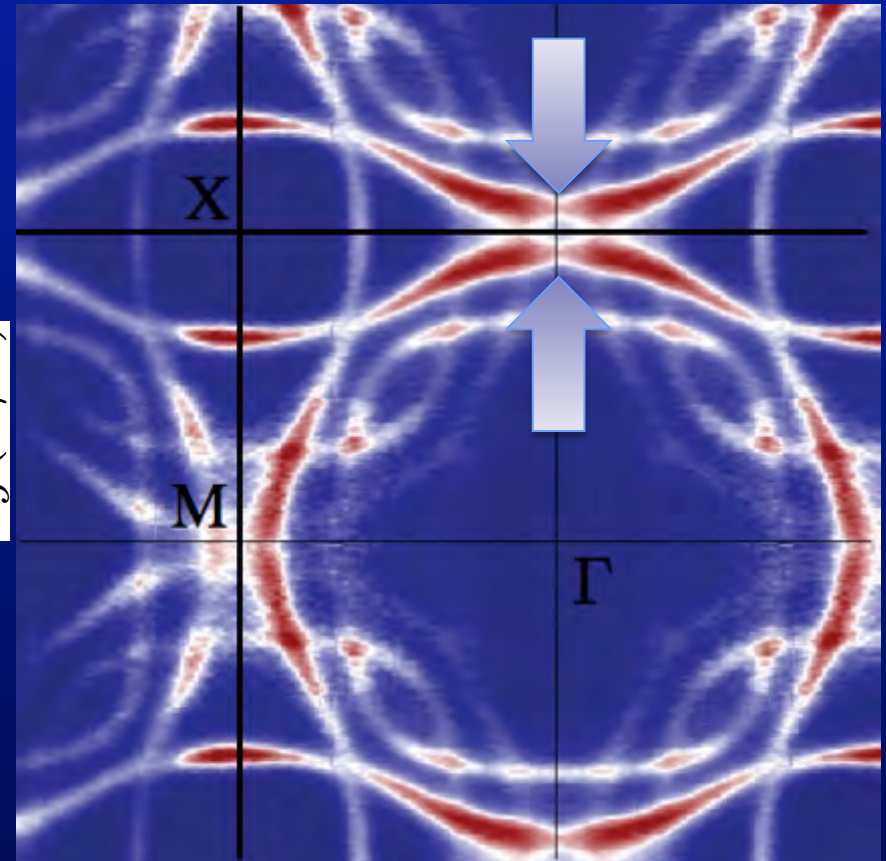
In-Plane Lattice Constant (Å)

Strain Control of Fermi Surface in Sr_2RuO_4

Unstrained Sr_2RuO_4

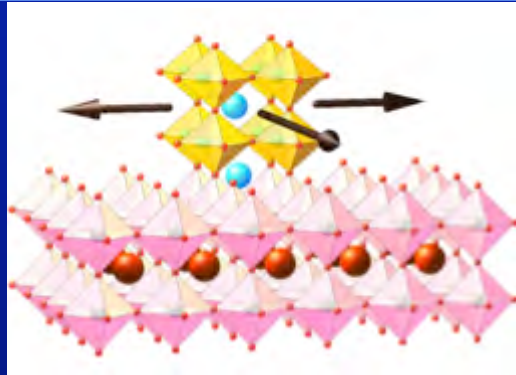


Sr_2RuO_4 on SrTiO_3 (+0.9%)

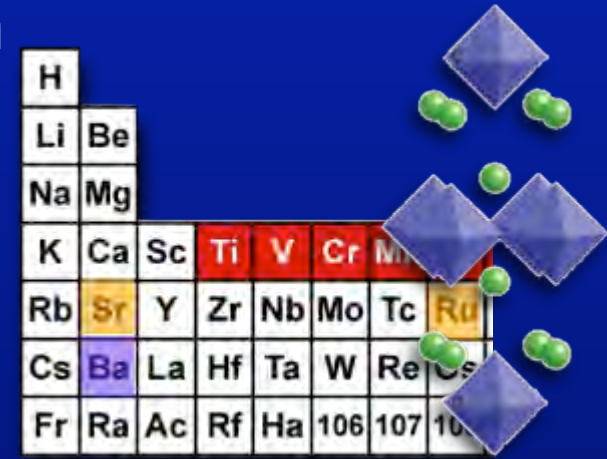
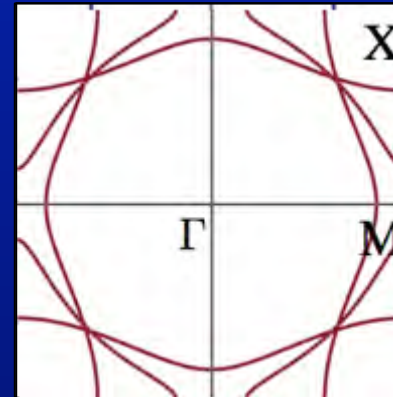


thin films still non-superconducting due to extreme sensitivity of spin-triplet SC to disorder, but low resistivities ($5 \mu\Omega\cdot\text{cm}$)

Pushing to Higher “strains” using Epitaxial Stabilization of Ba_2RuO_4



Lifshitz transition



Periodic table highlighting transition metals and their oxides, including DyScO_3 and GdScO_3 .

bulk
 Sr_2RuO_4

SrTiO_3

DyScO_3

GdScO_3

3.860

3.905
(+0.9%)

3.942
(+2.1%)

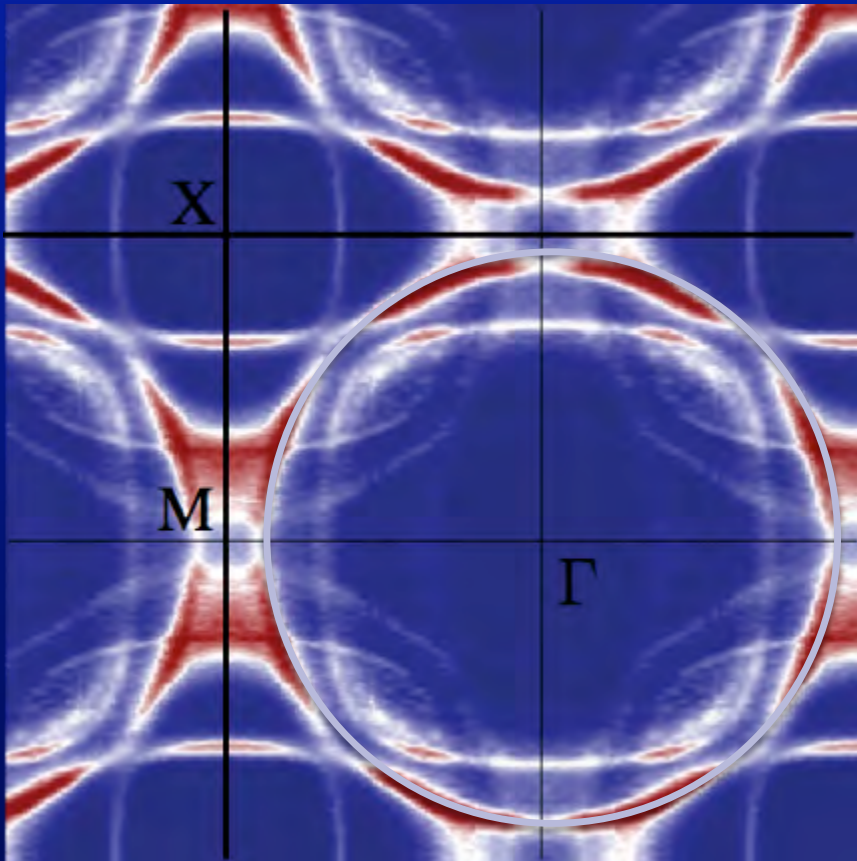
3.960
(+2.6%)

In-Plane Lattice Constant (\AA)

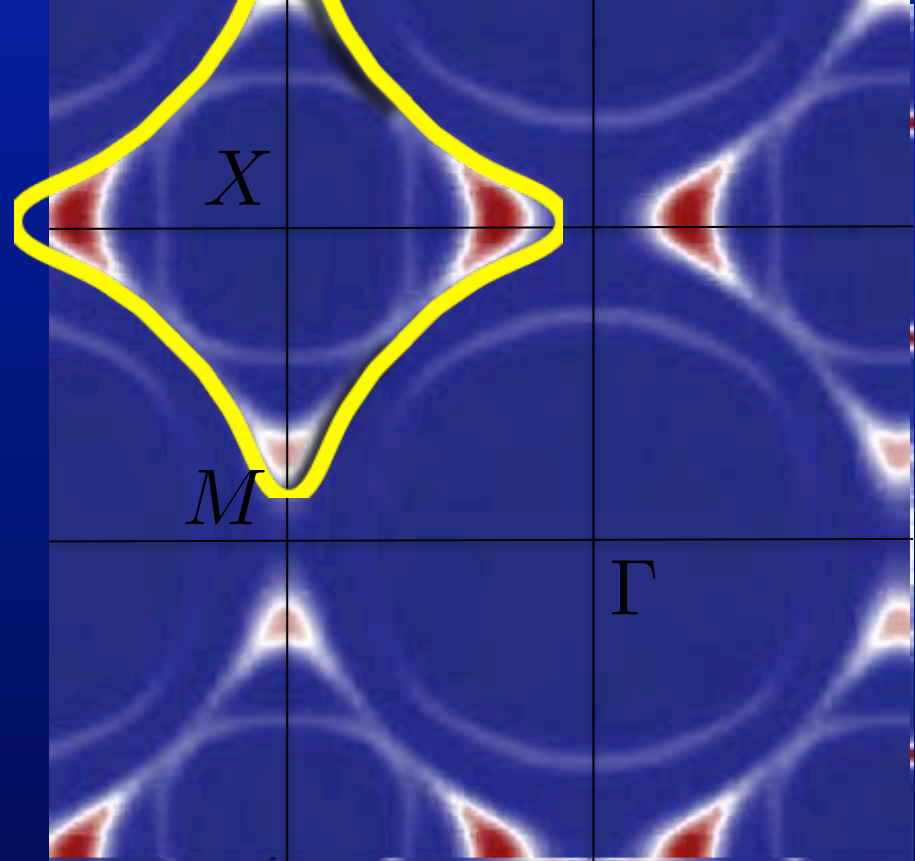
- Ba_2RuO_4 is isoelectronic and isostructural to Sr_2RuO_4
- Metastable in bulk, but can be epitaxially stabilized

Strain Control of Fermi Surface in Sr_2RuO_4

Unstrained Sr_2RuO_4

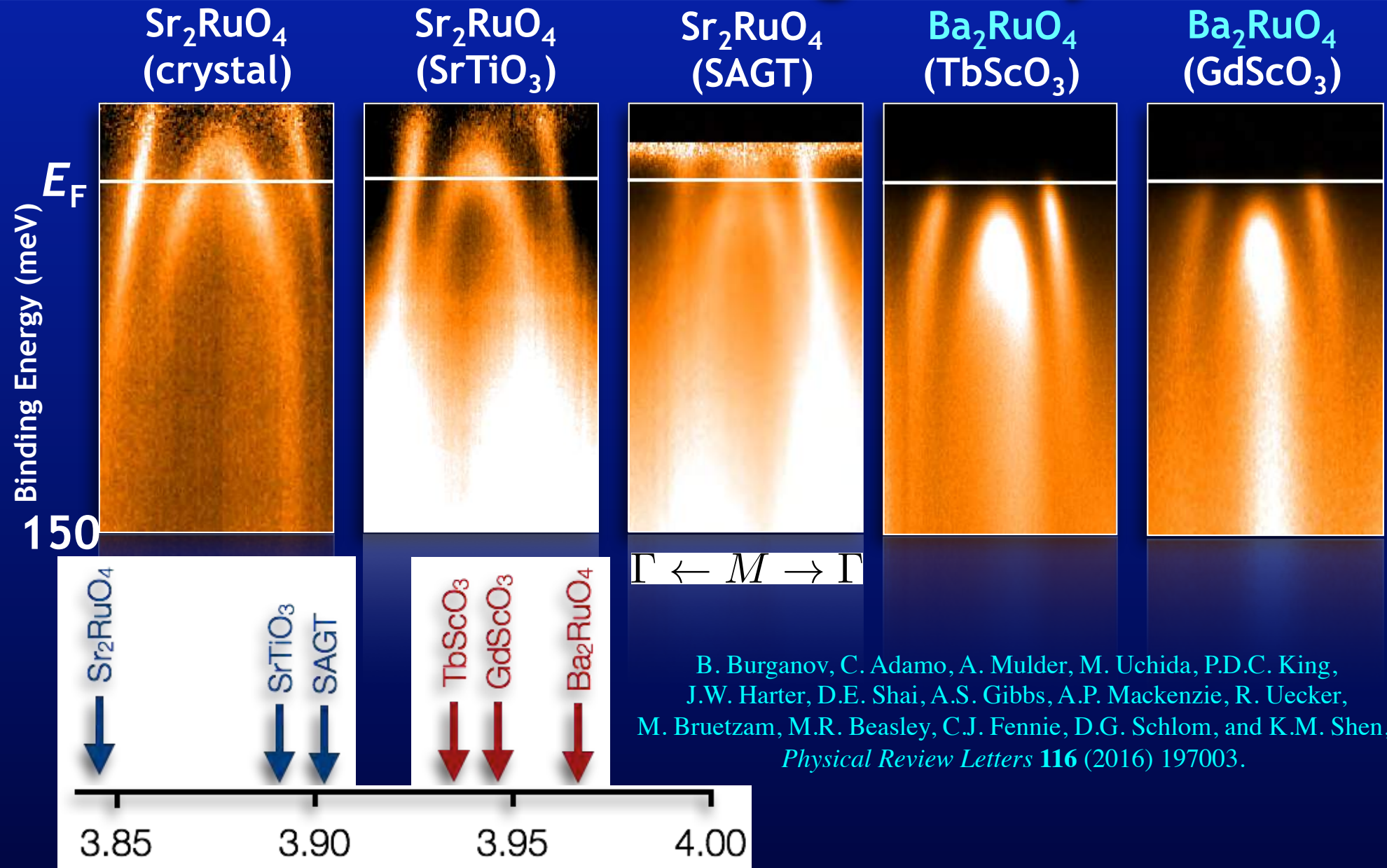


$\text{Ba}_{0.8}\text{Sr}_{1.2}\text{RuO}_4$ on GdTiO_3 (+0.8%)



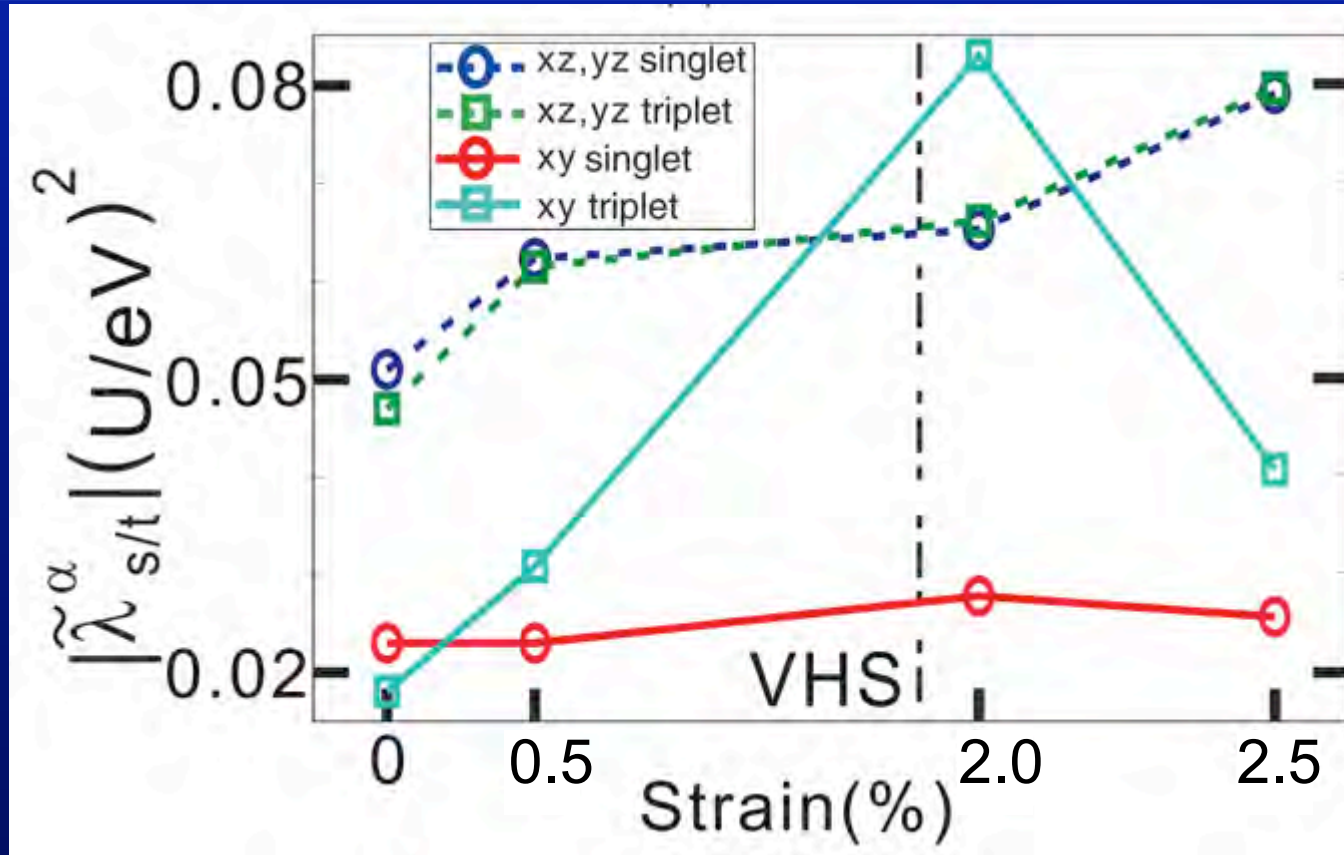
Large epitaxial strains turn the large electron-like Fermi surface closed around Γ to a hole-like Fermi surface closed around X

Strain Control of Band Structure and van Hove singularity



B. Burganov, C. Adamo, A. Mulder, M. Uchida, P.D.C. King, J.W. Harter, D.E. Shai, A.S. Gibbs, A.P. Mackenzie, R. Uecker, M. Bruetzam, M.R. Beasley, C.J. Fennie, D.G. Schlom, and K.M. Shen, *Physical Review Letters* **116** (2016) 197003.

