

# Soft nanofluidics

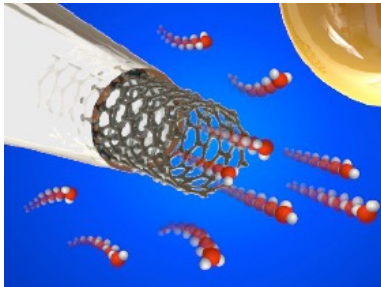
Anne-Laure Bianco

# Sub-nanometric transport in water

Anomalous behaviors reported :

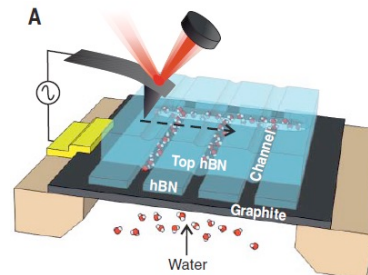
- in confined systems (<1nm)
- in the vicinity of van der Waals materials (graphene, hBN)

## Friction



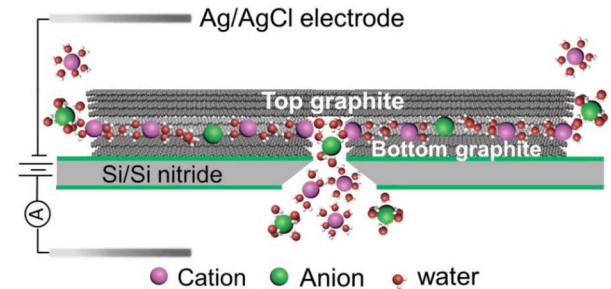
Secchi et al. 2016

## Electric permittivity



Fumagalli et al. 2018

## Ionic transport



Esfandiar et al. 2017

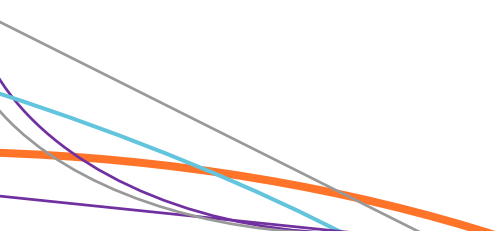
Effect of the confinement and/or of the material?

# Motivations

## Limitations of model experimental systems

- Not versatile (one size, one material)
- Difficult to build from a technological point of view
- Far from real systems for applications

—————▶ **Soft Matter!**



# Soft films as nanofluidic channels

Benjamin Franklin's experiment (1774)



Clapham pond before a drop of oil is added



Clapham pond after a drop of oil  
(with contaminants) is added

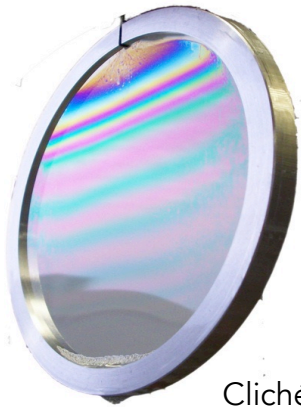
<https://edu.rsc.org/download?ac=11854>

Liquid film of molecular thickness!



# Soft films as nanofluidic channels

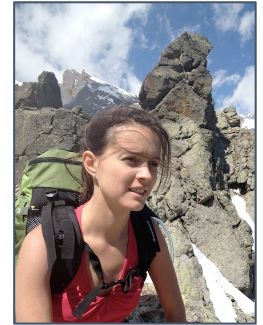
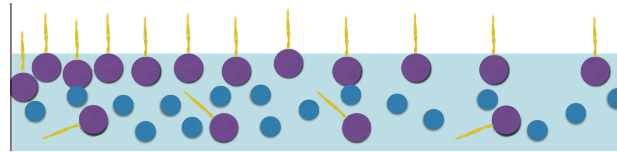
## Soap films



Cliché P. Petit

**Thickness: 5-100 nm**

Complex boundary conditions, deformability



O. Bonhomme

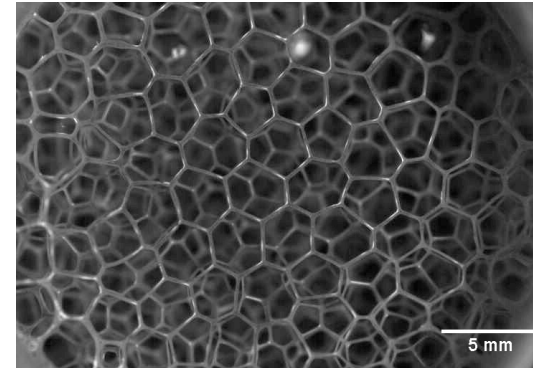
## Condensed films



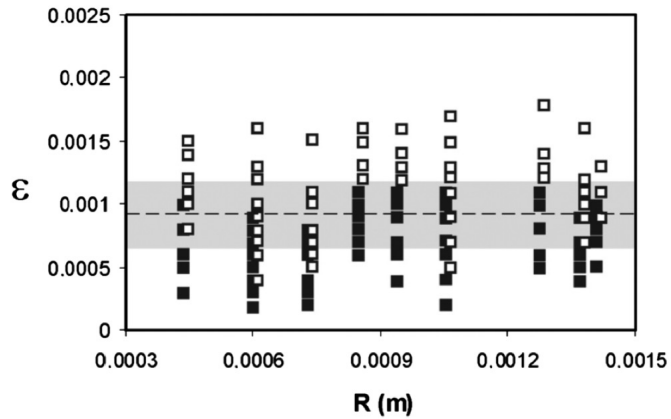
**Thickness: 0.2-3 nm**

Ultimate confinement

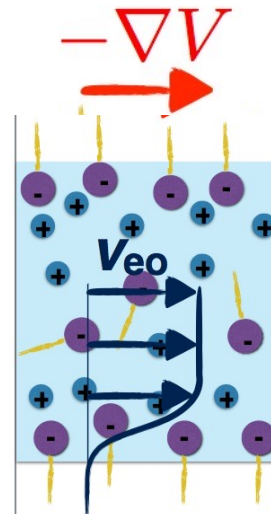
# Soap films: motivations



Stable: wet foams



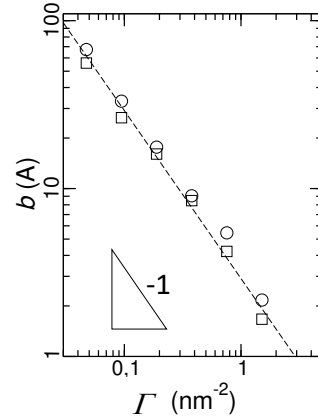
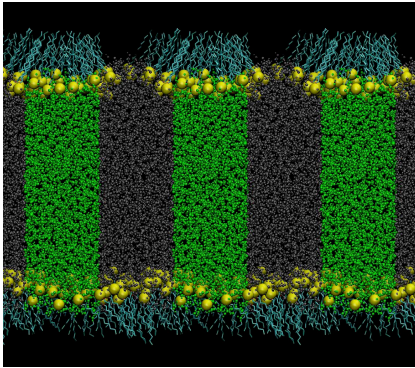
Unstable: dry foams



**Electroosmosis!**

# Complex boundary condition

- **Slippage** depends on surfactant concentration (MD simulations)



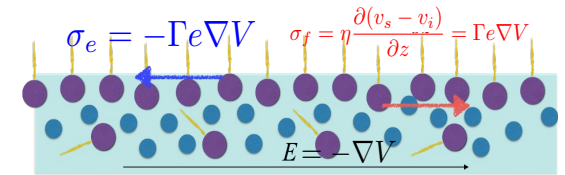
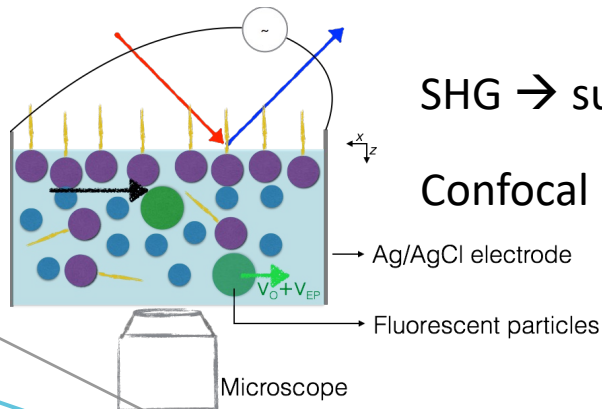
$$\lambda = 3\pi\eta a\Gamma$$

