

NANOMEDICAMENTS: MATERIAUX ET METHODES DE PREPARATION

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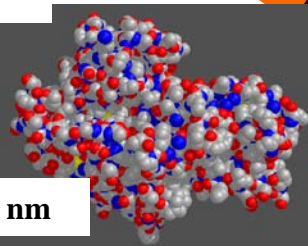
2009-2010

L'ECHELLE DES TAILLES



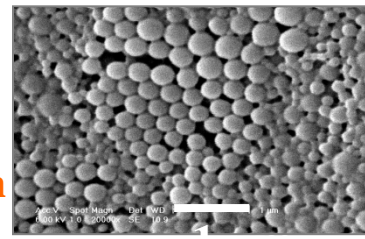
1 nm

Molecule < 10 nm

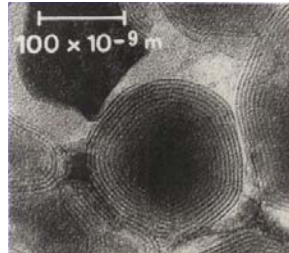


Protein 10 - 20 nm

10 nm

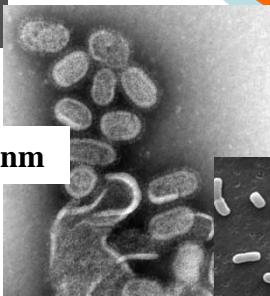


nanoparticules



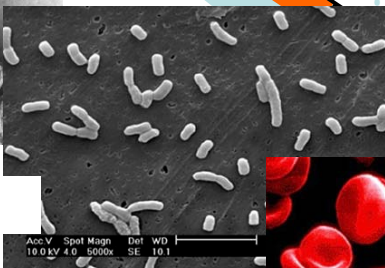
Liposomes

100 nm

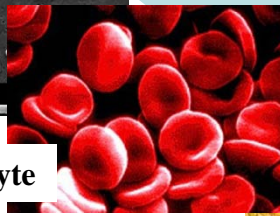


Virus 20 - 400 nm

1 µm



Bacteria 500 nm - 10 µm



Erythrocyte

10 µm



Pollen 10 - 100 µm

100 µm

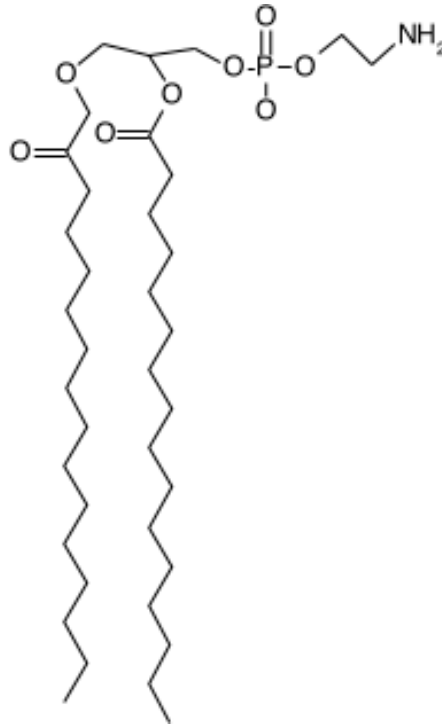
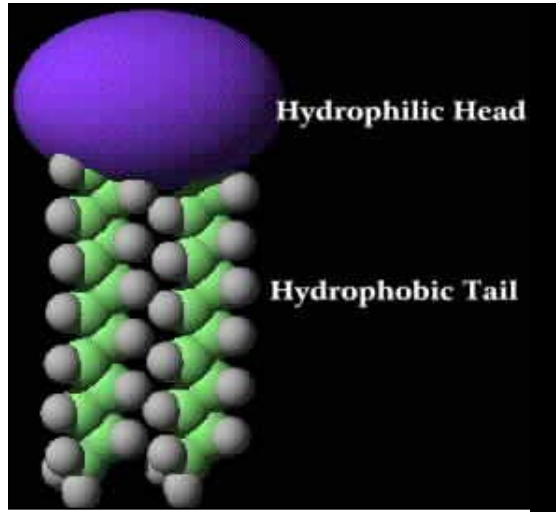


Sand 100 µm - 5 mm

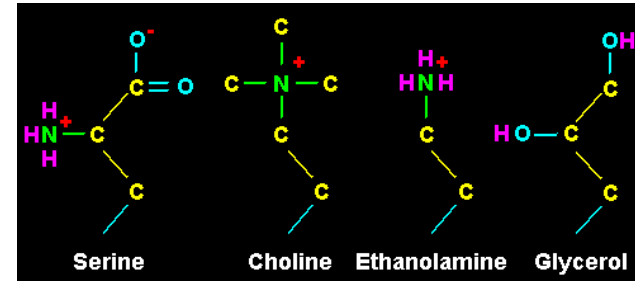
LIPOSOMES

SYSTEMES MOLECULAIRES
ORGANISES

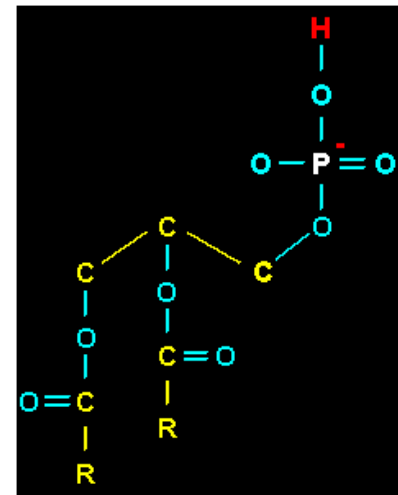
Phospholipids



Polar Head Groups



Three carbon glycerol



STRUCTURE DES PHOSPHOLIPIDES

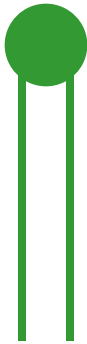
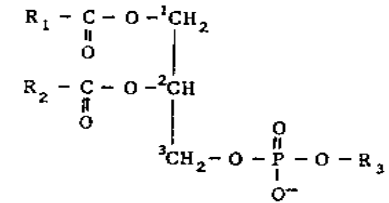
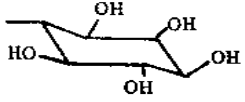


Tableau 1
STRUCTURE CHIMIQUE DES PRINCIPAUX PHOSPHOLIPIDES
(EXTRAIT DE [1])

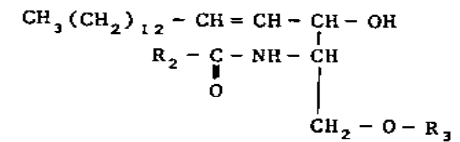
GLYCEROPHOSPHOLIPIDES



R_1, R_2 chaînes aliphatiques

R_3 :	$-\text{CH}_2 - \text{CH}_2 - \overset{+}{\text{N}}(\text{CH}_3)_3$	phosphatidylcholines	PC
	$-\text{CH}_2 - \text{CH}_2 - \overset{+}{\text{NH}}_3$	phosphatidyléthanolamine	PE
	$-\text{CH}_2 - \underset{\text{COO}^-}{\text{CH}} - \overset{+}{\text{NH}}_3$	phosphatidylsérine	PS
	$-\text{CH}_2 - \underset{\text{OH}}{\text{CH}} - \text{CH}_2\text{OH}$	phosphatidylglycérol	PG
	$-\text{H}$	acide phosphatidique	PA
		phosphatidylinositol	PI

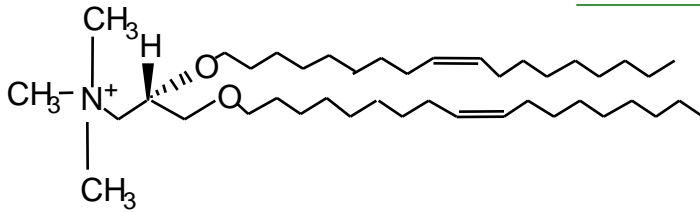
SPHINGOLIPIDES



R_2 , chaîne aliphatique

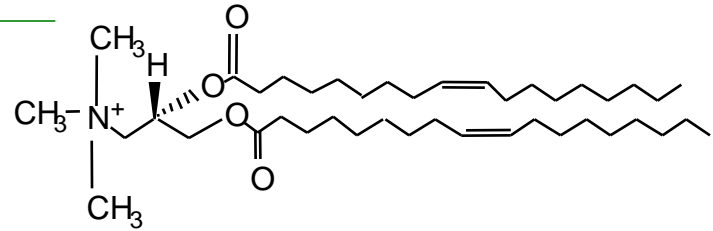
R_3 :	$\bar{\text{P}}\text{O}_3 - \text{CH}_2\text{CH}_2 - \overset{+}{\text{N}}(\text{CH}_3)_3$	sphingomyéline	SM
	$\beta\text{Glc} - \beta\text{Gal} - \beta\text{Gal} - \text{Nac} \dots$ NeuNAc	ganglioside	GM

CATIONIC LIPIDS



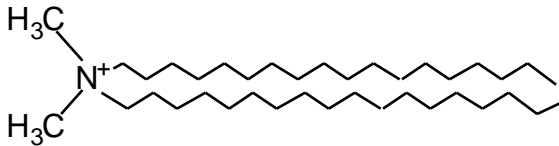
DOTMA

N-(1-(2-3-Dioleoyloxy)propyl)-N,N,N-trimethyl ammonium chloride



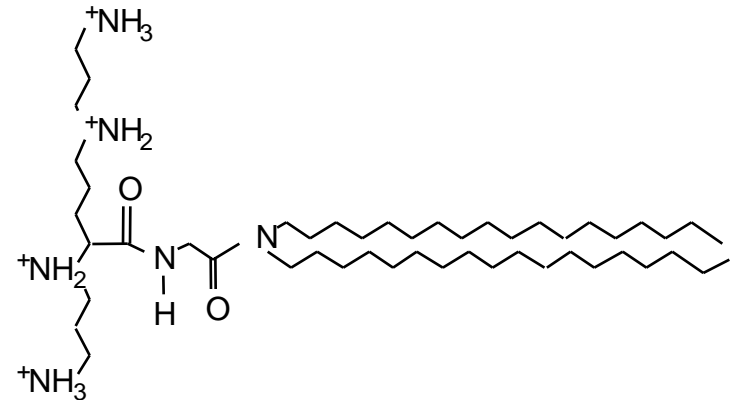
DOTAP

1,2-Dioleoyl-3-trimethylammonium-propane



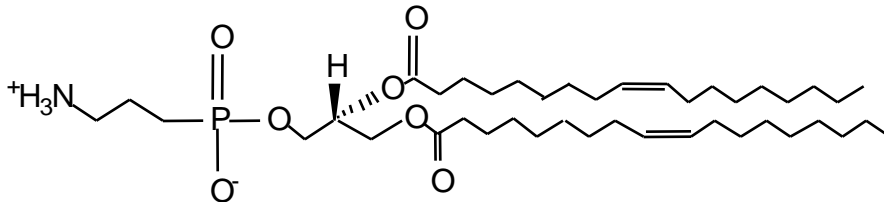
DDAB

Dimethyldioctadecylammonium bromide



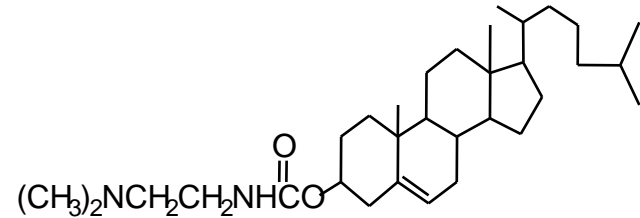
DOGS

Dioctadecylamidoglycyl spermine



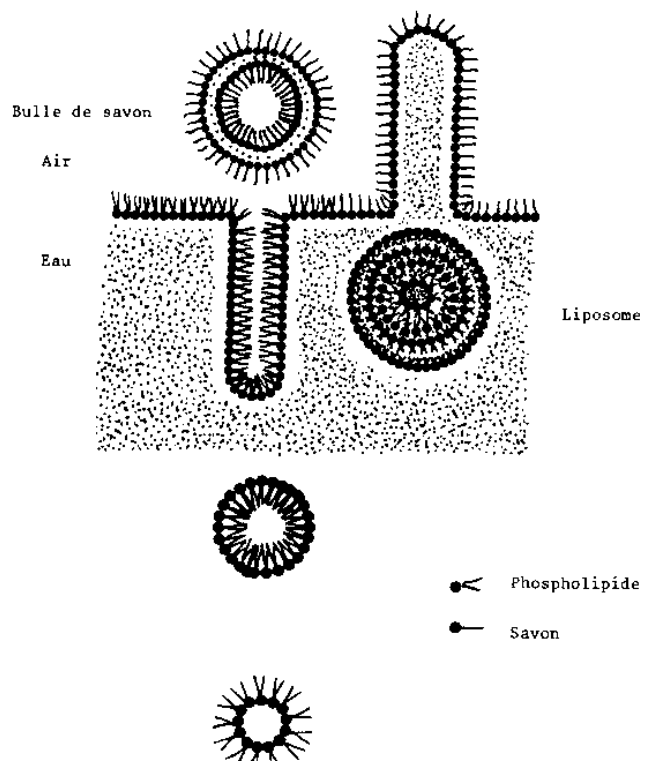
DOPE

1,2-Dioleoyl-sn-Glycero-3-Phosphoethanolamine



DC-Cholesterol

ORGANISATION SUPRAMOLECULAIRE DES PHOSPHOLIPIDES



Critical Packing parameter ($v/a_p l_c$)	Critical packing shape	Structure Formed
$< 1/3$	Cone	Spherical micelle
$1/3 - 1/2$	Truncated cone	Cylindrical micelle
$1/2 - 1$	Truncated cone	Flexible bilayers
~ 1	Cylinder	Planar bilayers
> 1	Inverted truncated cone	Inverted micelles

TEMPERATURE DE TRANSITION DE PHASE DES PL

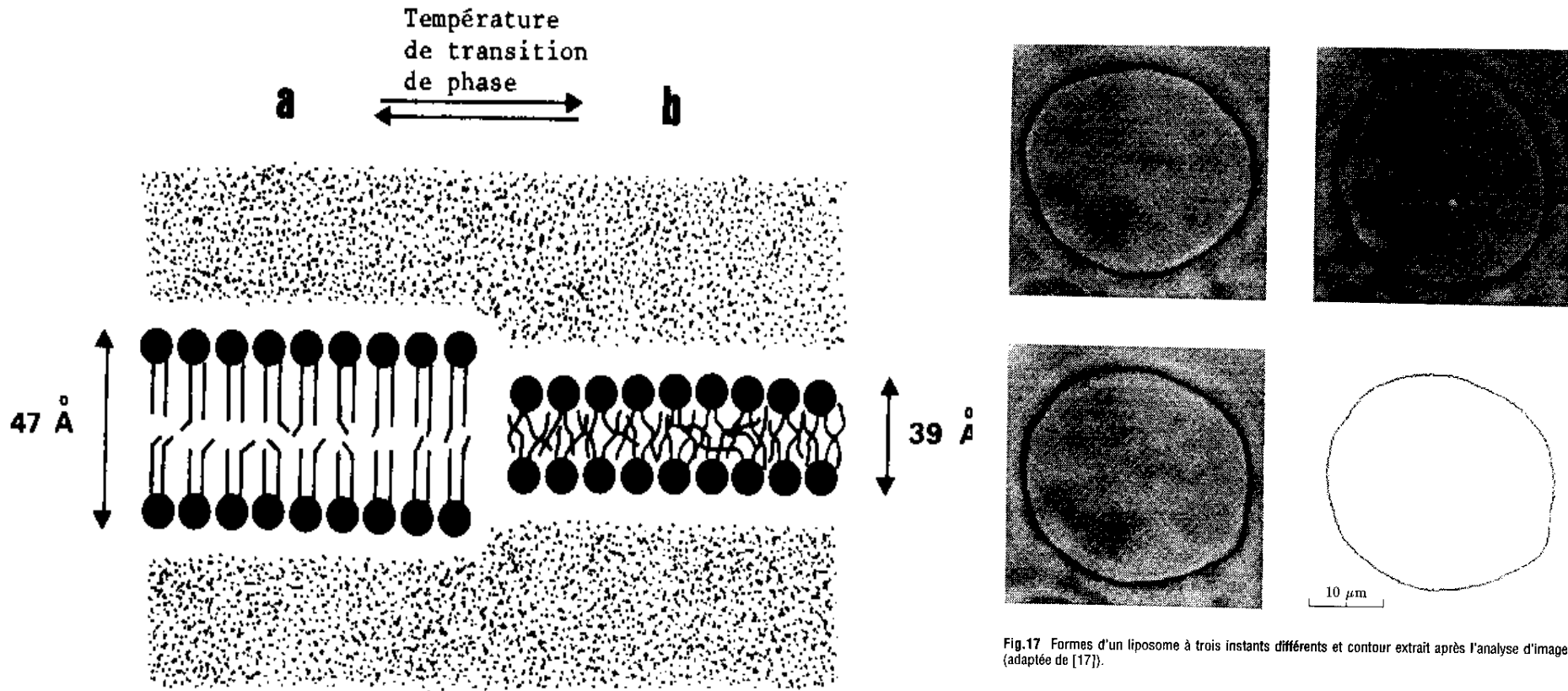
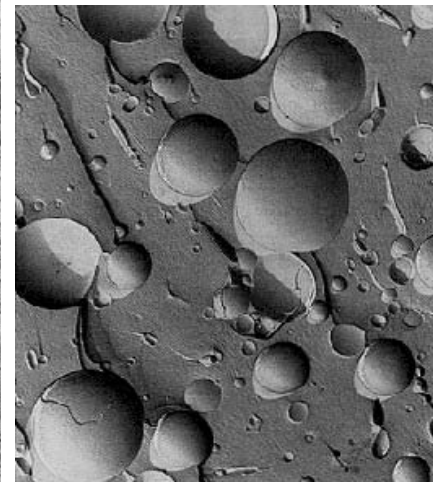
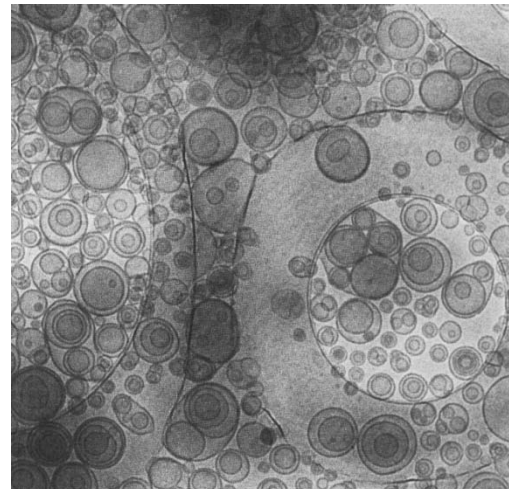
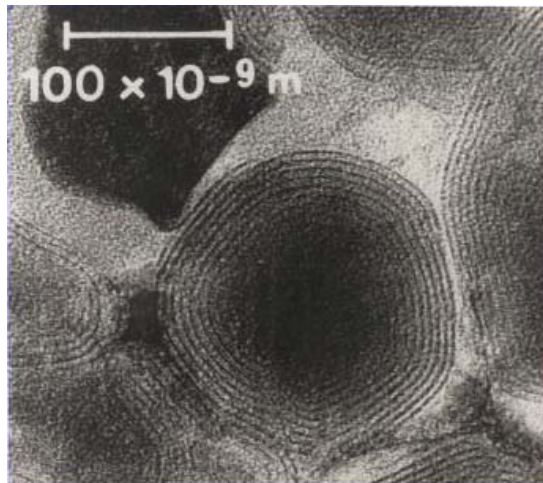
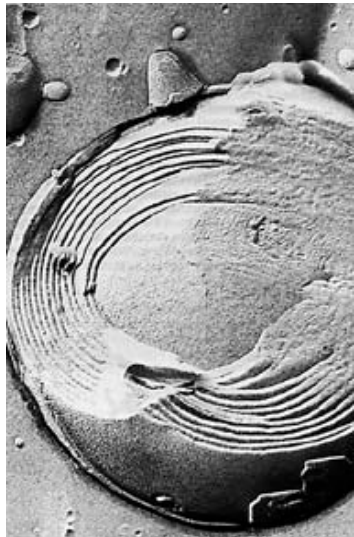
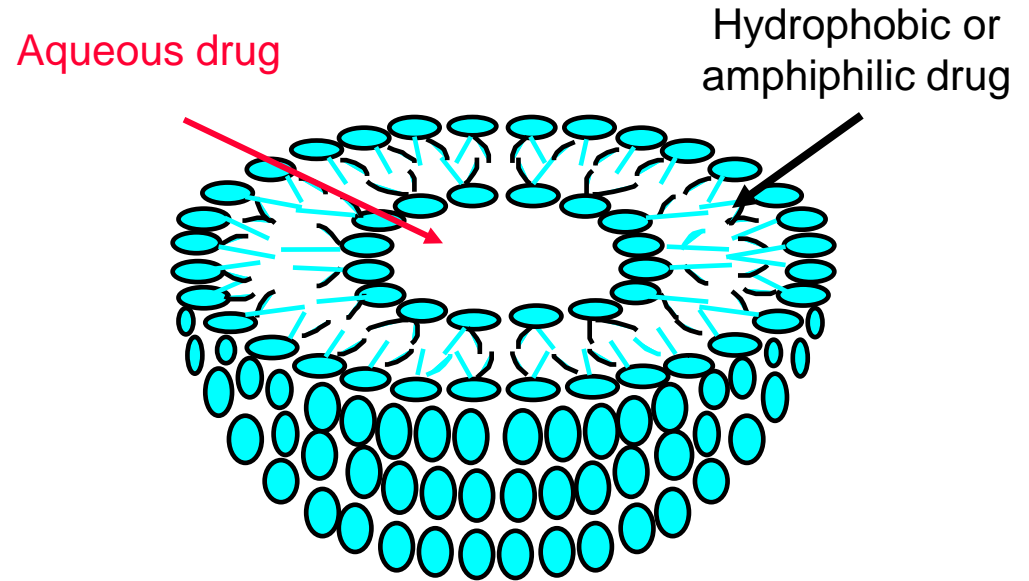


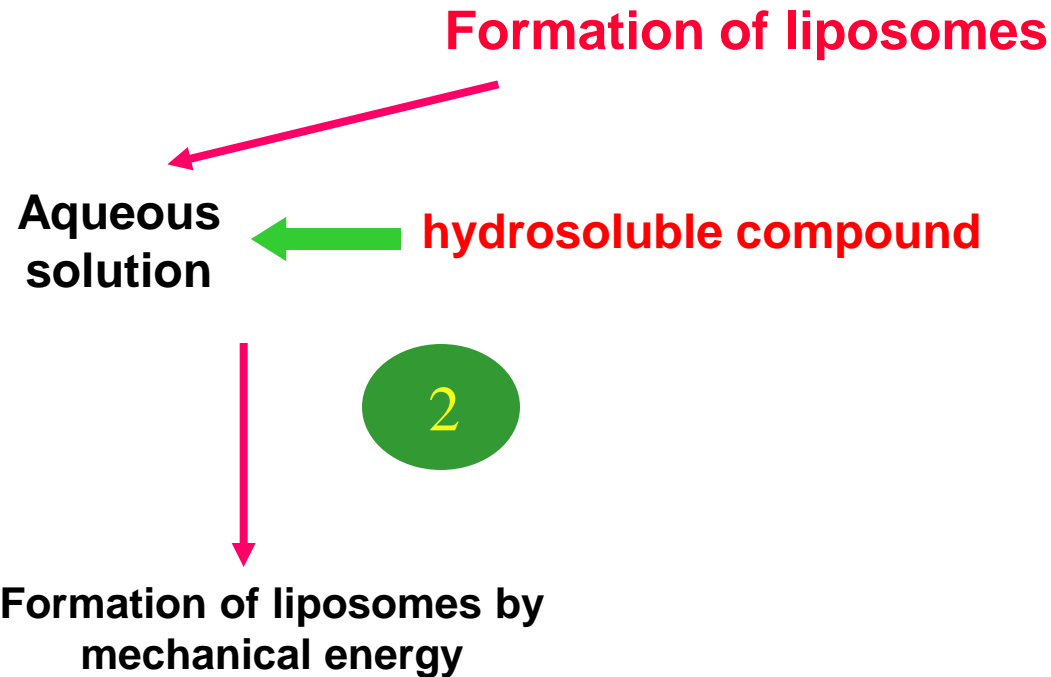
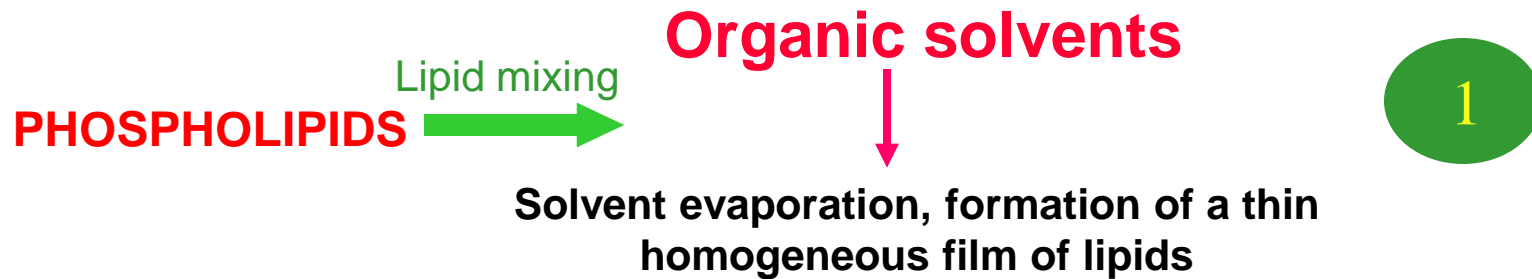
Fig.17 Formes d'un liposome à trois instants différents et contour extrait après l'analyse d'image (adaptée de [17]).