Theory Predicts Enhancement of Spin-triplet Superconductivity



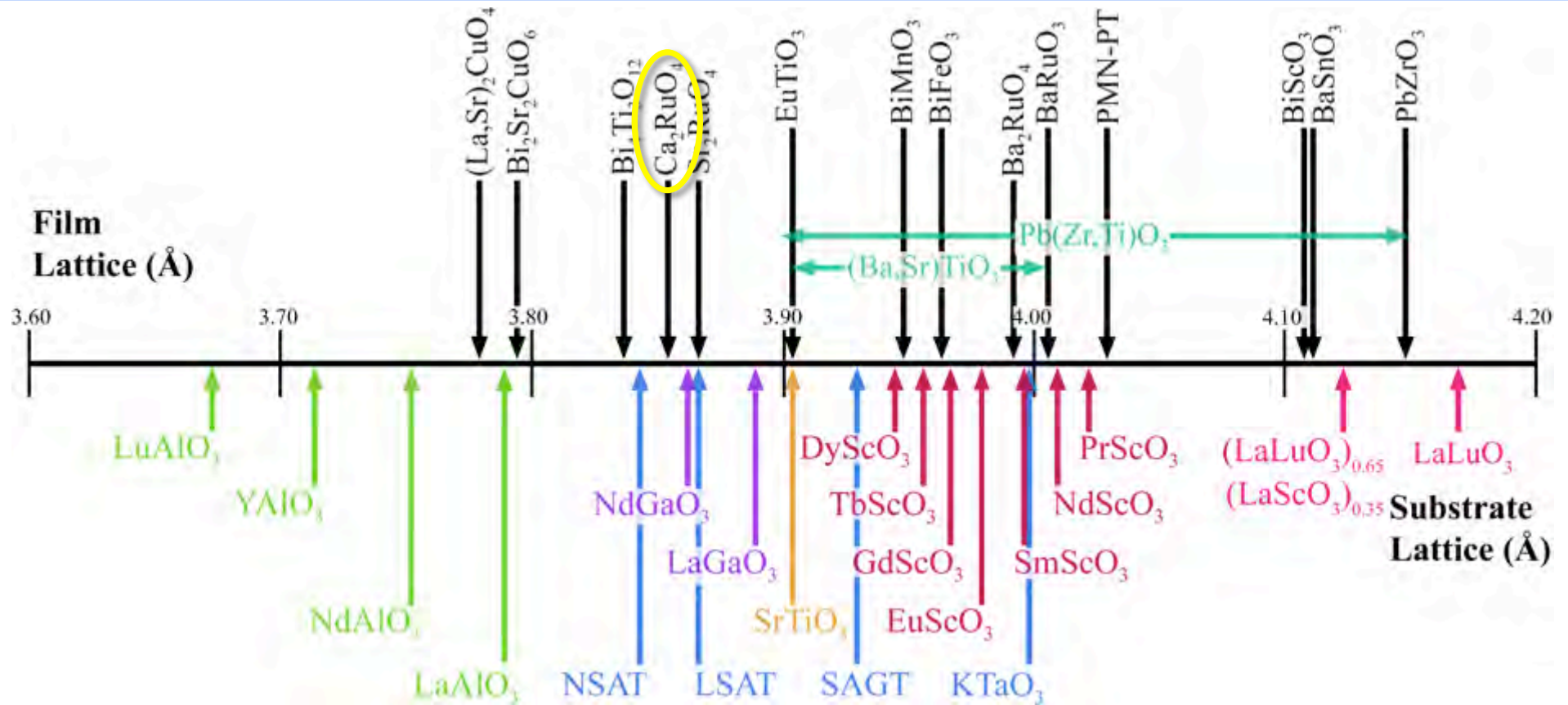
Eun-Ah Kim



Craig Fennie

Spin-triplet superconductivity is predicted to be strongly enhanced when the van Hove singularity is brought near the Fermi level (see arXiv:1604.06661)

Commercial Perovskite Substrates



[110] DyScO_3 , $d = 32 \text{ mm}$

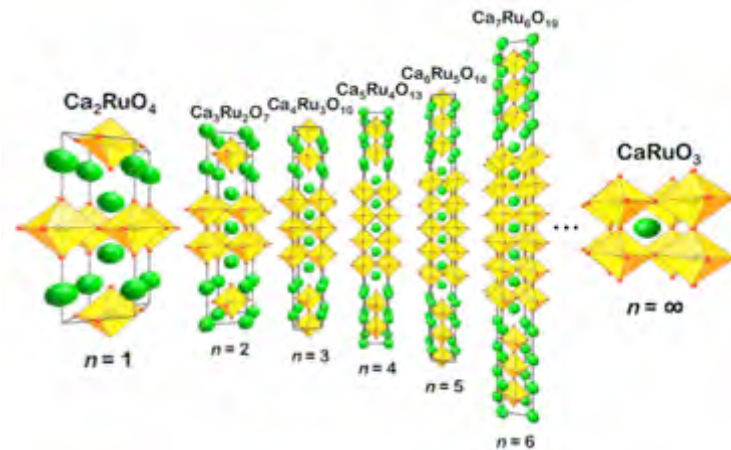
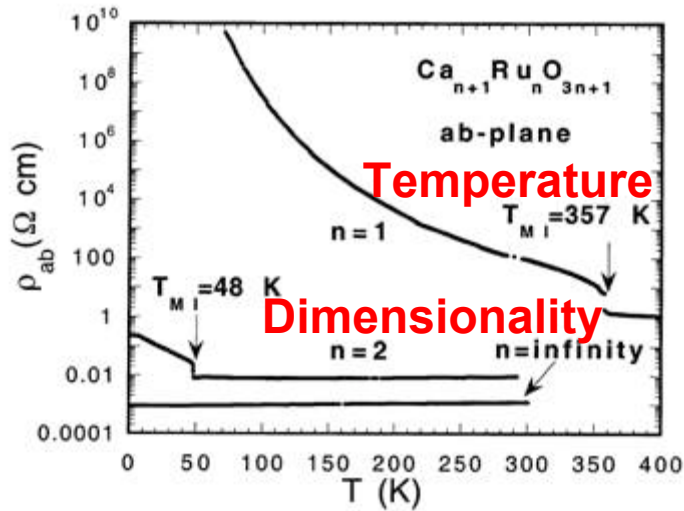


[110] GdScO_3 , $d = 32 \text{ mm}$

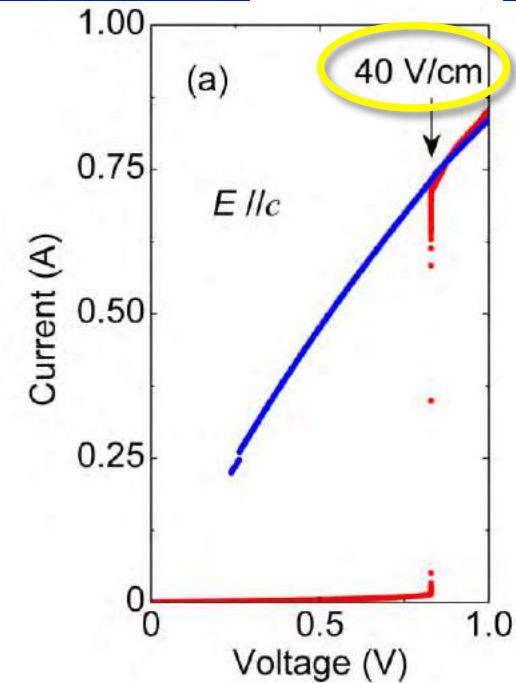


D.G. Schlom, L.Q. Chen, C.J. Fennie, V. Gopalan, D.A. Muller, X.Q. Pan, R. Ramesh, and R. Uecker, "Elastic Strain Engineering of Ferroic Oxides," *MRS Bulletin* **39** (2014) 118-130.

Ca₂RuO₄—a very Tunable System



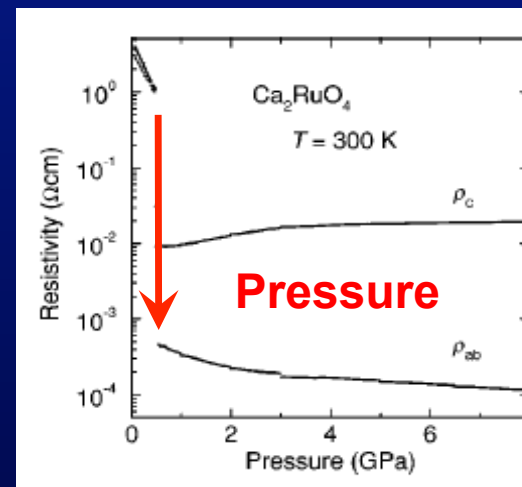
Electrical



G. Cao, C.S. Alexander, S. McCall, and J.E. Crow, *Mater. Sci. Eng., B* **63** (1999) 76–82.

Metal-Insulator Transition in Ca₂RuO₄ Triggered by

- Temperature
- Dimensionality
- Pressure
- Voltage/current



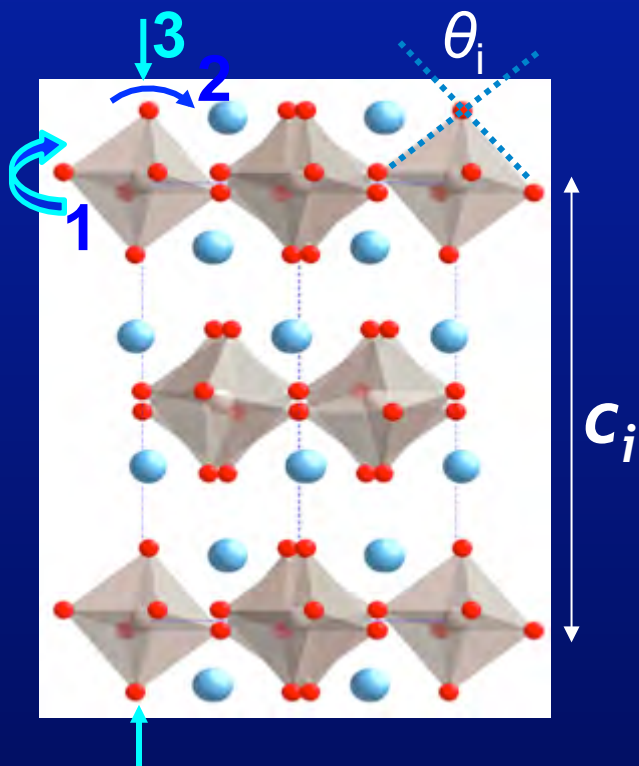
F. Nakamura, M. Sakaki, Y. Yamanaka, S. Tamaru, T. Suzuki, and Y. Maeno, *Sci. Rep.* **3** (2013) 1–6.

F. Nakamura, T. Goko, M. Ito, T. Fujita, S. Nakatsuji, H. Fukazawa, Y. Maeno, P. Alireza, D. Forsythe, and S. R. Julian *Phys. Rev. B* **65**, (2002) 220402.