Friction on half-sphere

$$b = \frac{1}{3\pi\Gamma a}$$

Joly, Detchevry, Biance, PRL 2014

- Surfactant **distribution** at interface

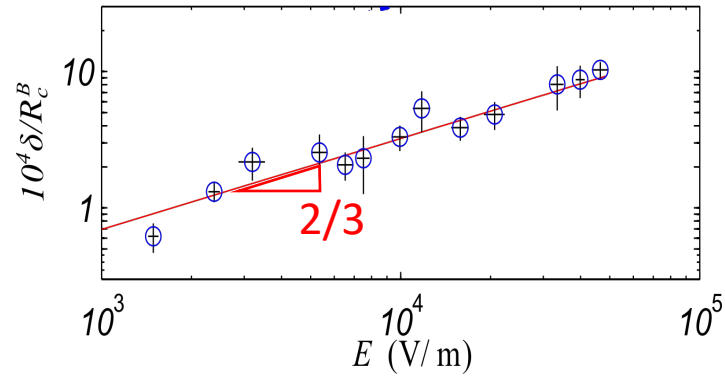
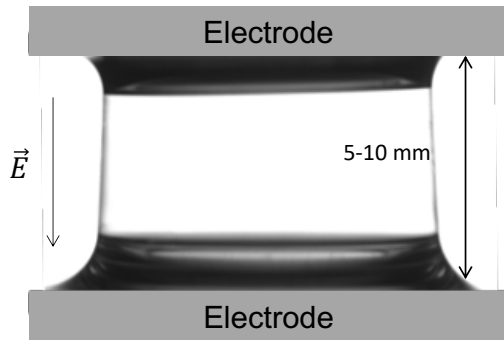


No Marangoni stress,  
no net force on the surfactants

Similar to hydrophobic surfaces,  
Huang et al., PRL 2007

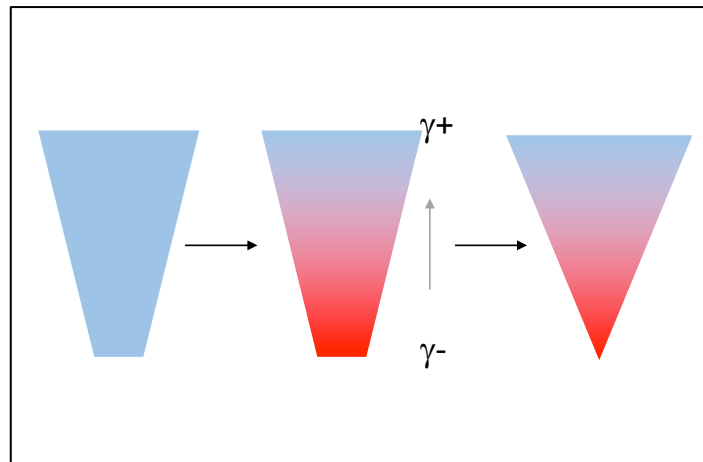
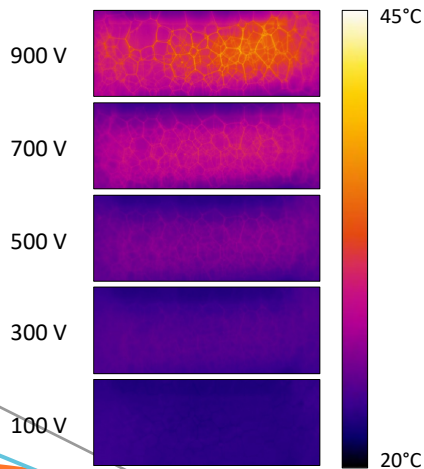
# Soap films and foams

- **Deformable systems – non-linearities**



Bonhomme, Liot, Bianco, Bocquet, PRL 2013

- **Coupling with thermal effects (liquid foams)**



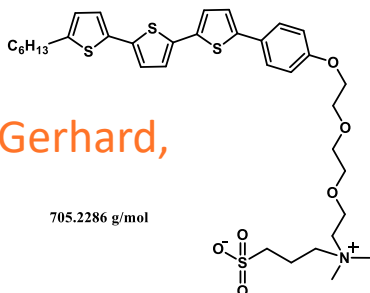
Asymmetry + Joule effect

Enhanced electroosmosis  
(X100)

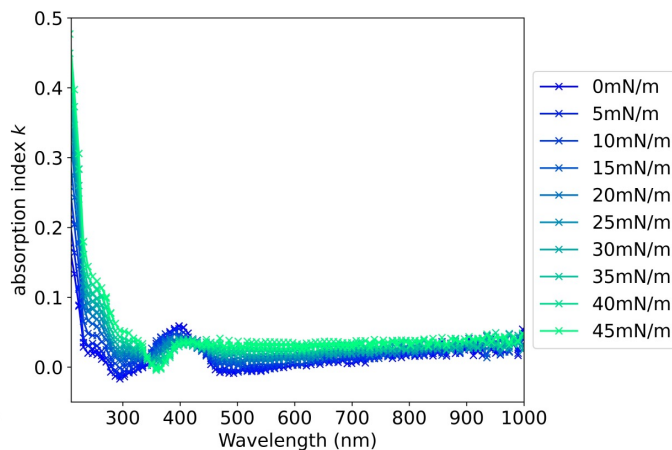
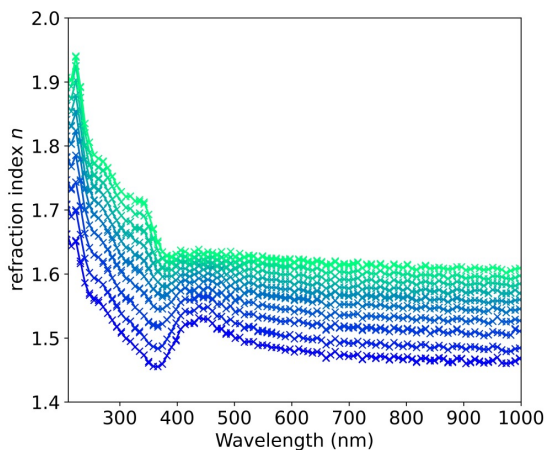
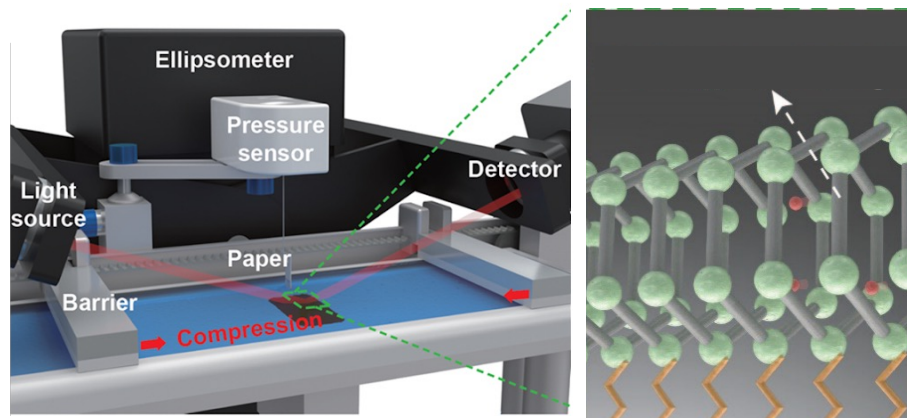
Bonhomme, Peng, Bianco, PRX 2021

- Electronic surfactants**

Isham Idriss,  
Institut Charles Gerhardt,  
Montpellier



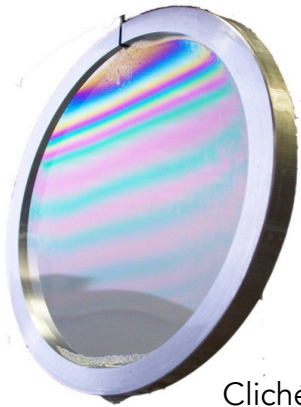
## Spectrometric ellipsometry



Blue shift,  
 $\pi$ - $\pi$  stacking,  
electronic transport?