WHAT IS A LIPOSOME ?

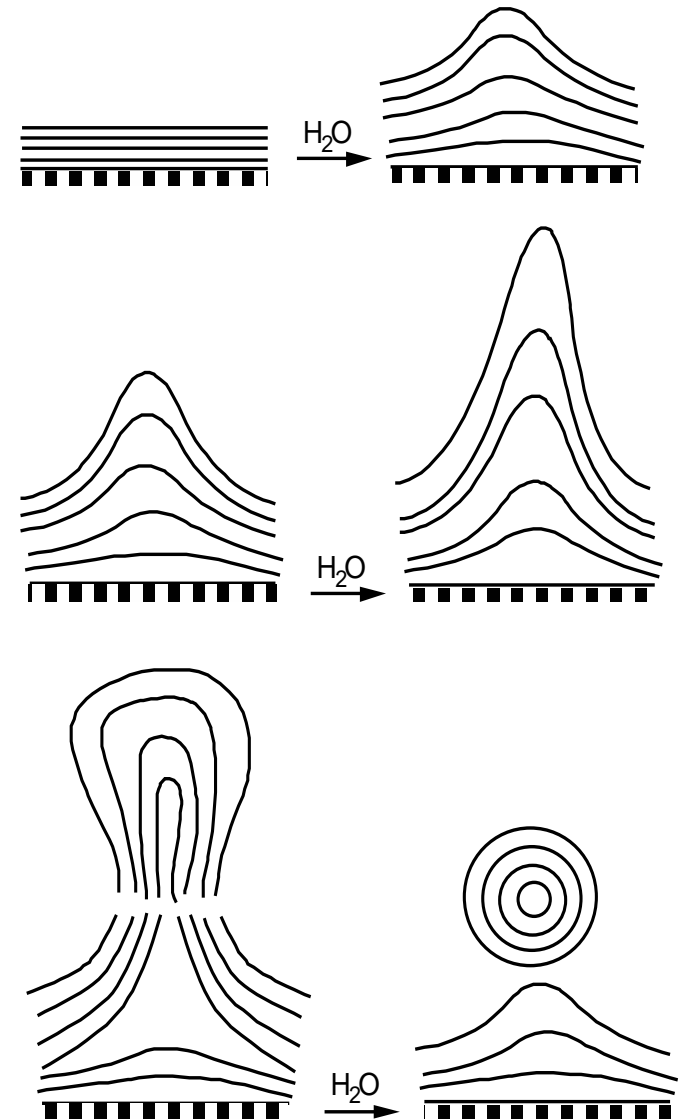
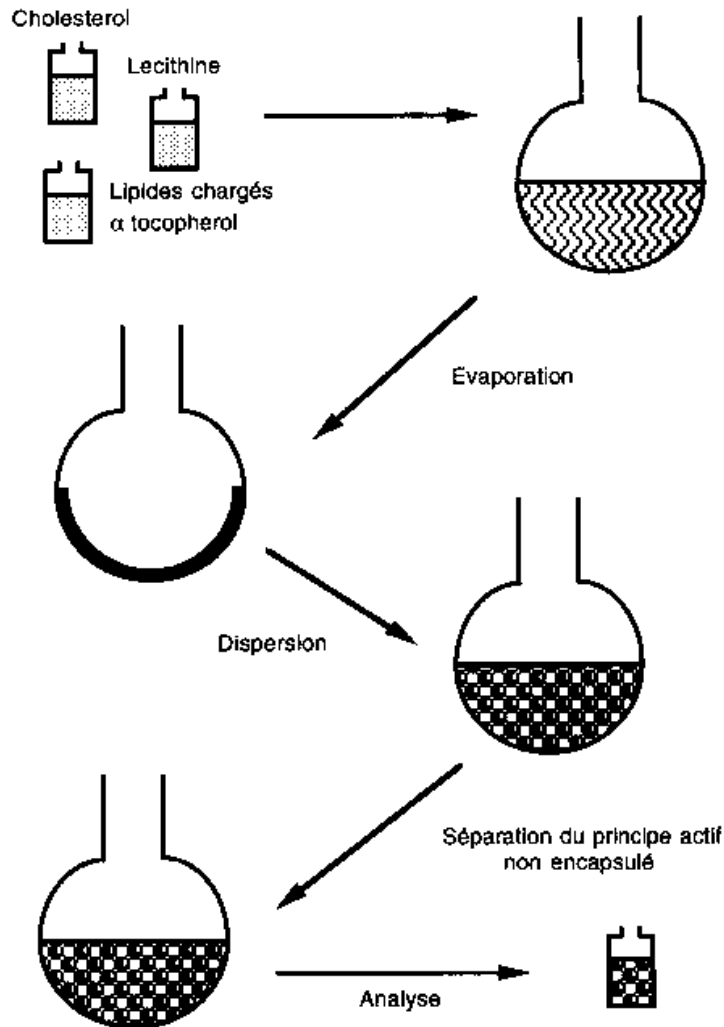
- Spherical
- Diameter less than 1, 000 nm
- A vesicle
- An aqueous phase
- Made of amphiphilic molecules
- One or more bilayer membrane



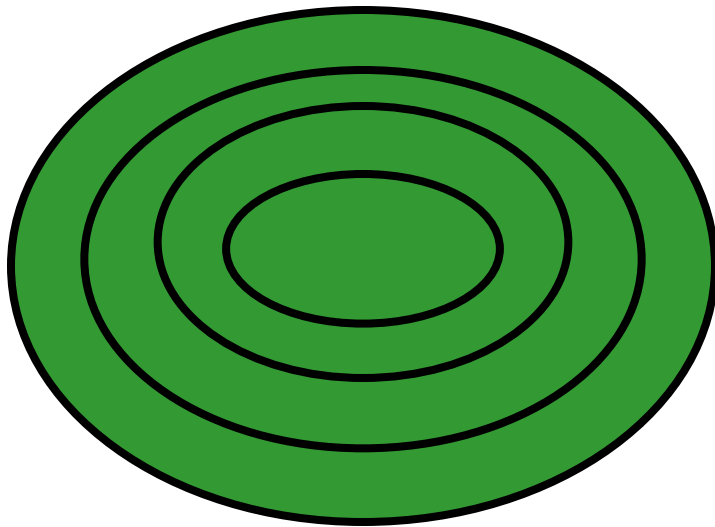
PRINCIPLE OF LIPOSOME FORMATION



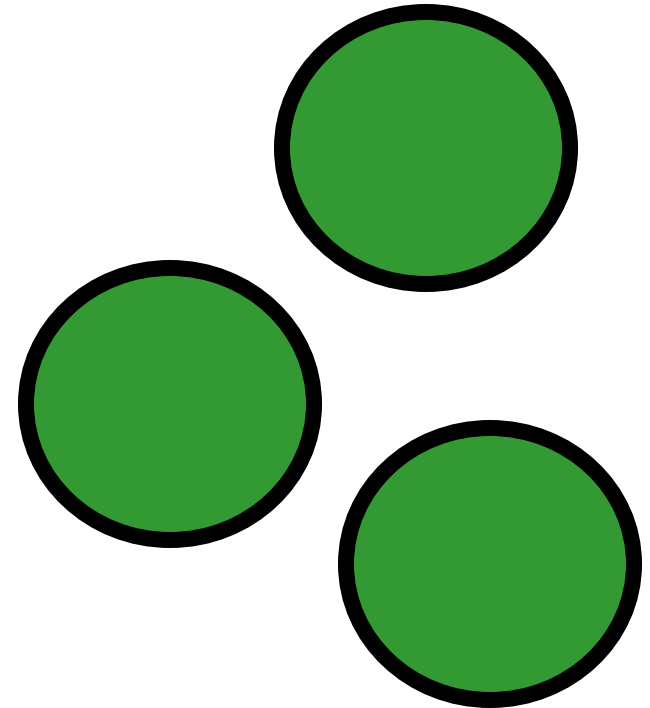
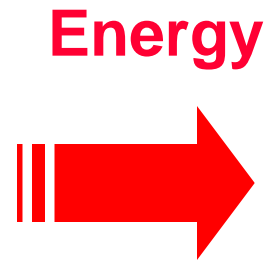
PREPARATION DES LIPOSOME PAR LA METHODE DE BANGHAM



LIPOSOMES HOMOGENEIZATION



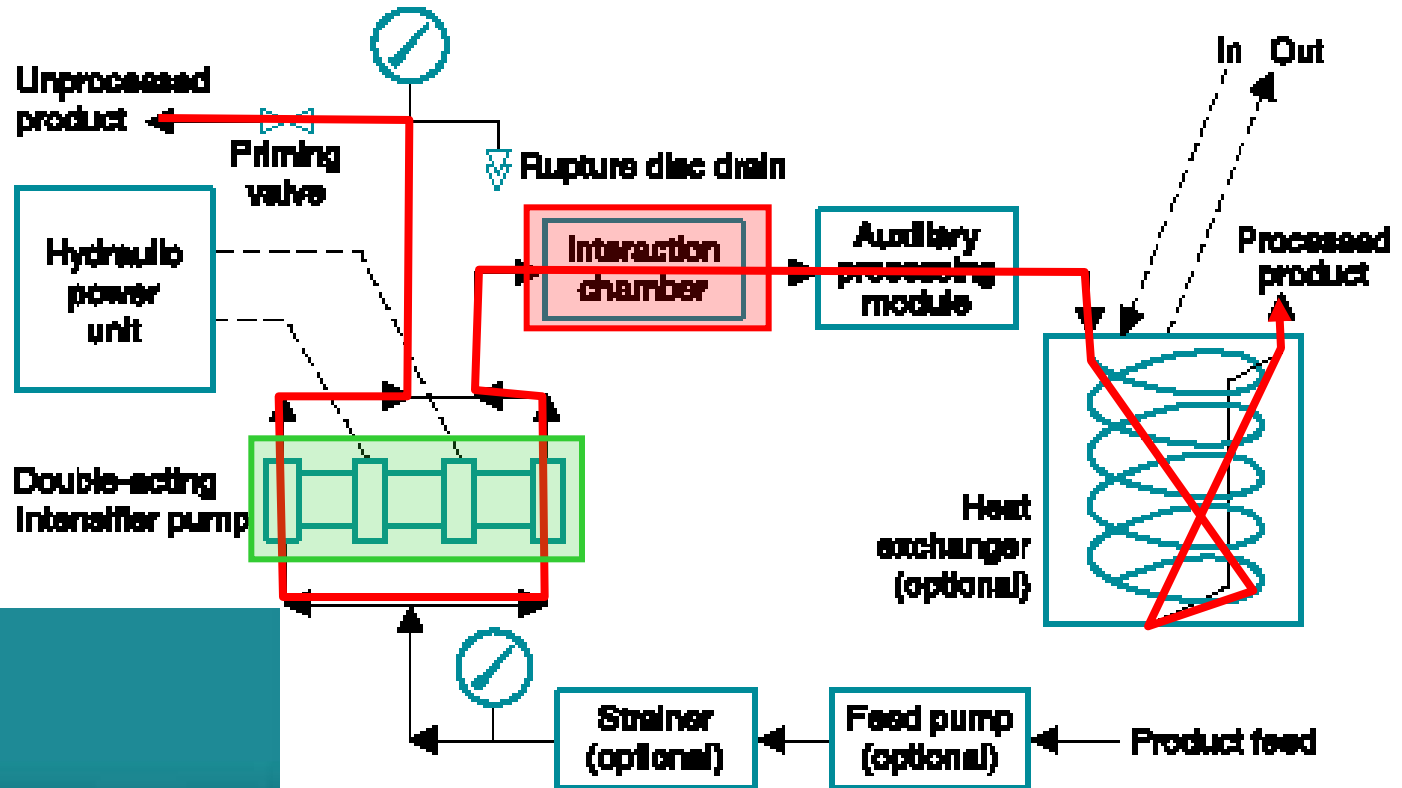
Multilamellar



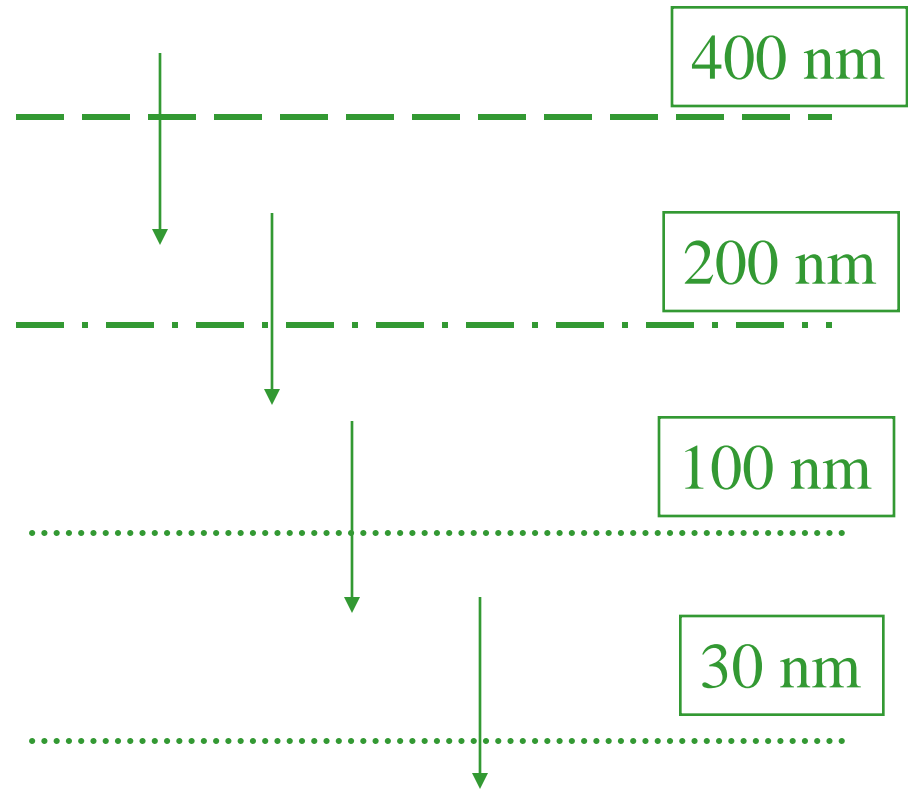
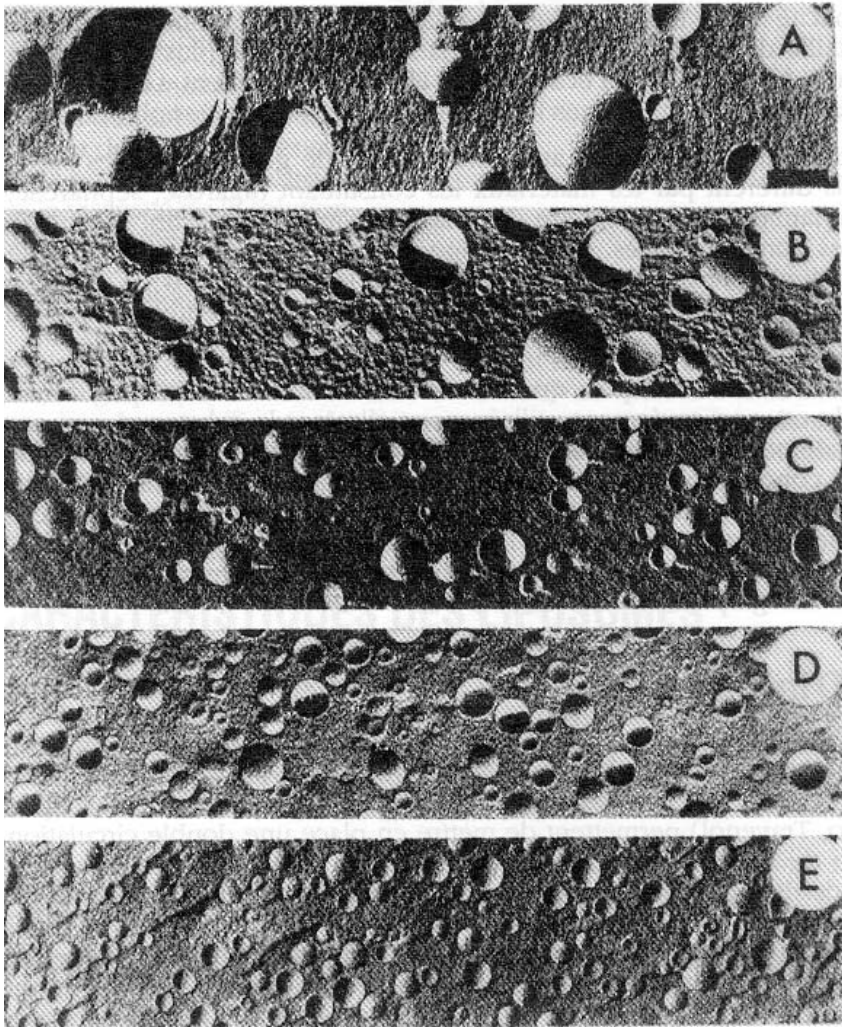
Unilamellar

Sources of Energy: Pressure, Extrusion, Ultrasounds etc.

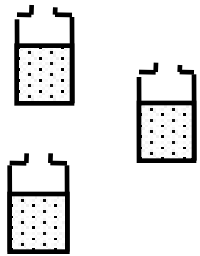
MICROFLUIDIZER



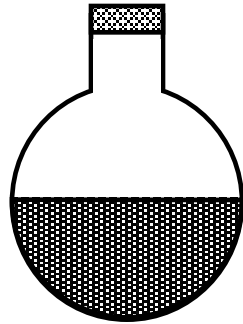
EXTRUSION



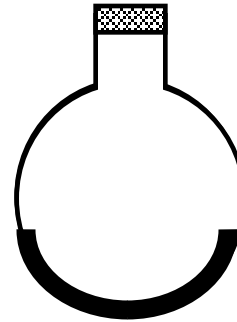
PREPARATION BY REVERSE PHASE EVAPORATION



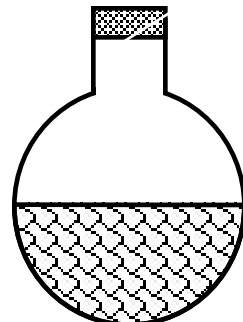
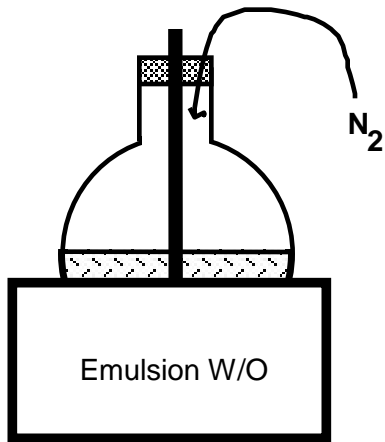
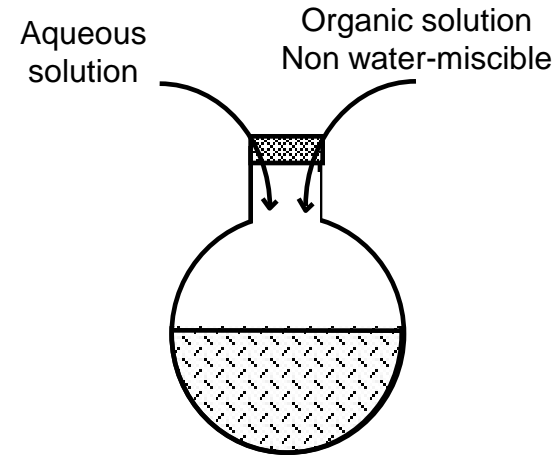
Lipids in organic solvent



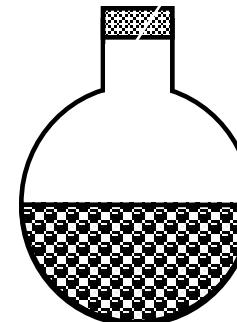
Solvent evaporation



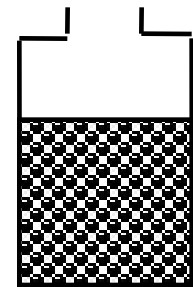
Lipid film formation



Formation of a gel

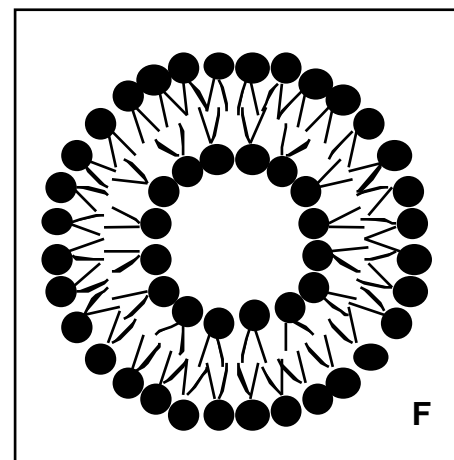
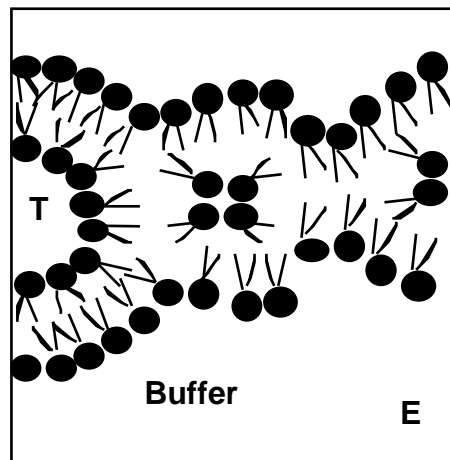
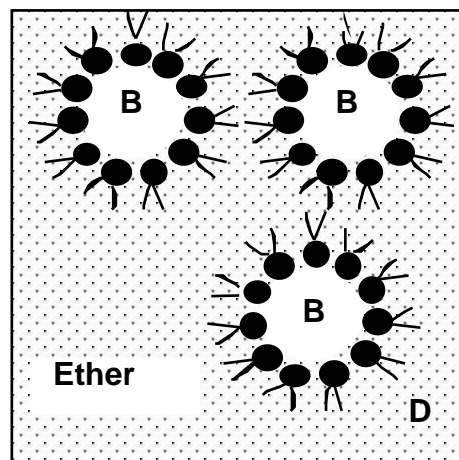
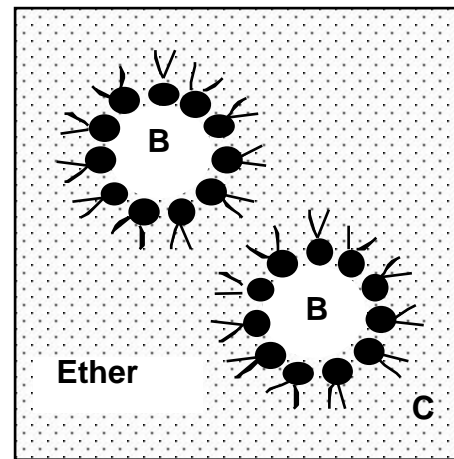
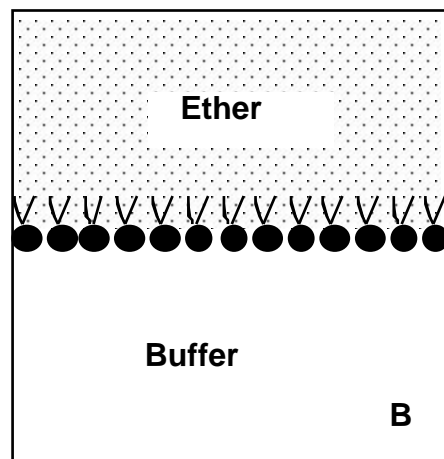
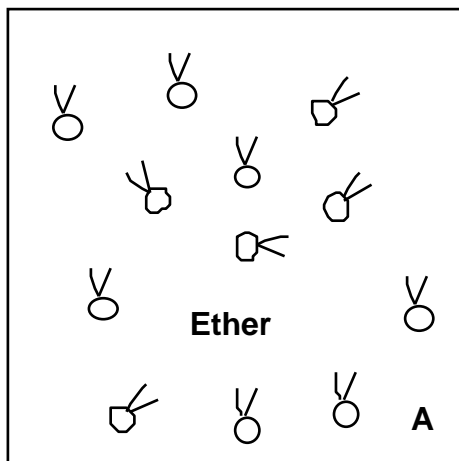