Ca₂RuO₄—Phase Transition at T_{MIT}

F. Nakamura, M. Sakaki, Y. Yamanaka,
S. Tamaru, T. Suzuki, and Y. Maeno,
Sci. Rep. **3** (2013) 1–6.

Insulator

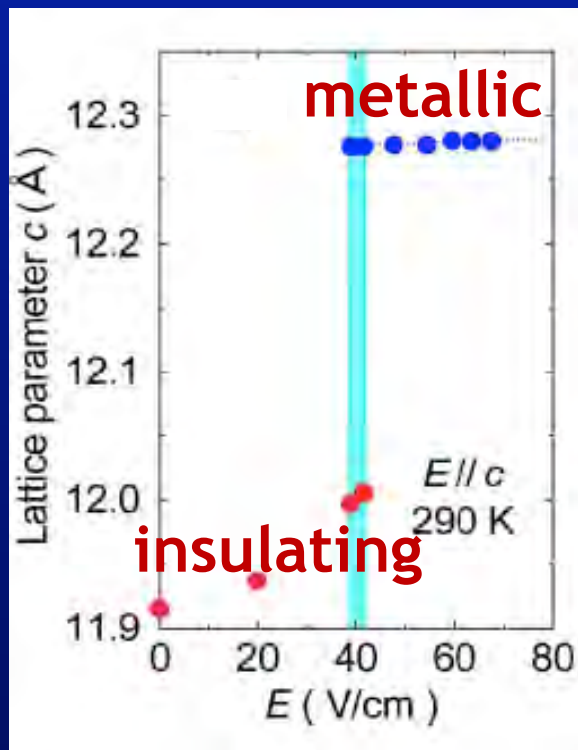


$$\theta_i = 91.3^\circ$$

$$a = 5.410 \text{ \AA}$$

$$b = 5.492 \text{ \AA}$$

$$c = 11.96 \text{ \AA}$$



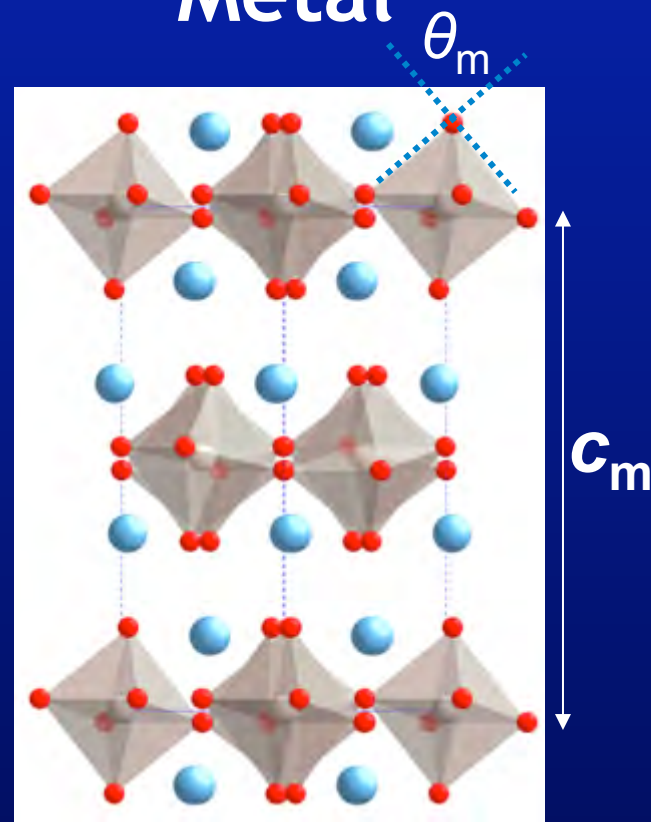
$$\theta_i > \theta_m$$

$$\Delta a = -0.9\%$$

$$\Delta b = -2.6\%$$

$$\Delta c = 2.5\%$$

Metal



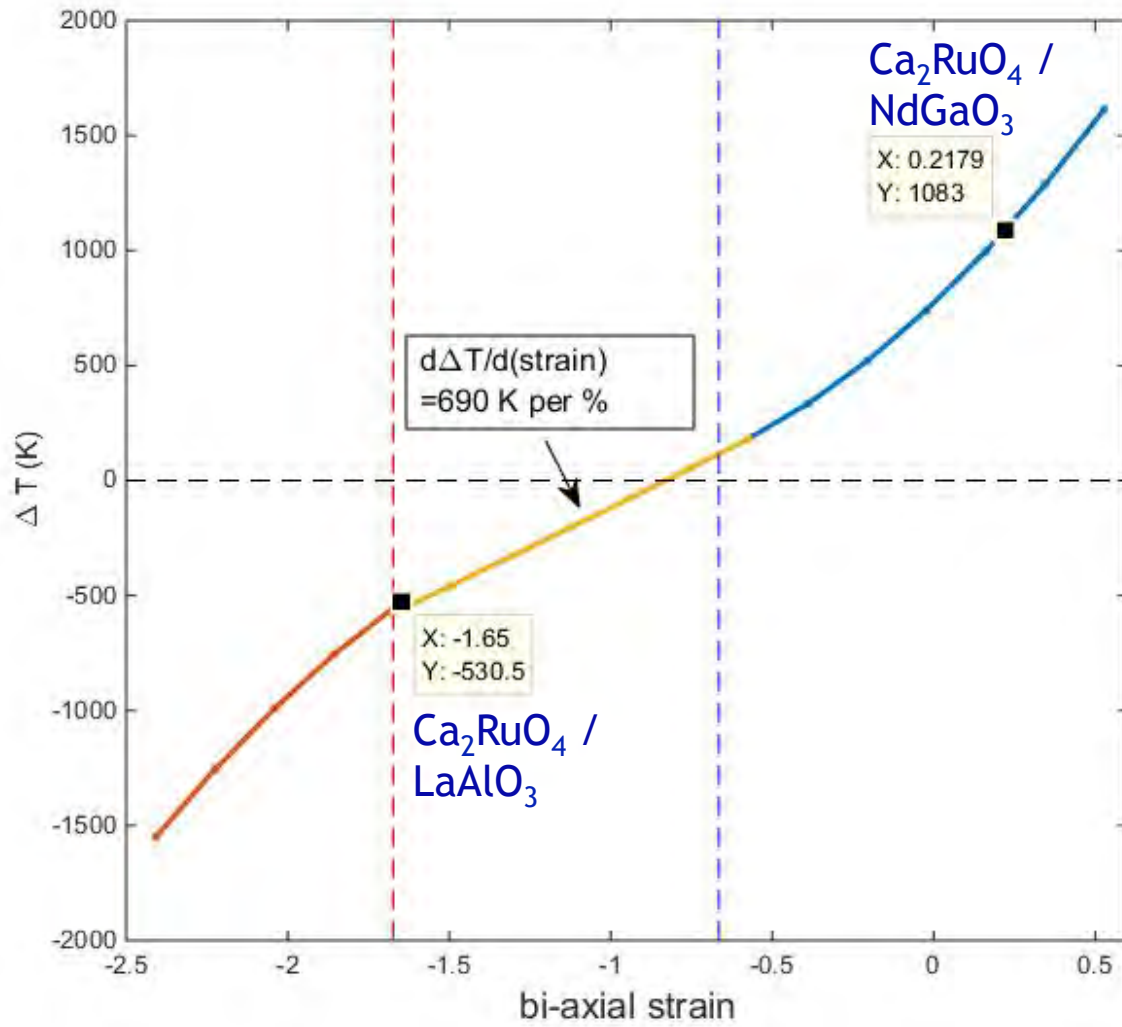
$$\theta_m = 87.3^\circ$$

$$a = 5.361 \text{ \AA}$$

$$b = 5.351 \text{ \AA}$$

$$c = 12.26 \text{ \AA}$$

Ca₂RuO₄—Effect of Strain on T_{MIT}



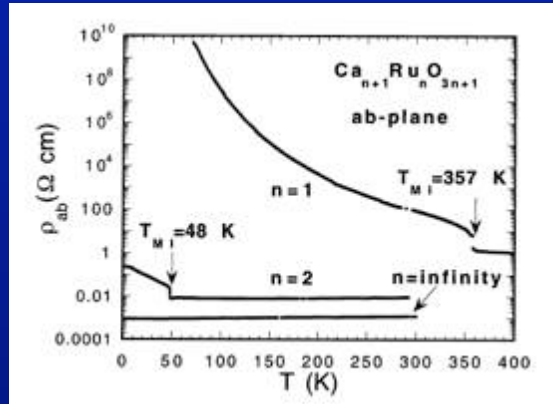
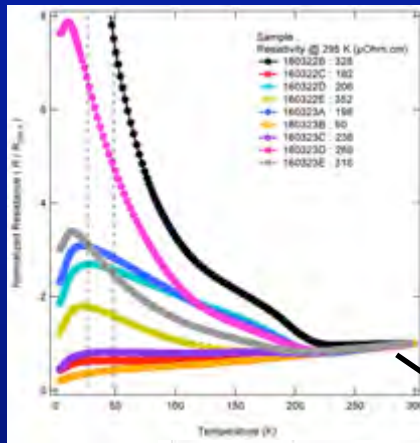
Andy Millis

$$\frac{\partial T_{MIT}}{\partial \epsilon_{\text{biaxial}}} = 69,000 \text{ K}$$

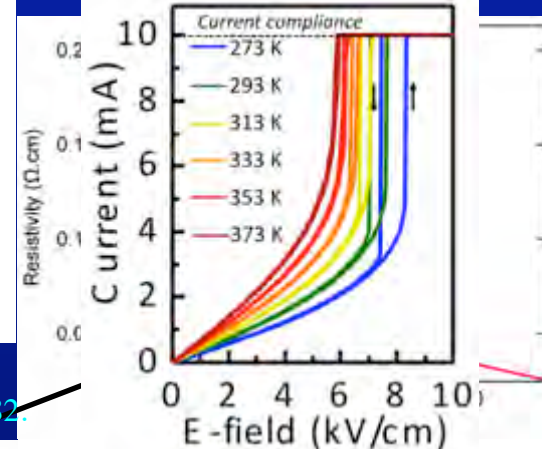
Transport on Strained Ca_2RuO_4

Bulk - unstrained

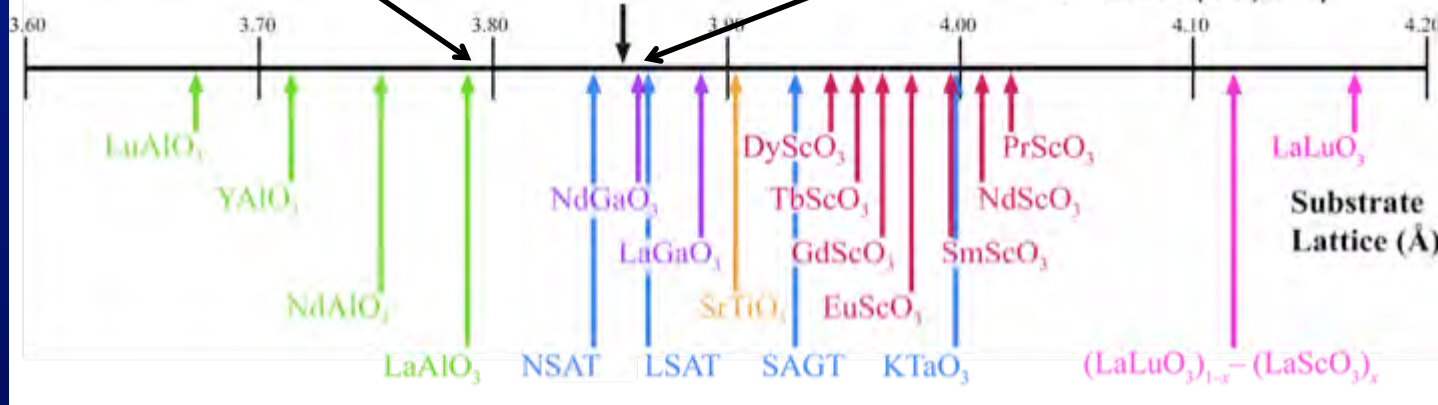
Compressive



Tensile



G. Cao, C.S. Alexander, S. McCall, and J.E. Crow, *Mater. Sci. Eng., B* 63 (1999) 76–82.

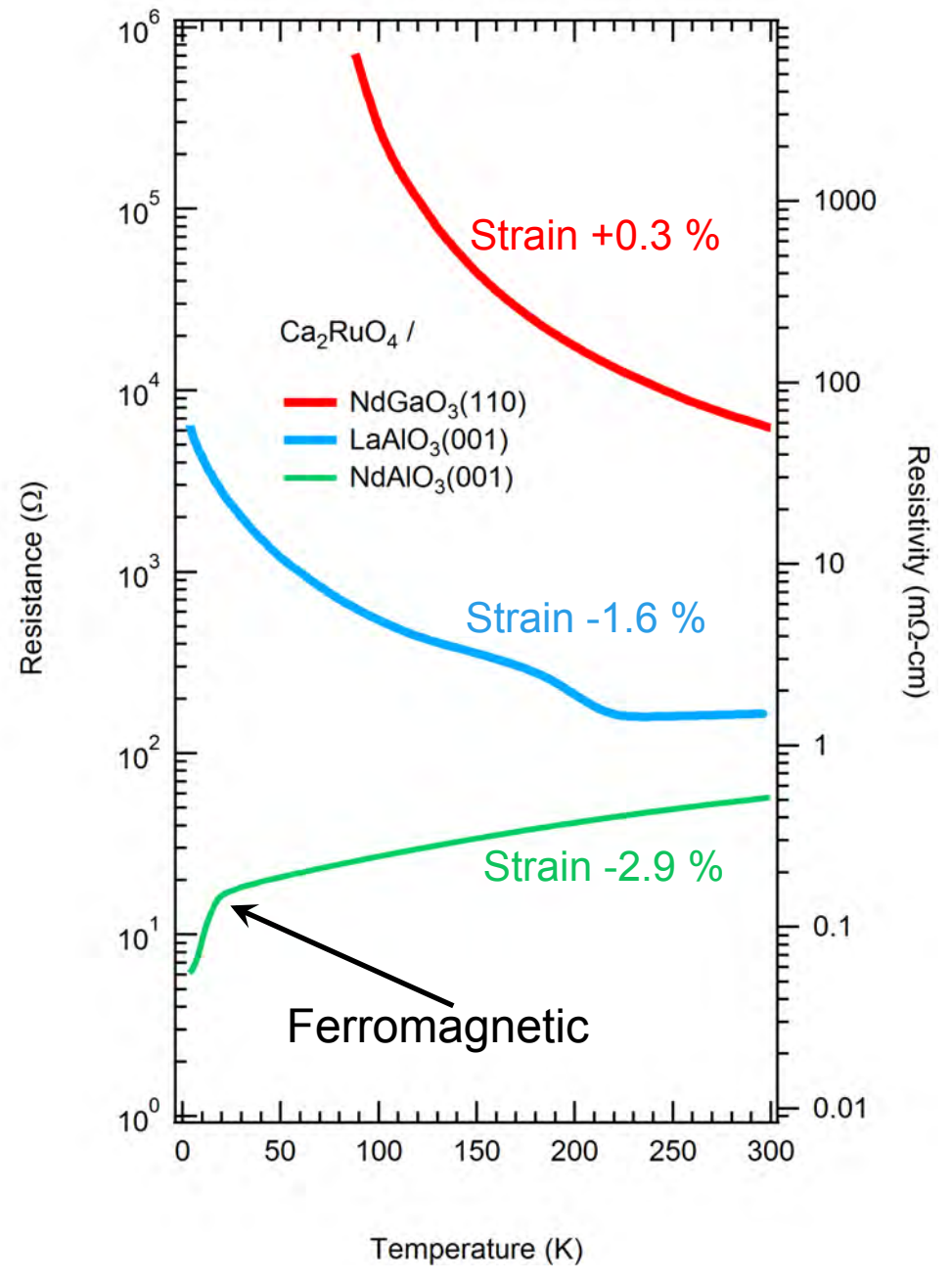


Bulk – T_{MIT} at 360 K

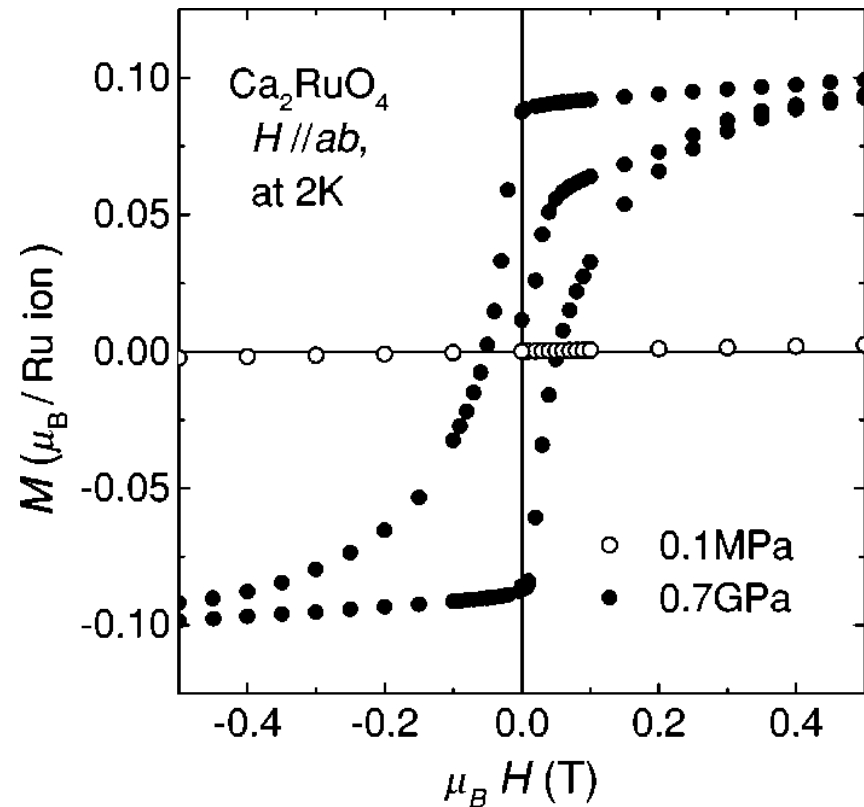
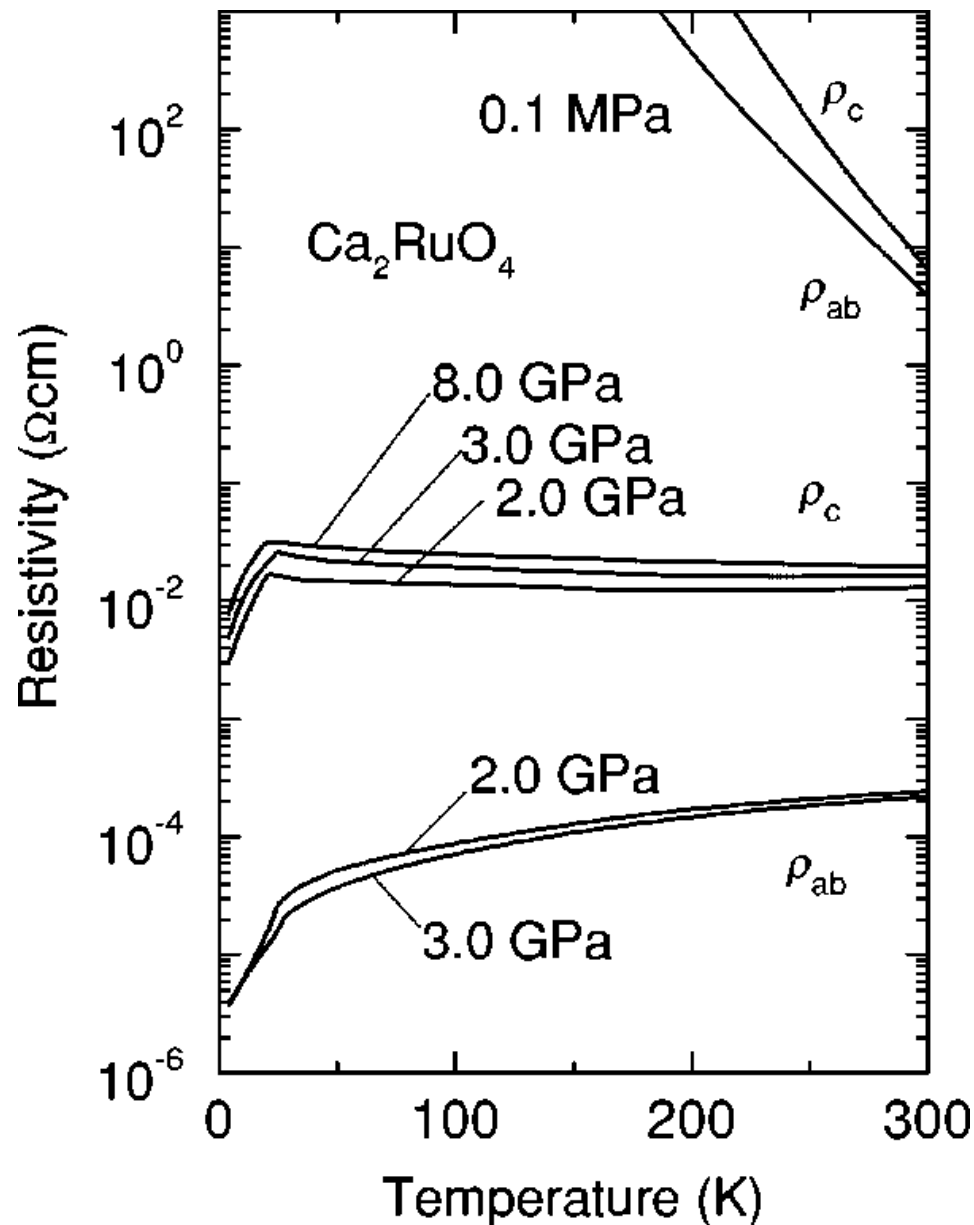
0.3% tensile – T_{MIT} above 650 K

1.6% compressive – T_{MIT} suppressed or shifted to lower T

Transport on Strained Ca_2RuO_4



Ca₂RuO₄ under Hydrostatic Pressure



F. Nakamura, T. Goko, M. Ito, T. Fujita, S. Nakatsuji, H. Fukazawa, Y. Maeno, P. Alireza, D. Forsythe, and S. R. Julian, *Phys. Rev. B* **65** (2002) 220402.