# Soft films as nanofluidic channels

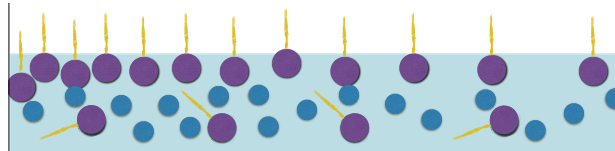
## Soap films



Cliché P. Petit

**Thickness: 5-100 nm**

Complex boundary conditions, deformability



## Condensed films

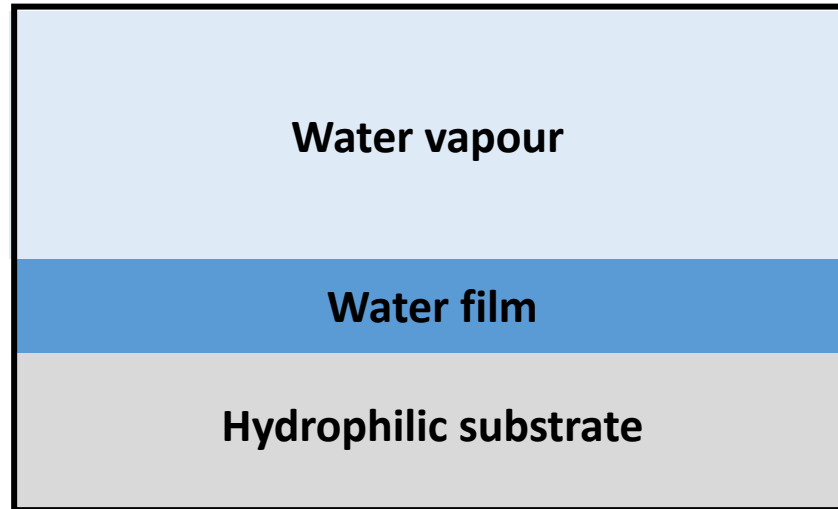


**Thickness: 0.2-3 nm**

Breakdown of continuous description:  
Conductance measurements



## Condensed film on a substrate



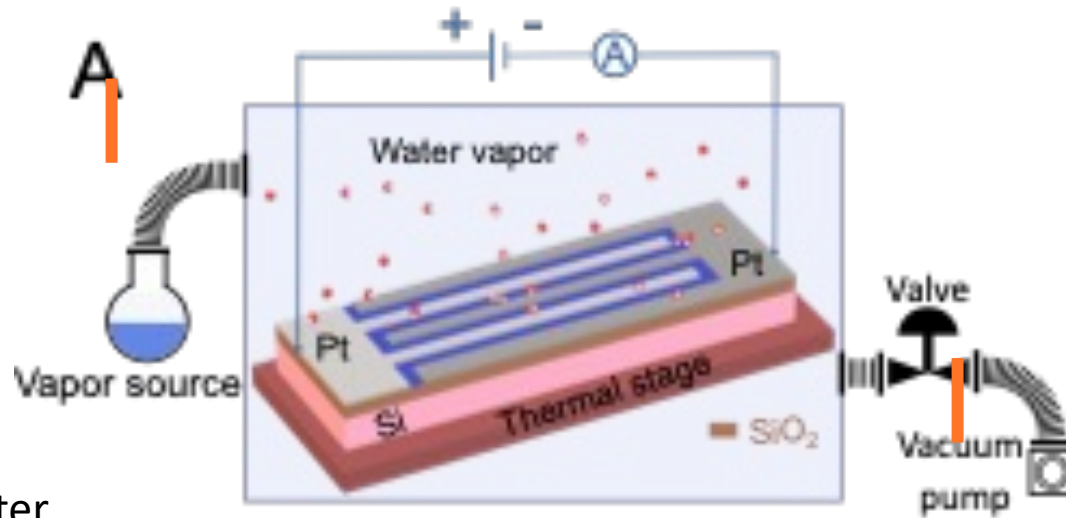
Spreading parameter

$$S = \gamma_{SV} - (\gamma_{SL} + \gamma_{LV}) \geq 0$$

# Humidity control

2: introduction of vapour at pressure

$$p_{vap} = p_{sat}(T_{source})$$



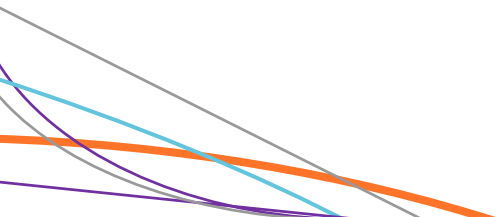
1: vacuum

Thermostated water

$T_{source}$

3: variation of substrate temperature

$$RH[\%] = \frac{p_{vap}}{p_{sat}(T_{sub})} \times 100$$





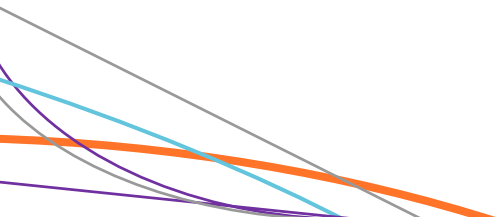
# Condensed film experiments: set-up

Temperature controller



Thermostated  
bath

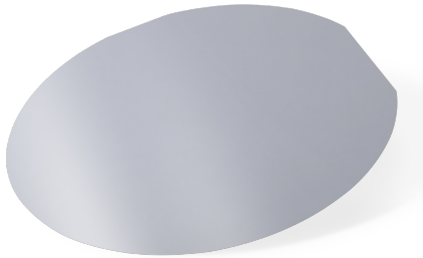
Vacuum pump



# Condensed film thickness: ellipsometry

## Material = Silica

- Silicon wafer with deposited layer of  $\text{SiO}_2$



## Cleaning

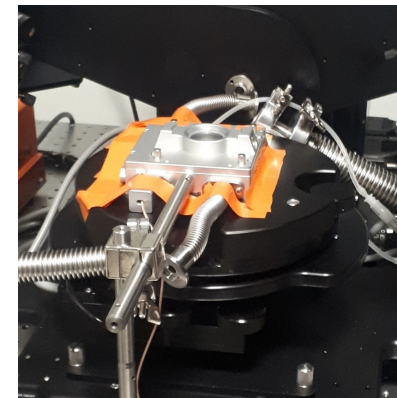
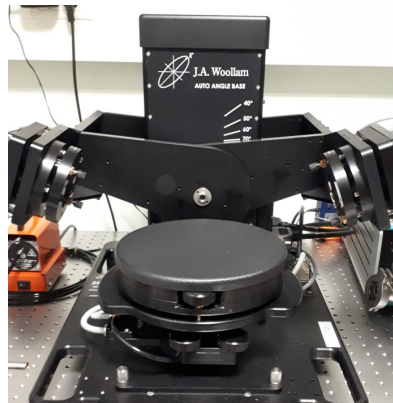
### *Substrate*

- Detergent + ultrasound
- 3 min UV Ozone plasma

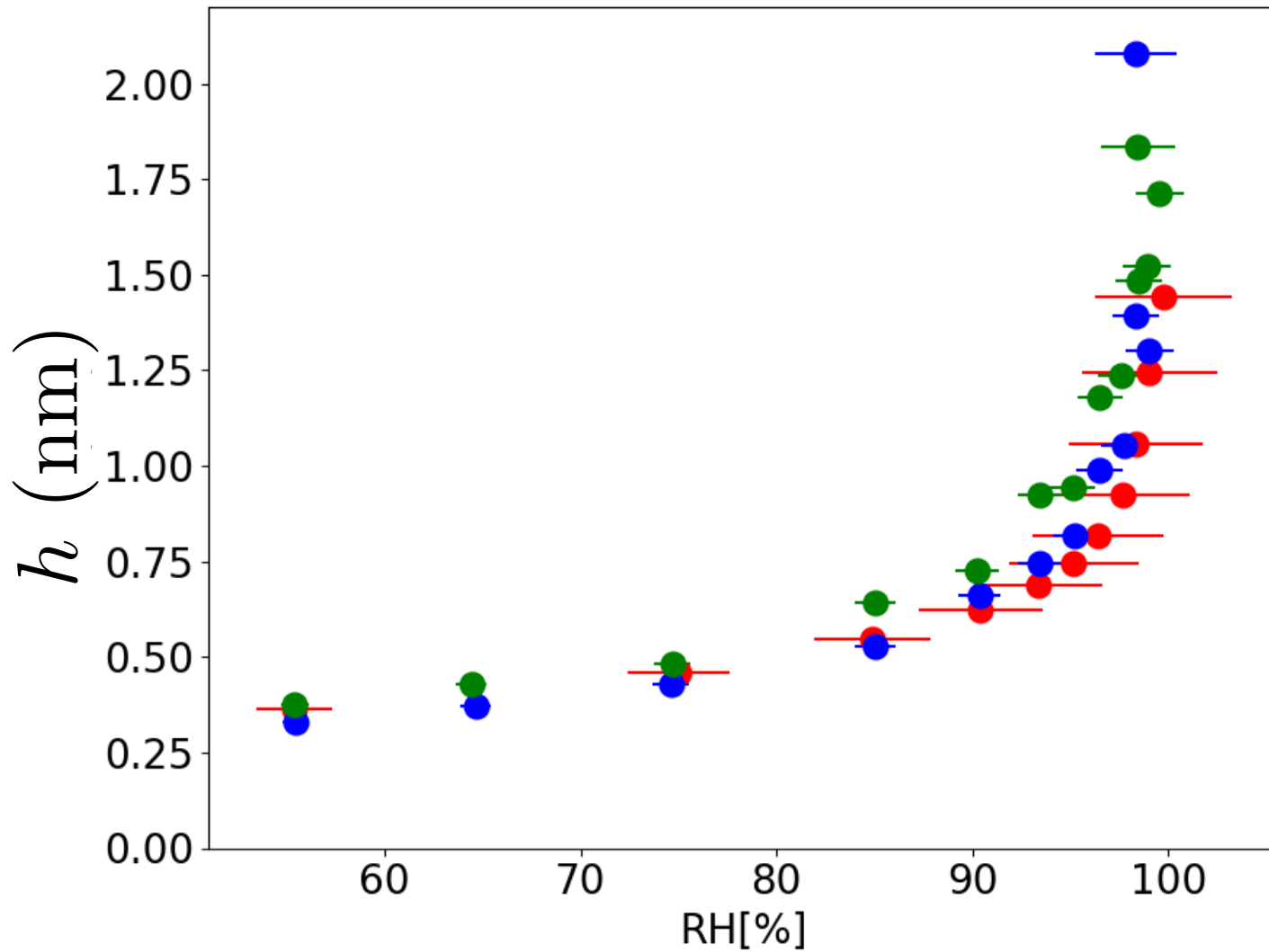
### *Cell and set-up*

- Vacuum (10 Pa) during one week to remove dusts

## Thickness measurements: Ellispometry



# Film thickness



Control of humidity  $\rightarrow$  control of film thickness

# Liquid-substrate interactions

- Film thickness set by the **interaction energy (disjoining pressure)**

$$\pi(h) = \rho_l kT \log(\text{RH})$$

- Disjoining pressure origins

**Wetting: statics and dynamics,**  
**P. G. de Gennes, Rev. Mod. Phys. 57, 827, 1985**

-Electrostatics (counter-ions only)