Disruption of the gel

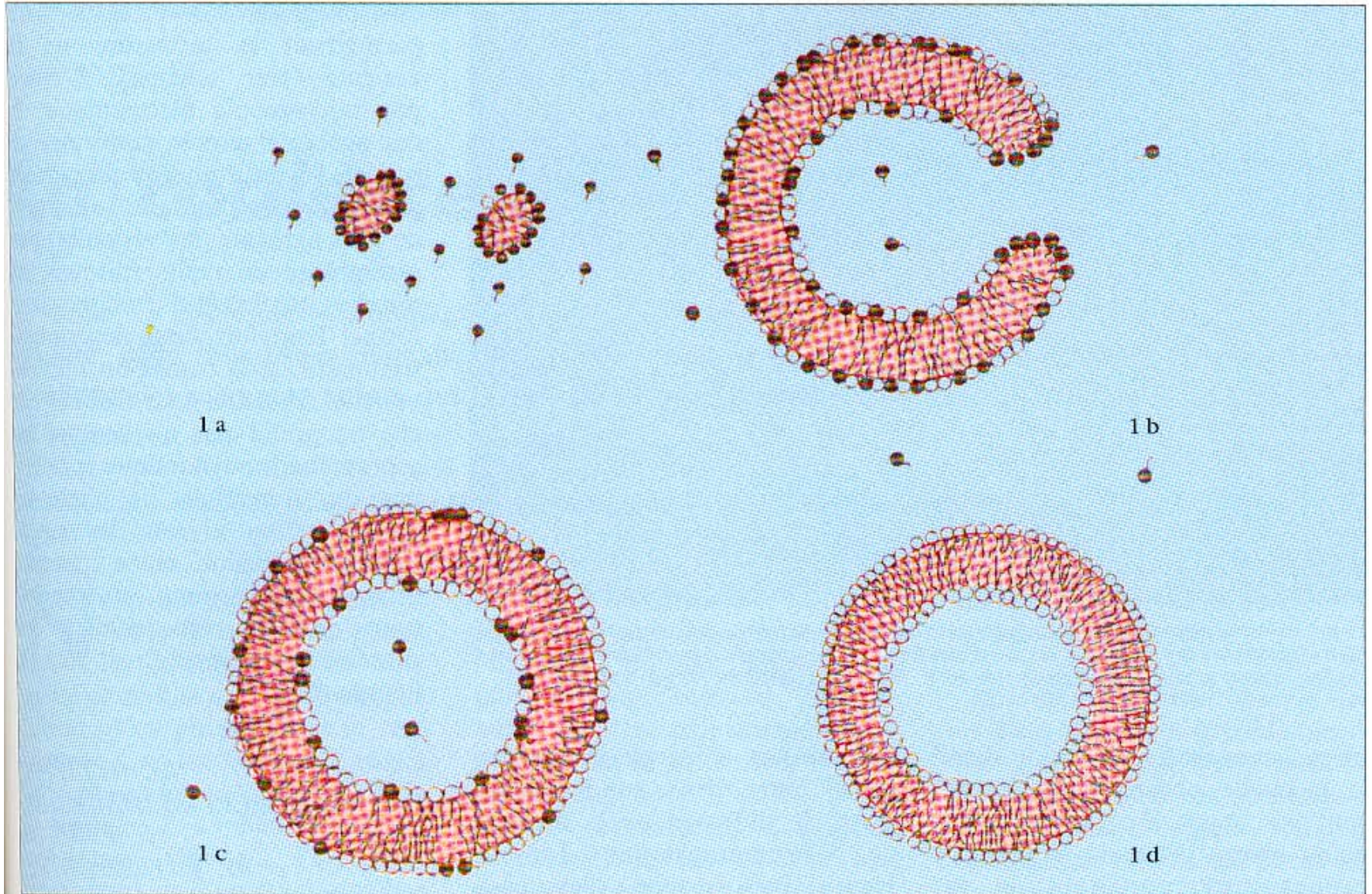


Liposomes formation

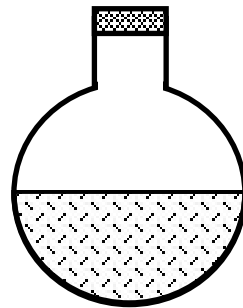
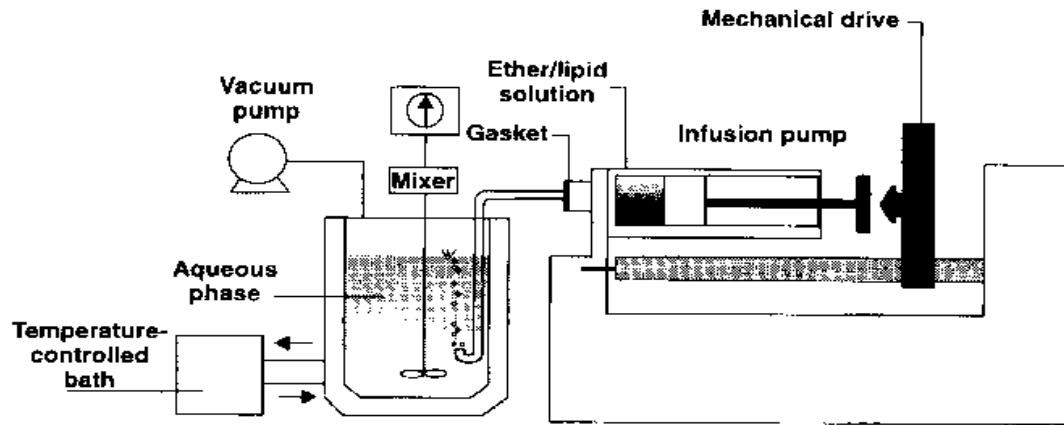
PREPARATION BY REVERSE PHASE EVAPORATION



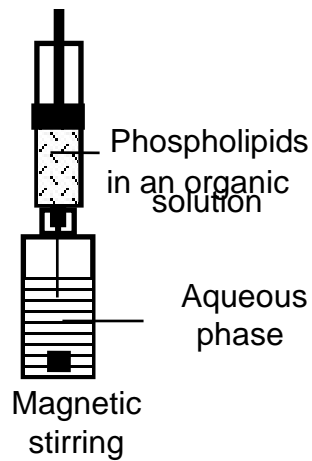
PREPARATION DES LIPOSOMES PAR DIALYSE



PREPARATION DES LIPOSOMES PAR INJECTION DE SOLVANTS



Lipids are dissolved
In a water-miscible solution



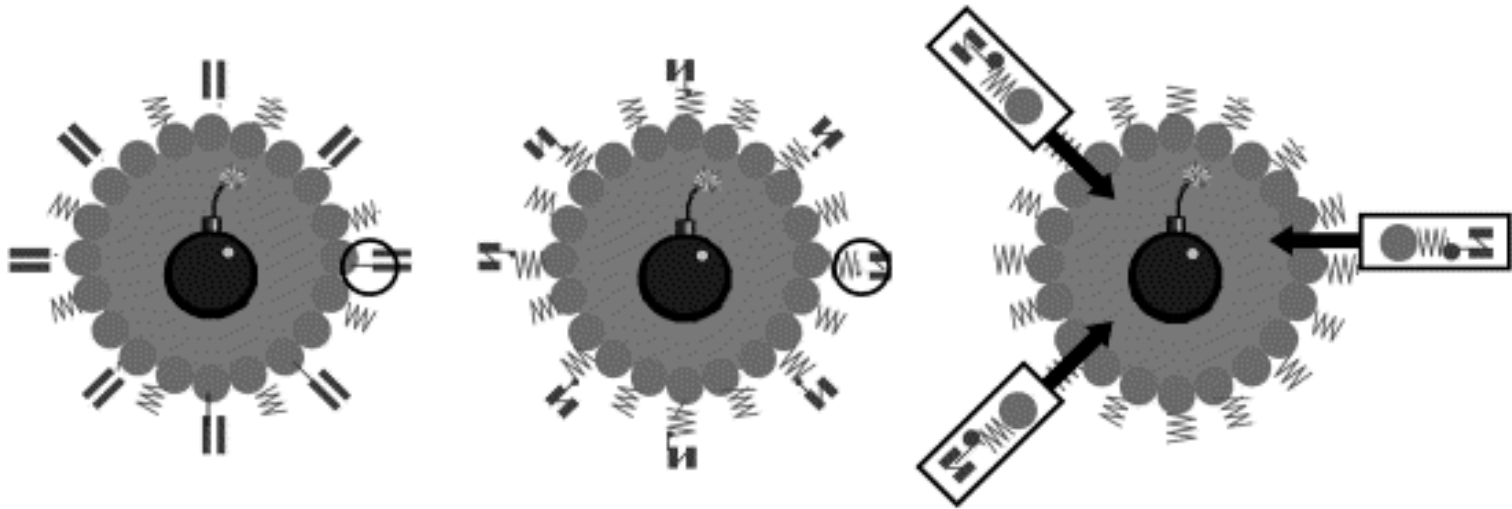
CONJUGATION STRATEGIES USED FOR CONSTRUCTION OF ANTI-HER 2 IMMUNOLIPOSOMES

Park JW et al., Journal of Controlled Release 74, 95-113, 2001

A. Ls-MAb Linkage

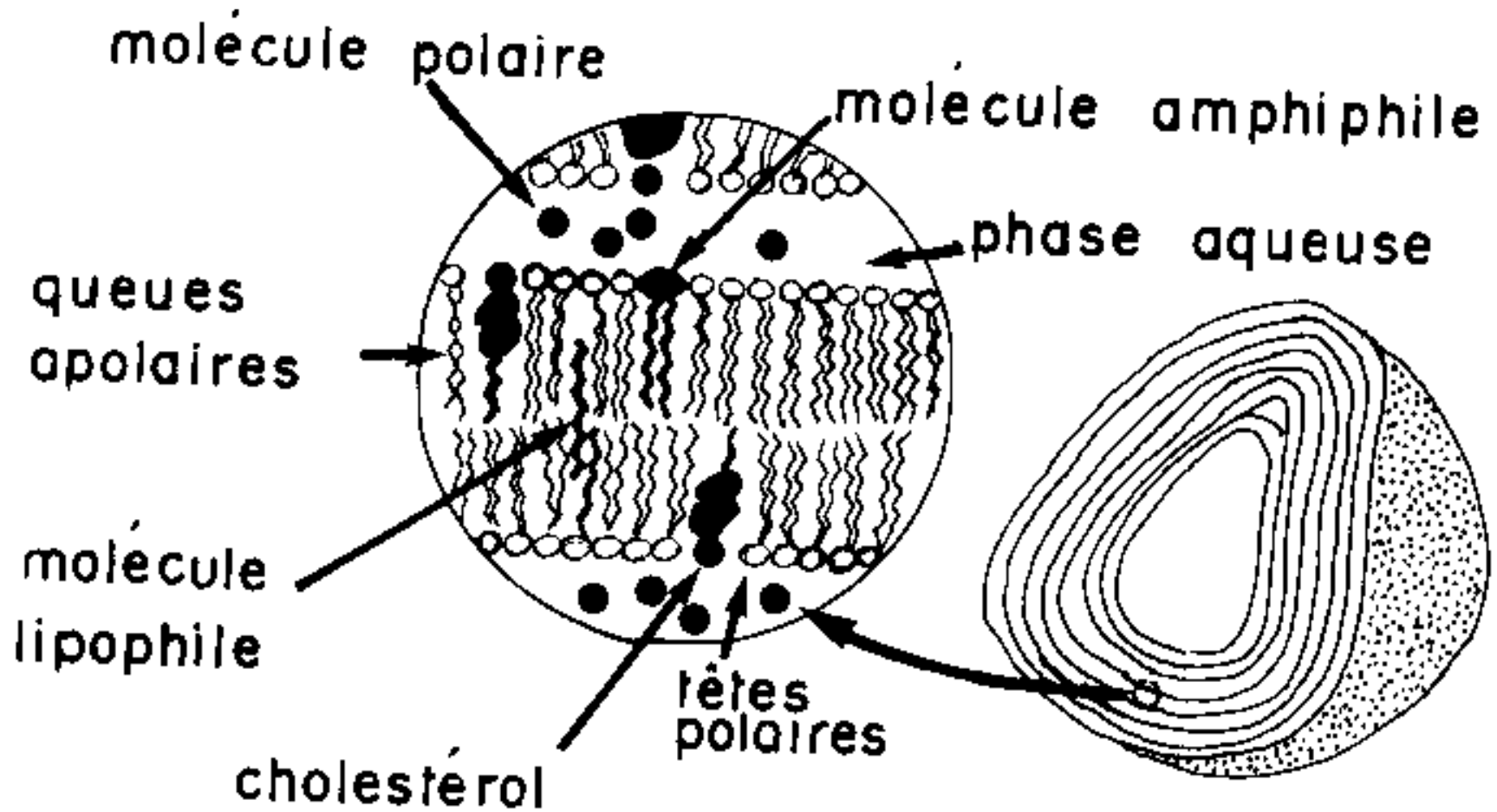
B. PEG-MAb Linkage

C. Micellar Incorporation

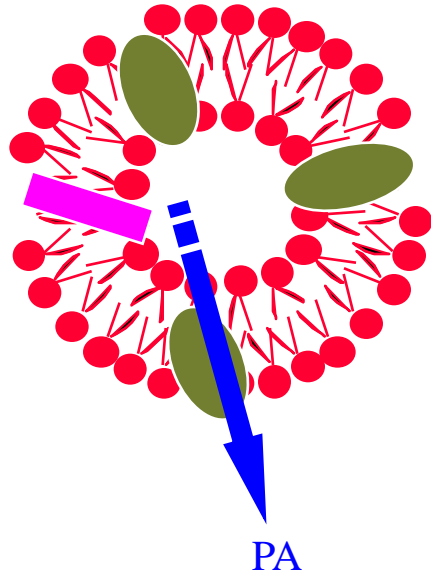


- SUV (70–100 nm) with PEG2000-derivatized disteoylphosphatidylethanolamine (PEG-PE). Anti-HER2 MAb fragments consisted of rhuMAb HER2 (trastuzumab)-Fab', scFv C6.5, scFv F5, or variants of these. MAb fragments all contained a C-terminal cysteine for covalent conjugation.
- (Left) Ls-MAb linkage. MAb fragments were conjugated to maleimide-terminated phosphatylethanolamine (M-PE) at the liposome surface.
- (Middle) PEG-MAb linkage. MAb fragments were conjugated to maleimide-terminated PEG-PE (M-PEG-PE)
- (Right) Micellar incorporation. Preformed liposomes lacking functional sites for conjugation were converted into immunoliposomes by incorporation of modified MAb fragments coupled to M-PEG-PE, forming micelles which incorporate into liposomes

L' ENCAPSULATION DANS LES LIPOSOMES

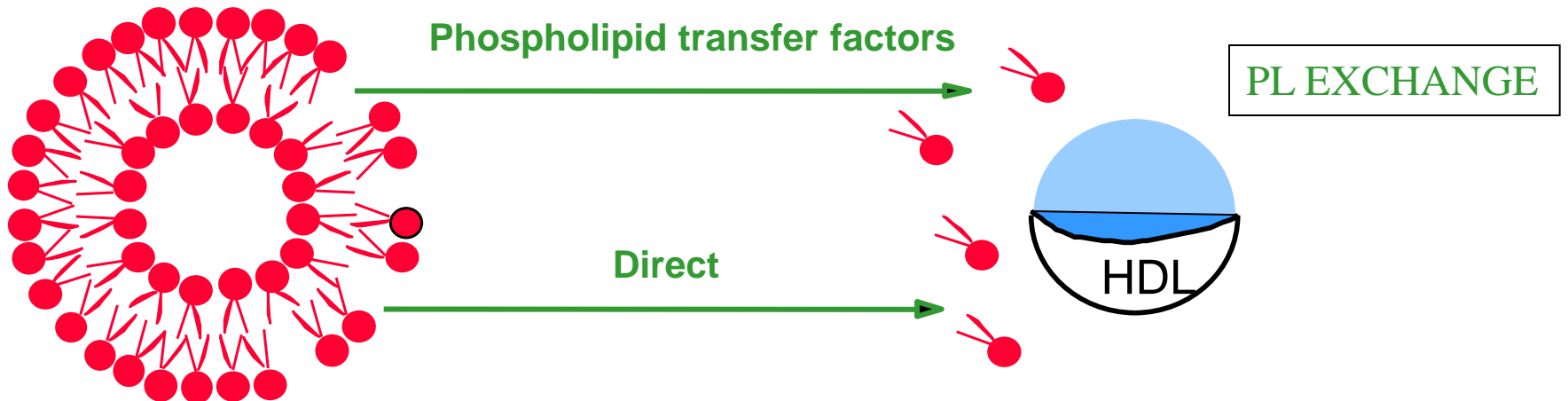


LIPOSOME STABILITY

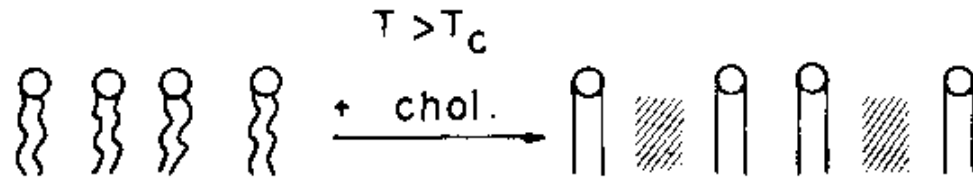
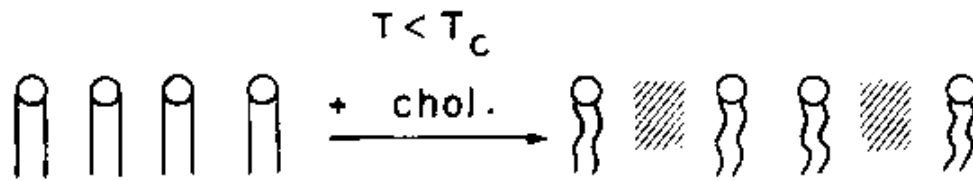


- ✓ Fibronectin
- ✓ Albumin
- ✓ Lipoproteins
- ✓ Complement
- ✓ Immunoglobulins

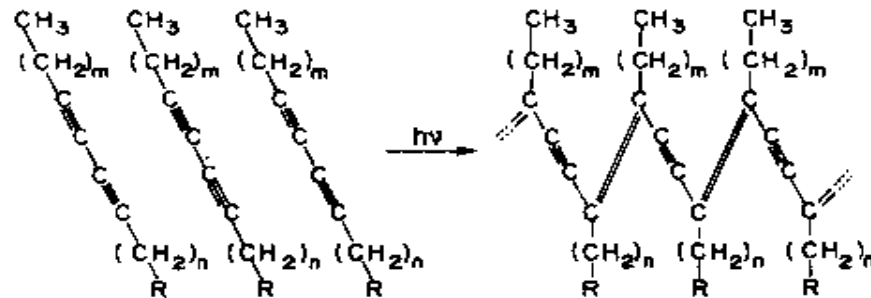
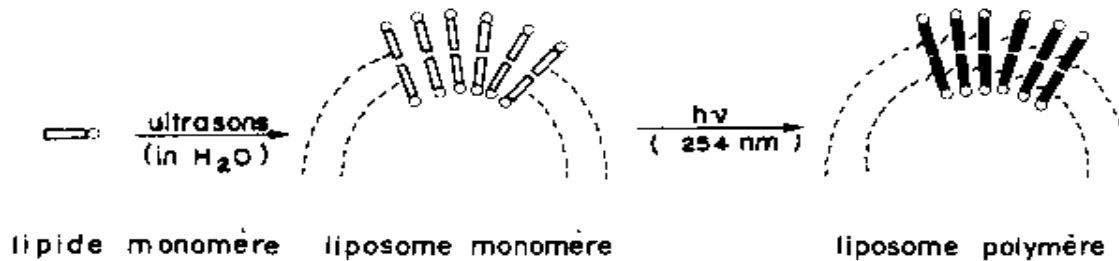
DRUG LEAKAGE



LIPOSOMES STABILIZATION



CHOLESTEROL



POLYMERIZABLE PL

NANOPARTICULES

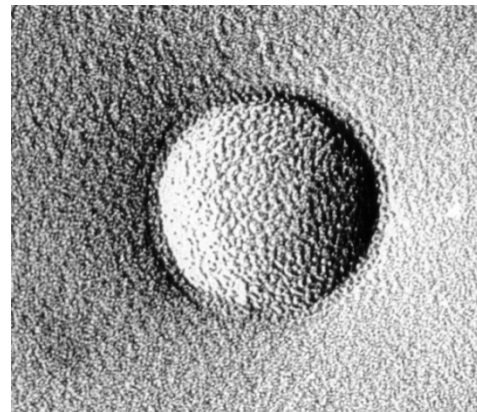
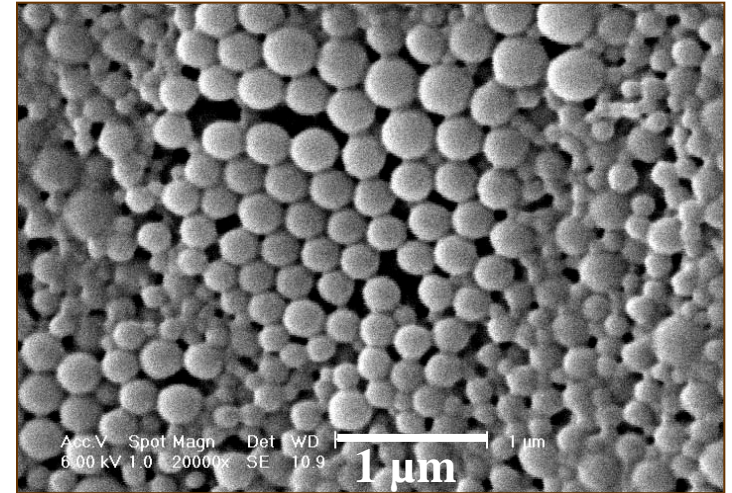
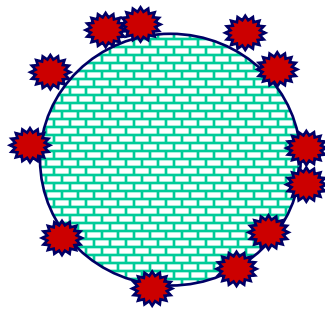
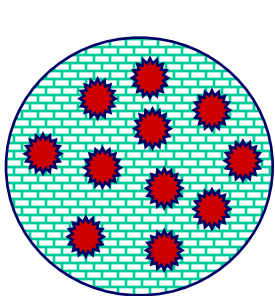
Nanospheres et Nanocapsules

NANOPARTICULES: DEFINITION

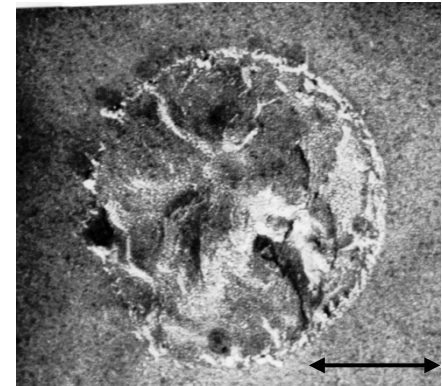
NANOPARTICULES: structures supramoléculaires solides ultradispersées, généralement à base de polymères, ayant une taille inférieure au micron