Unleashing Hidden Properties

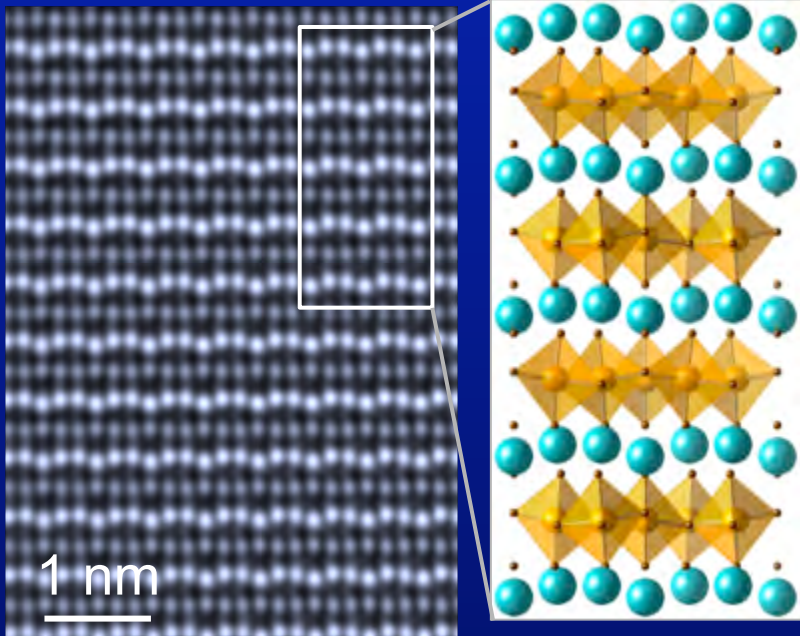
- Toolkit Enabling Materials-by-Design of Oxide Thin Films
- Ruthenates— BaRuO_3 , $(\text{Ca}, \text{Sr}, \text{Ba})_2\text{RuO}_4$ altering band structure and properties
- EuTiO_3
strongest ferromagnetic ferroelectric
- $(\text{LuFeO}_3)_9 / (\text{LuFe}_2\text{O}_4)_1$ Superlattices
strongest ferromagnetic ferroelectric
at room temperature

Room-Temperature Multiferroics

Material	M_s and P_s	T_N and T_C	Synthesis
BiFeO_3	$M_s \sim 0.03 \mu_B/\text{Fe}$ $P_s \sim 95 \mu\text{C}/\text{cm}^2$	$T_N \sim 643 \text{ K}$ $T_C \sim 1100 \text{ K}$	
BiCoO_3	$M_s = 0$ (AFM) $P_{s,\text{theory}} \sim 170 \mu\text{C}/\text{cm}^2$	$T_N \sim 470 \text{ K}$ $T_C > 520 \text{ K}$	60,000 atm
ScFeO_3 (corundum polymorph)	$M_s \sim 0.01 \mu_B/\text{Fe}$ $P_{s,\text{theory}} \sim 3 \mu\text{C}/\text{cm}^2$	$T_N \sim 356 \text{ K}$ $T_{C,\text{theory}} \sim 1300 \text{ K}$	60,000 atm or Epitaxial Stabilization
$[(\text{Ca}_{0.6}\text{Sr}_{0.4})_{1.15}$ $\text{Tb}_{1.85}\text{Fe}_2\text{O}_7]_{0.83}$ — $[\text{Ca}_3\text{Ti}_2\text{O}_7]_{0.17}$	$M_s \sim 0.009 \mu_B/\text{Fe}$ $P_{s,\text{theory}} \sim 10 \mu\text{C}/\text{cm}^2$	$T_N \sim 330 \text{ K}$ $T_C \sim 430 \text{ K}$	

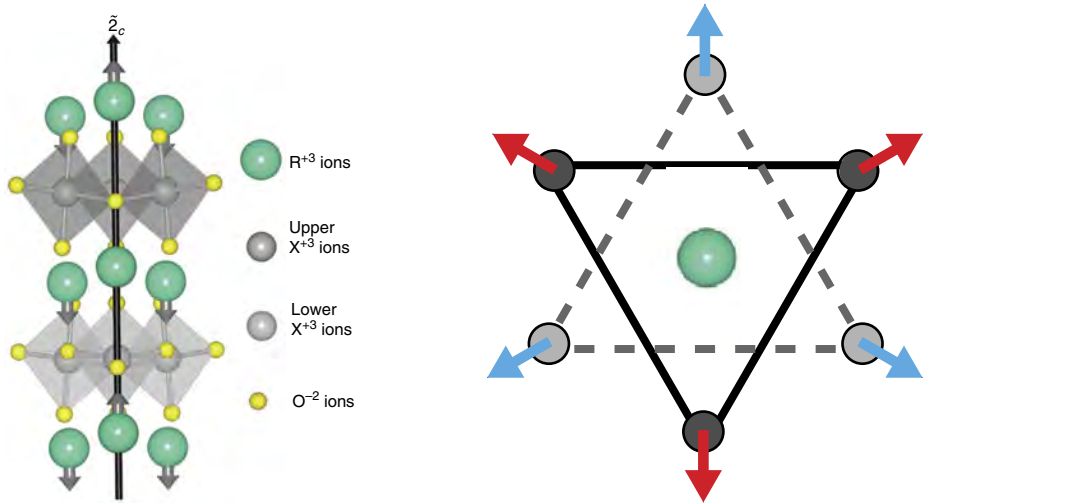
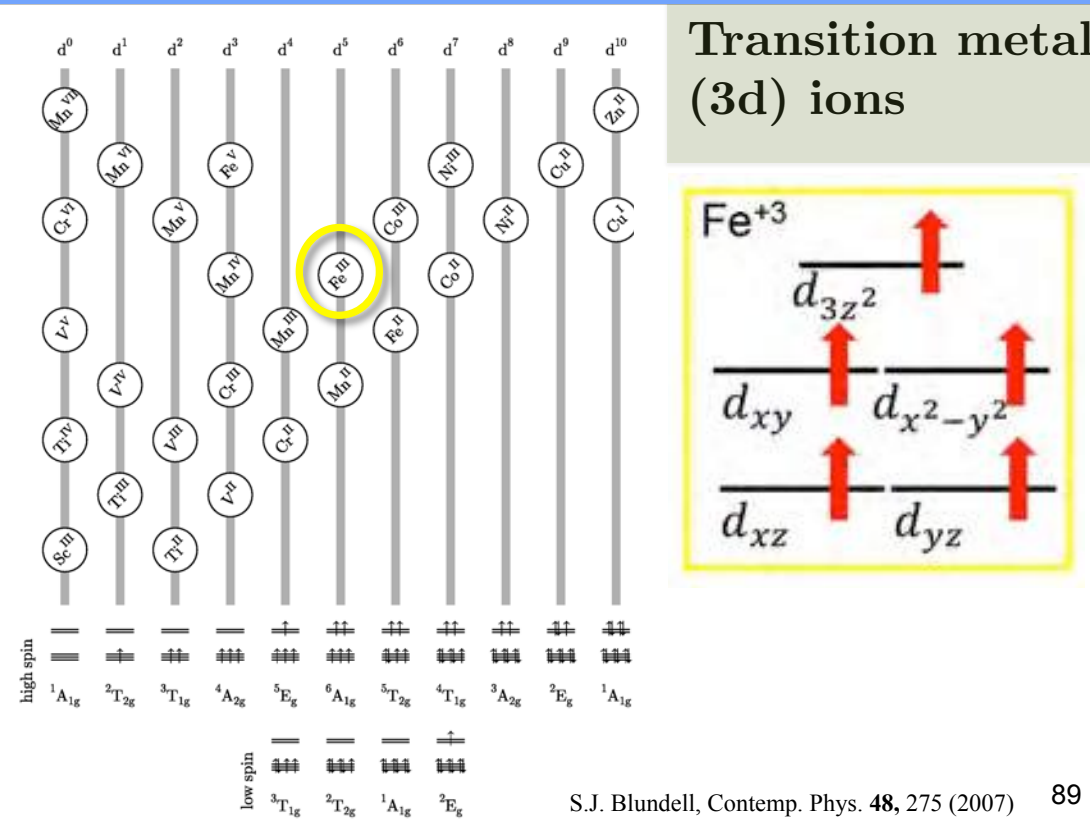
The Idea: $(\text{LuFeO}_3)_m / (\text{LuFe}_2\text{O}_4)_n$ Superlattices

LuFeO₃



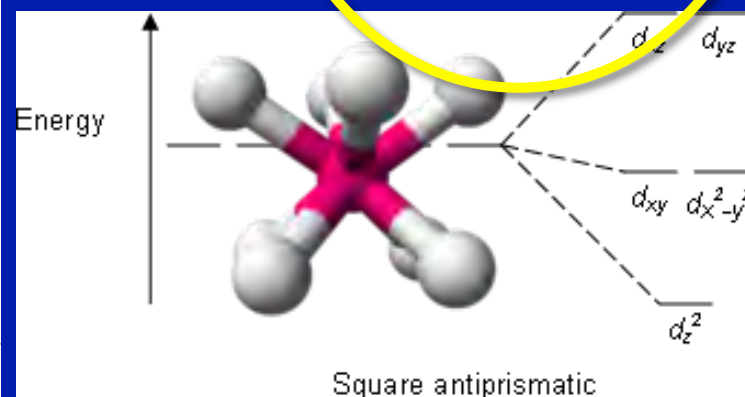
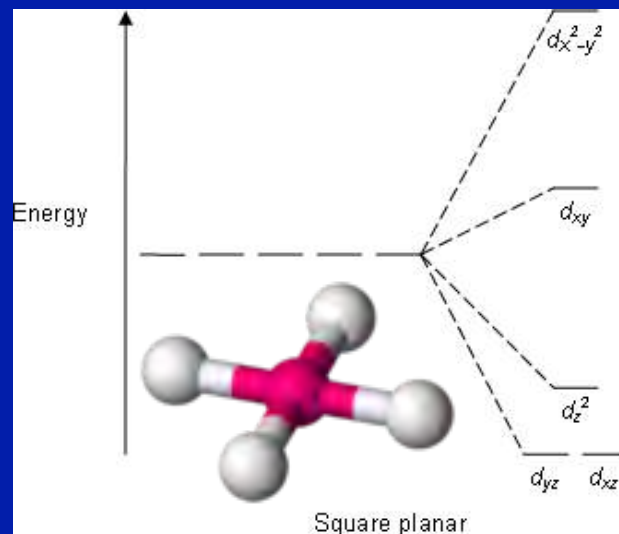
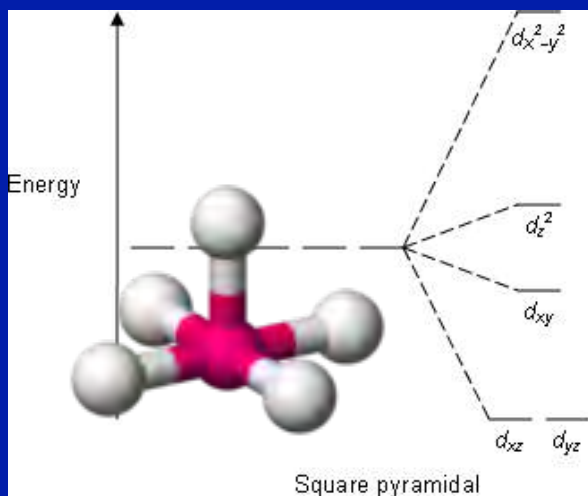
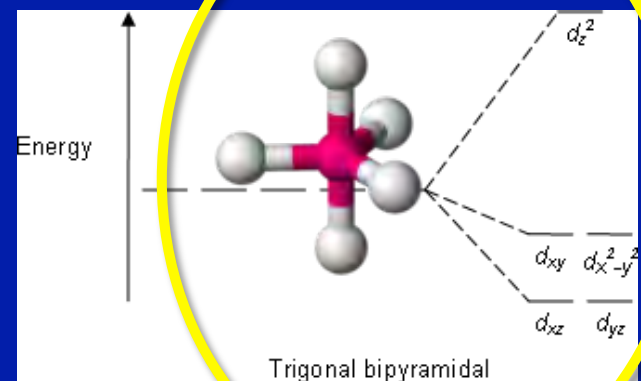
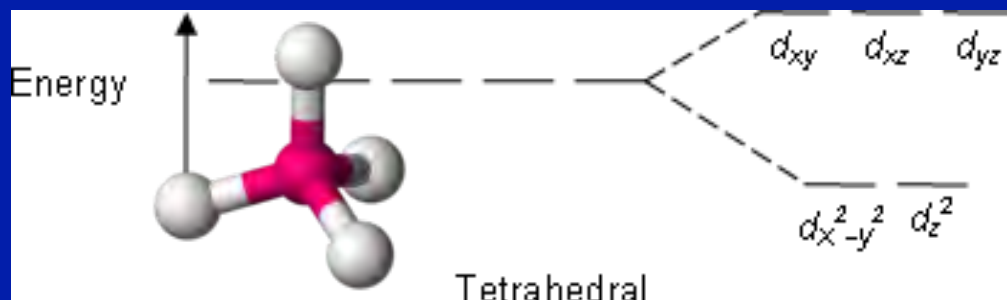
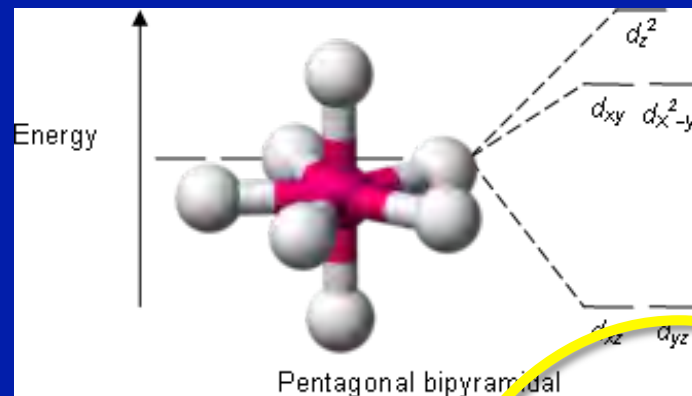
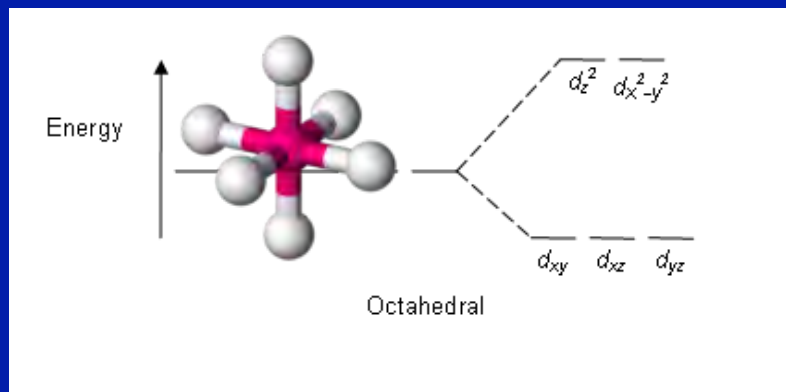
Canted AFM, $T_N \sim 147$ K
Ferroelectric, $T_C \sim 1050$ K

Bossak, A.A. *et al. Chem. Mater.* **16**, 1751 (2004)
 Wang, W. *et al. Phys. Rev. Lett.* **110**, 237601 (2013)
 Moyer, J. A. *et al. APL Mater.* **2**, 12106 (2014).
 H. Das, A. L. Wysocki, Y. Geng, W. Wu and
 Disseler, S. M. *et al. Phys. Rev. Lett.* **114**, 217601 (2015)
 C. J. Fennie, "Bulk Magnetoelectricity in the
 Hexagonal Manganites and Ferrites,"
Nature Communications **5**, 2998 (2014).