D. Andelman, Soft condensed matter physics in  
molecular and cell biology, 6, 2006

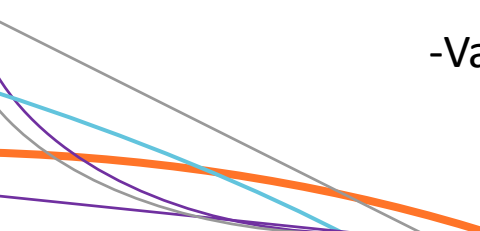
-Adsorption layer model

R. Pashley, Journal of colloids Int. Science, 78, 1980

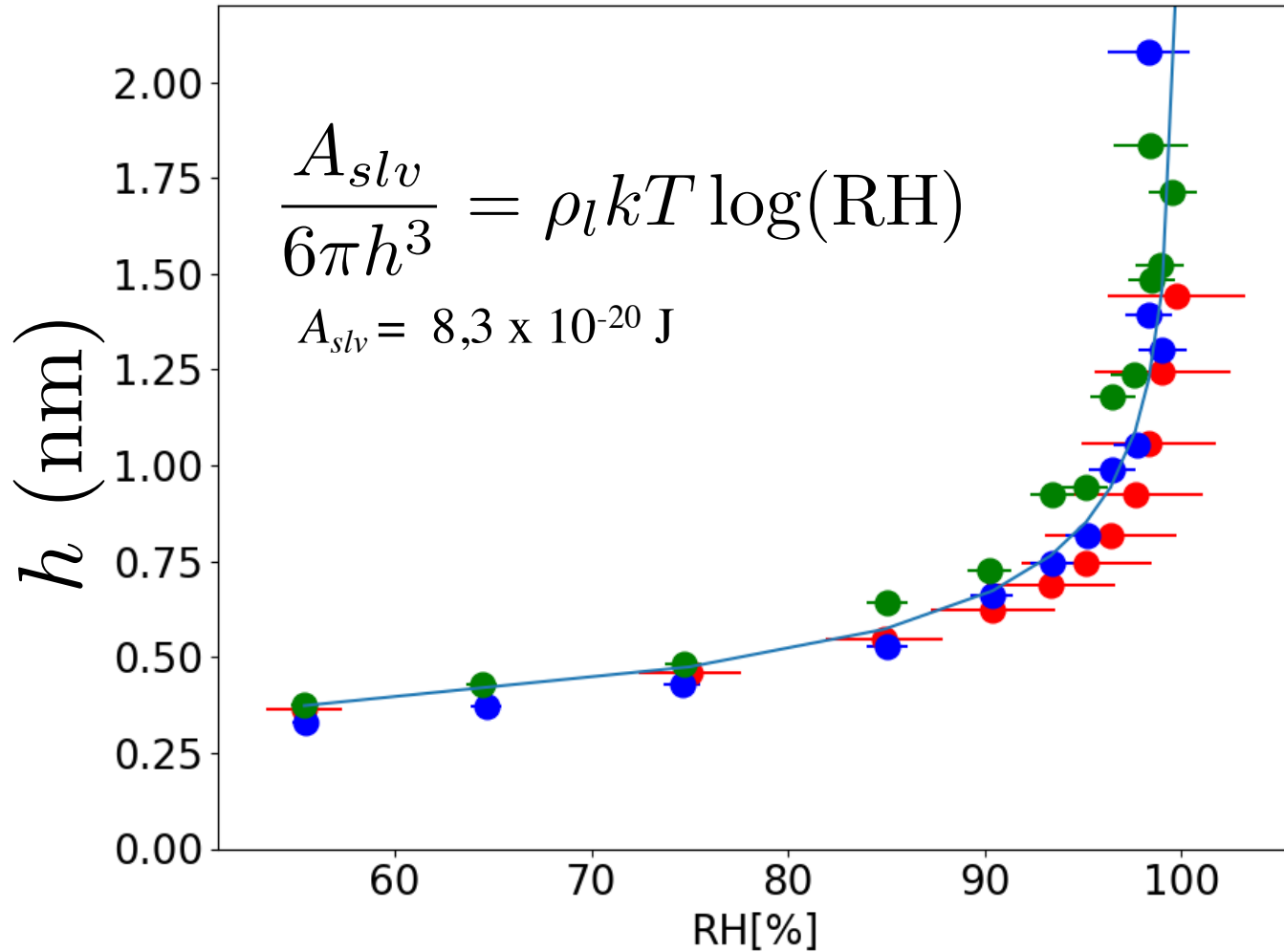
$$\left. \begin{array}{l} \text{ } \\ \text{ } \\ \text{ } \end{array} \right\} \pi(h) \sim \frac{1}{h}$$

-Van der Waals

$$\pi(h) = \frac{A_{slv}}{6\pi h^3}$$

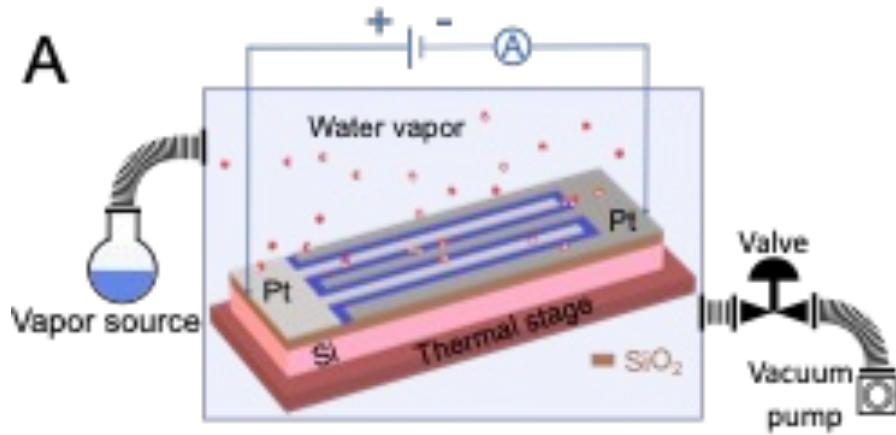


# Film thickness: discussion

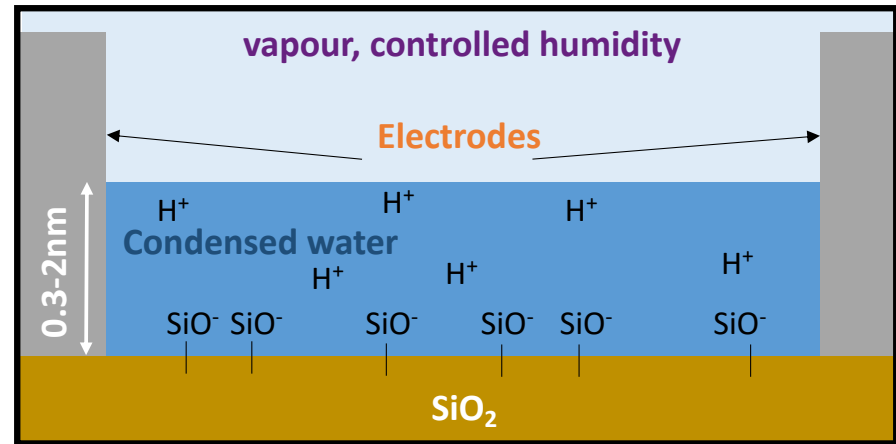
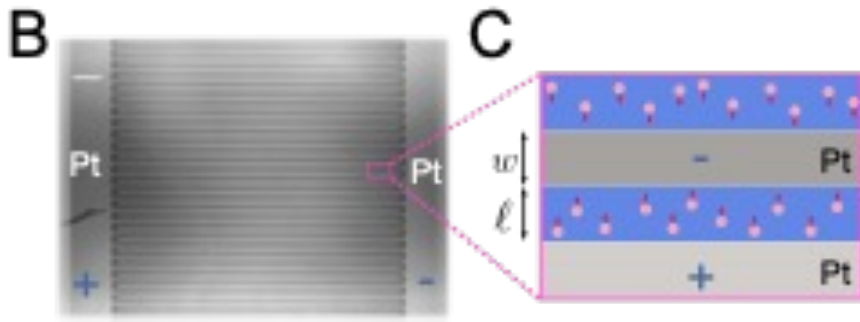