Ces objets peuvent être de type matriciels (NANOSPHERES) ou vésiculaires (NANOCAPSULES)

Le médicament vectorisé peut s'associer à ces structures sous différents états physiques

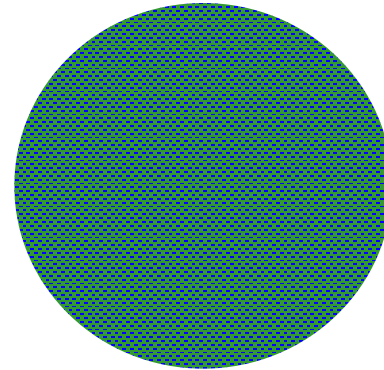
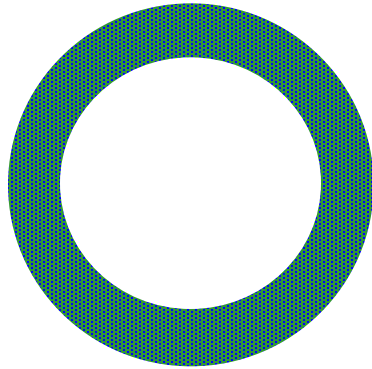


Nanospheres



Nanocapsules

NANOPARTICLES PREPARATION

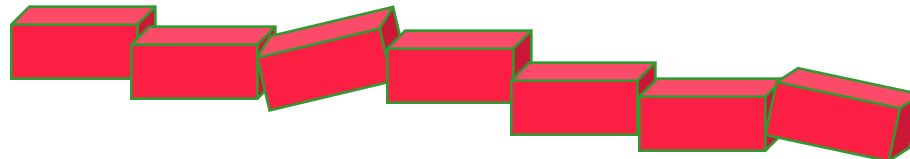


- natural macromolecules
- synthetic polymers

monomers



polymers

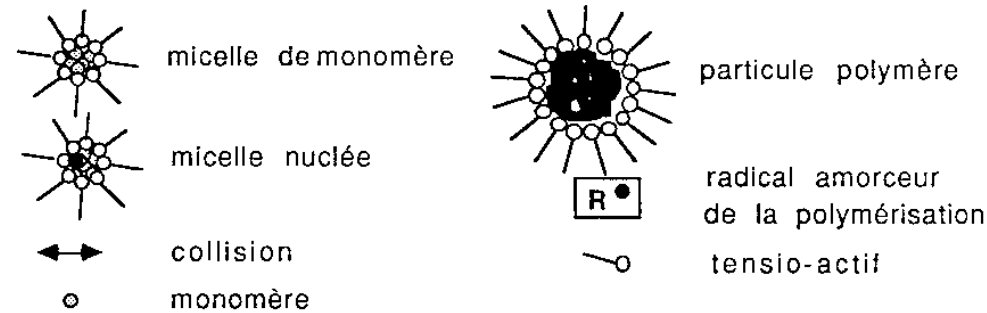
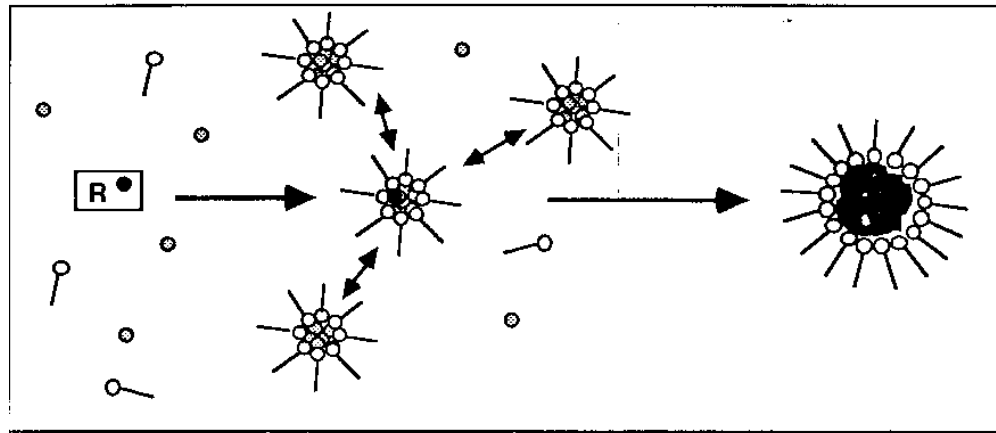


PREPARATION DE NANOPARTICULES PAR POLYMERISATION

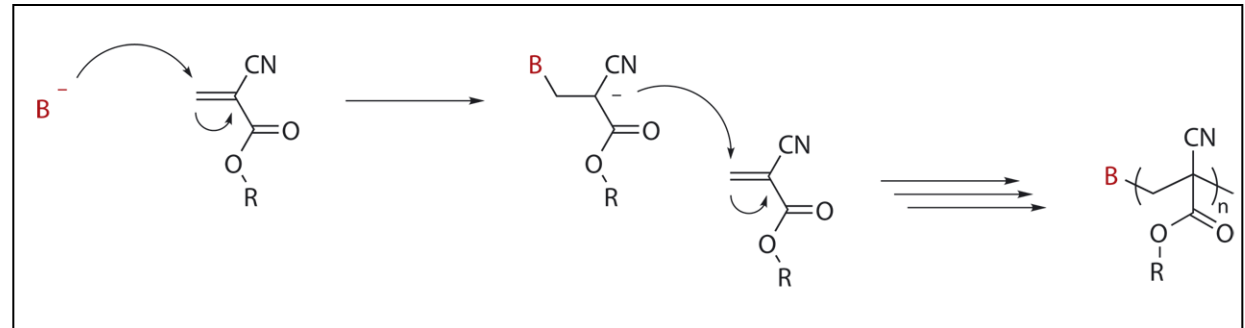
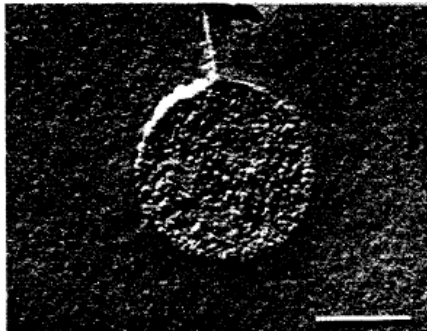
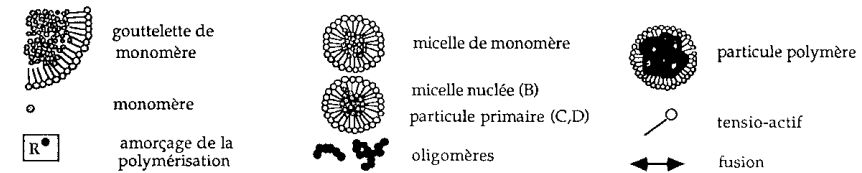
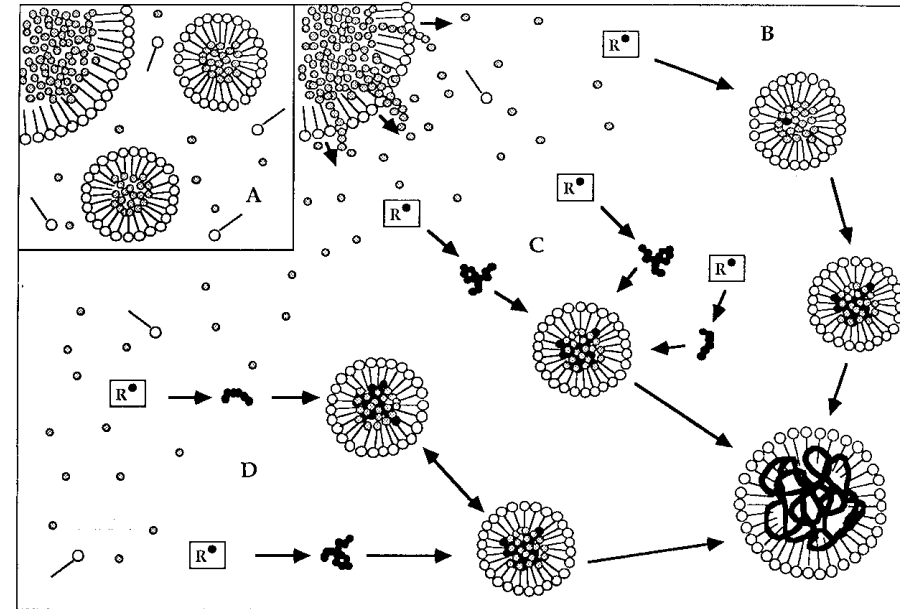
Polymérisation en émulsion et en suspension

PREPARATION DE NANOSPHERES PAR POLYMERISATION

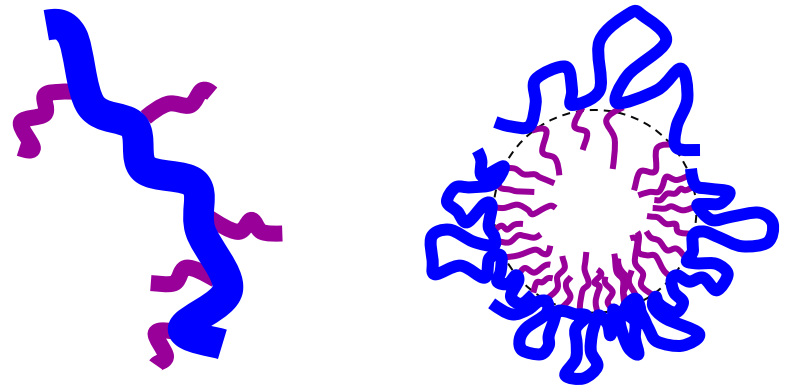
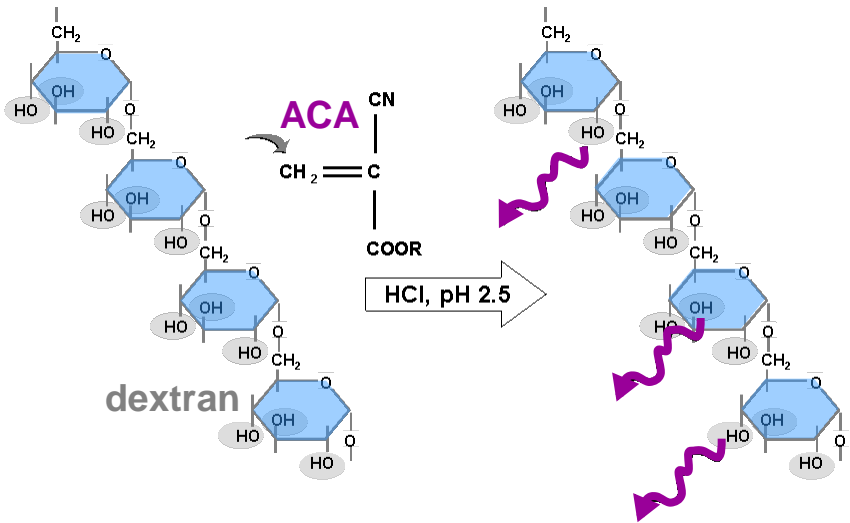
En suspension



En émulsion



A



POLYMERIZATION

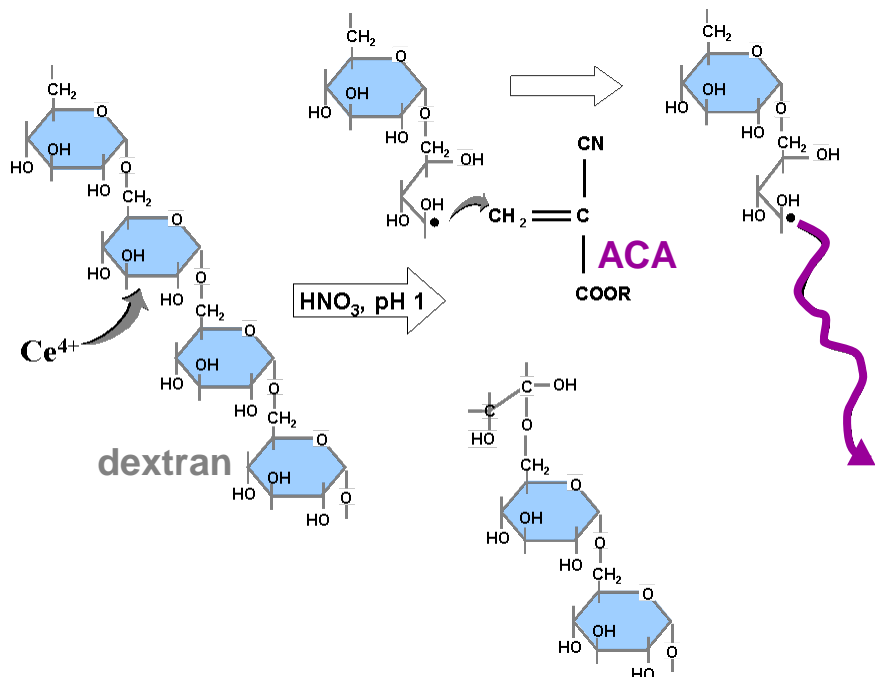
INITIATION

ELONGATION

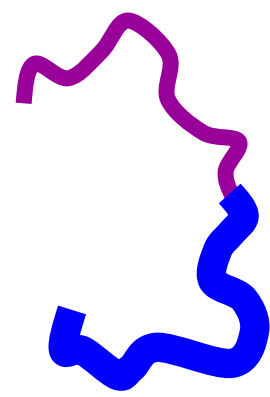
COPOLYMERS

NANOPARTICLES

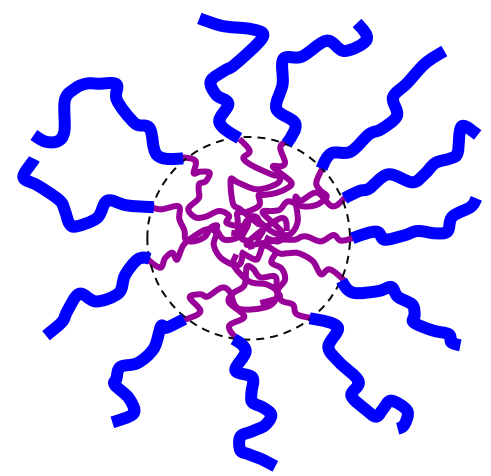
B



Structure ?

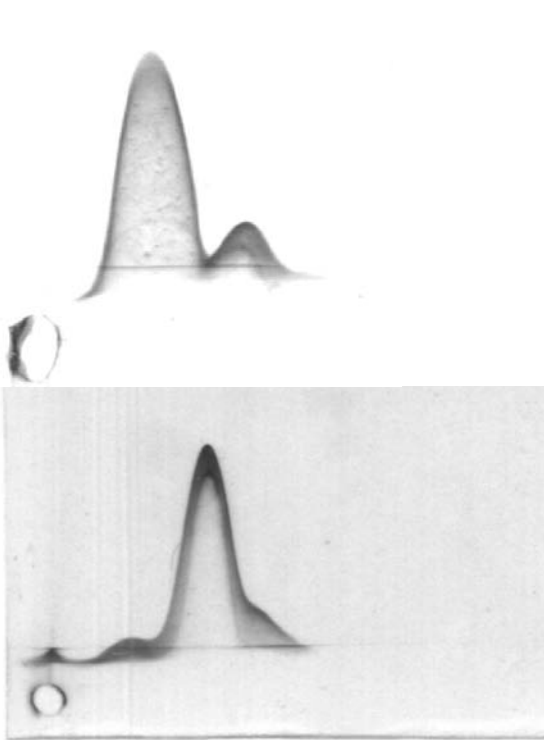


Arrangement des chaînes en surface ?

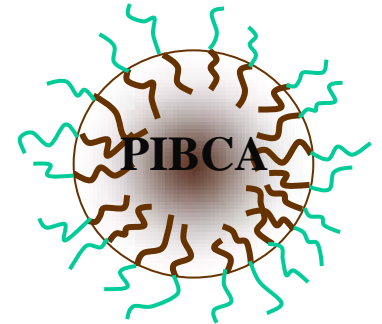


POLYMERISATION ANIONIQUE ET RADICALAIRE ET ACTIVATION DU COMPLEMENT

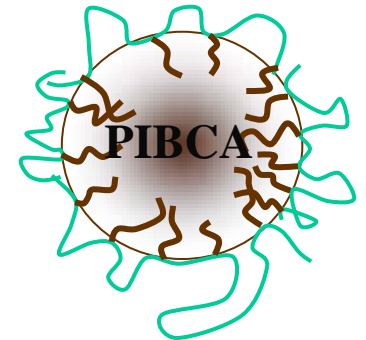
Chauvierre et al., Cell Mol Biol, 50, 233-239 (2004)



Polymérisation radicalaire
Non activateur



Polymérisation anionique
Activateur

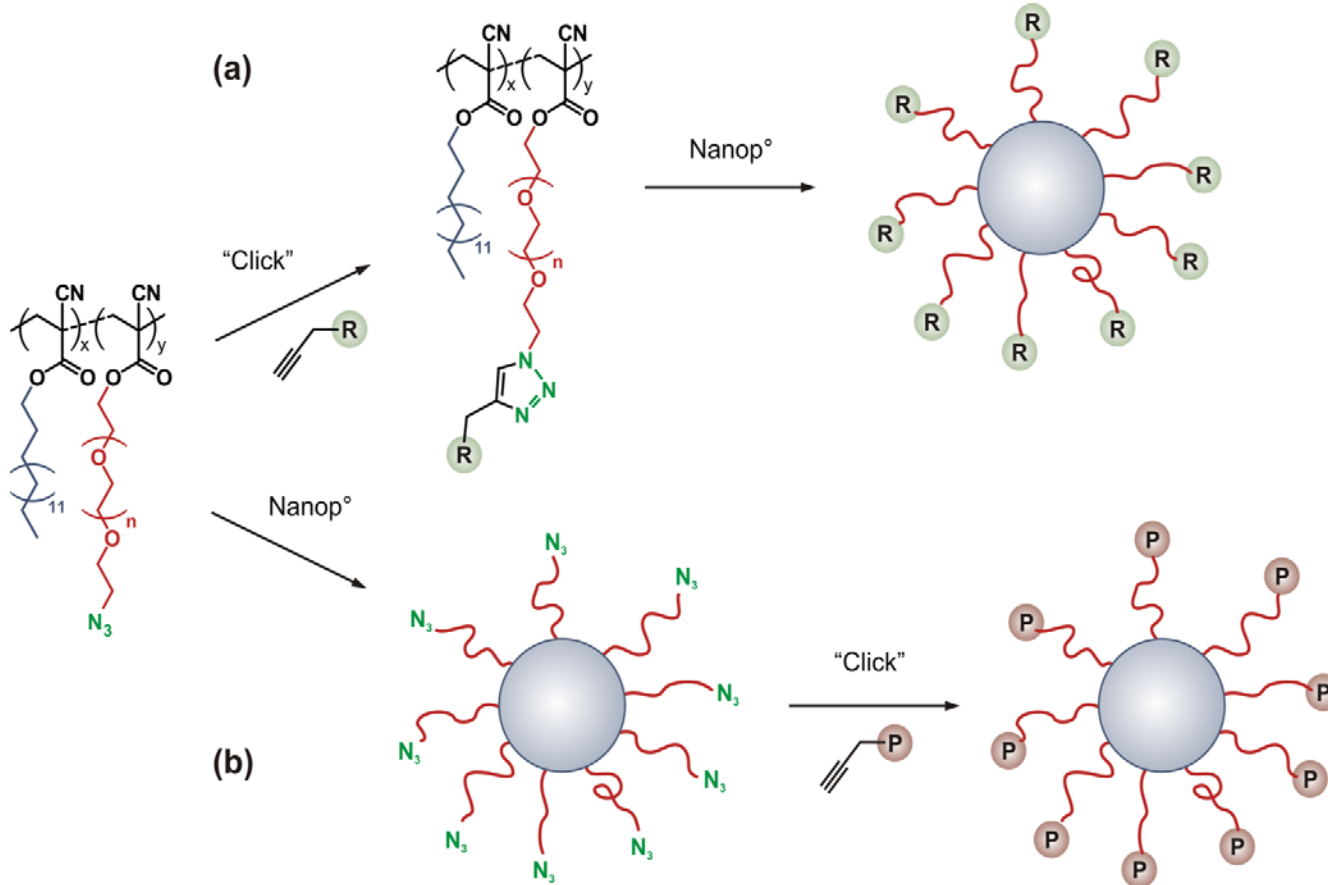


Dextrane 71 000KD



CLICK CHEMISTRY » POUR LE COUPLAGE DE LIGANDS A LA SURFACE DE NANOPARTICULES DE PACA

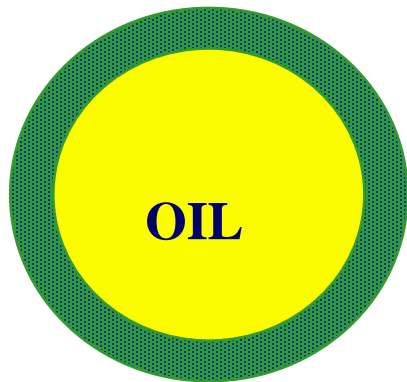
Nicolas J et al., *Macromolecules*, in press, 2008
Nicolas J et al, *Macromolecules*, **41**, 3758-3761 (2008)