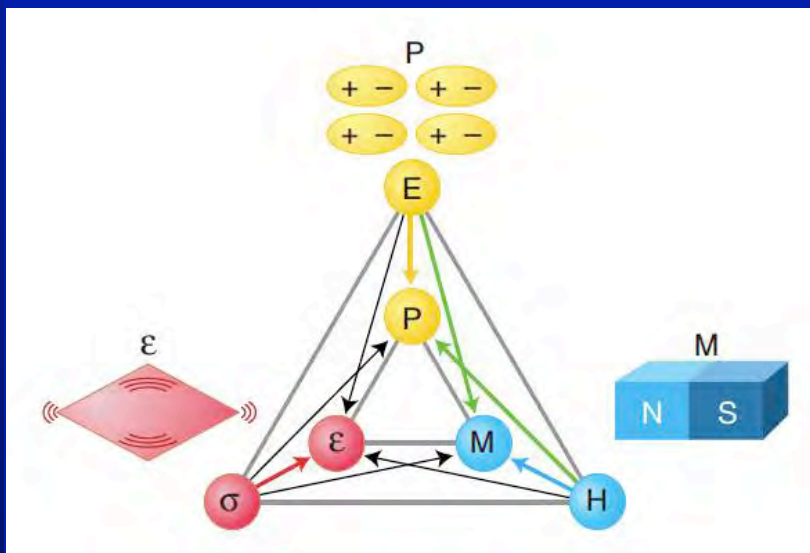
From Lecture #2

Other environments



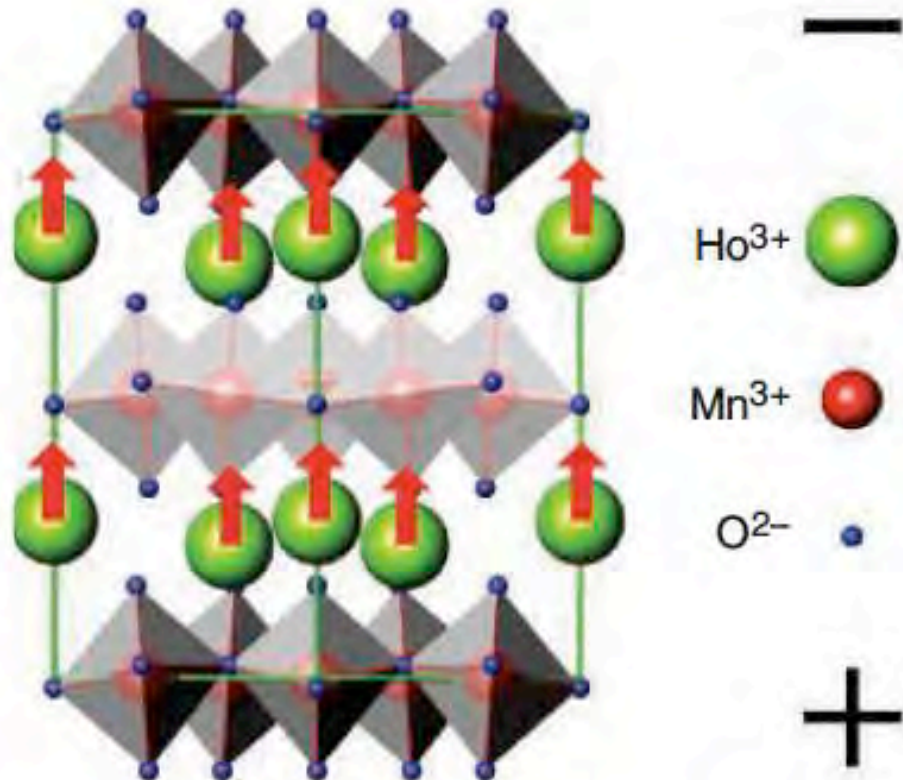
From Lecture #1

Ferroelectricity + Magnetism: *Multiferroics*



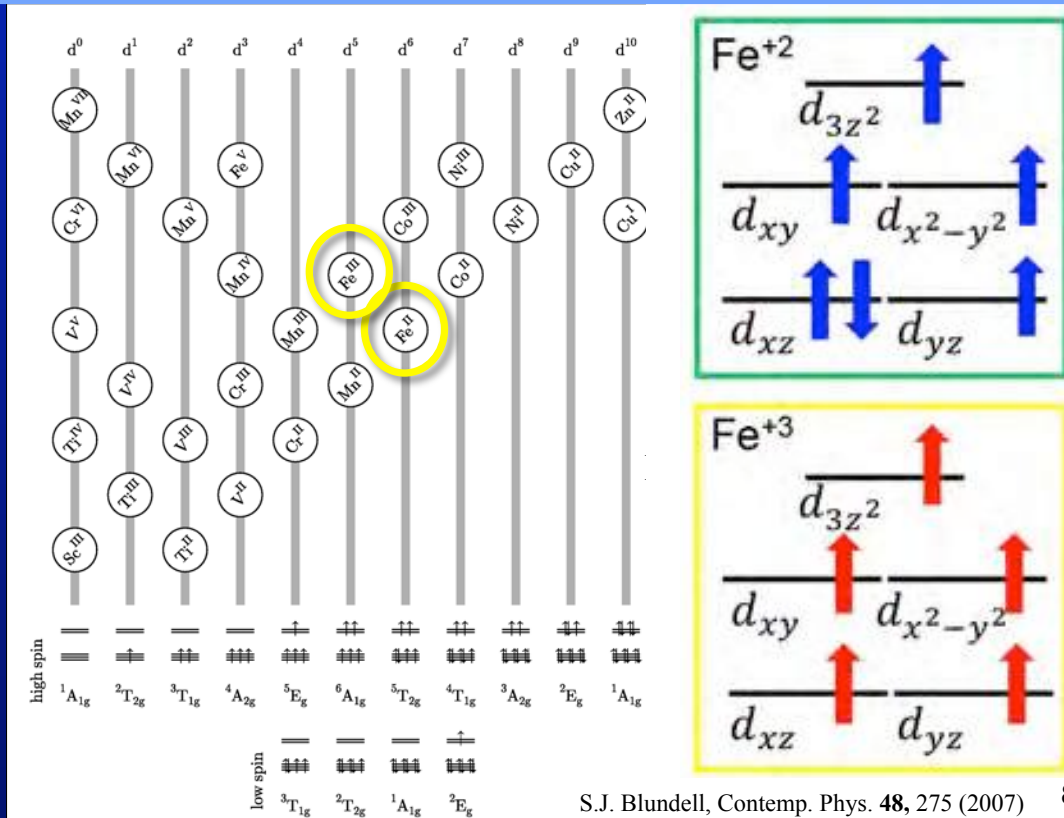
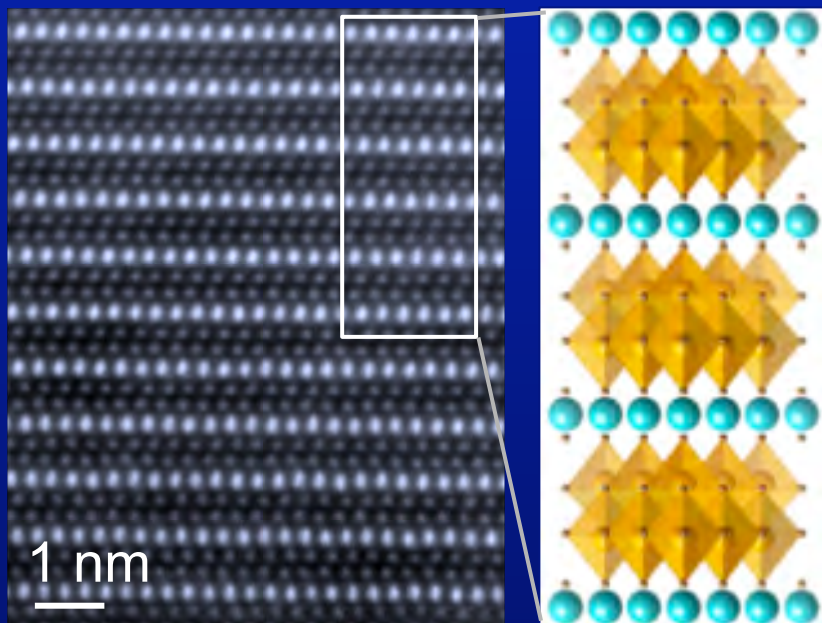
Phase control in ferroics and multiferroics. The electric field E , magnetic field H , and stress σ control the electric polarization P , magnetization M , and strain ϵ , respectively. In a ferroic material, P , M , or ϵ are spontaneously formed to produce ferromagnetism, ferroelectricity, or ferroelasticity, respectively. In a multiferroic, the coexistence of at least two ferroic forms of ordering leads to additional interactions. In a magnetoelectric multiferroic, a magnetic field may control P or an electric field may control M (green arrows).

N.Spaldin, Science, 2005



Structure of multiferroic HoMnO_3 . Hexagonal HoMnO_3 is ferroelectric, because the oxygen bipyramids surrounding each Mn^{3+} ion are tilted and shifted relative to the Ho^{3+} ions. It is also magnetic, with ferromagnetic alignment of the Ho^{3+} magnetic moments combined with antiferromagnetic Mn^{3+} ordering. Therefore, hexagonal HoMnO_3 is multiferroic.

The Idea: $(\text{LuFeO}_3)_m / (\text{LuFe}_2\text{O}_4)_n$ Superlattices



Ferrimagnetic, $T_C \sim 240$ K

Ikeda, N. *et al.* *Nature* **436**, 1136–1138 (2005).

Iida, J. *et al.* *J. Phys. Soc. Jpn.* **62** (1993) 1723-1735.

Non-Polar

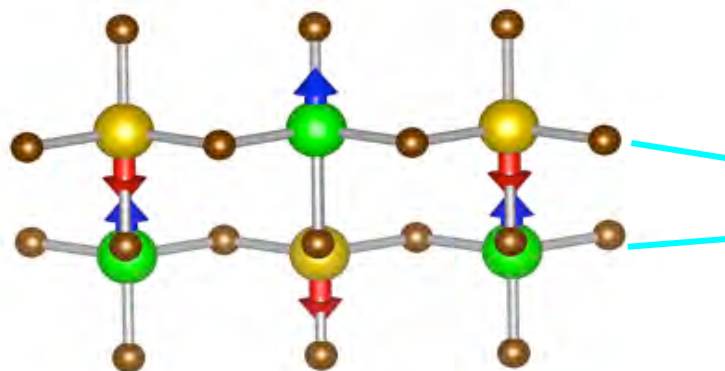
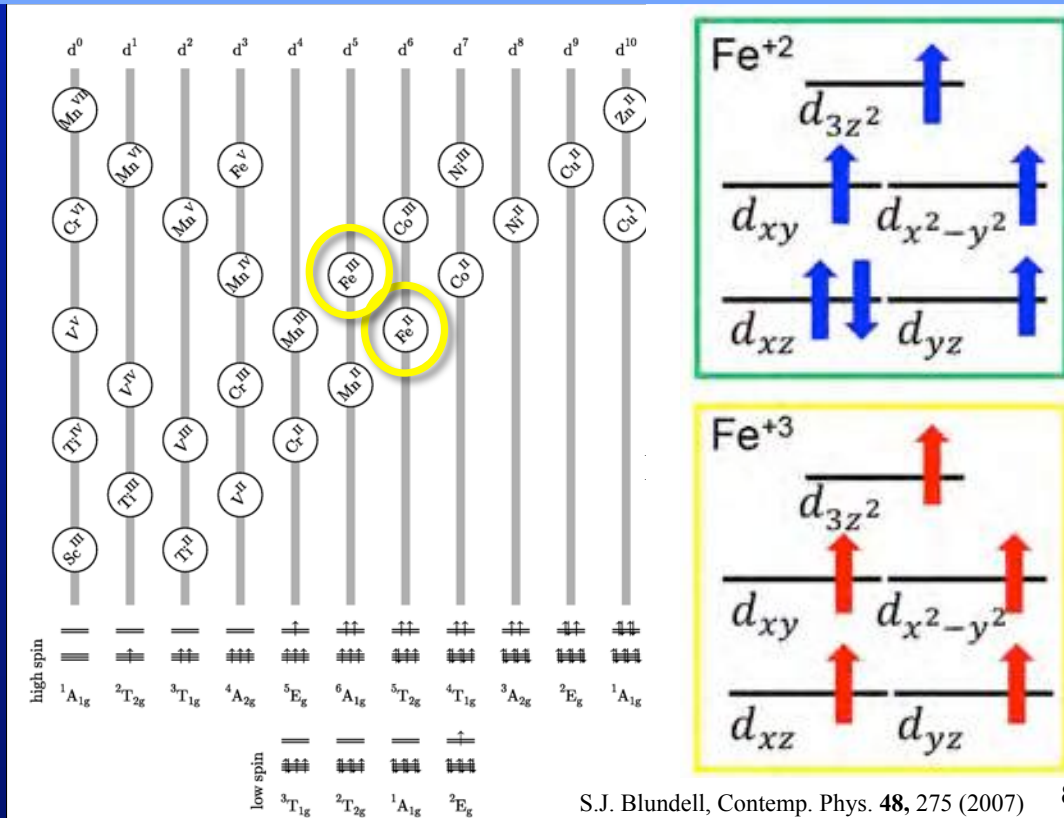
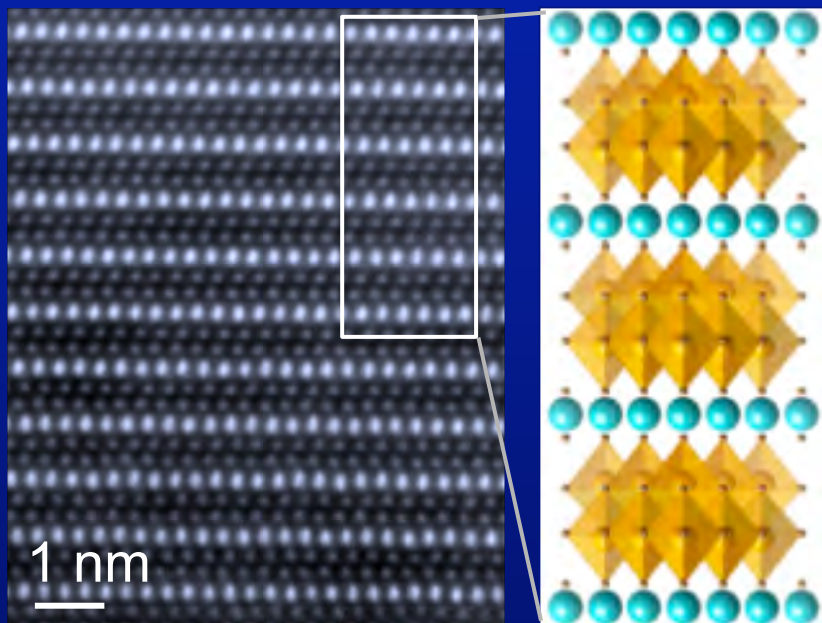
De Groot, J. *et al.* *Phys. Rev. Lett.*

108, 187601 (2012).

S.J. Blundell, *Contemp. Phys.* **48**, 275 (2007)

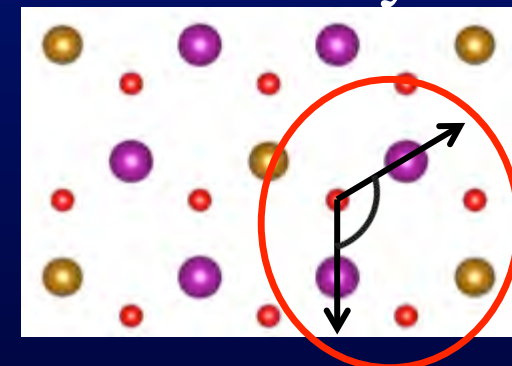
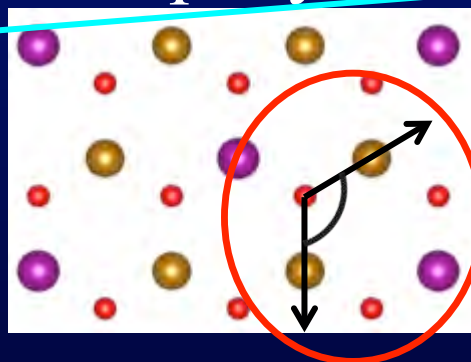
The Idea: $(\text{LuFeO}_3)_m / (\text{LuFe}_2\text{O}_4)_n$ Superlattices

LuFe_2O_4



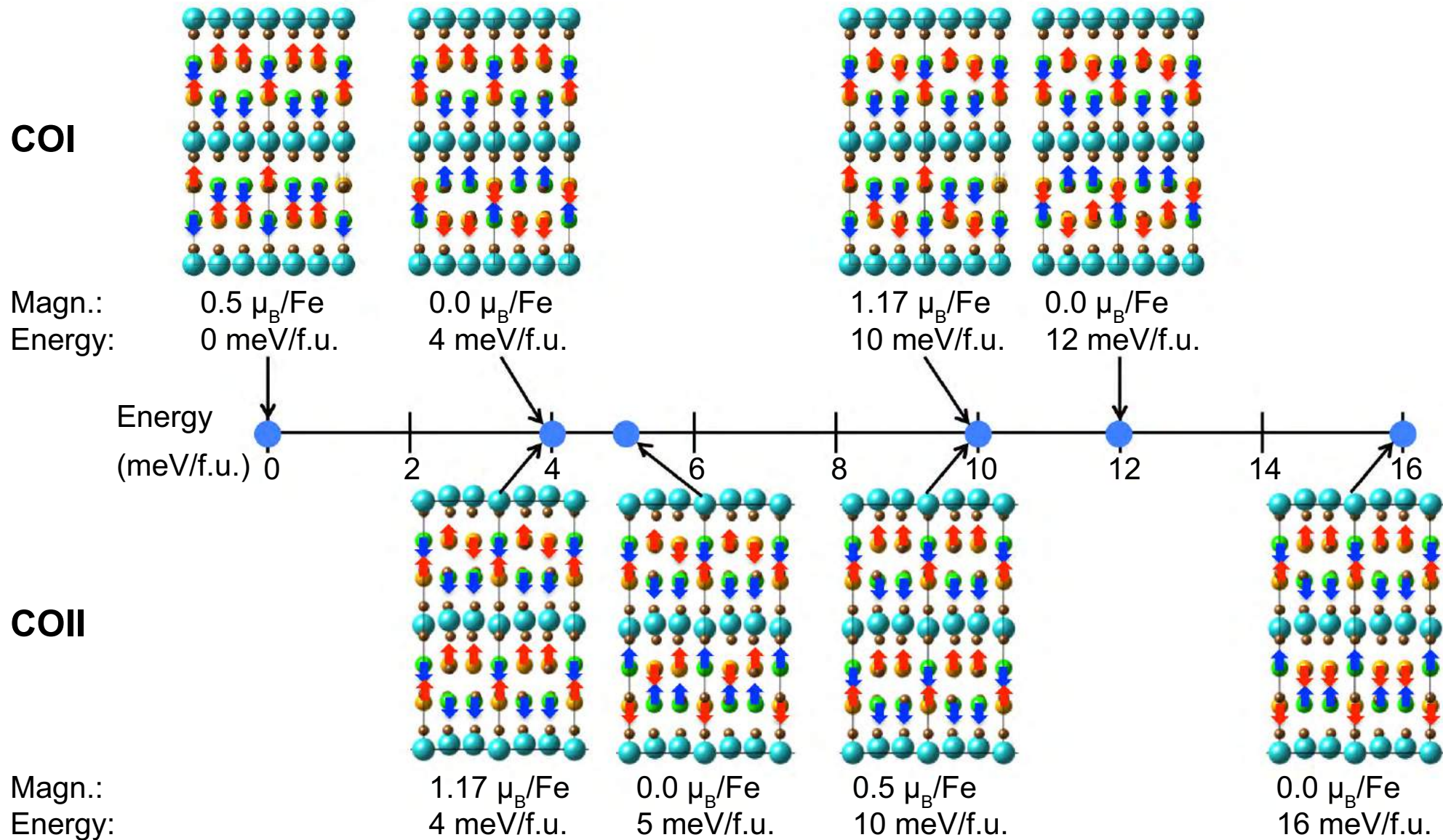
$$M = g\mu_B \left(\frac{3S_{\text{Fe}^{3+}}}{5/2} - \frac{3S_{\text{Fe}^{2+}}}{4/2} \right) / 6 = 0.5 \mu_B / \text{Fe}$$

In-plane interactions
top layer → bottom layer



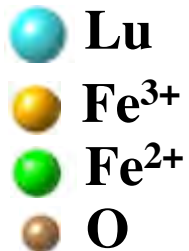
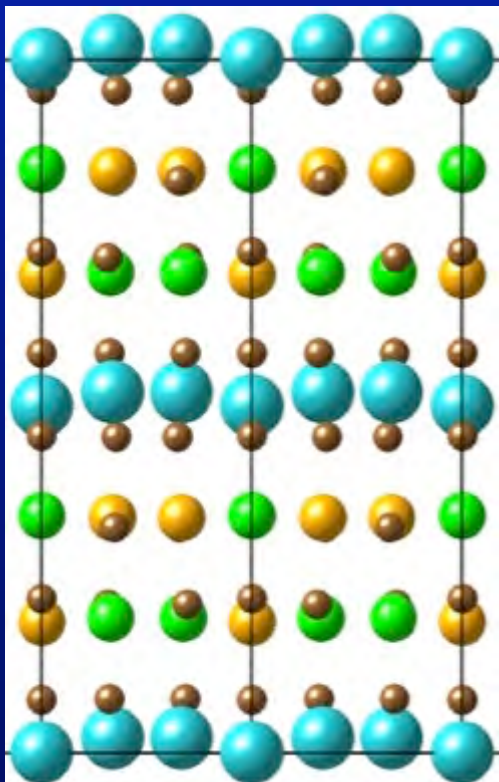
First-Principles DFT: LuFe_2O_4