Calculated value of the silica/water/air Hamaker constant:  $1-5 \times 10^{-20} \text{ J}$

# Electrical measurements

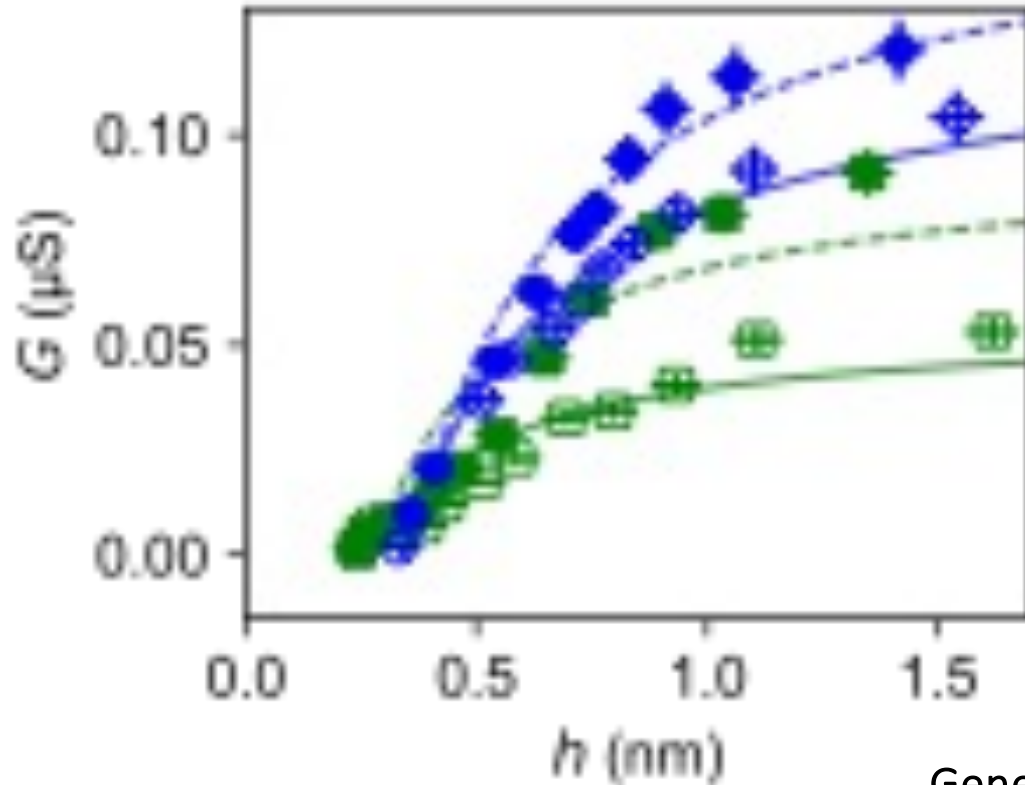


- $L_{tot} = 78$  cm
- Applied AC voltage, 10Hz, 1V



Side view

# Conductance as a function of film thickness



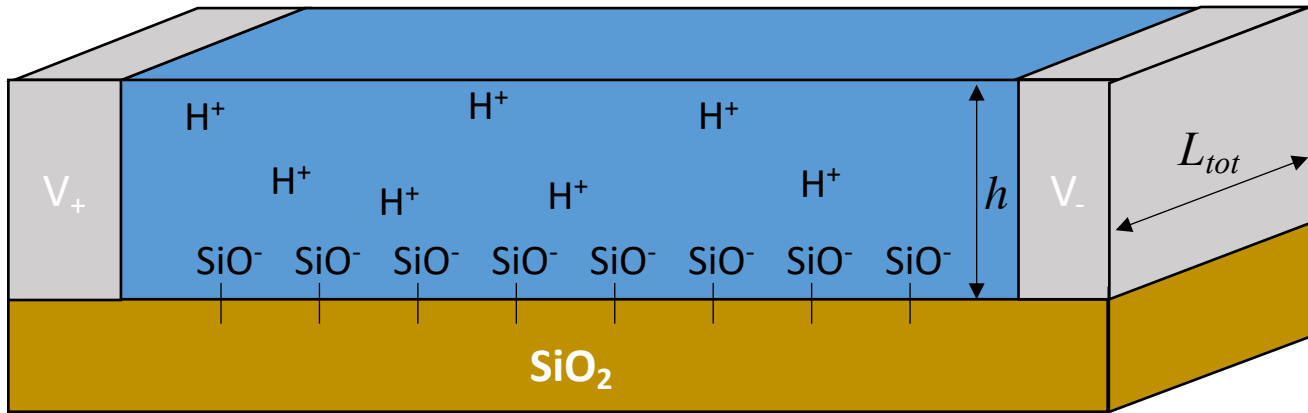
$T_{\text{source}} = 6^\circ\text{C}$

$T_{\text{source}} = 16^\circ\text{C}$

Generic Features:

- Non-linear Increase with  $h$
- Conductance saturation

# Conductance: continuous regime



Hypotheses:

- (i) Zero charge at the water/vapor interface
- (ii) Ionic transport due to counter-ions ( $\text{H}^+$ )

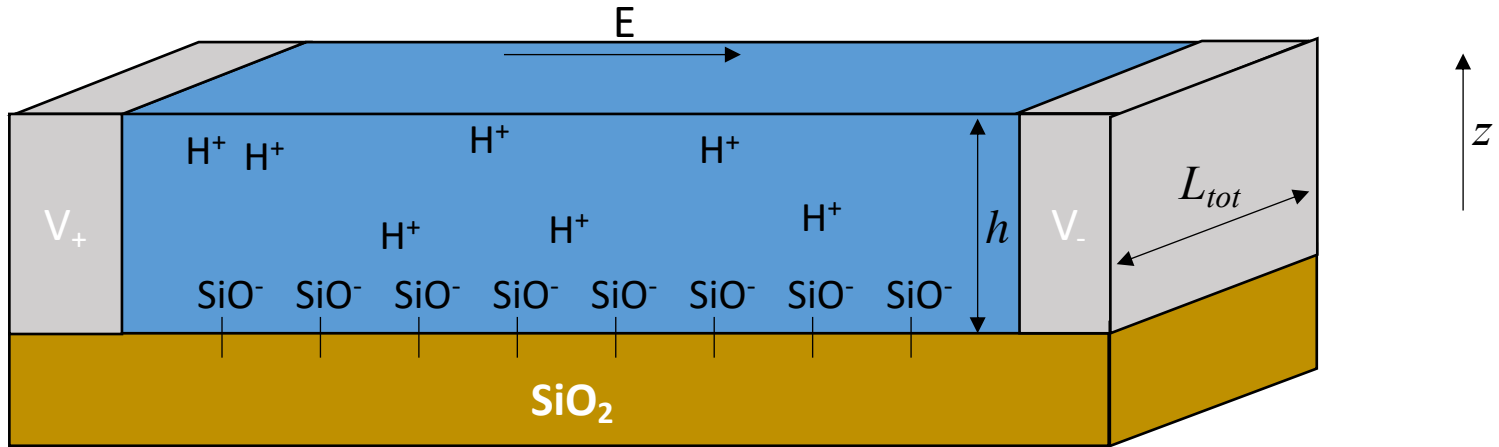
$$I = e \times L_{tot} \int_0^h \rho_+(z) u_+(z) dz$$