PREPARATION DE NANOCAPSULES PAR POLYMERISATION INTERFACIALE

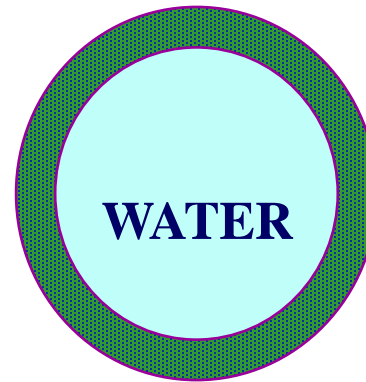
PREPARATION DE NANOCAPSULES PAR POLYMERISATION INTERFACIALE

O/W

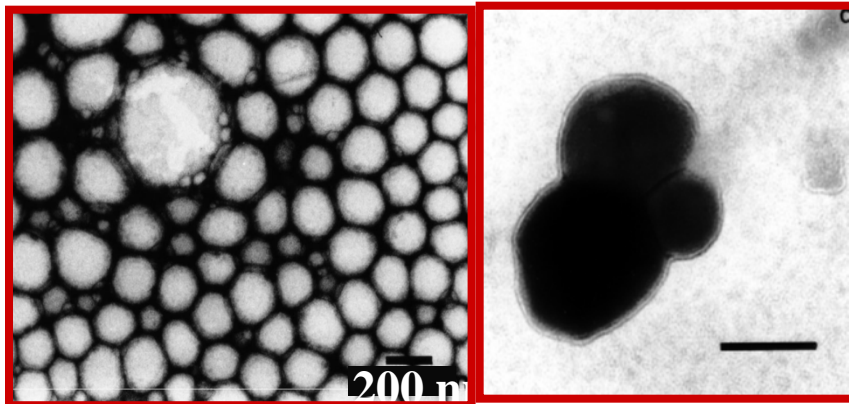


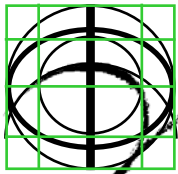
→ PA lipophiles

W/O



→ Oligonucleotides
→ Peptides

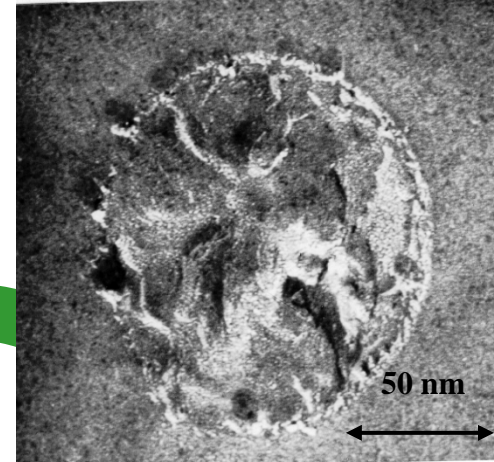
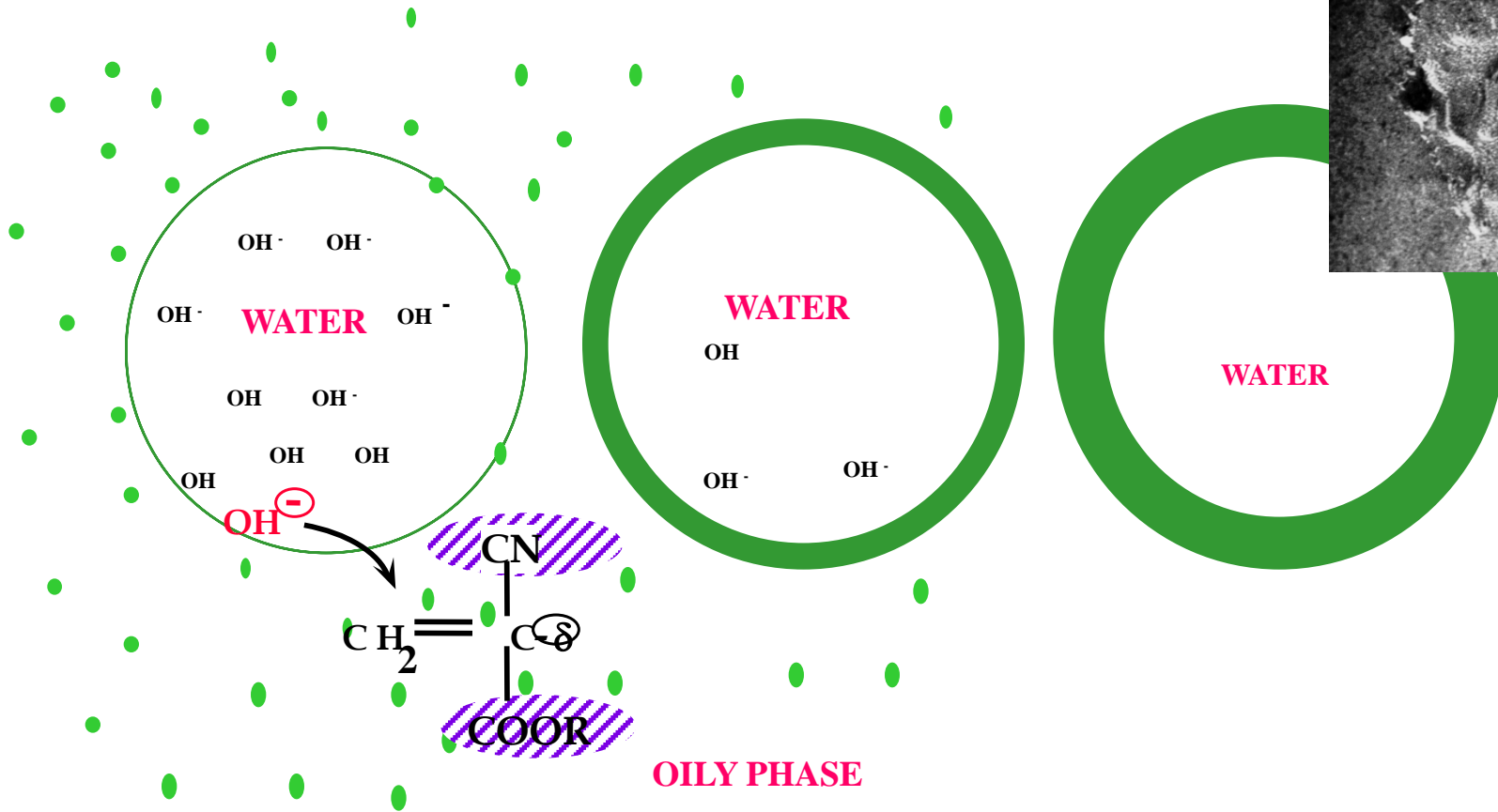




PACA NANOCAPSULE FORMATION

ibert, J.R. Bertrand, E. Fattal, F. Subra, H. Pinto-Alphandary, C. Malvy, C. Auclair, P. Couvreur, BBRC, 2001

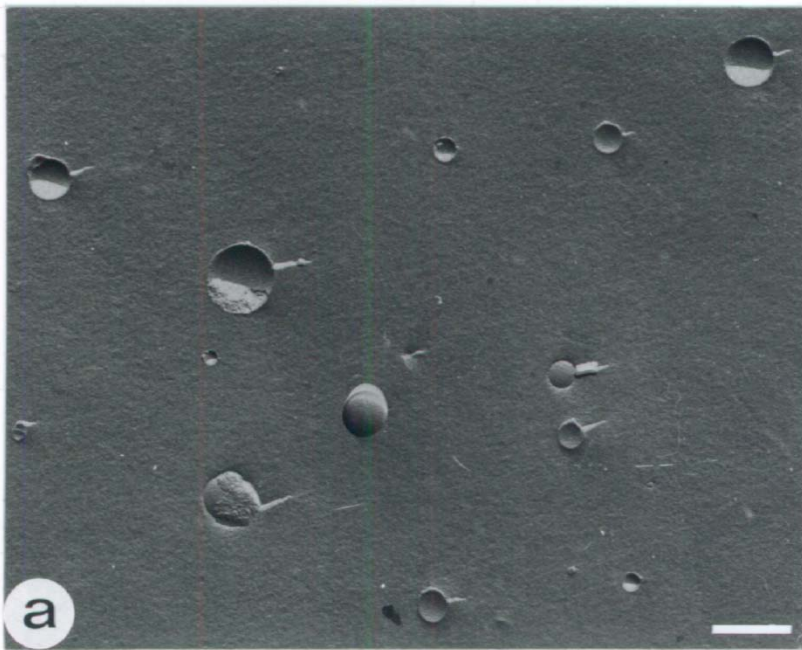
UMR CNRS 8612



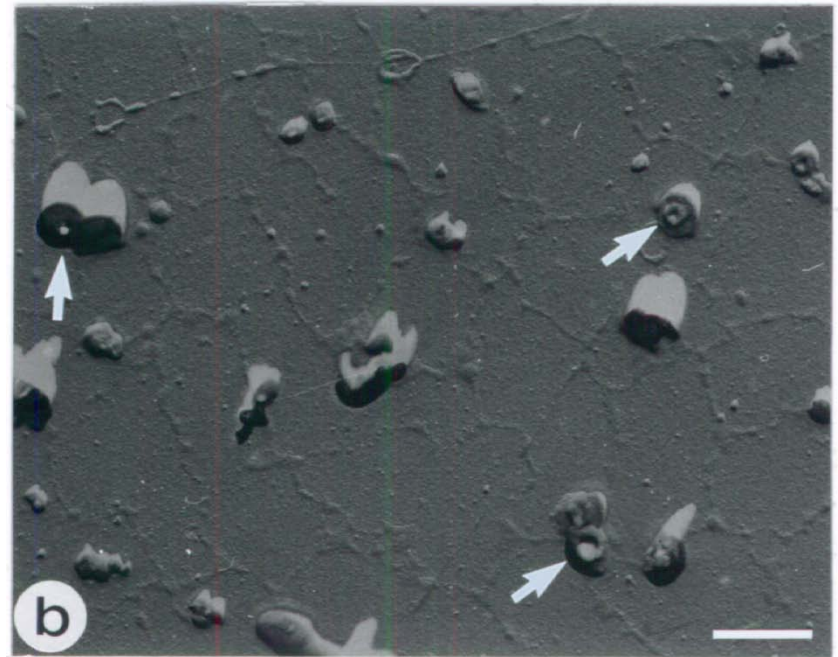
50 nm

NANOCAPSULES WITH AN AQUEOUS CORE/ FREEZE FRACTURE

Lambert et al., Pharm. Res., 2000



a) Before drying



b) After drying

NANOCAPSULES PREPARATION BY INTERFACIAL POLYCONDENSATION

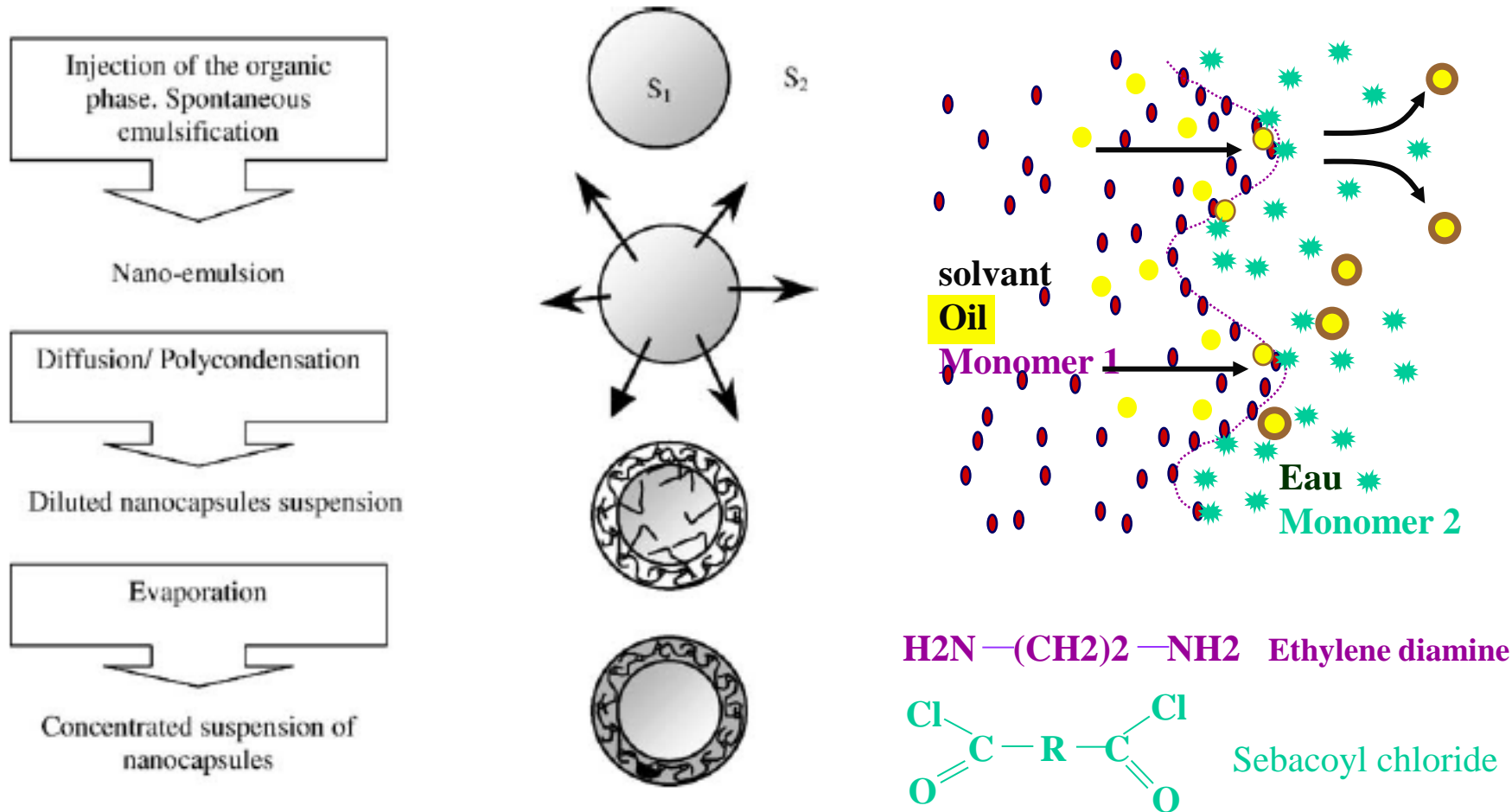
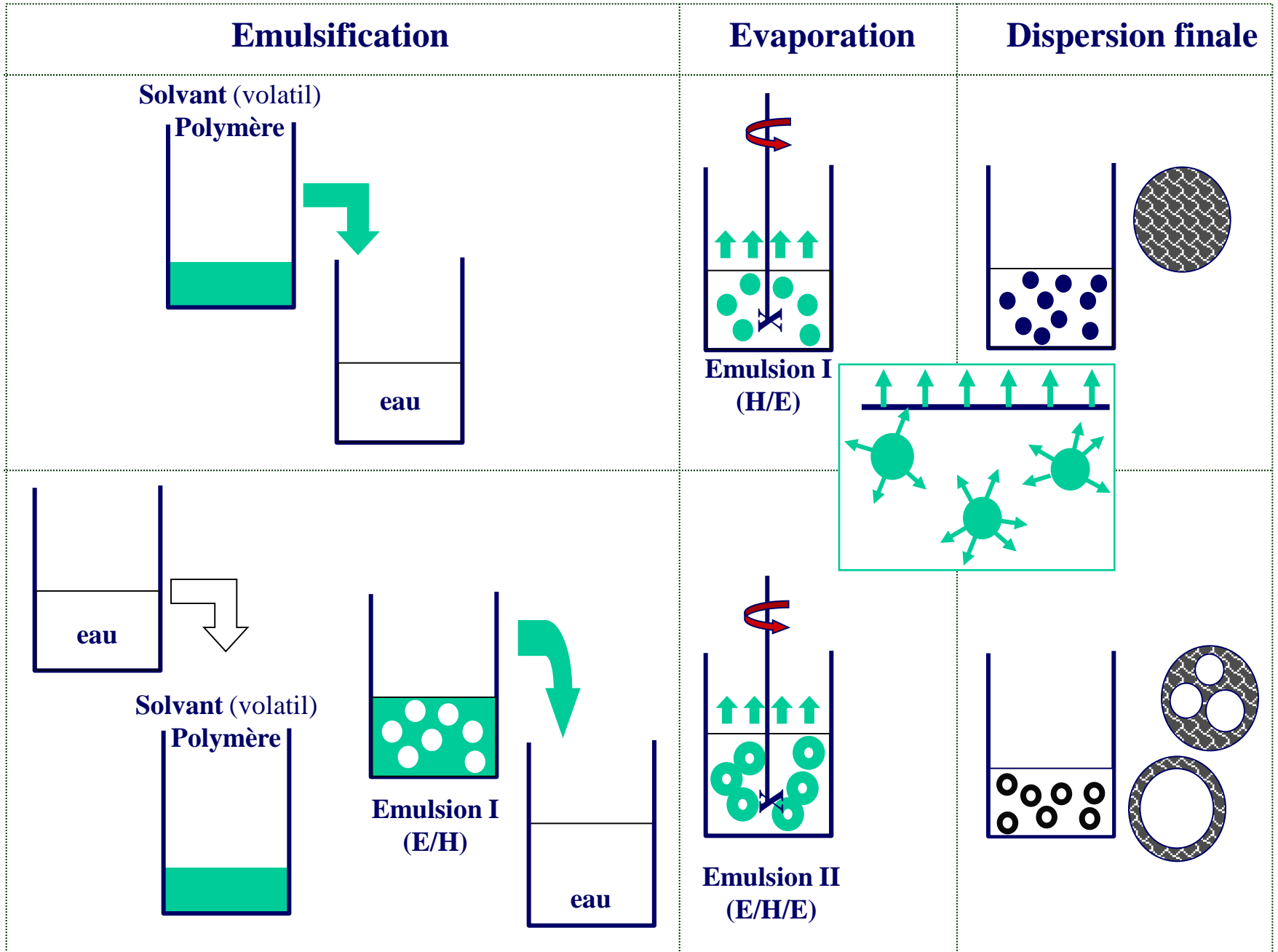


Fig. 1. Mechanism of nanocapsules preparation using the new interfacial polycondensation technique (Montasser et al., 2001).

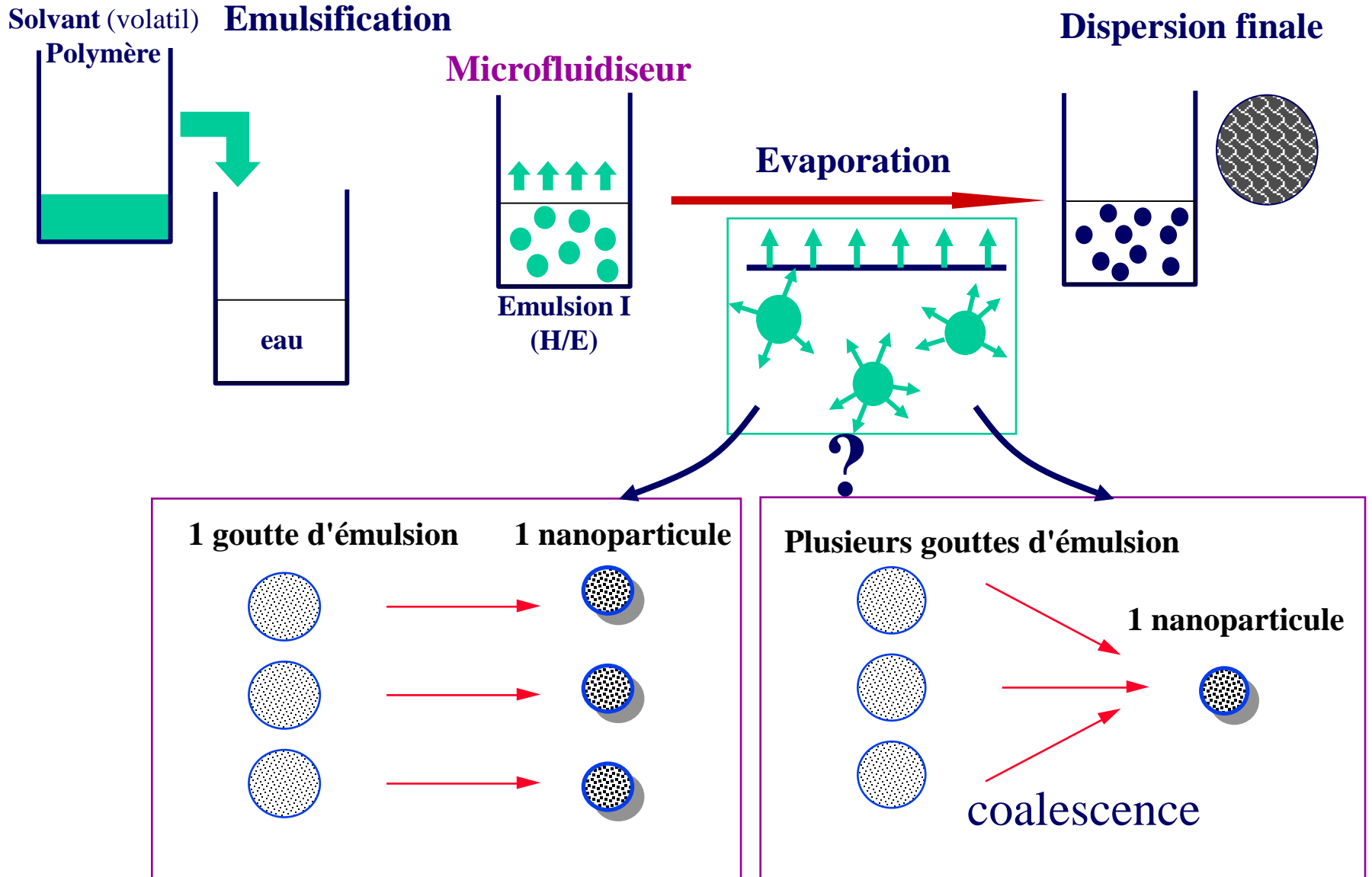
PREPARATION DE NANOPARTICULES A PARTIR D'UN POLYMERE PREFORME

Emulsion évaporation de solvant
Emulsion extraction de solvant

EMULSIFICATION-EVAPORATION DE SOLVANT

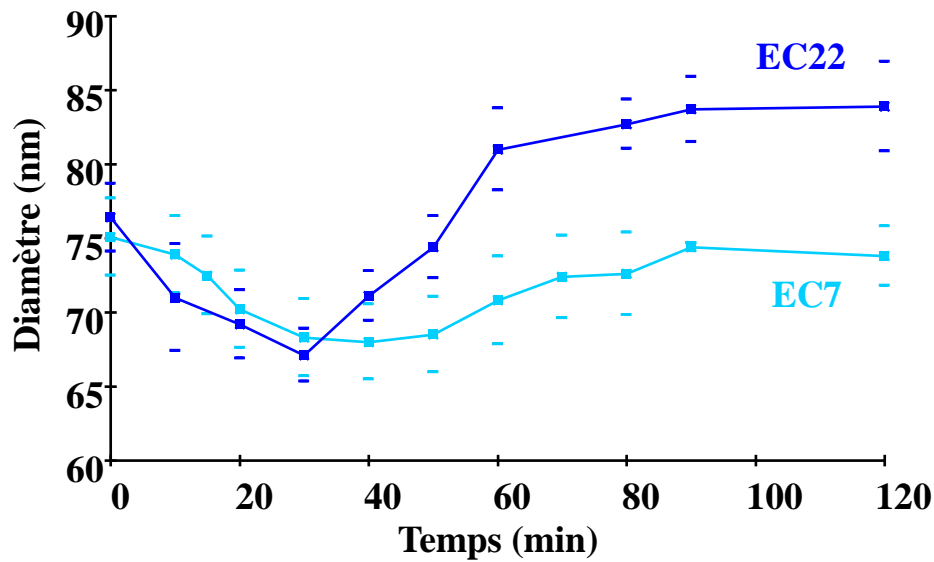


Emulsification – solvent evaporation

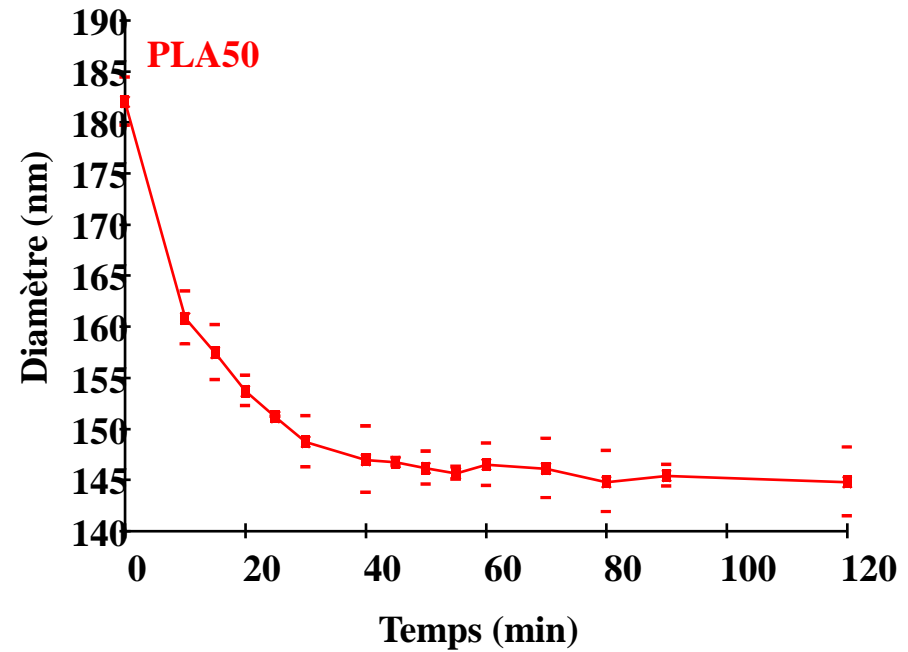


MISE EN EVIDENCE DE LA COALESCENCE

Desgouilles et al., Langmuir 2003



Evaporation



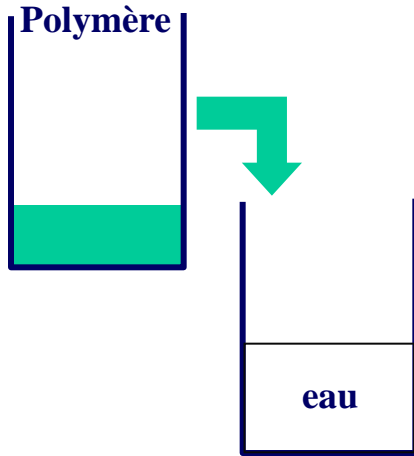
Evaporation

EMULSIFICATION-EVAPORATION DE SOLVANT

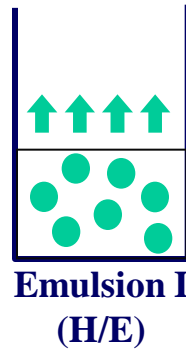
Desgouilles et al., Langmuir 2003

Solvant (volatil)