Spin Configurations



First-Principles DFT: LuFe_2O_4

COII



Energy:

$\sim 4 \text{ meV/Fe}$

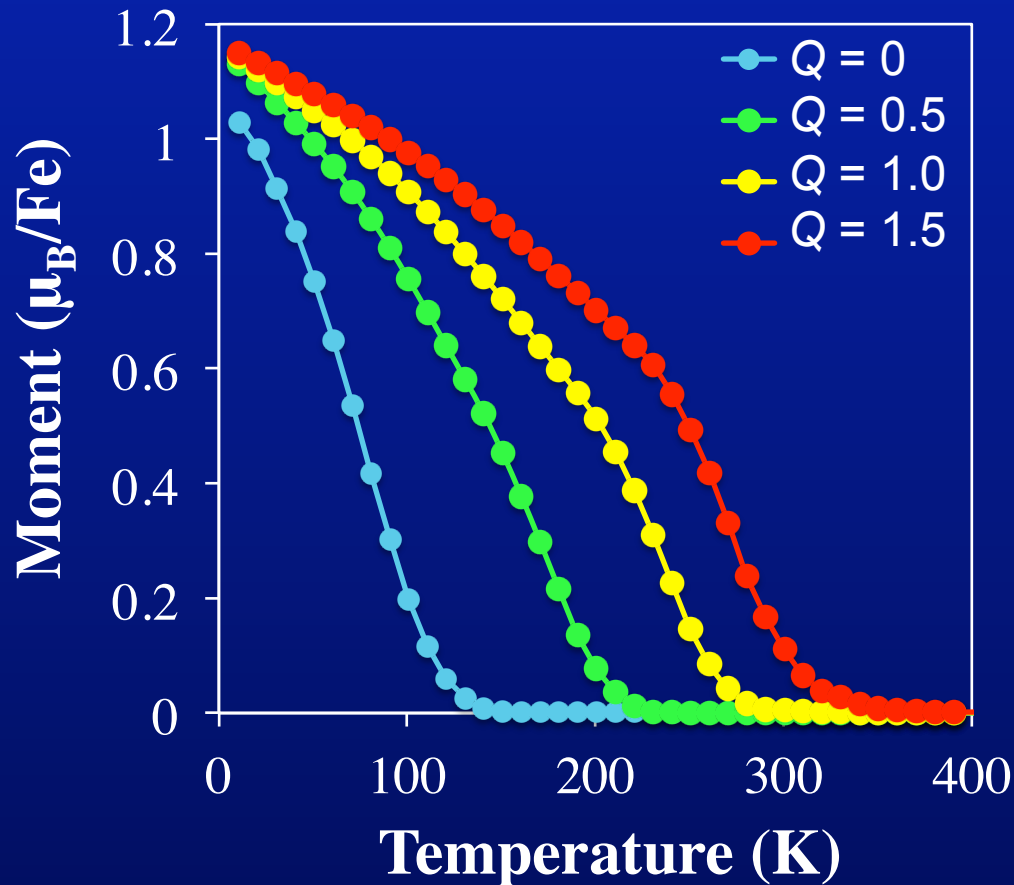
Group:

Cm

Moment:

$\sim 1.16 \mu_B/\text{Fe}$

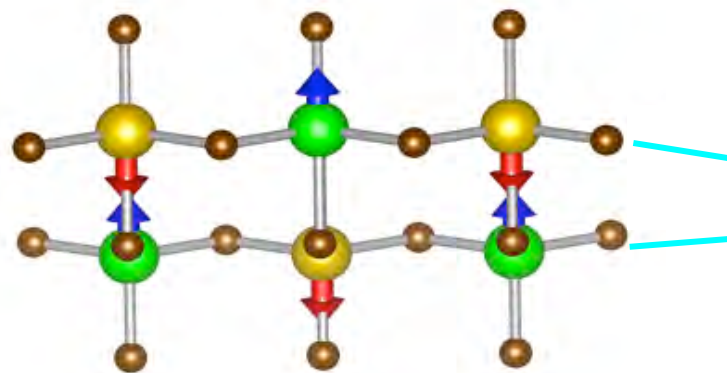
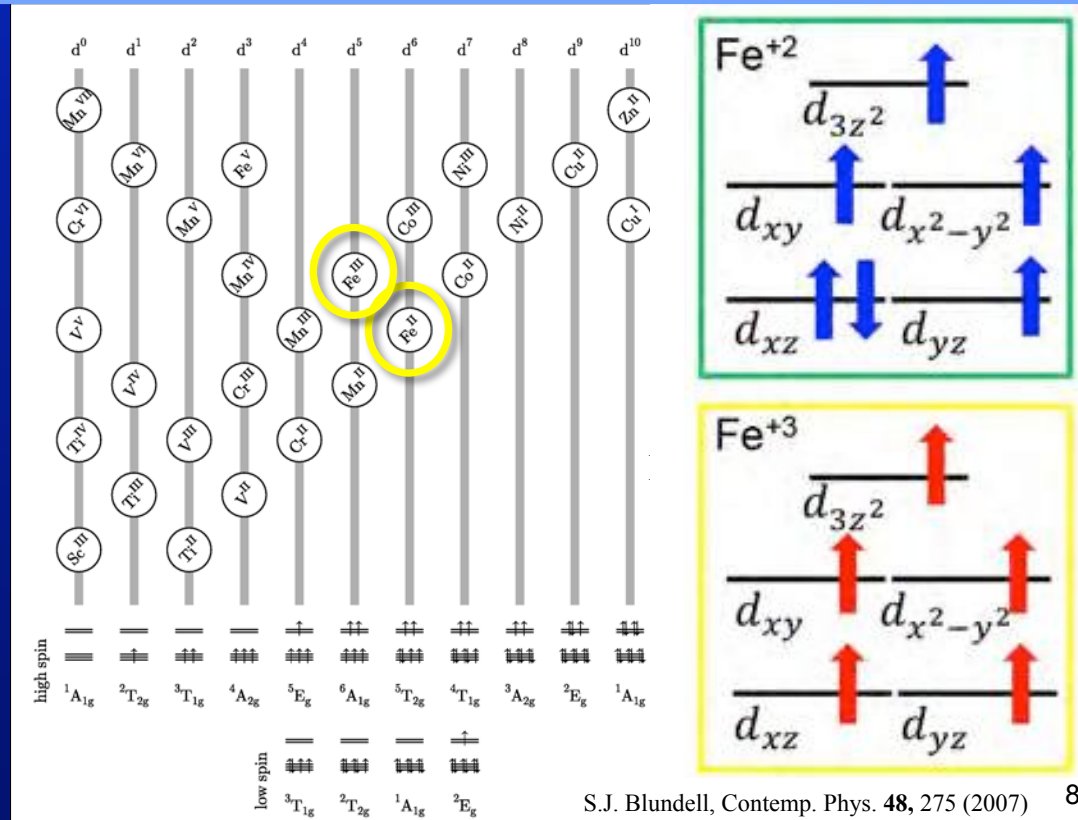
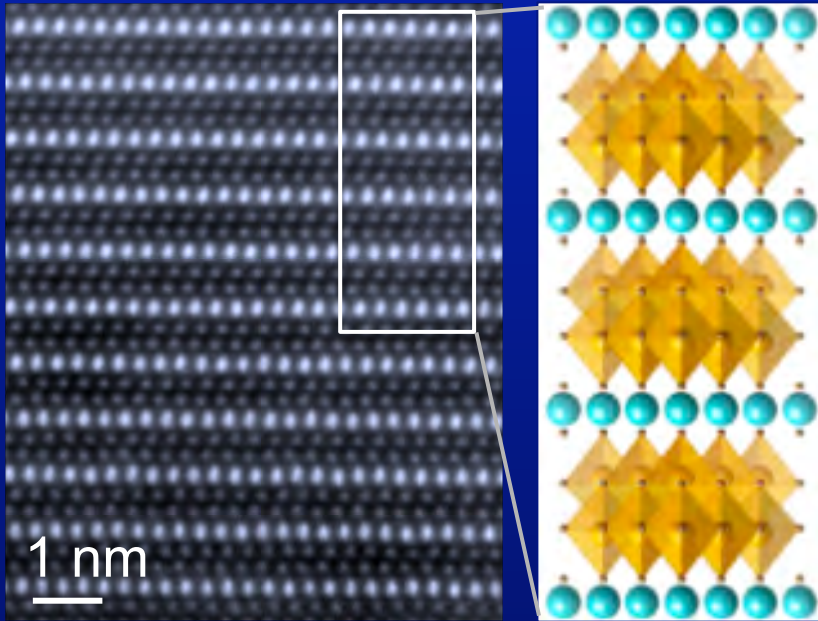
Ferroelectric



Lu Distortions (“ Q ”) increase magnetic transition in LuFe_2O_4 to above room-temperature.

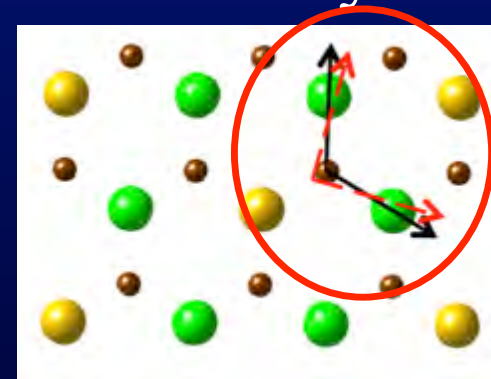
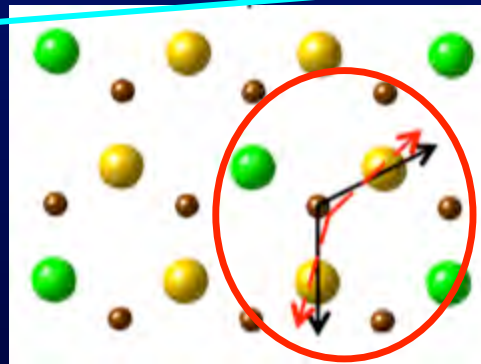
The Idea: $(\text{LuFeO}_3)_m / (\text{LuFe}_2\text{O}_4)_n$ Superlattices

LuFe_2O_4



$$M = g\mu_B \left(\frac{3S_{\text{Fe}^{3+}}}{5/2} - \frac{3S_{\text{Fe}^{2+}}}{4/2} \right) / 6 = 0.5 \mu_B / \text{Fe}$$

In-plane interactions
 top layer → bottom layer



Lu₂Fe₃O₇: A Natural (LuFeO₃)₁ / (LuFe₂O₄)₁ Superlattice

Lu₂Fe₃O₇
T_{C,FM} ~ 270 K

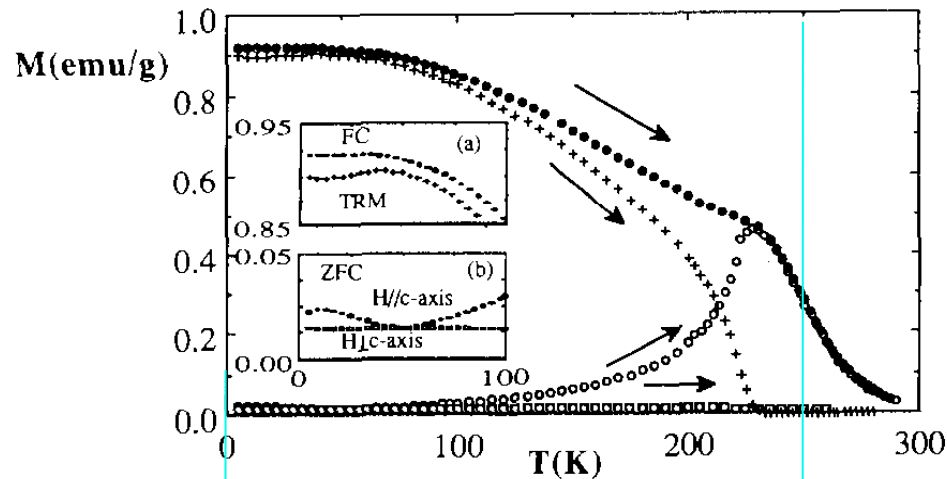
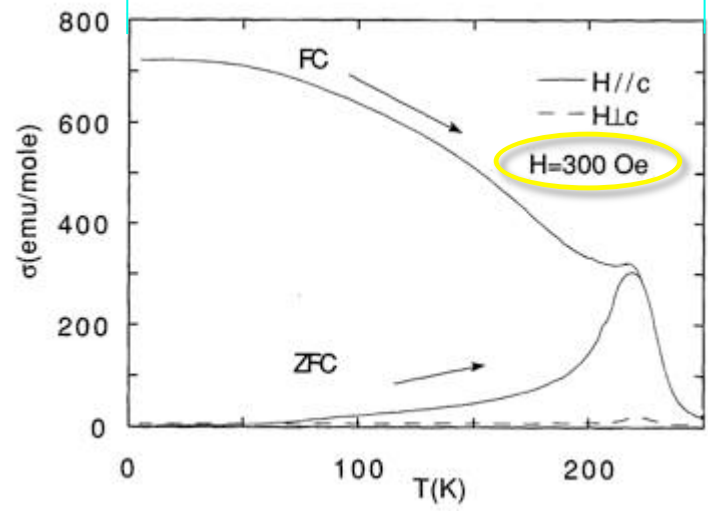


Fig. 1. Thermomagnetization curves of single crystal Lu₂Fe₃O₇ in 300 Oe. Circles and squares show those of the

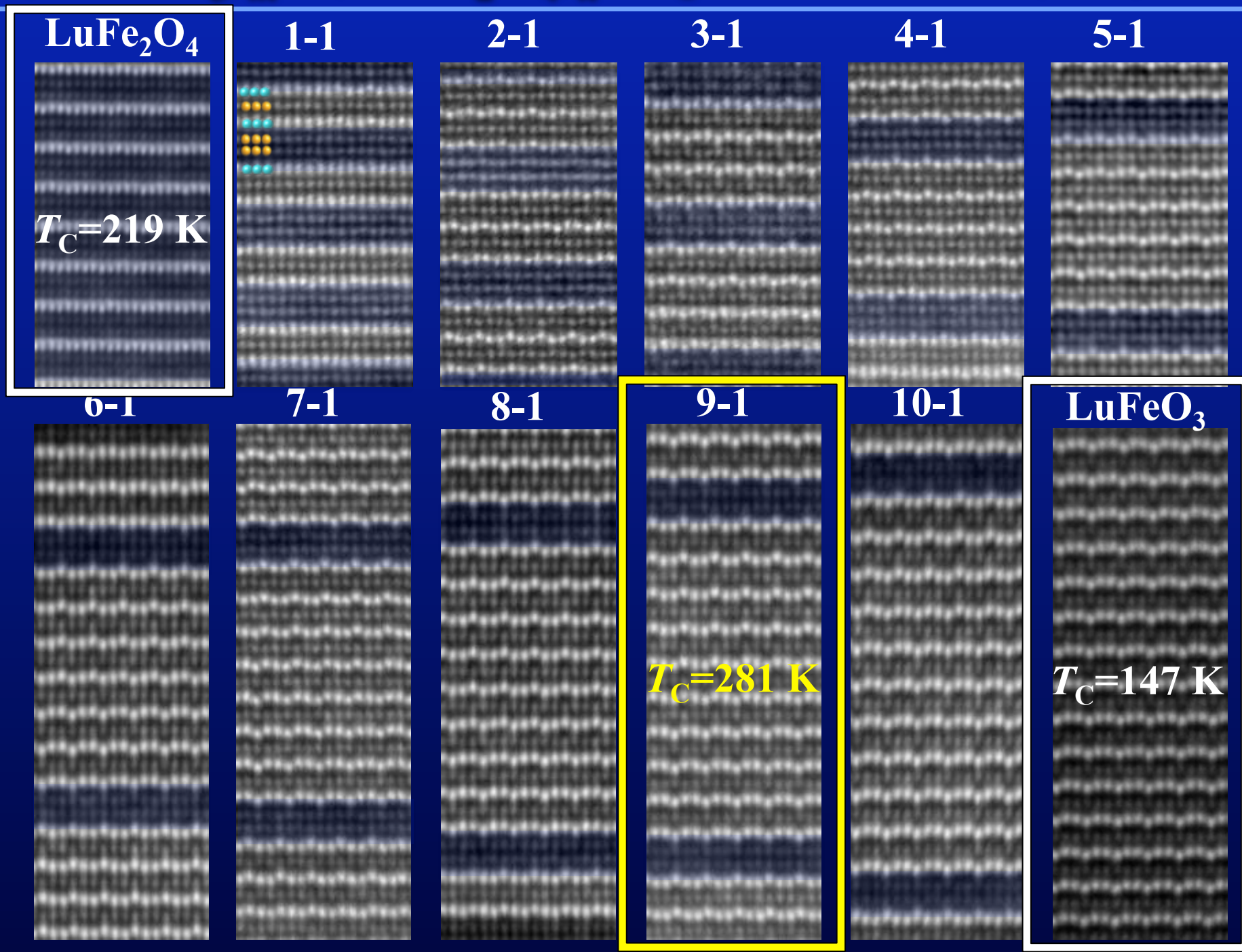
LuFe₂O₄
T_{C,FM} ~ 240 K



J. Iida, M. Tanaka, Y. Nakagawa,
S. Funahashi, N. Kimizuka,
and S. Takekawa,
Journal of the Physical Society of Japan **62**
(1993) 1723-1735.

J. Iida, M. Tanaka, and S.
Funahashi,
“Magnetic Property of Single
Crystal Lu₂Fe₃O₇,”
J. Magn. Magn. Mater. **104-107**
(1992) 827-828.