$$\rho_+(z)?$$

$$u_+(z)?$$



# Conductance: continuous regime



$$I = e \times L_{tot} \int_0^h \rho_+(z) u_+(z) dz$$

$$u_+(z) \simeq e\mu_+ E$$

+

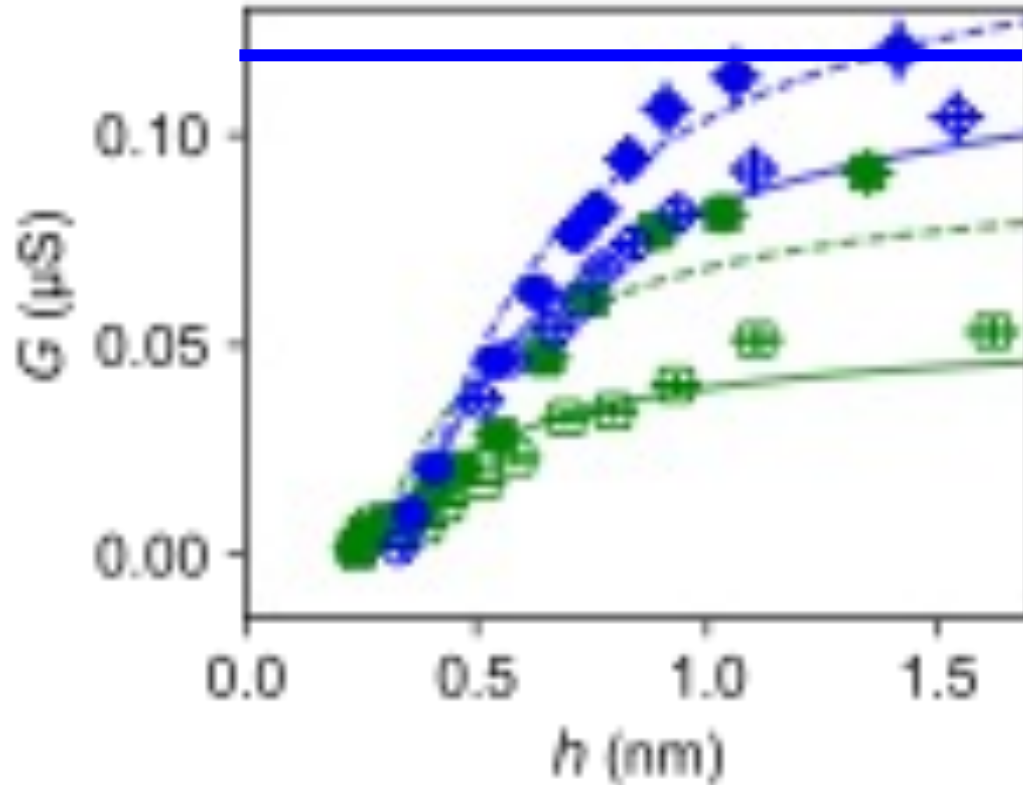
Electroneutrality

$$\int_0^h \rho(z) dz = [\text{SiO}^-]_s$$

$$I \simeq eL_{tot} \times [\text{SiO}^-]_s \times e\mu_+ E$$

(Independent of  $h$  if  $[\text{SiO}^-]_s$  is constant)

## Conductance: continuous regime

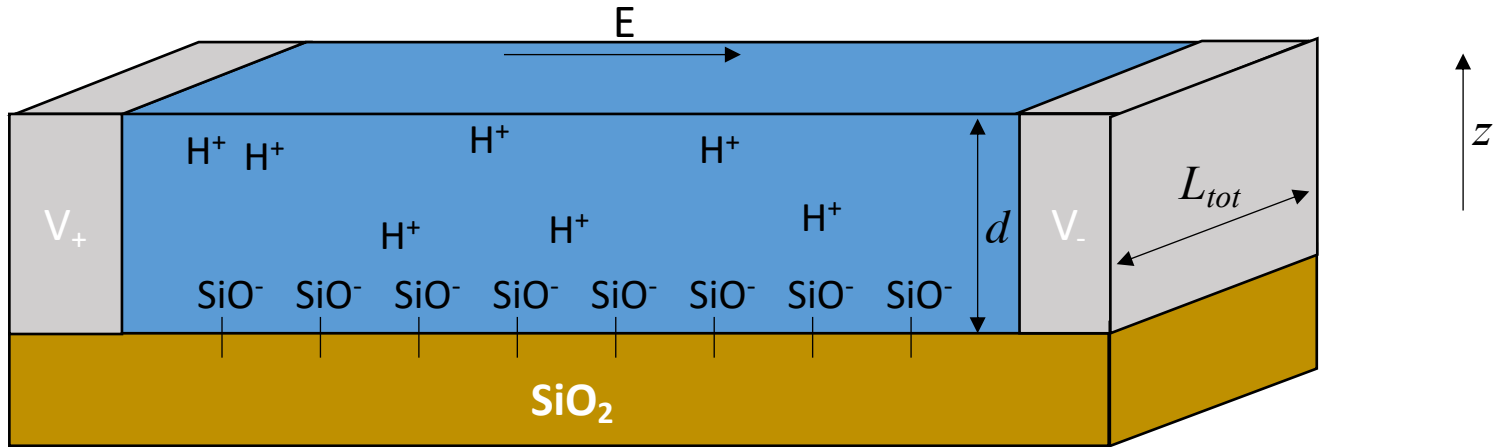


# constant charge

- constant potential
  - charge regulation
- Not satisfying*

Increase with  $h$  in the confined regime?

# #Hypothesis: effect of surface friction

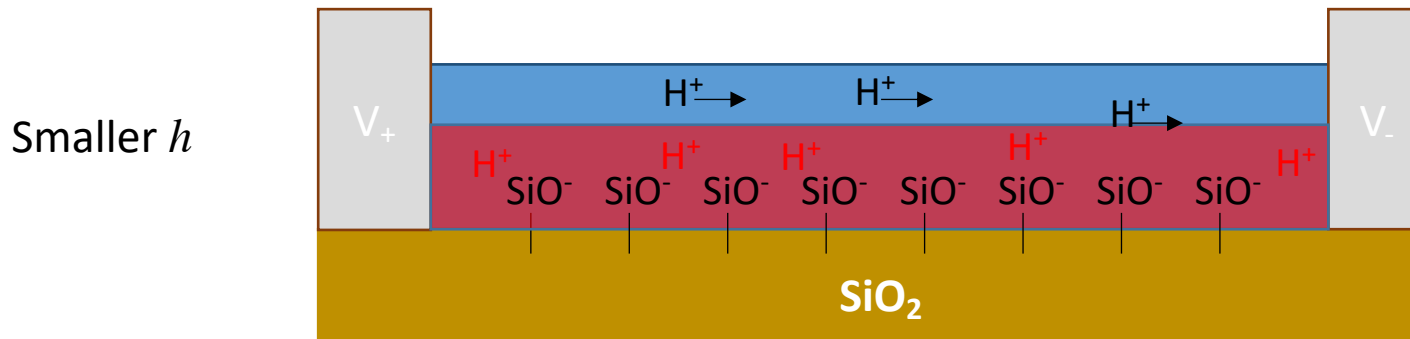
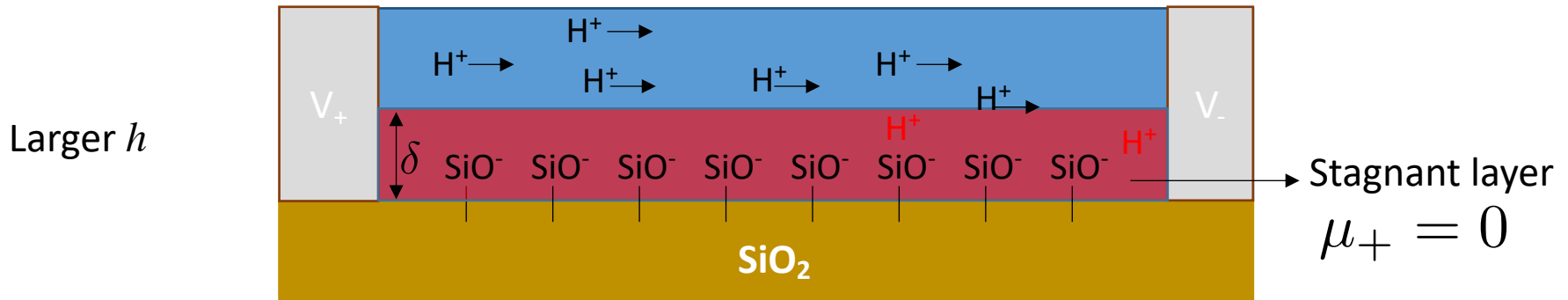


$$I = e \times L_{tot} \int_0^h \rho_+(z) u_+(z) dz$$

$$u_+(z) \simeq e \mu_+(z) E$$

**Stagnant layer of ions?**

# Naïve model: chemical layer with immobile ions



Stagnant layer  $\rightarrow$  conductance increases with  $h$

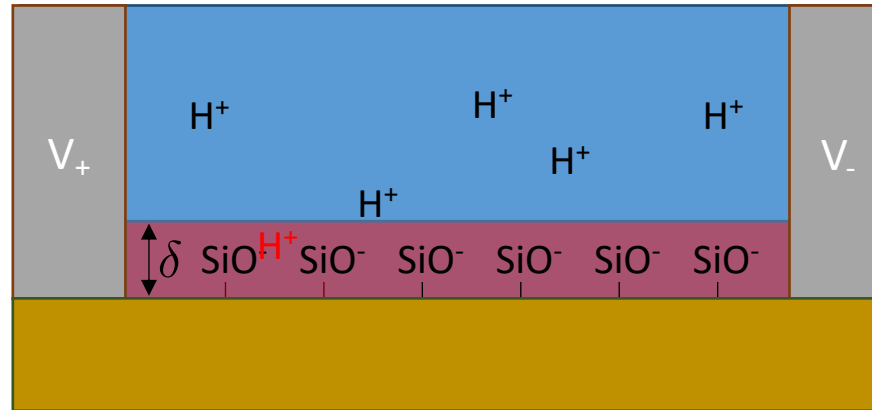
# Detailed analysis: stagnant layers model

$$I \simeq E e L_{tot} \times [\text{SiO}^-]_s \times \mu_+ \times \frac{h - \delta}{h}$$