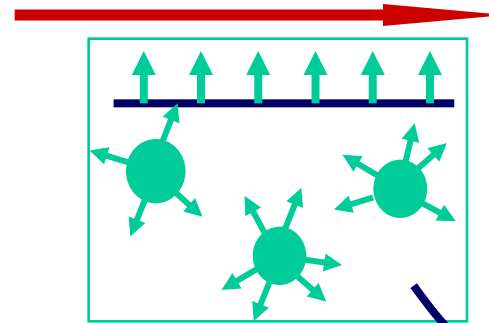
Emulsification



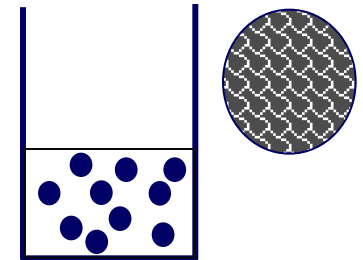
Microfluidiseur



Evaporation

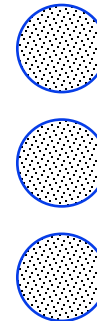


Dispersion finale



$$d_g \propto \mu_d^\alpha$$

Plusieurs gouttes d'émulsion



1 nanoparticule

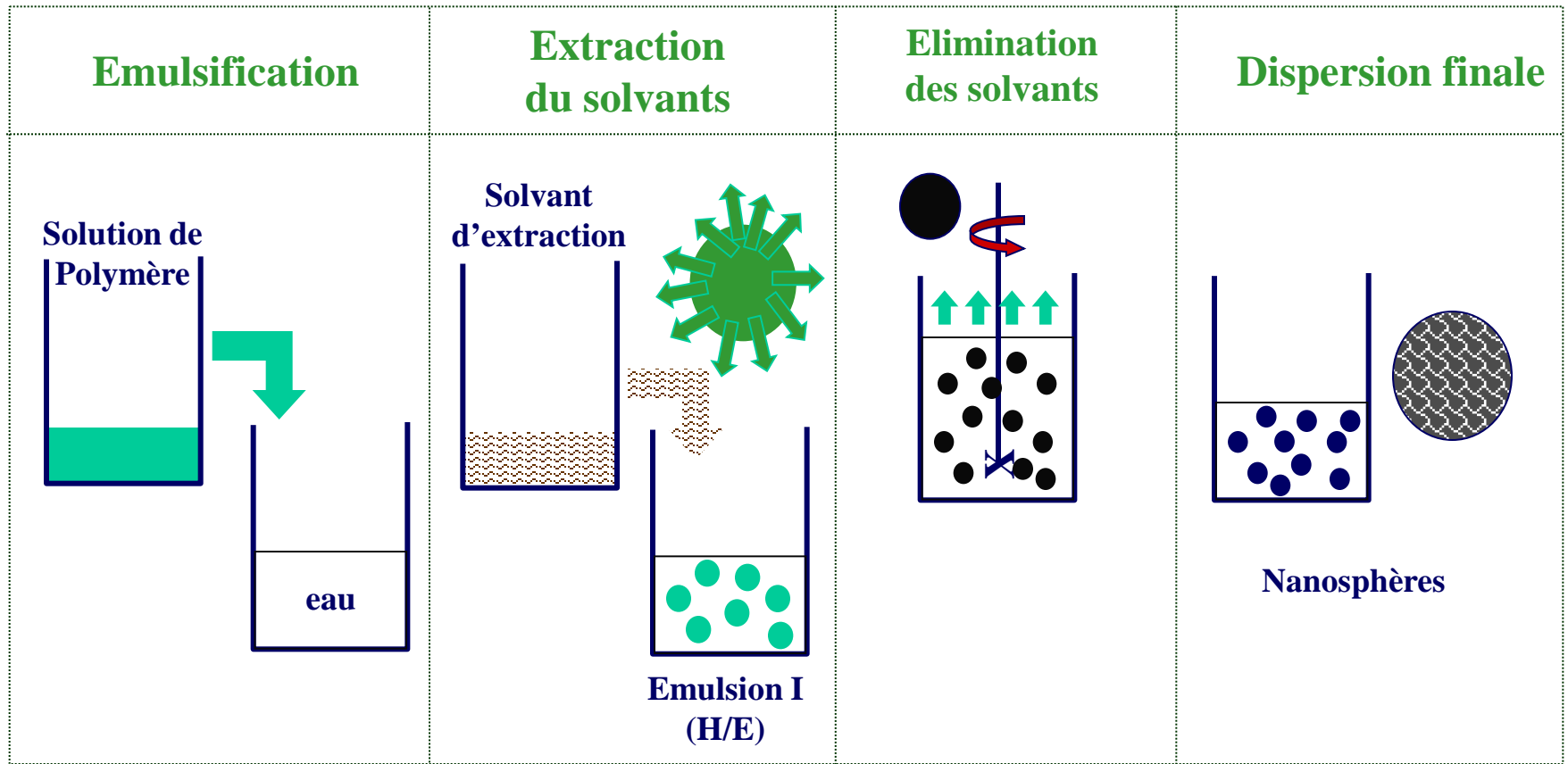


coalescence

Fc Facteur de coalescence
 d g diamètre des gouttes
 d p diamètre des particules
 C concentration massique du polymère
 ρ_0 densité
 μ facteur de viscosité

$$F_c = \frac{\rho_0}{C} \left(\frac{dp}{dg} \right)^3$$

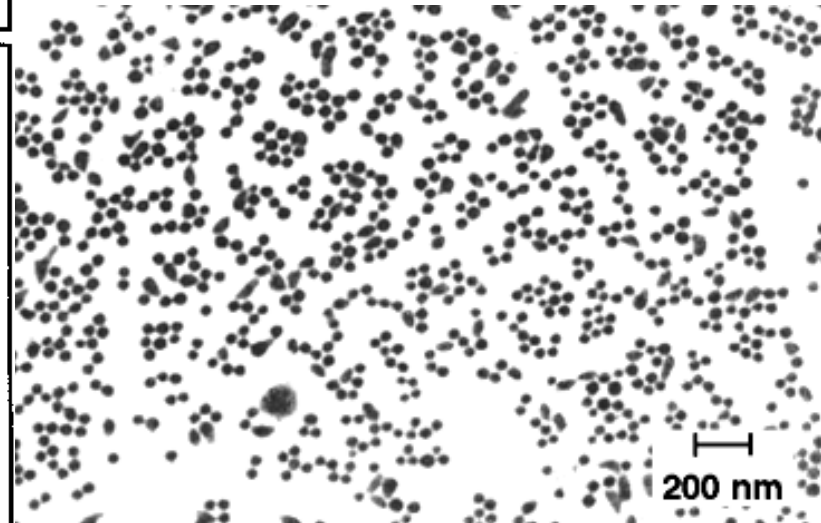
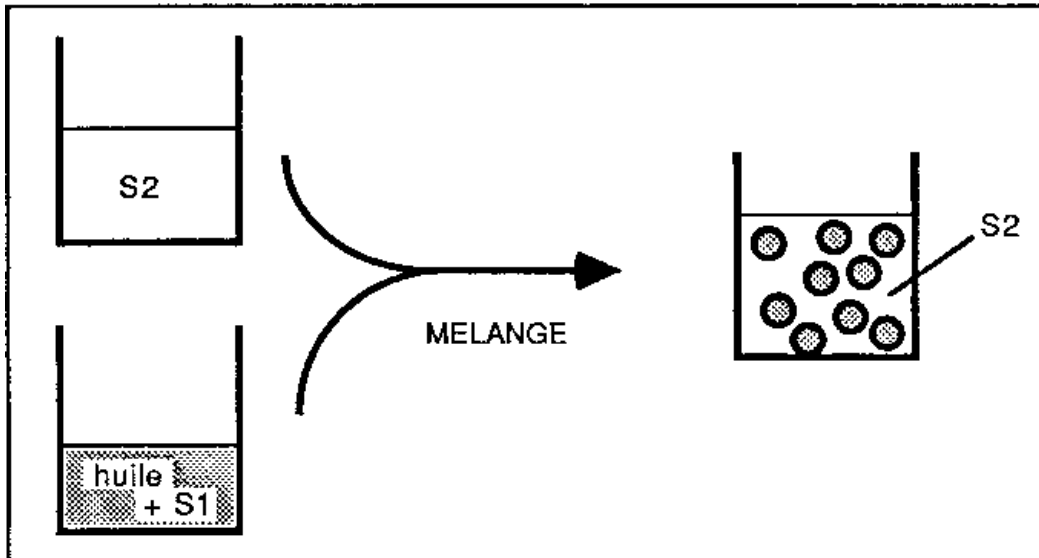
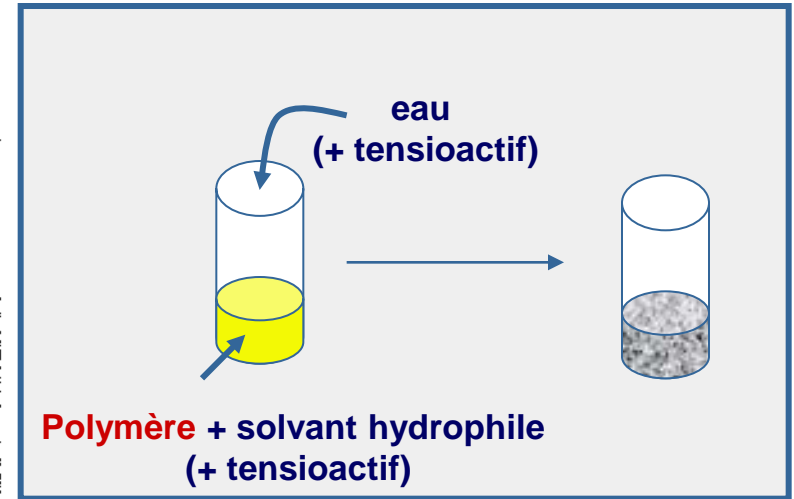
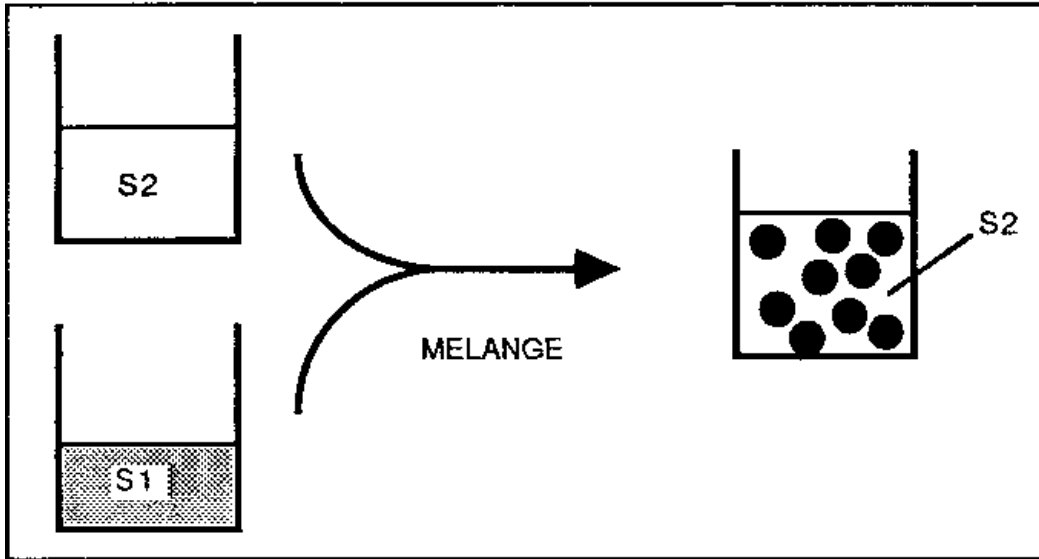
EMULSIFICATION-EXTRACTION DE SOLVANT



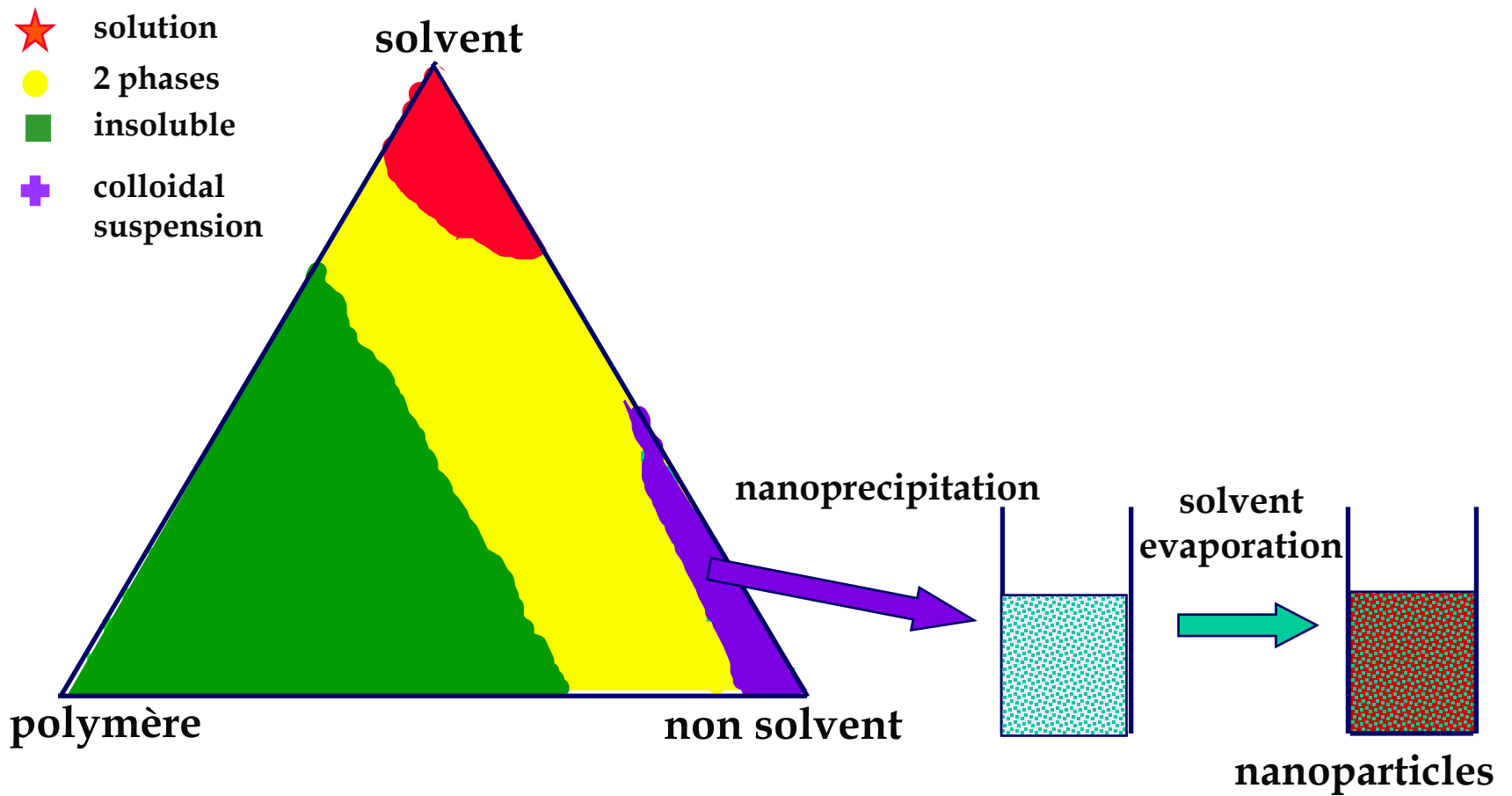
PREPARATION DE NANOPARTICULES A PARTIR D'UN POLYMERE PREFORME

Nanoprécipitation

NANOPRECIPITATION: PRINCIPE GENERAL

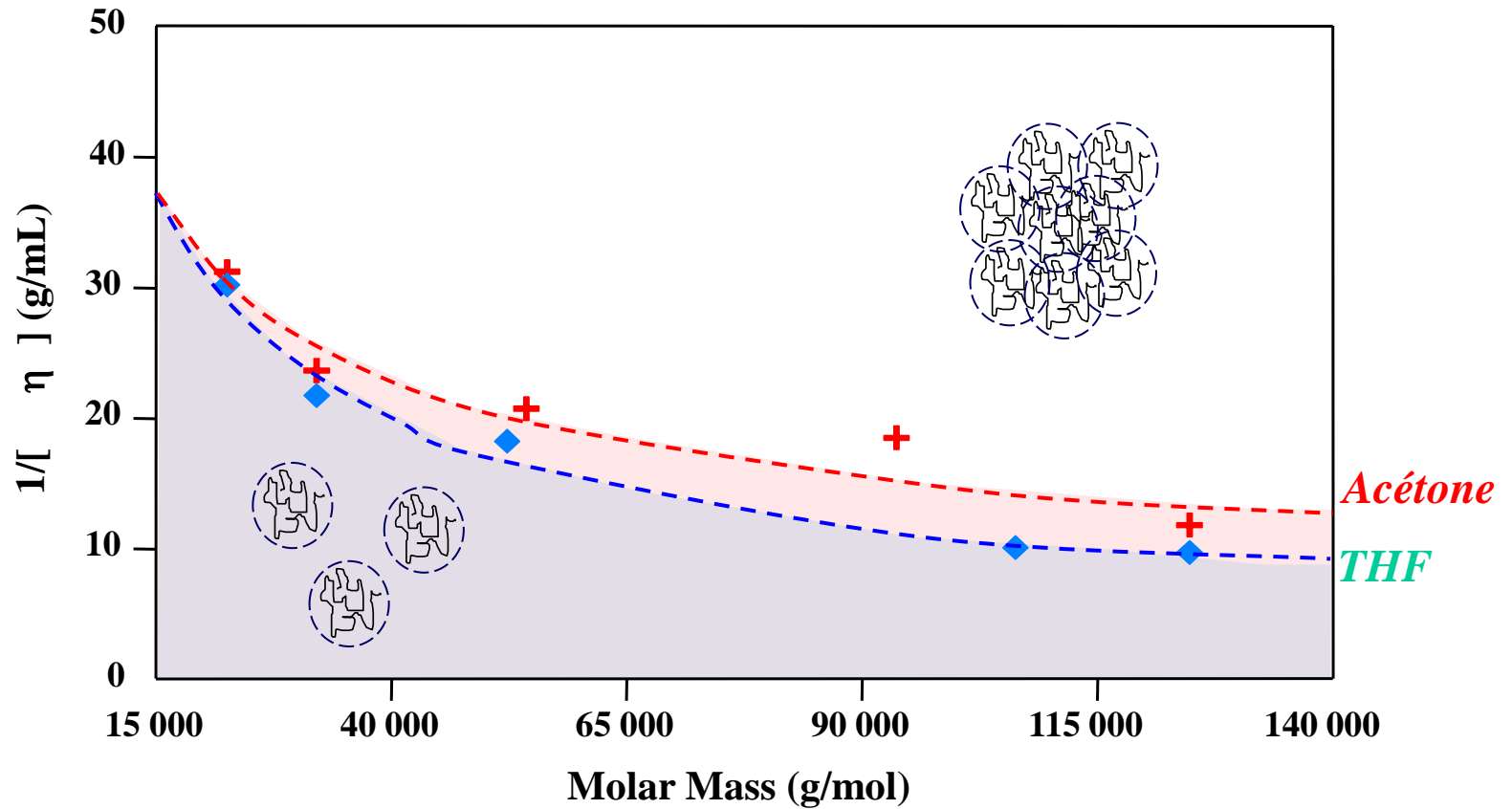


NANOPRECIPITATION ET DIAGRAMME DE PHASE



INFLUENCE DU POIDS MOLECULAIRE ET DE LA VISCOSITE

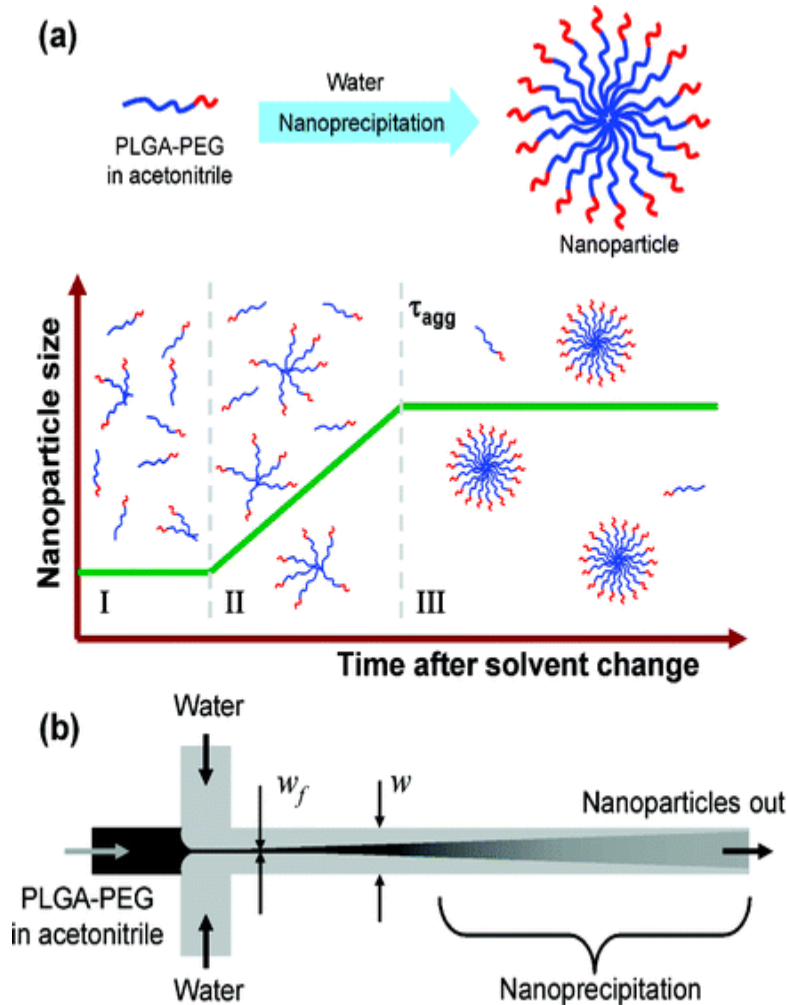
Legrand et al. Int. J. Pharm 2007



Viscosimétrie Capillaire

APPLICATION OF MICROFLUIDICS TO THE PREPARATION OF PLA NANOPARTICLES BY NANOPRECIPITATION

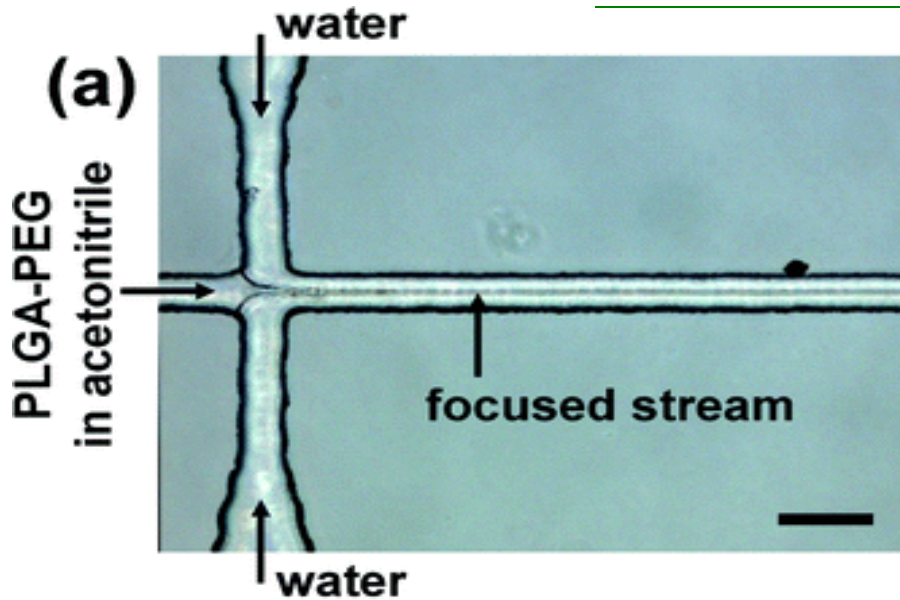
Karnik R et al., Nanoletters, 8, 2906-2912, 2008



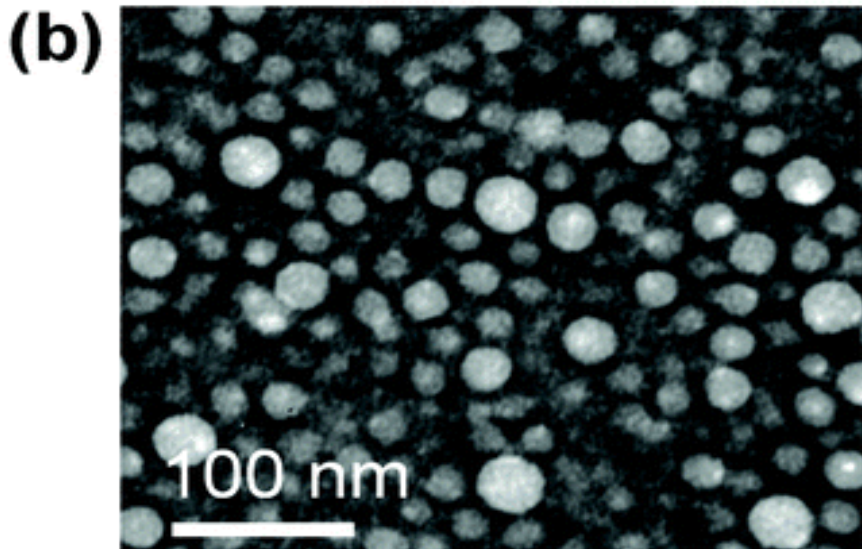
- (a) The process occurs in three stages involving nucleation of nanoparticles, growth through aggregation, and results in kinetically locked nanoparticles after a characteristic aggregation time scale τ_{agg} .
- (b) The process of mixing can be carried out in a microfluidic device using hydrodynamic flow focusing, where the polymer stream is focused into a thin stream between two water streams with higher flow rates.