$(\text{LuFeO}_3)_m / (\text{LuFe}_2\text{O}_4)_n$ Superlattices on (111) YSZ



Ferroelectric Puckering *enhances* Ferrimagnetism

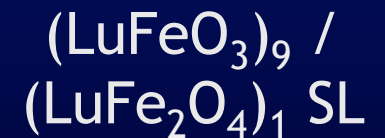
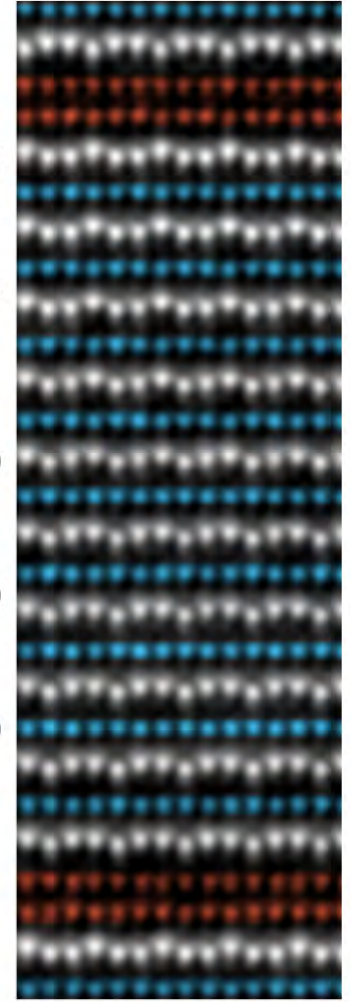
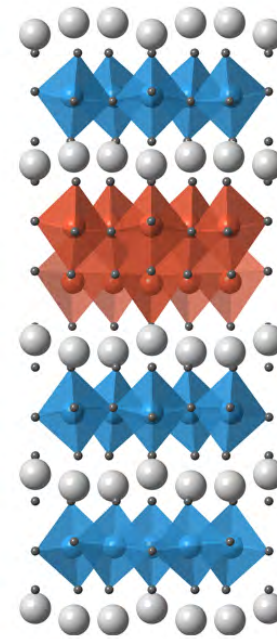
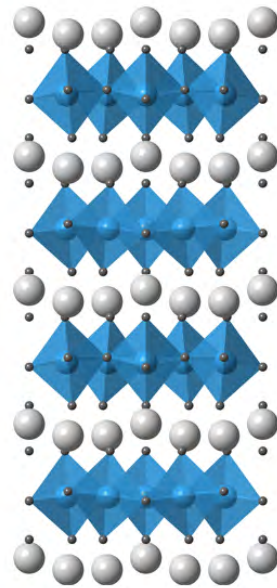
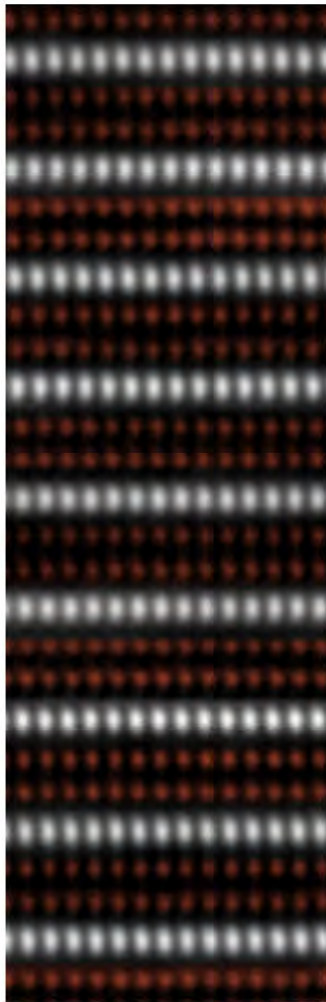
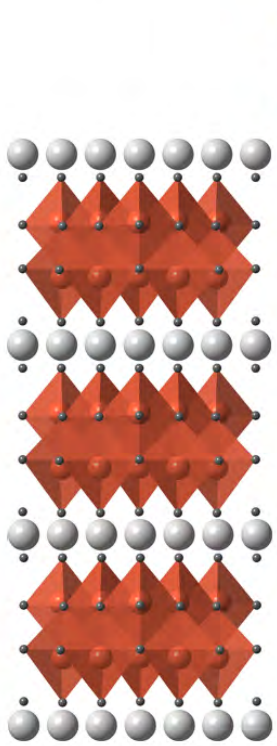
Ferrimagnetic

+

Ferroelectric

=

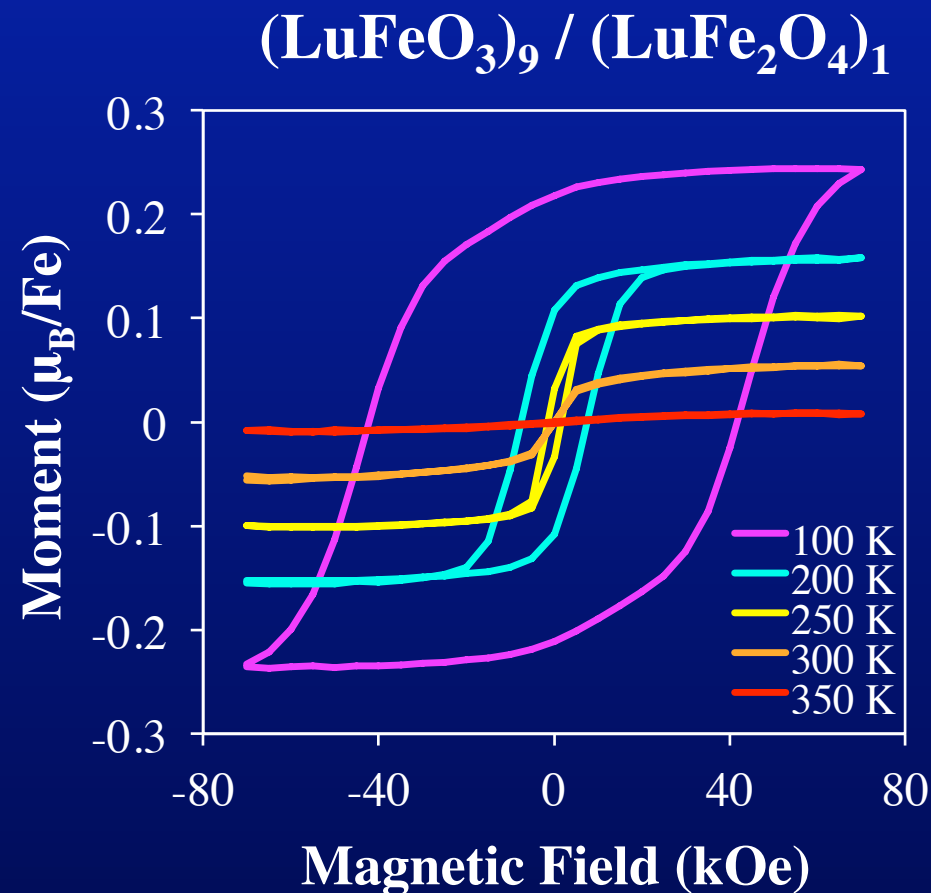
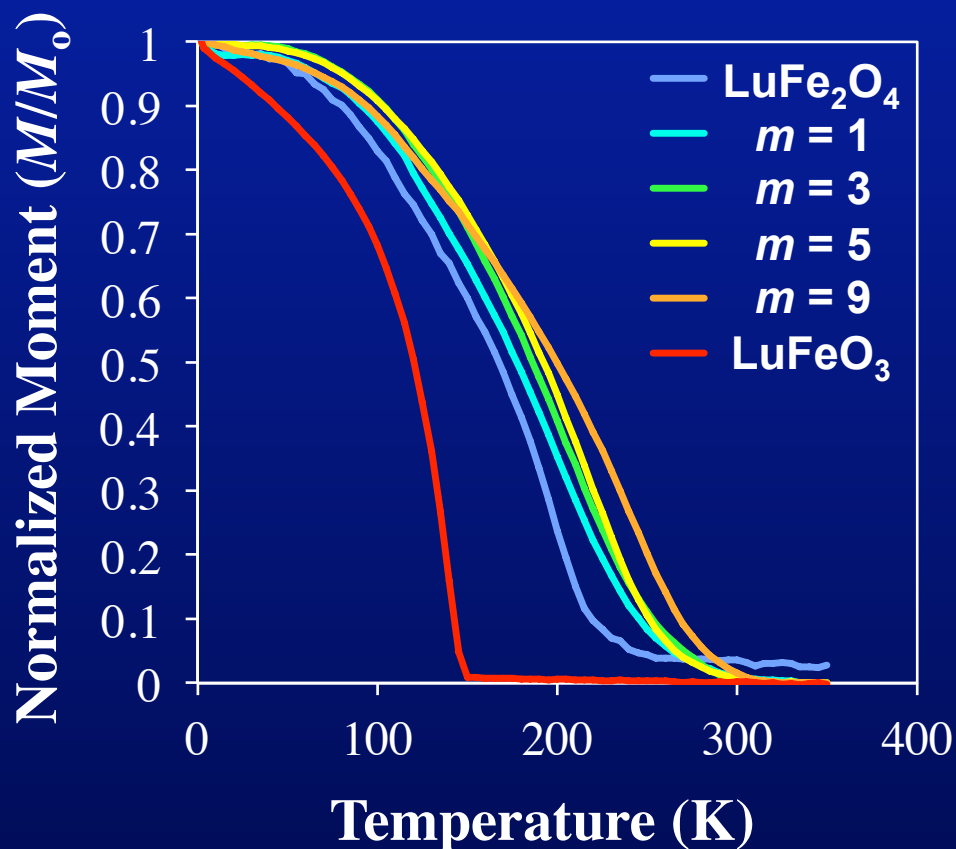
Multiferroic



Quantifying Magnetic Moment:



SQUID Magnetometry

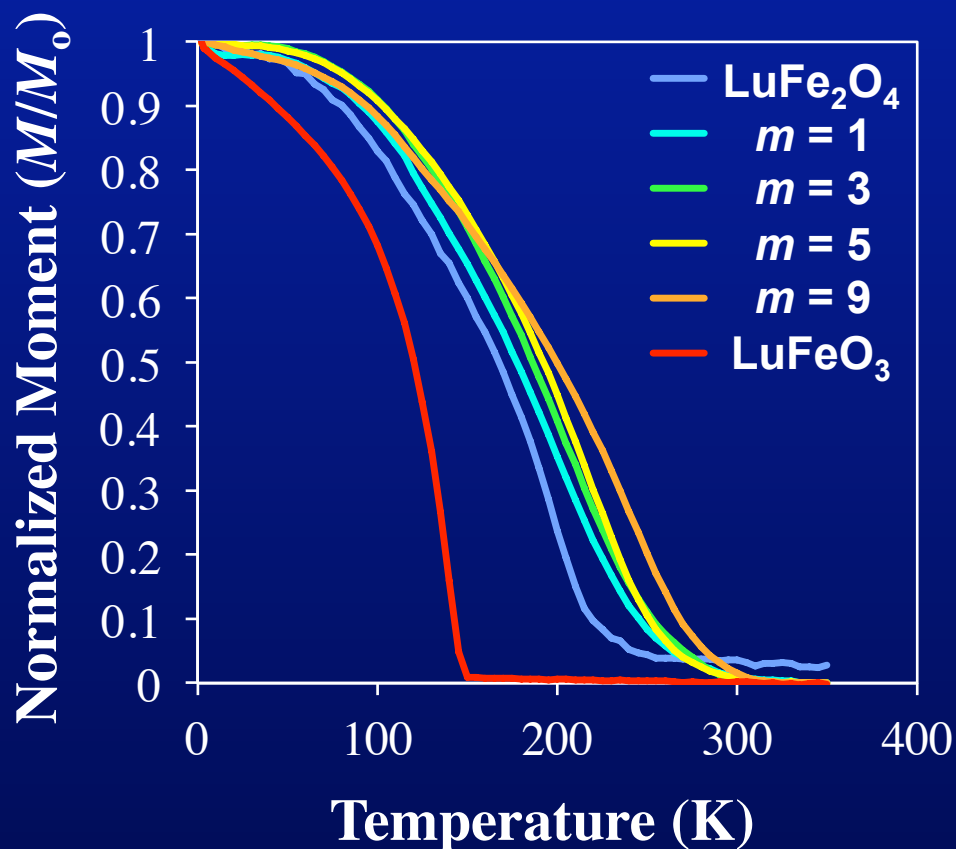


All $(\text{LuFeO}_3)_m / (\text{LuFe}_2\text{O}_4)_1$ superlattices have a higher T_C than both LuFeO₃ and LuFe₂O₄

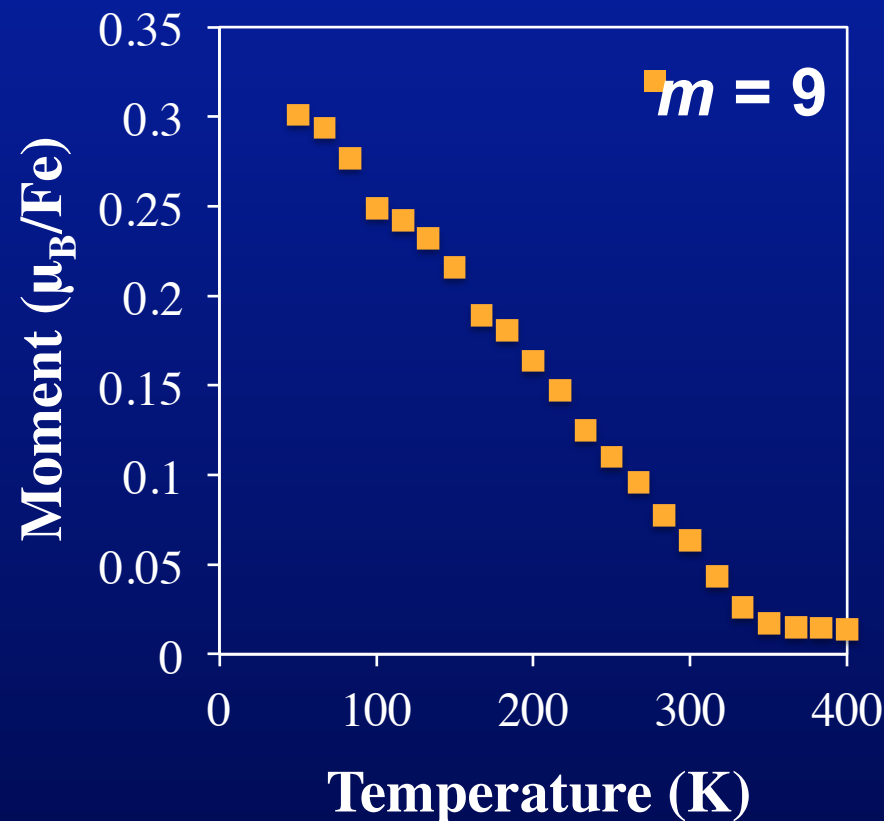
Quantifying Magnetic Moment:



SQUID Magnetometry

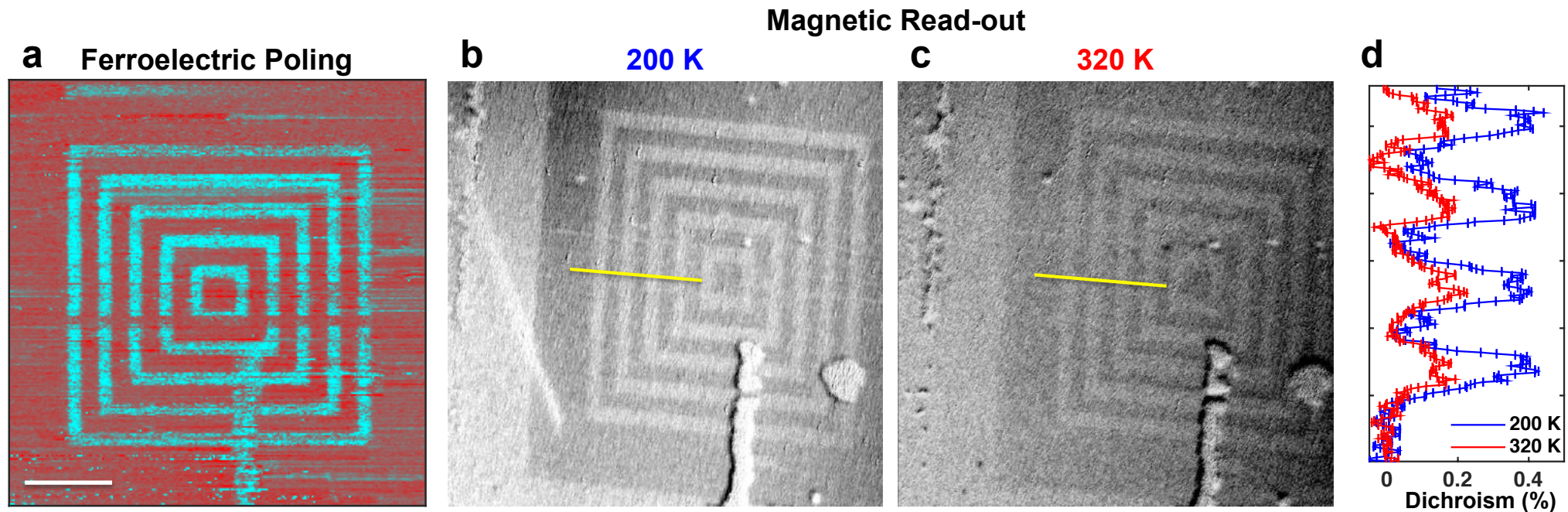


(LuFeO₃)₉ / (LuFe₂O₄)₁



Ferromagnetic fluctuations in (LuFeO₃)₉ / (LuFe₂O₄)₁ superlattice persist above room-temperature