Stagnant layer

$$\mu_+ = 0$$

$$v = 0$$

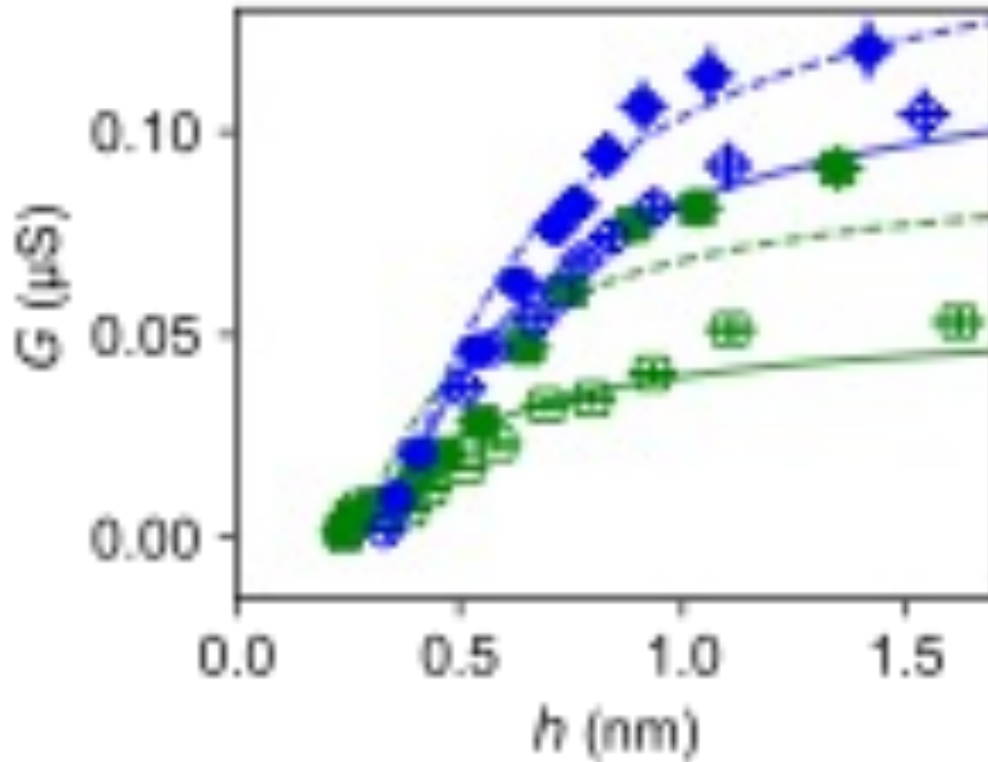


$$I = E \frac{4\epsilon K}{2\pi\ell_B\eta} \left( \underbrace{\frac{\ell_B}{3r_h} \tan\left(K \frac{h - \delta}{2}\right)}_{\text{electrophoretic contribution}} + \underbrace{\tan\left(K \frac{h - \delta}{2}\right) - \frac{K\delta}{2}}_{\text{electroosmosis contribution}} \right)$$

$$Kh \times \tan(Kh) = \frac{h}{\ell_{GC}}$$

electrophoretic contribution    electroosmosis contribution

# Back to experiments



Fitting parameters

$$\delta = 0.3\text{nm}$$

Surface charge  $20 \mu\text{C}/\text{m}^2$   
 → OK with literature  
 (charge regulation)

*Stagnant layer*

→ *In contrast with so-called Stern layer*

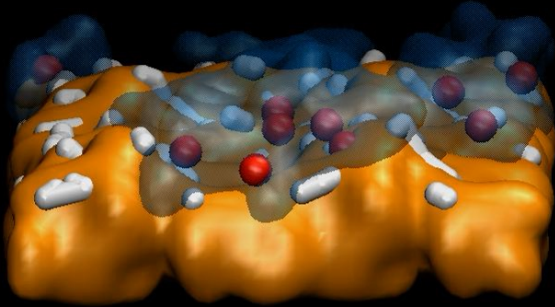
→ *In agreement with statics and hydrodynamics*

Georges, S Millot, J Loubet, A Tonck, 1993.  
 M Cieplak, J Koplik, JR Banavar, Phys. Rev. Lett., 2001.

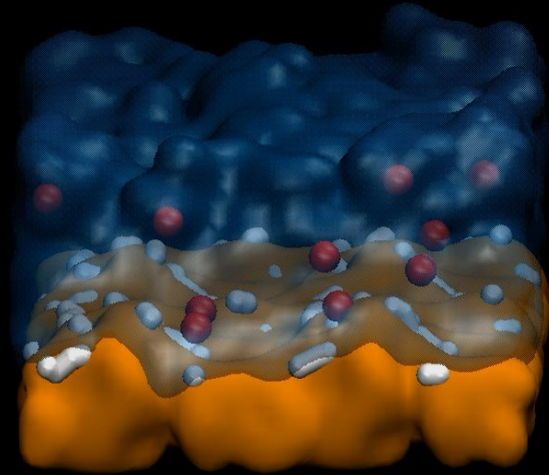
Allemand, Zhao, Vincent, Fulcrand, Joly, Ybert, Bianco, PNAS (in press)

# Origins of the stagnant(s) layer?

$h=0.284$  nm



$h=1.73$  nm



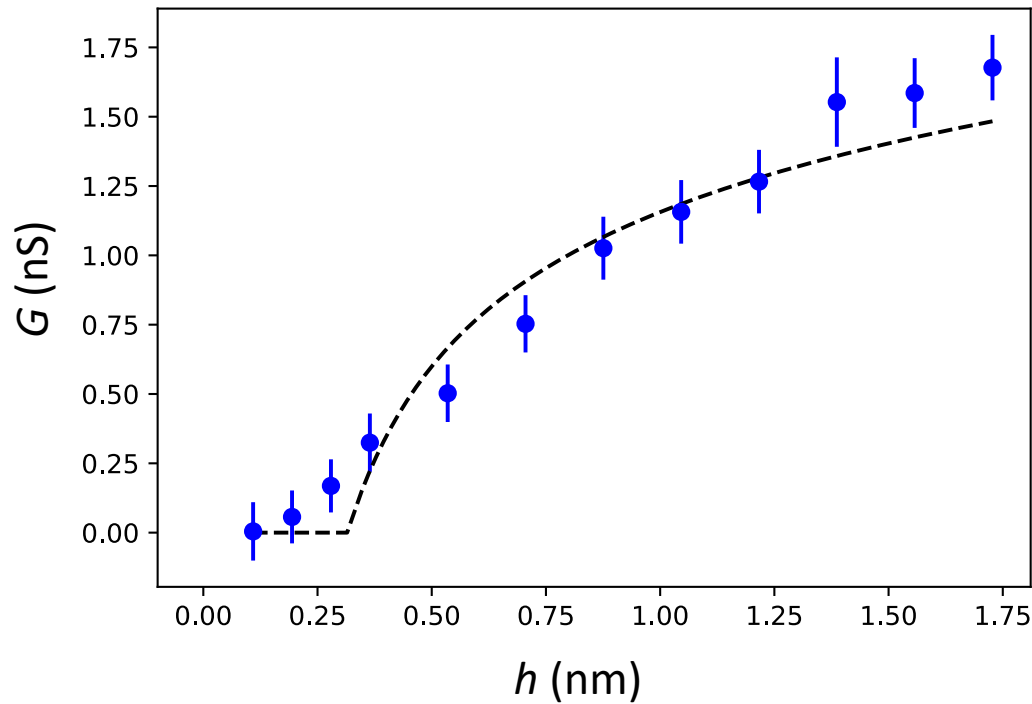
Fixed ionized groups  
 $[\text{SiO}^-]_s=10$

-Silanol -Silica  
-Cation -Water

MD simulations  
 $dt=50\text{ps}$

# Conductance measurements: MD

$$I \simeq E e L_{tot} \times [\text{SiO}^-]_s \times \mu_+ \times \frac{h - \delta}{h}$$