MICROFLUIDICS FOR THE PREPARATION OF PLA NANOPARTICLES BY NANOPRECIPITATION

Karnik R et al., Nanoletters, 8, 2906-2912, 2008



(a) A microfluidic device for hydrodynamic flow focusing of polymeric nanoparticles in water. Scale bar 50 μm .

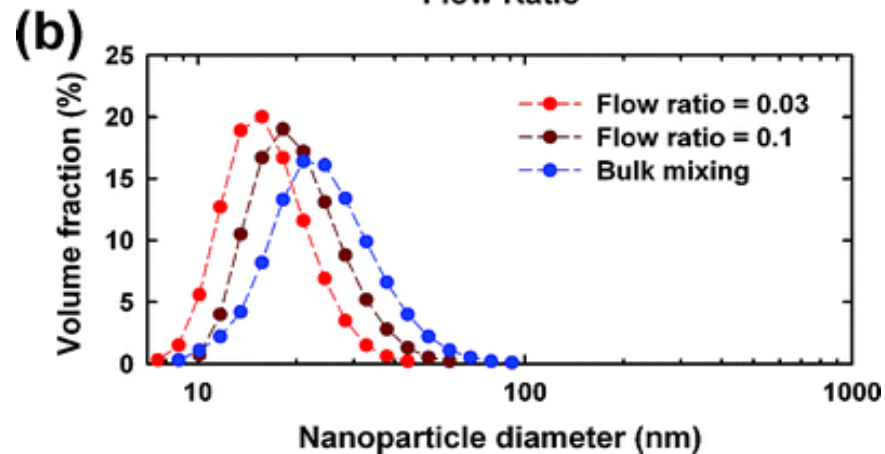
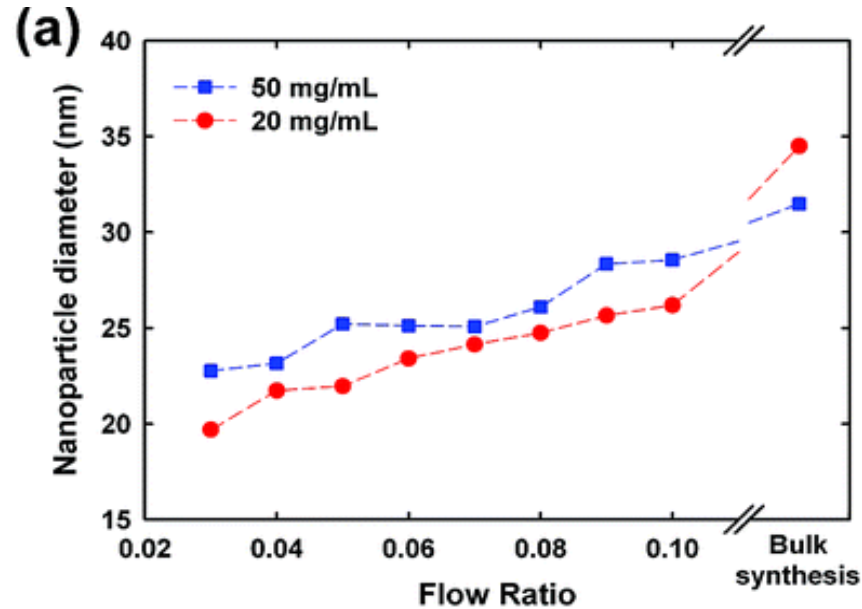


(b) TEM image of nanoparticles synthesized by nanoprecipitation of PLGA15K-PEG3.4K by hydrodynamic flow focusing

EFFECT OF FLOW RATIO ON NANOPARTICLE SIZE

Karnik R et al., Nanoletters, 8, 2906-2912, 2008

$$\text{Flow ratio} = \frac{\text{Flow rate of the acetonitrile polymer solution}}{\text{Flow rate of water}}$$

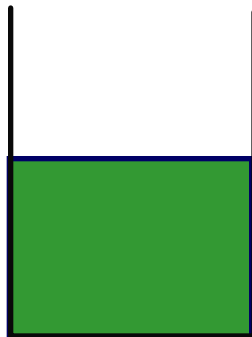


PREPARATION DE NANOPARTICULES A L'AIDE DE METHODES SPECIFIQUES REPOSANT SUR CERTAINES PROPRIETES PARTICULIERES DES POLYMERES

- Gélification
- Complexes d'inclusion
- Formation de complexes électrolytiques (polyplexes)

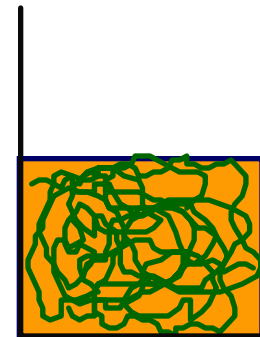
PREPARATION DE NANOPARTICULES PAR GELIFICATION

Repose sur une propriété particulière
de certains polysaccharides



Solution

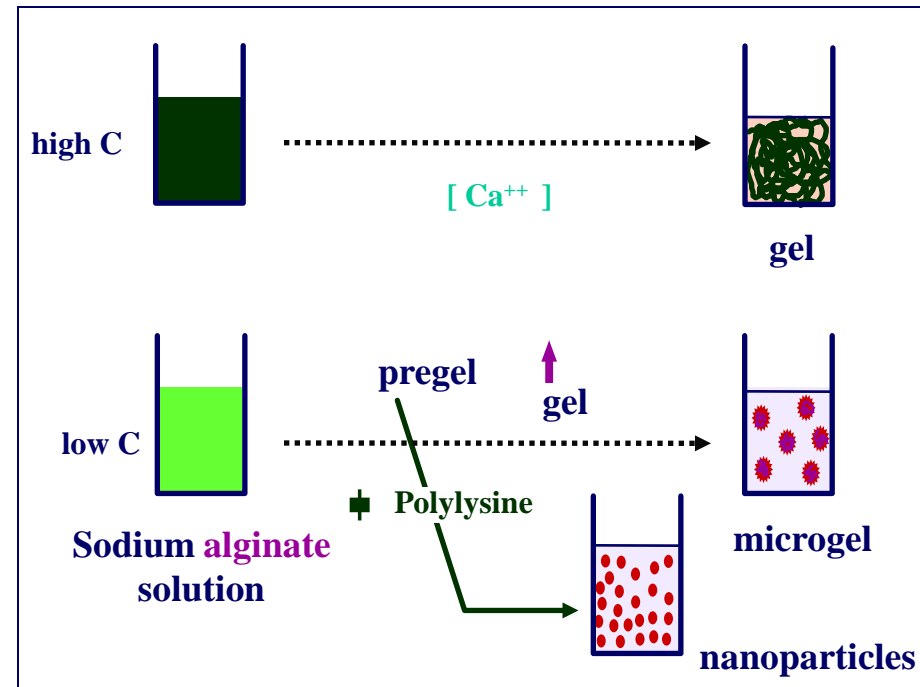
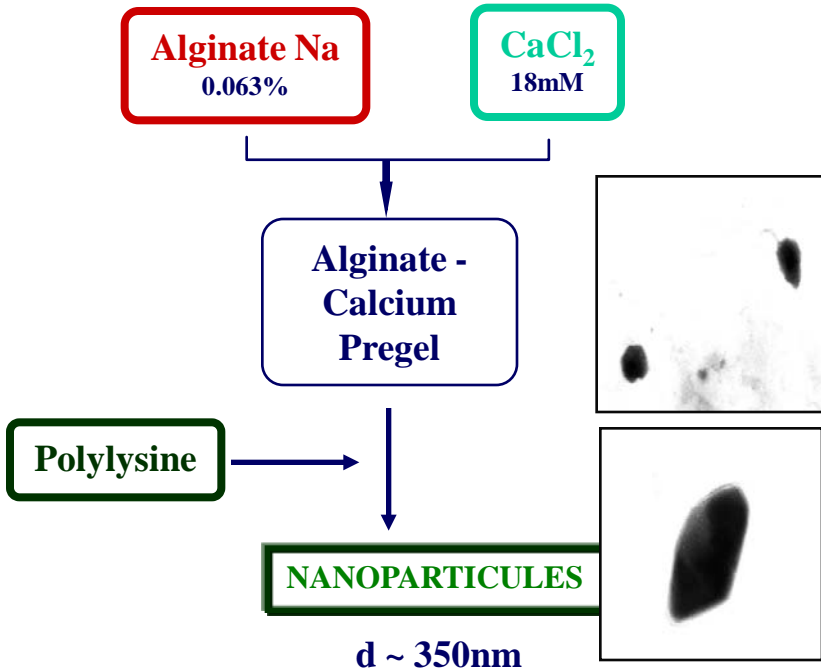
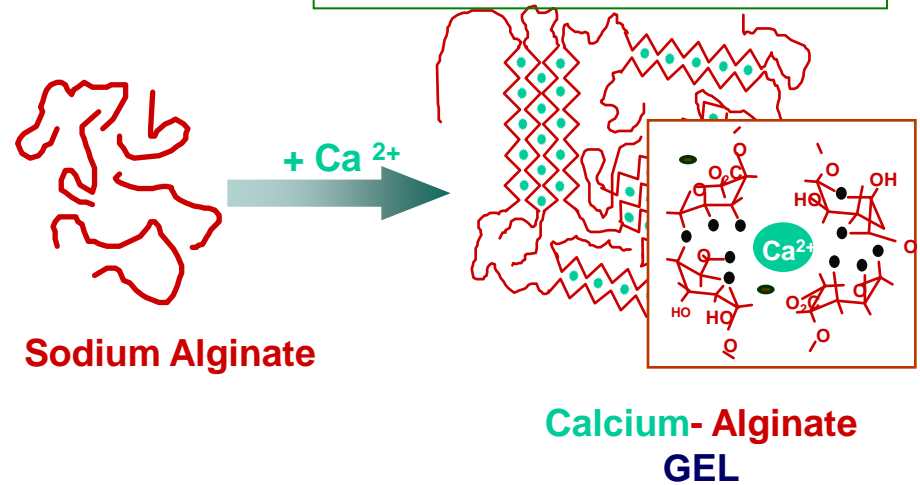
Changement de propriétés
physico-chimiques



Gel

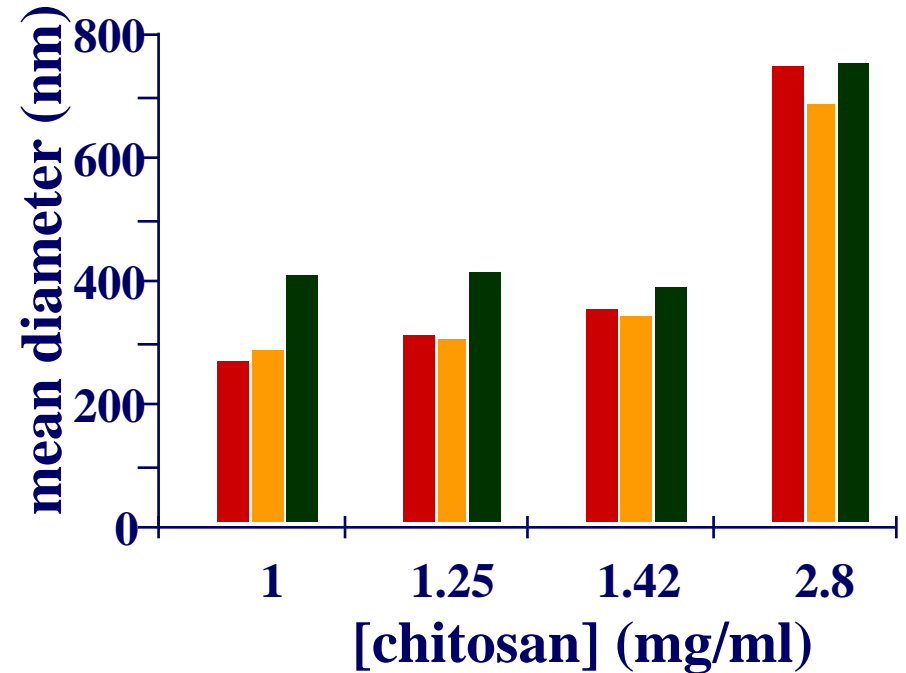
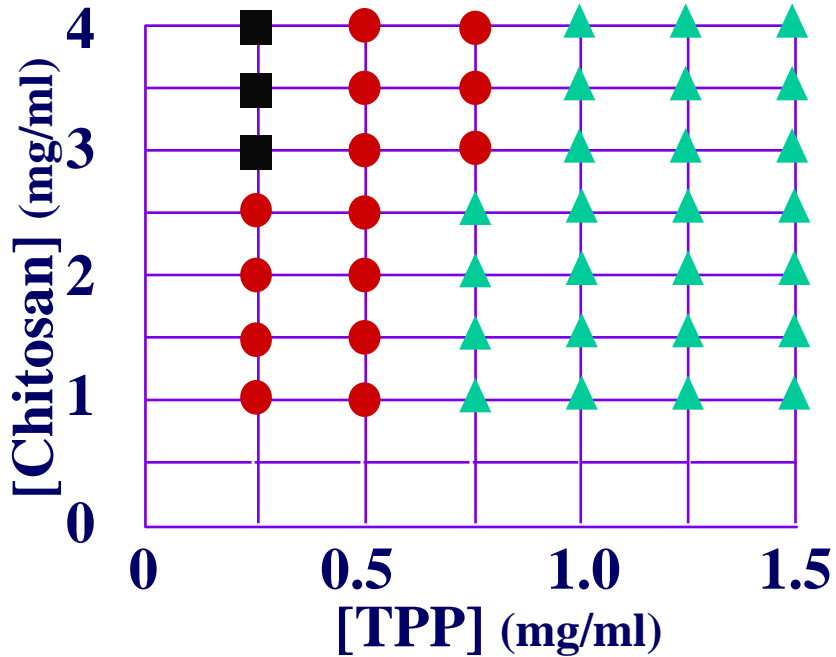
NANOPARTICULES D'ALGINATE

Rajaonarivony et al., J. Pharm. Sci., 1993



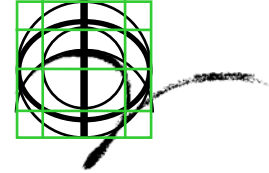
NANOPARTICULES DE CHITOSAN

Calvo et al., J. Applied Polym., 63, 125-132, 1997



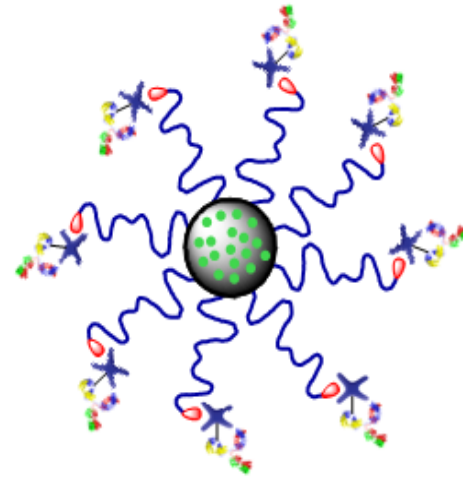
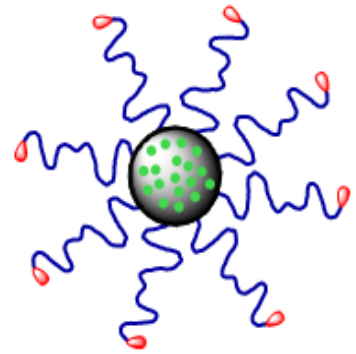
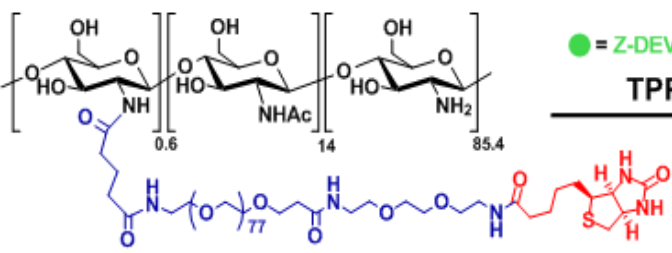
■ solution
● nanoparticles
▲ aggregates

polyphosphate (mg/ml)
■ 0.21 ■ 0.28 ■ 0.43



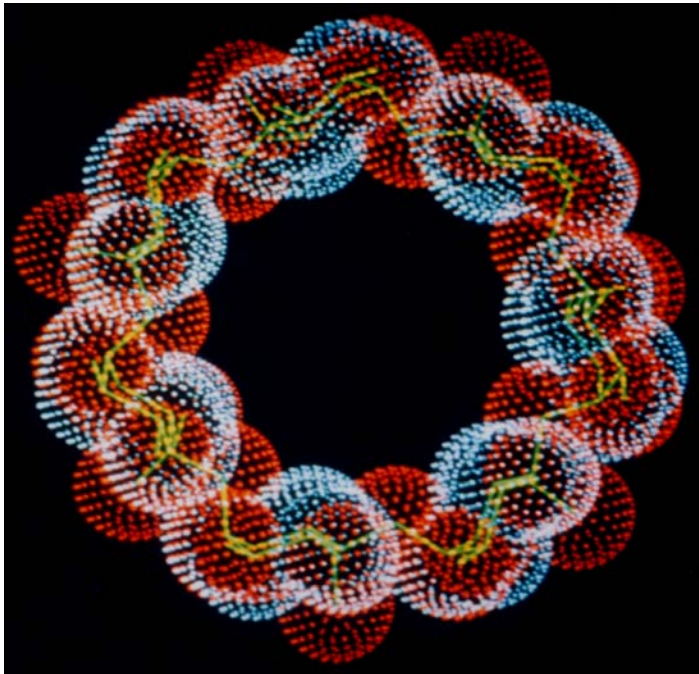
EXEMPLE DE GREFFAGE D'UN ANTICORPS A LA SURFACE DE NANOPARTICULES DE CHITOSAN PAR L'APPROCHE BIOTINE/AVIDINE

Y. Aktaş et al. , Bioconj. Chem., 16, 1503-1511 (2005)



Chitosan-PEG-Biotine

PREPARATION DE NANOPARTICULES PAR AUTO-ASSEMBLAGE

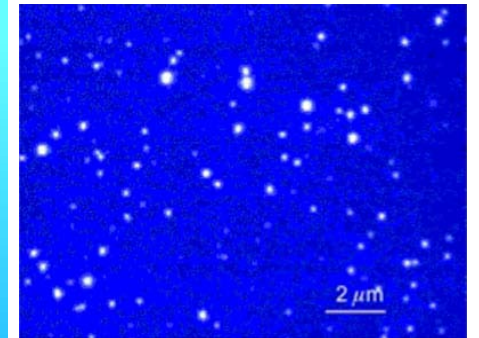
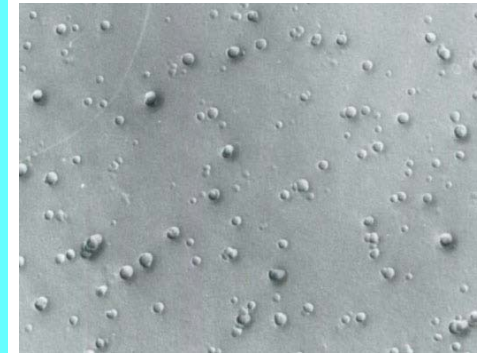
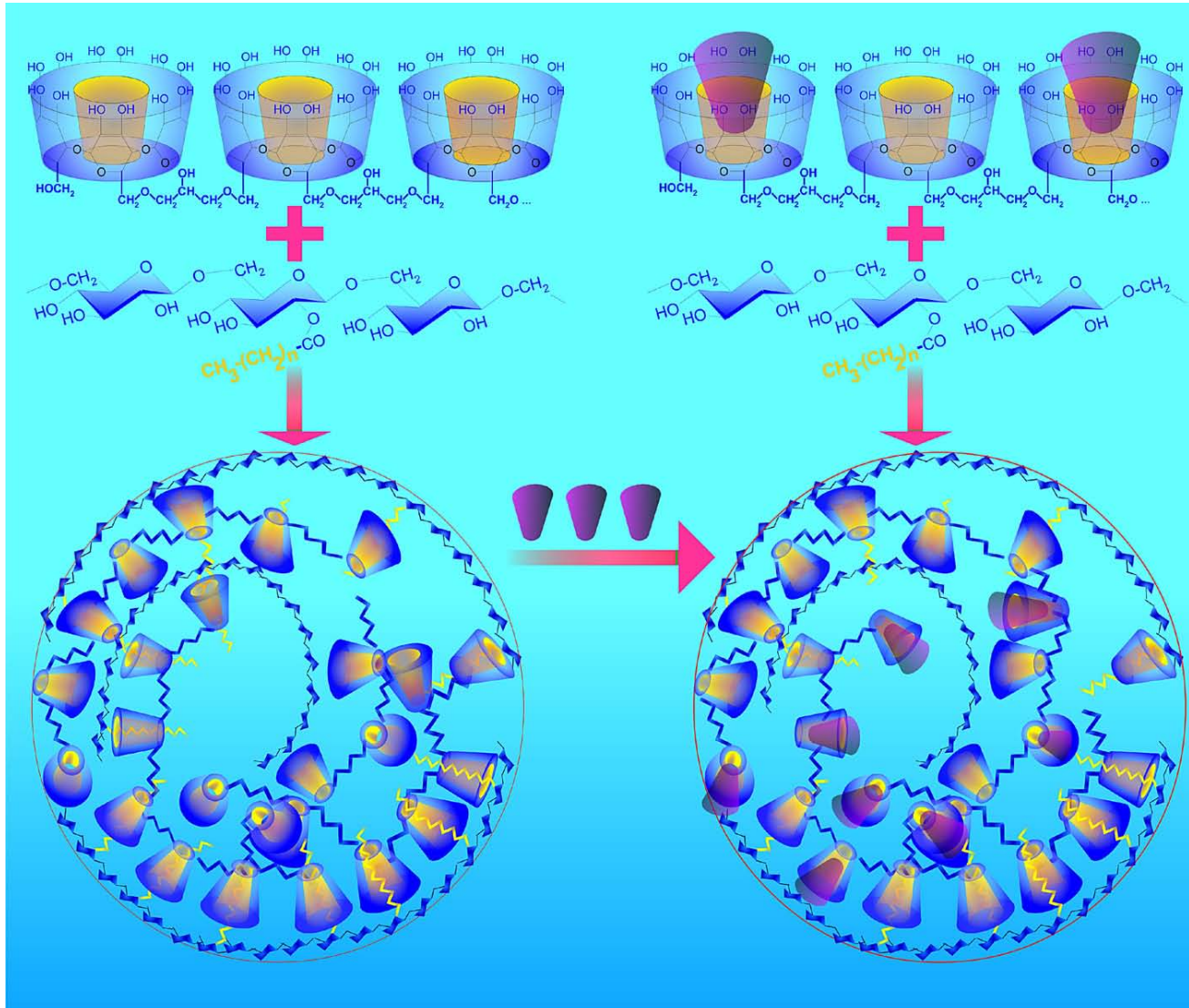