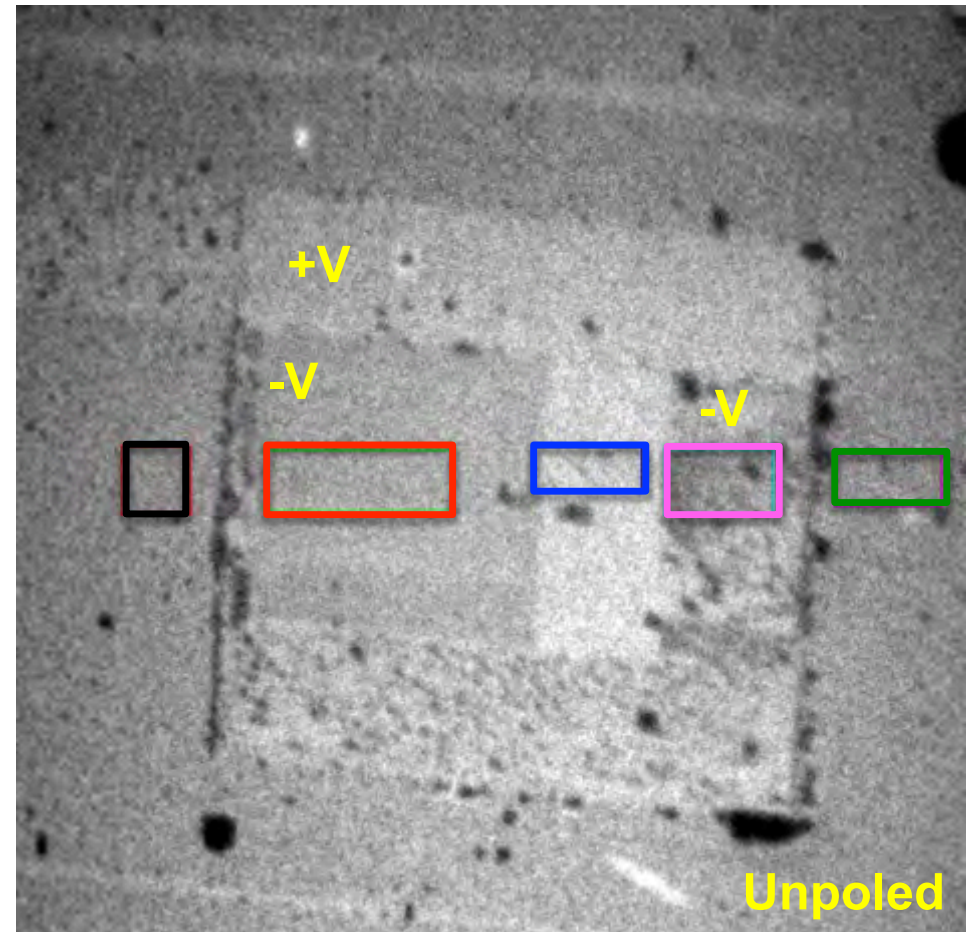
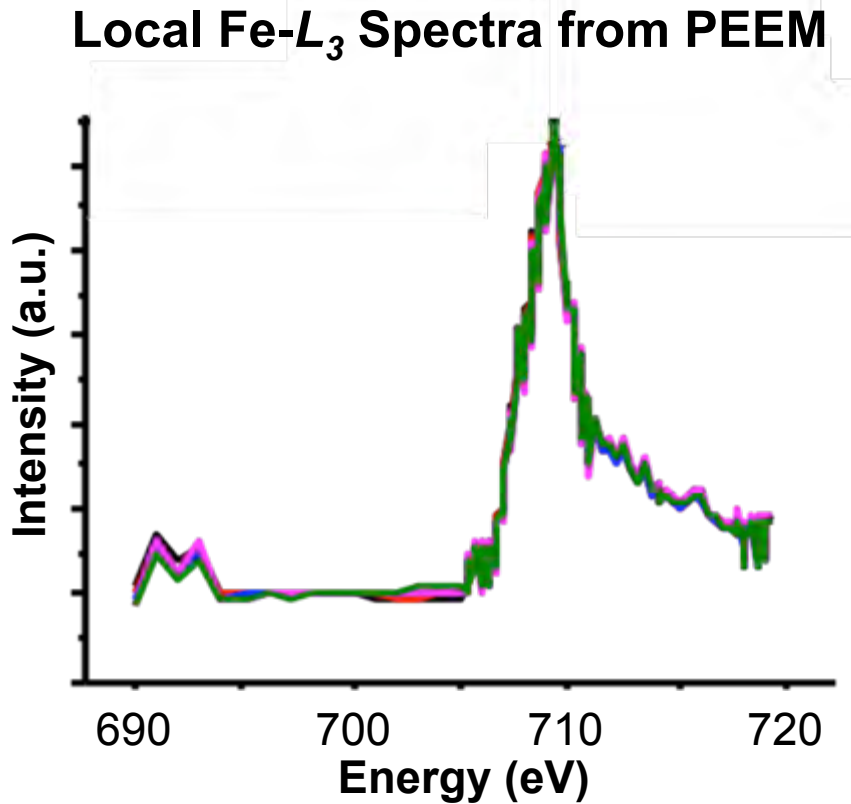
Polarization coupled to Magnetism!



$(\text{LuFeO}_3)_9 / (\text{LuFe}_2\text{O}_4)_1$ superlattice

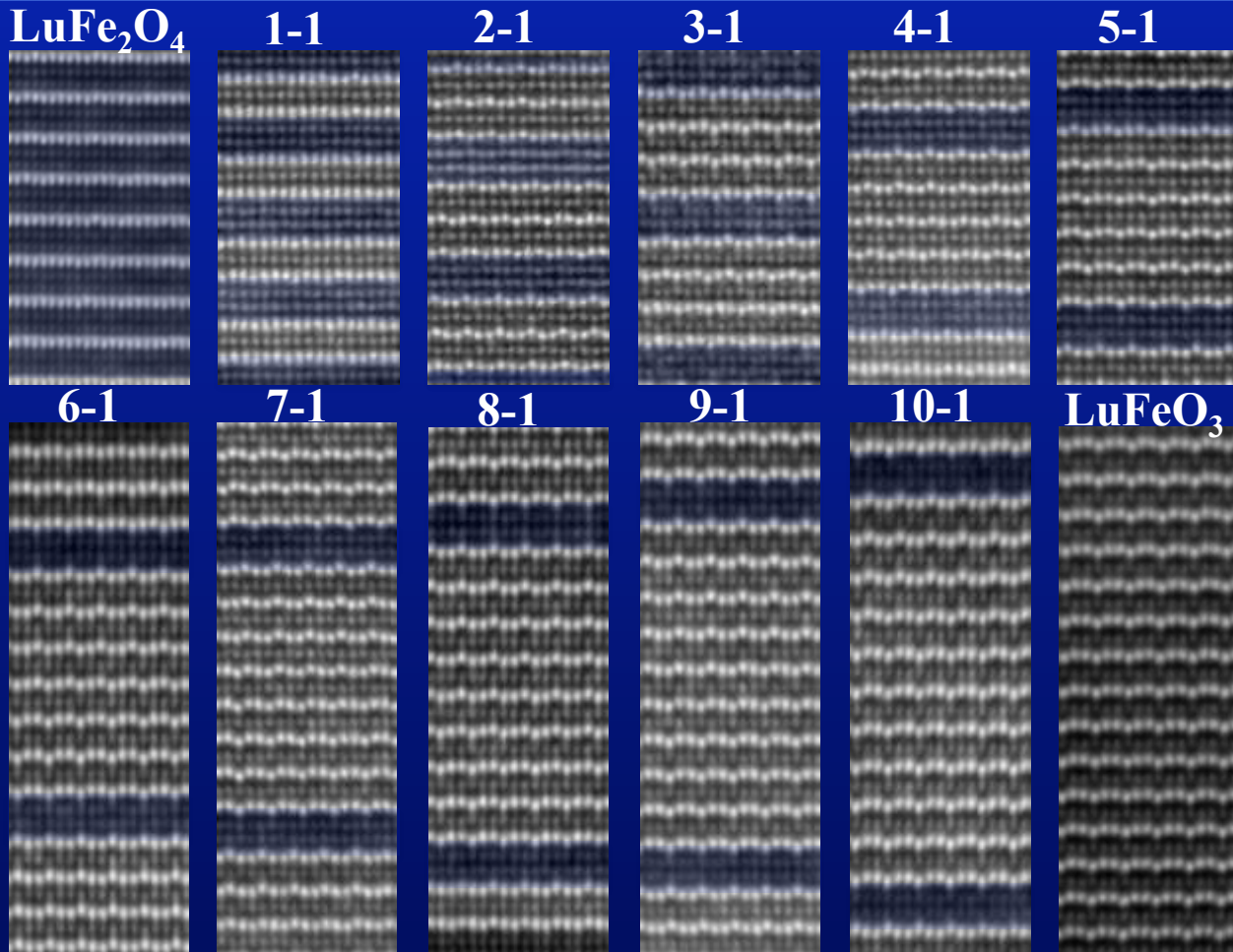
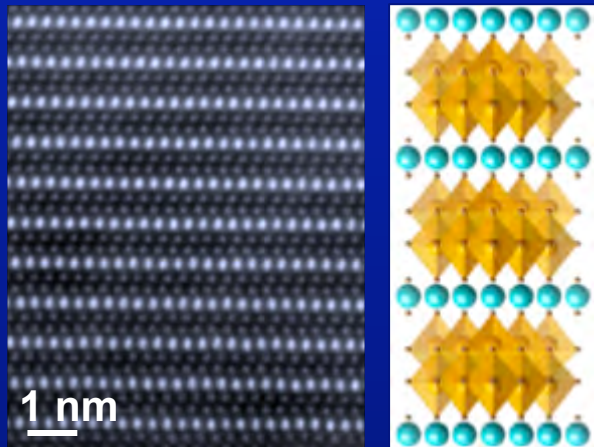
J.A. Mundy, C.M. Brooks, M.E. Holtz, J.A. Moyer, H. Das, A.F. Rébola, J.T. Heron, J.D. Clarkson, S.M. Disseler, Z. Liu, A. Farhan, R. Held, R. Hovden, E. Padgett, Q. Mao, H. Paik, R. Misra, L.F. Kourkoutis, E. Arenholz, A. Scholl, J.A. Borchers, W.D. Ratcliff, R. Ramesh, C.J. Fennie, P. Schiffer, D.A. Muller, and D.G. Schlom, *Nature* **537** (2016) 523–527.

Electrical Poling Does Not Induce Local Modulation in Fe Valence

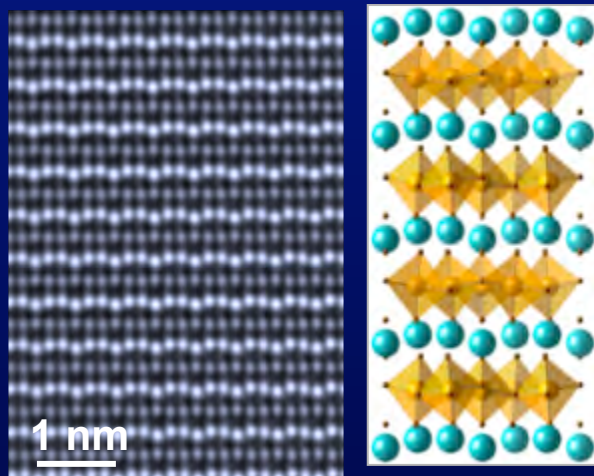


- PEEM is used to locally record Fe valence spectra demonstrating no detectable shift in the spectra (e.g. valence modulation) post-poling.
- XMCD-PEEM from the same region showed distinct magnetic order between the poled regions in this sample.

Highest T_C Ferrimagnetic Ferroelectric



LuFe_2O_4
(a ferrimagnet) +



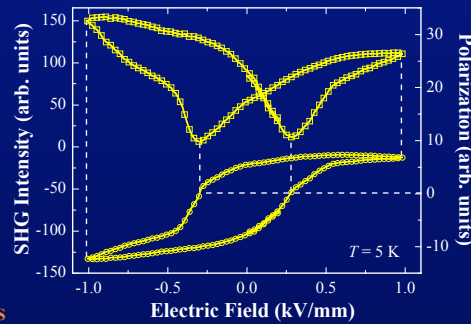
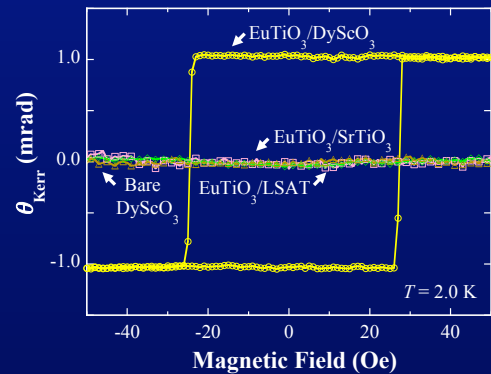
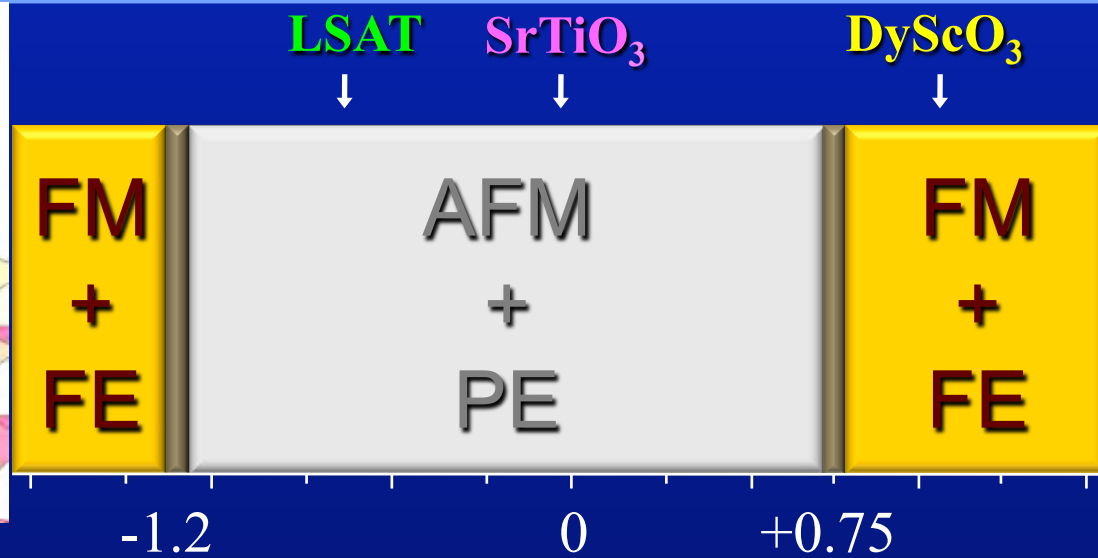
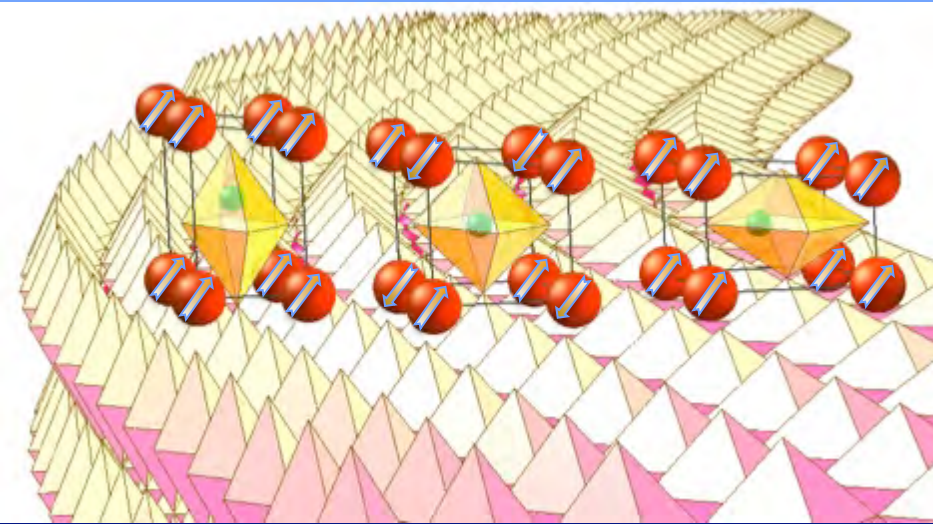
LuFeO_3
(a geometric ferroelectric)

$(\text{LuFeO}_3)_m / (\text{LuFe}_2\text{O}_4)_n$ superlattices

Ferrimagnetic Ferroelectric

(puckering and polarization-doping from ferroelectric *enhance* ferrimagnetism)

Strongest Ferromagnetic Ferroelectric



Biaxial Strain (%), ϵ_s

C.J. Fennie and K.M. Rabe
Physical Review Letters **97** (2006) 267602.

22 nm thick EuTiO₃ (a boring dielectric) **+** **1.1% Strain** (by growing it commensurately) **=** **Multiferroic** (1000× stronger than prior ferromagnetic ferroelectrics)

J.H. Lee, L. Fang, E. Vlahos, X. Ke, Y.W. Jung, L.F. Kourkoutis, J-W. Kim, P.J. Ryan, T. Heeg, M. Roeckerath, V. Goian, M. Bernhagen, R. Uecker, P.C. Hammel, K.M. Rabe, S. Kamba, J. Schubert, J.W. Freeland, D.A. Muller, C.J. Fennie, P. Schiffer, V. Gopalan, E. Johnston-Halperin, and D.G. Schlom, *Nature* **466** (2010) 954-958.

Conclusions

Using Thin-Film Alchemy:

- Hidden Ground States can be Accessed
- Superior Properties shown to Exist
- Theory + Synthesis is a Powerful Combination (Materials by Design)
- Imagine what Properties Await!