Fitting parameters

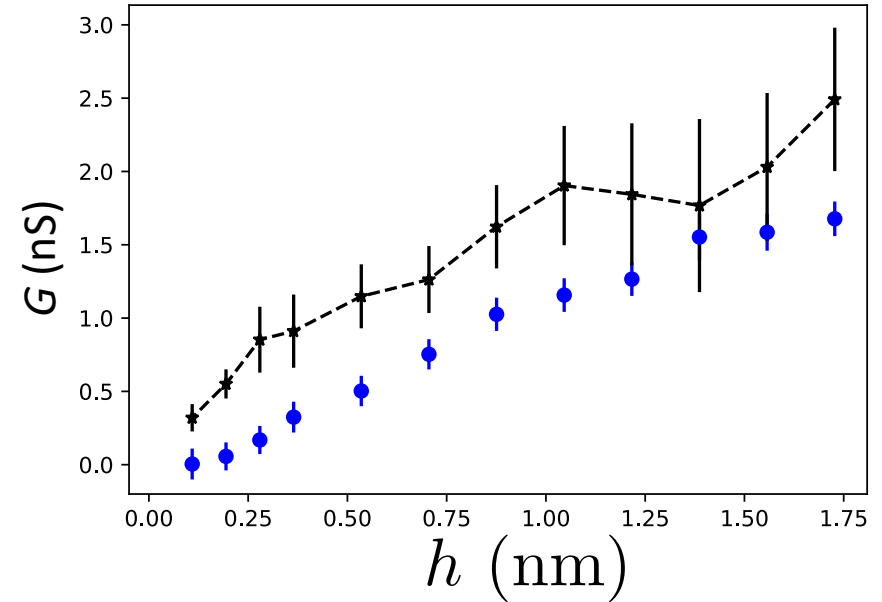
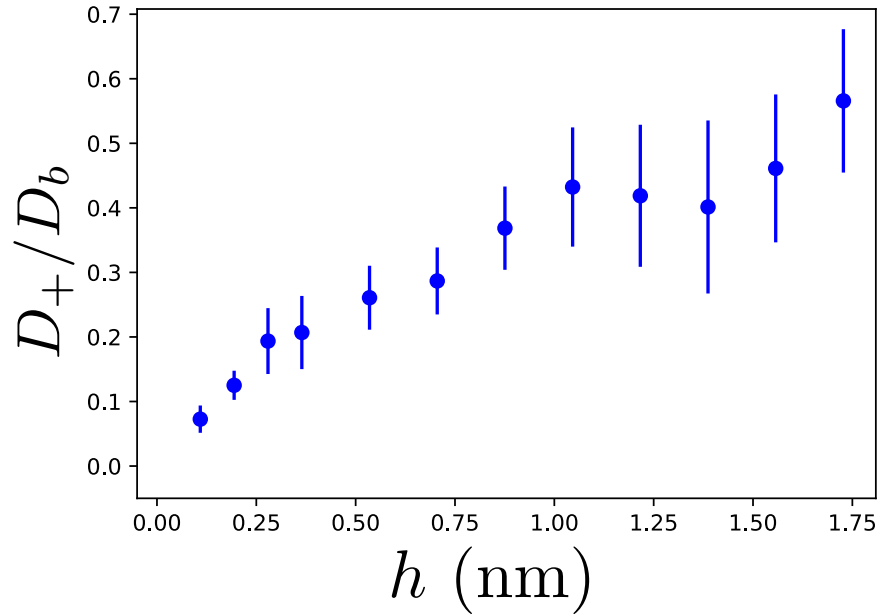
$$\delta = 0.3\text{nm}$$

$[\text{SiO}^-]_s = 3$  (instead of 10)

→ Mismatch



# Diffusion coefficient of the ions: MD



$$D_+(z) = \mu_+(z)kT$$

$$u_+(z) = e\mu_+(z)E + u_{EO}$$

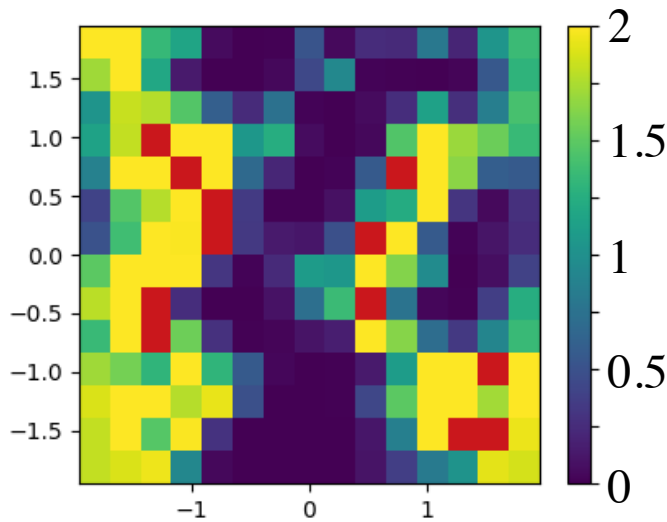
$$I = e \times L_{tot} \int_0^h \rho_+(z)u_+(z)dz$$

- Good agreement

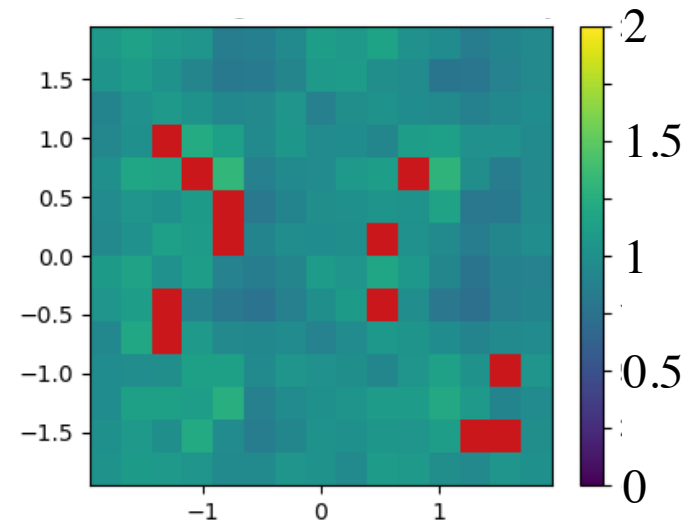
# Origins of the hindered diffusion layer

Normalized local time averaged thickness maps  
(top view on the film)

Thin film  
 $h = 0.284 \text{ nm}$



Thick film  
 $h = 1.78 \text{ nm}$



■ SiO<sub>2</sub> group

**Dry spots on neutral zones?**

# Take-home messages

❖ Experiments to measure **transport in subnanometric** water films.

❖ Break-down of continuous description **< 0.3 nm**.

→ *Effect of the material?*

(*F. Violla, M. Zhao*)

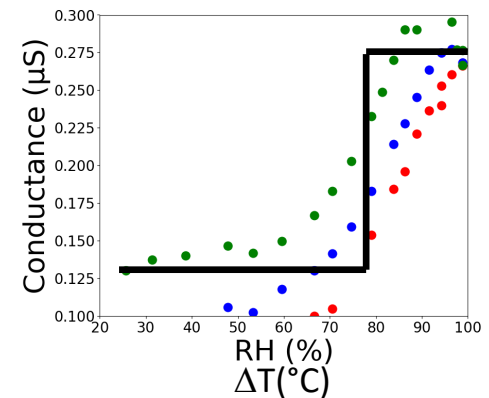
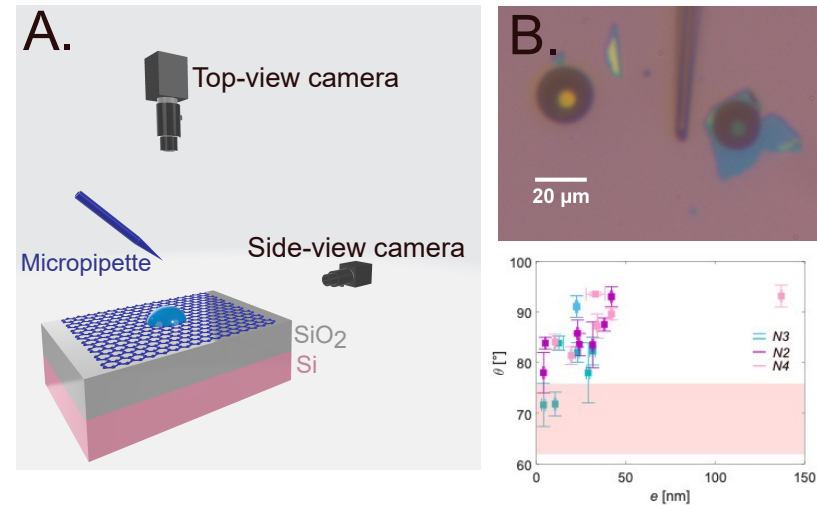
→ *MD with reactive interface*

❖ Design of an **ionic gate**, activated by humidity or temperature.

→ *Optical activation?*

❖ Out-of-equilibrium response for humidity gradients?

→ *Macroscopic results*



Thanks to my colleagues...



Aymeric  
Allemand



Menghua  
Zhao



Remy  
Fulcrand



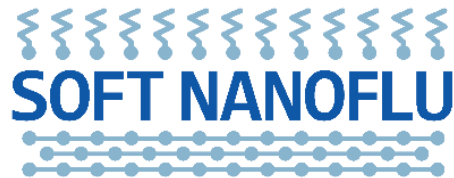
Laurent  
Joly



Christophe  
Ybert



Olivier  
Vincent



Team Liquid and Interfaces