Utilisation des propriétés d'inclusion
des Cyclodextrines

NOVEL NANOTECHNOLOGY BASED ON THE “KEY AND LOCK” CONCEPT

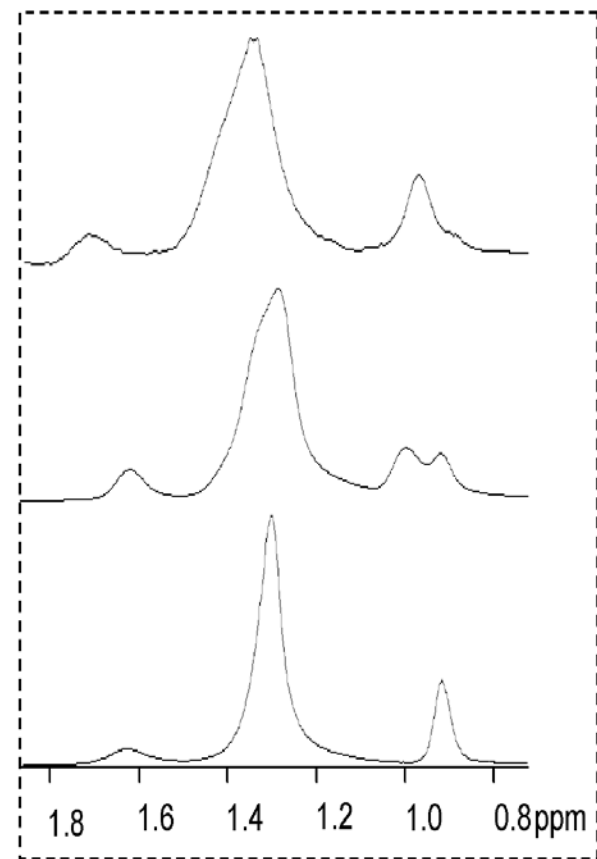
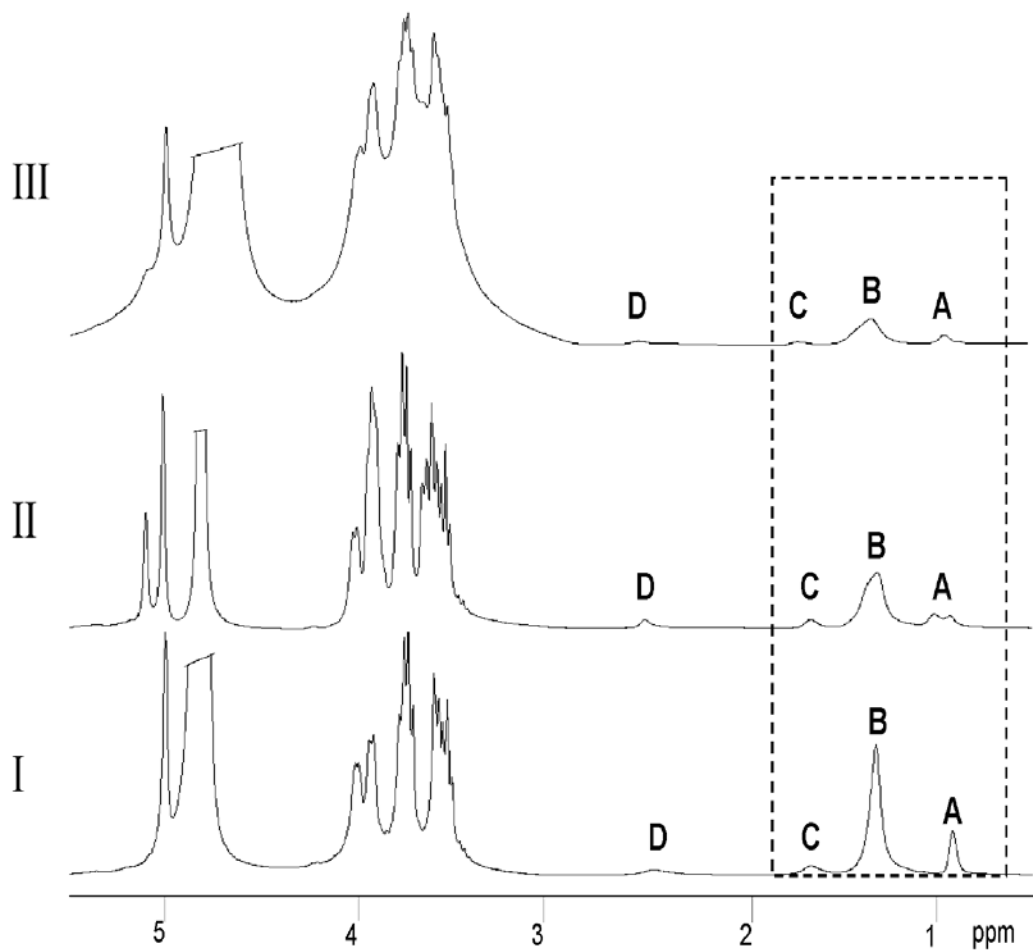
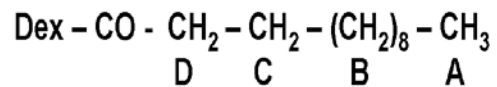
R. Gref et al., French Patent N° 02 08766 (2002)

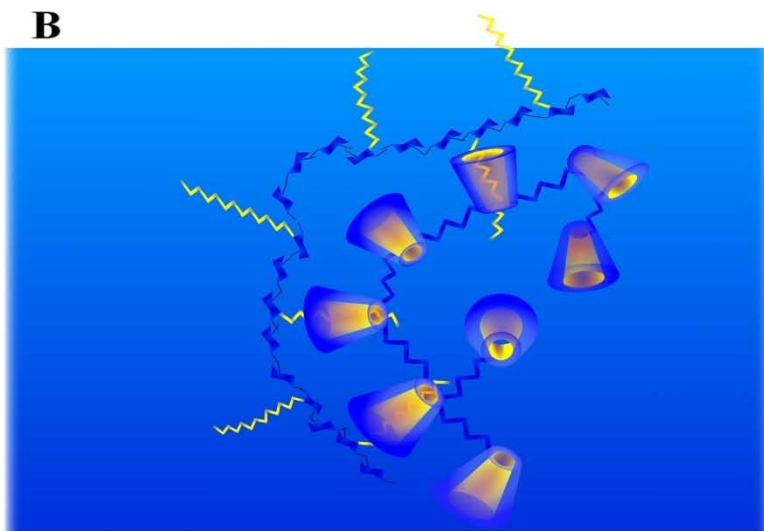
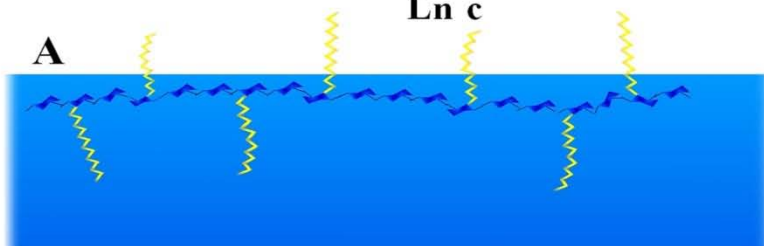
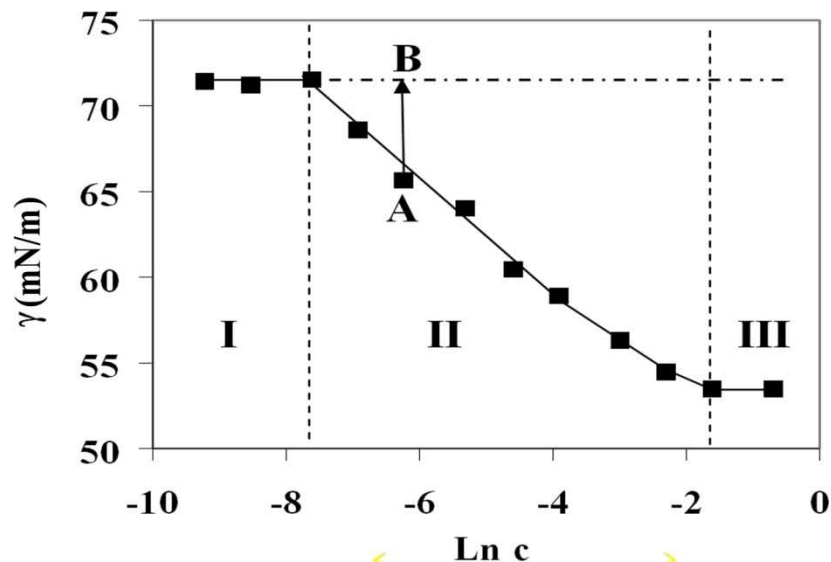
R. Gref et al., J Control. Rel., 2006



^3H -RMN

R. Gref et al., J Control. Rel., 2006





**HYDROPHOBIC
DEXTRAN REDUCES
 γ (air/water)
BUT THE ADDITION
OF
POLY-CD RESTAURE
THE INITIAL VALUE
OF γ (A towards B)**

R. Gref et al., J Control. Rel., 2006

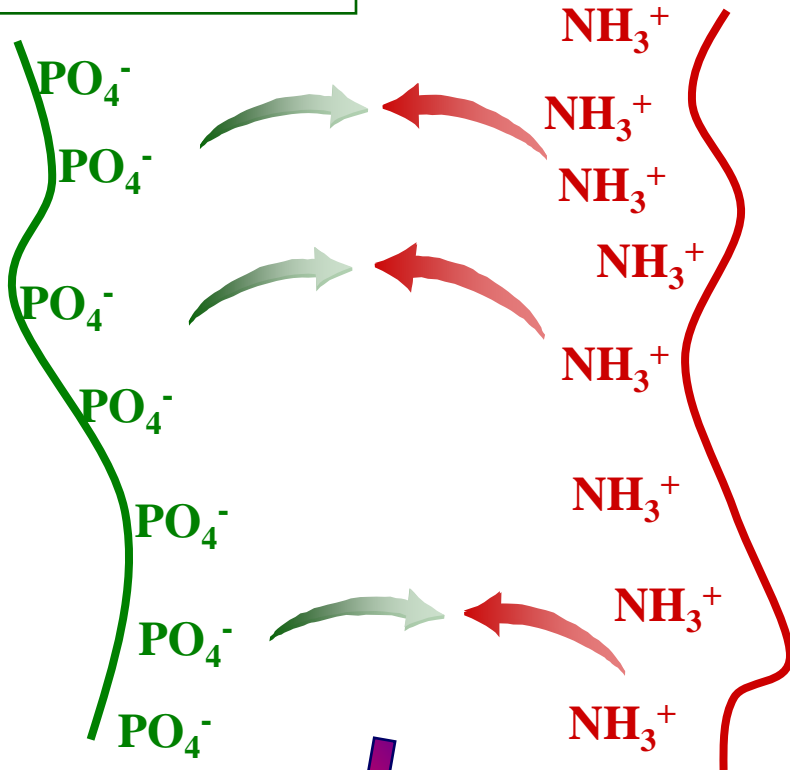
**PREPARATION DE
NANOPARTICULES PAR
INTERACTION ELECTROSTATIQUE**

DNA POLYPLEXES

Dunlap et al., Nucleic. Acid. Res. 25: 3095-3101, 1997

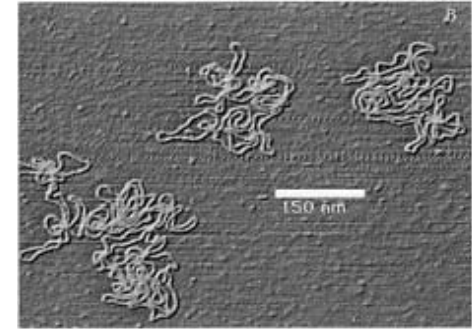
Nucleic acids

Chitosan
Poly(ethyleneimine)
Poly(aminoacid)

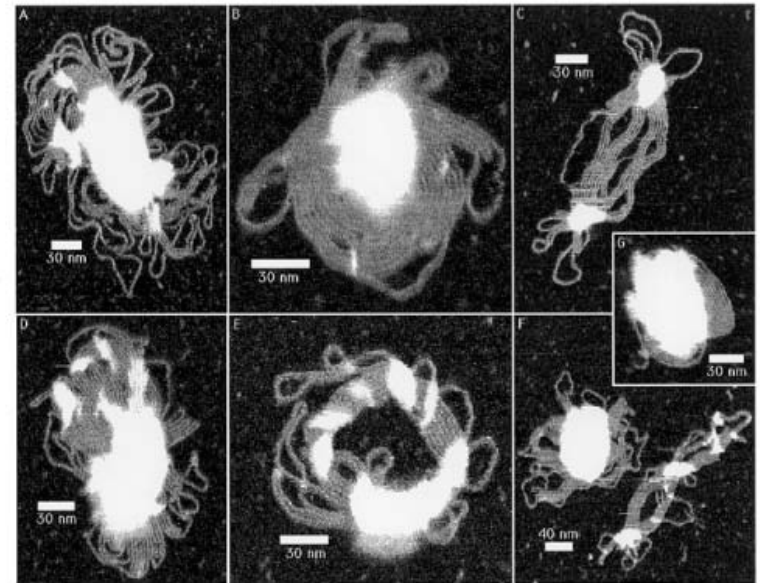


DNA condensation
Colloidal particles
(polyplexes) $d_H \sim 20-80\text{nm}$

AFM



Plasmid



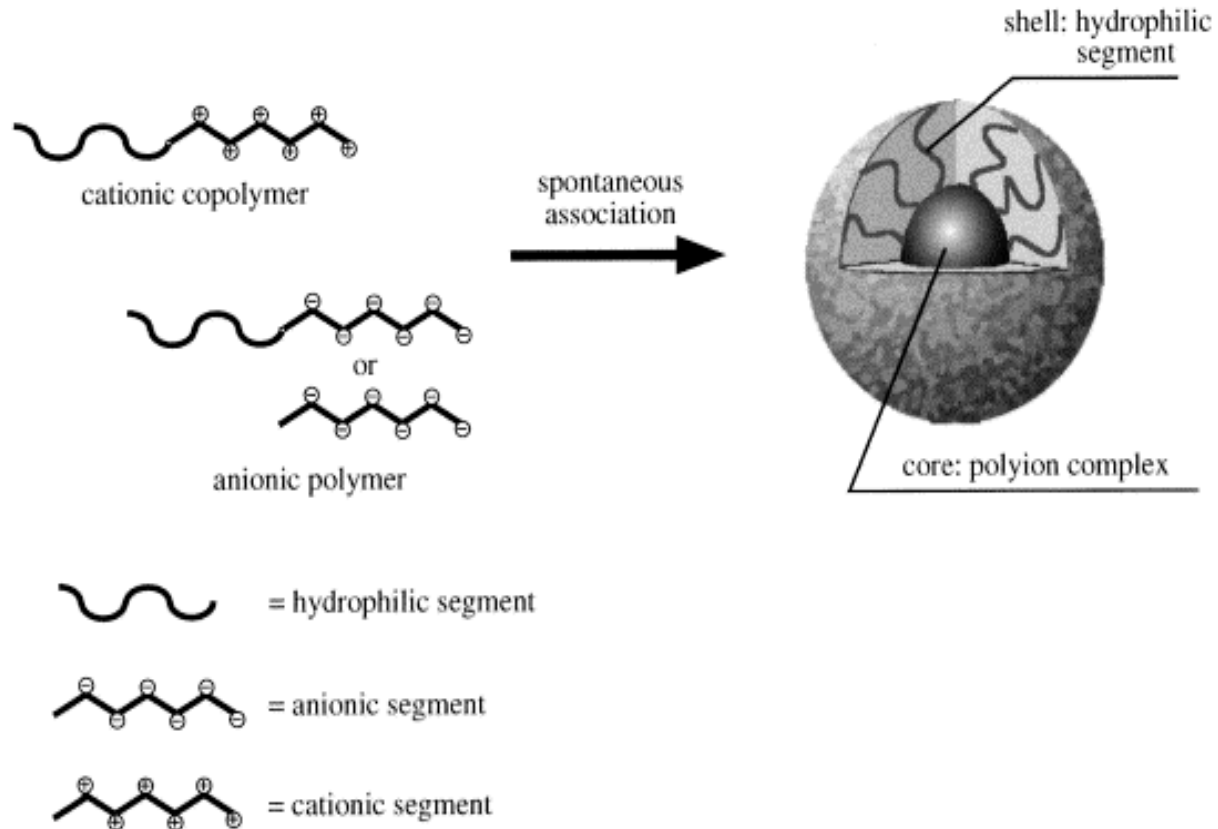
Plasmid condensation with PEI

MICELLES

Concentration Micellaire Critique
CMC

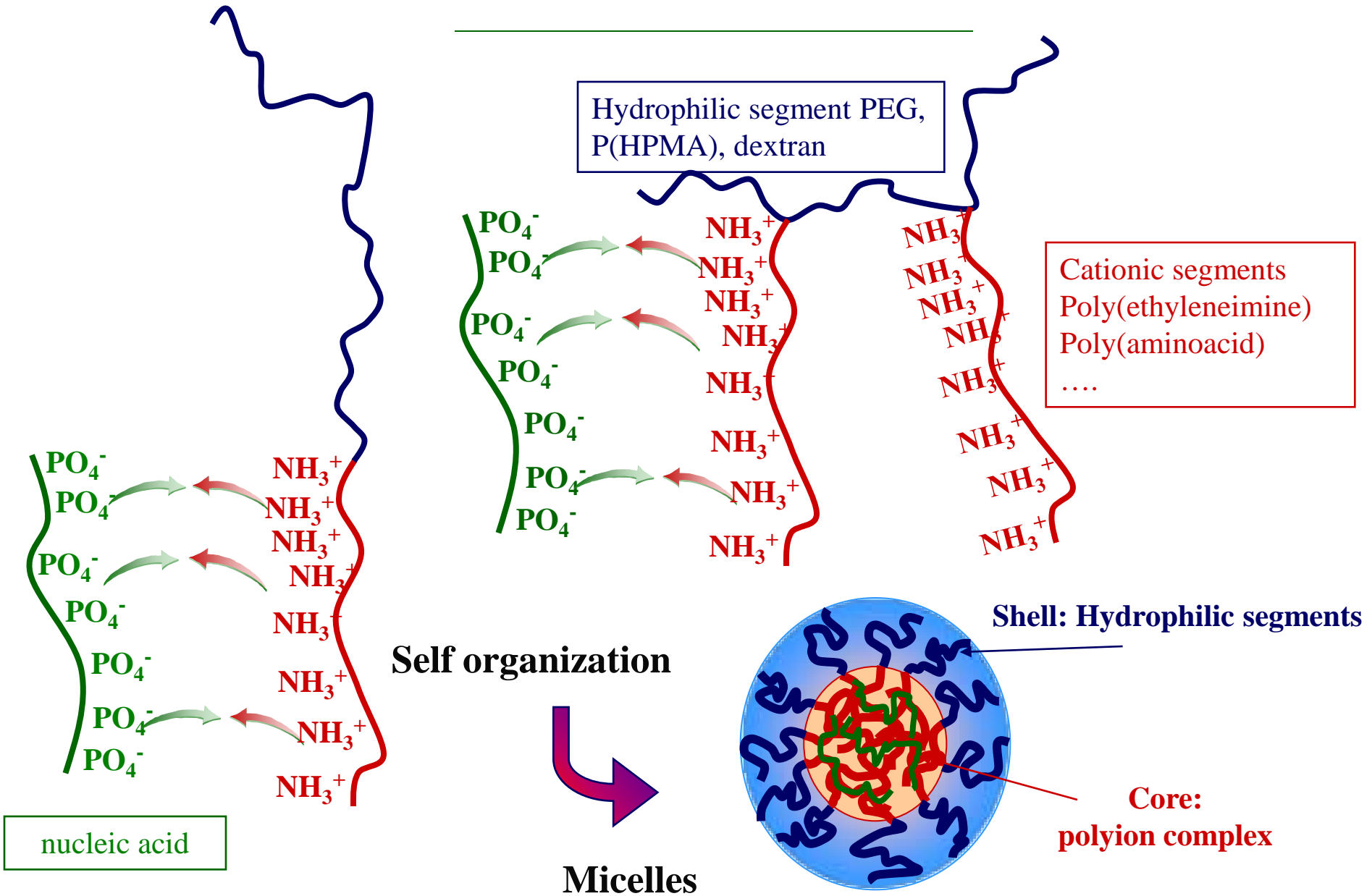
AMPHIPHILIC BLOCK COPOLYMER MICELLES

Kakizawa and Kataoka, Adv. Drug. Deliv. Rev. 54: 203-222, 2002

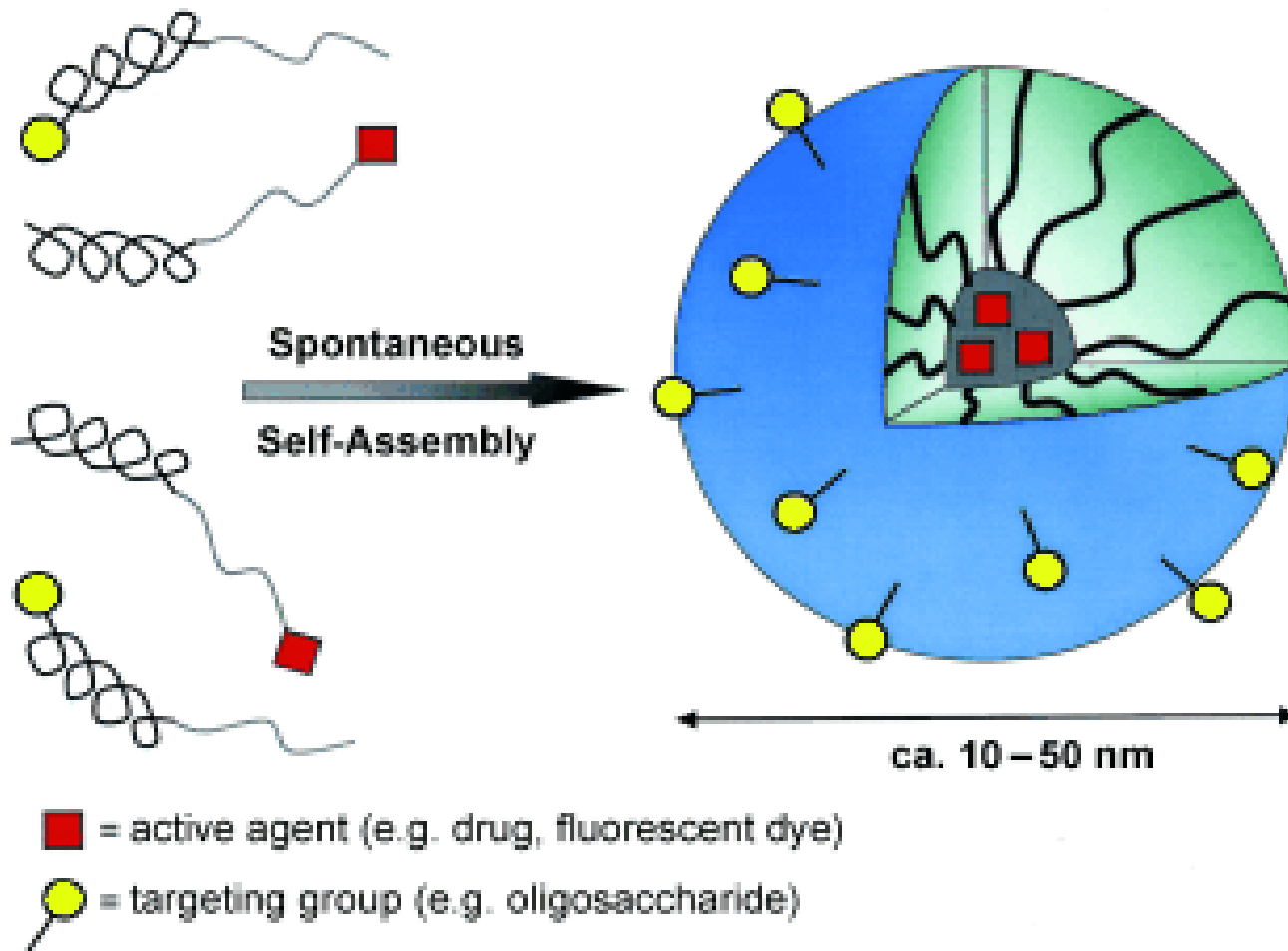


Formation of polyion complex micelles

BLOCK CO-POLYMER MICELLES



ADDRESSED BLOCK CO-POLYMER MICELLES



Typical examples of block copolymers are PEO-*b*-PPO, PEO-*b*-PCI, and PEO-*b*-PAsp.